

3-X Structural Model and Common Characteristics of Anomalous Thermal Transport: The Case of Two-Dimensional Boron Carbides

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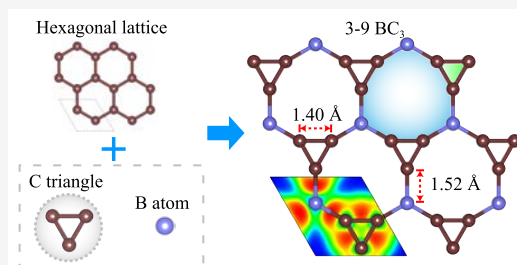


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Supporting Information

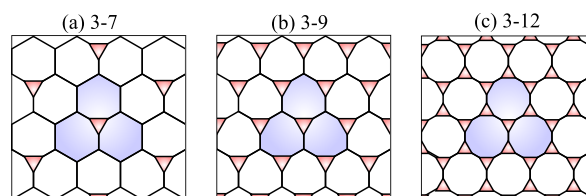
ABSTRACT: Improving the reliability of electronic devices requires effective heat management, and the key is the relationship between the thermal transport and temperature. Inspired by synthesized T-carbon and H-boron, the 3-X structural models are proposed to unify the two-dimensional (2D) multitriangle materials. Employing structural searches, we identify the stability of the 3-X configuration in 2D boron carbides as 3-9 BC₃ monolayer, which, unexpectedly, exhibits a linear thermal conductivity versus temperature, not the traditional $\sim 1/T$ trend. We summarize the common characteristics and explore why this behavior is absent in 3-9 AlC₃ and graphene via investigating the optical modes. We show that the linear behavior is a direct consequence of the special oscillation modes by the 3-X model associated with the largest group velocity. We find that 2D materials with such behavior usually share a relatively low thermal conductivity. Our work paves the way to deeply understand the lattice thermal transport and to widen nanoelectronic applications.



Boron and carbon are adjoining neighbors in the periodic table and can form a diversity of allotropic structures by the sp -, sp^2 - and sp^3 -hybridized bonds. With rapid progress in 2D materials, various allotropes are currently available, for example, carbon nanotubes,¹ fullerenes,² graphene,³ honeycomb borophene,⁴ triangular borophene,⁵ and other 2D sheets.^{6–8} Recently, T-carbon was proposed by substituting each atom in cubic diamond with a carbon tetrahedron⁹ and was verified by the synthesis of T-carbon nanowires via the laser irradiation method.¹⁰ A boron allotrope of cF-B₈ shares with T-carbon a similar structure composed of boron tetrahedrons.¹¹ In T-carbon-like structures, the triangle-dodecagon (3-12) configuration exists along the [110] direction. H-boron was further proposed by arranging the boron tetrahedron at each site of hexagonal diamond,^{12,13} where the 3-12 sketch appears along the [001] direction. However, the 3-12 monolayers remain unclear in 2D materials because of the too strong covalent bonds. Herein, inspired by T-carbon and H-boron, we first propose the 3-X ($X = 7, 8, 9, 10, 11$, or 12) structural models to unify the multitriangle configurations in 2D materials by putting the atomic triangles on sites of a hexagonal lattice. For example, the 3-9 configuration is available once three triangles are alternately arranged in a hexagonal ring. The sketches of 3-X ($X = 7, 9$, and 12) models are illustrated in Scheme 1.

Boron is short one electron, compared to carbon; thereby, a large difference exists in the bonding and structural properties.^{14–16} For example, cubic and hexagonal diamonds remain unstable when carbon atoms are substituted by boron.¹² The emphasis in this work is the 3-X configurations of 2D boron carbides by identifying the chemical stoichiometry, i.e., the “X”,

Scheme 1. Structural Sketches of 3-7, 3-9, and 3-12 Structural Models



to stabilize the 2D configurations, together with physical properties for further applications. Because the graphene and borophene are experimentally available, 2D 3-X boron carbides are promising to remain stable and thus to reflect the simplicity of planar clusters.

Structural properties are critical to the thermal conductivity κ .^{17–23} Most materials in either bulk or low-dimensional configuration display a temperature dependence of thermal conductivity as $\kappa \sim 1/T^\alpha$, where the α varies from 0.85 to 1.05,^{24–27} because the accumulated population of phonons by increasing temperature reduces the mean free paths; hence, the scattering processes are greatly activated. However, a linear

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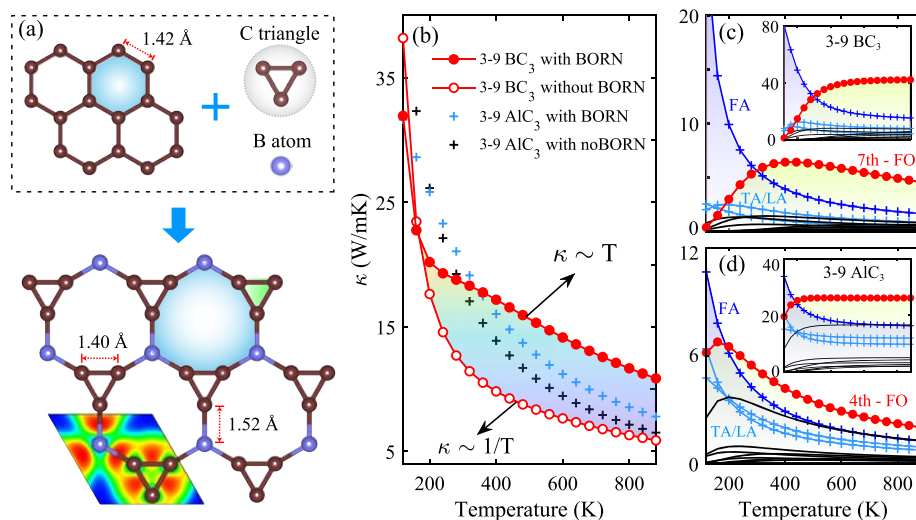


Figure 1. (a) 3-9 BC₃ configuration, where carbon triangles are alternately arranged in hexagonal lattice, together with the electron localization functions. (b) Temperature-dependent thermal conductivity in 3-9 BC₃ and AlC₃ monolayers. The absolute and percentage (inset panels) contributions from the phonon branches in (c) 3-9 BC₃ and (d) AlC₃ monolayers.

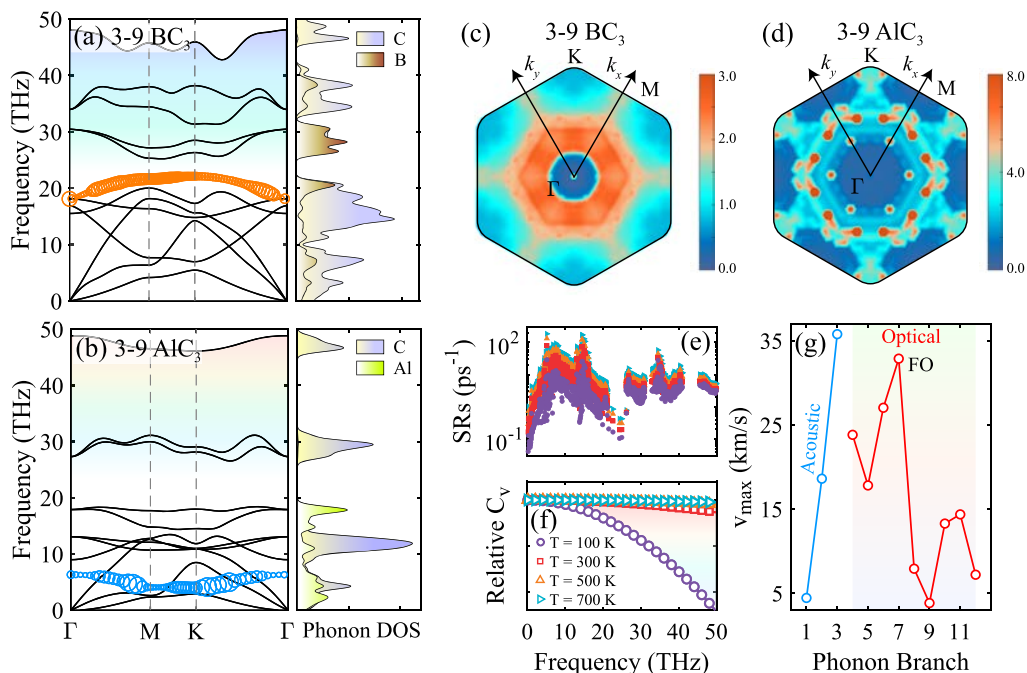


Figure 2. (a–d) Phonon spectra as well as density of states (DOS) and the scattering-rate distributions of the FO branch with the Born effective charges and the dielectric constants in the first Brillouin zone in 3-9 BC₃ and AlC₃. (e–g) The scattering rates and the heat capacity versus frequency at representative temperatures, and the maximum group velocity of each phonon branch in 3-9 BC₃.

thermal conductivity versus temperature relationship appears in hexagonal GaN monolayer by the large difference in atom mass and electronegativity.^{28,29}

In this Letter, we propose the 3-X structural models to unify the 2D multitriangle materials and identify the stability as 3-9 BC₃ monolayer in 2D boron carbides using the first-principles structural searches; the structural details are illustrated in Figure 1a. Unexpectedly, 3-9 BC₃ displays a linear thermal conductivity versus temperature, but the very close atom mass and electronegativity of B and C atoms contradict the mechanism of GaN monolayer.²⁸ Furthermore, large differences in atom mass and electronegativity clearly exist in the 3-9 AlC₃ monolayer, but this anomalous behavior is absent. We

summarize the common characteristics of the linear behavior and point out that 2D materials with this behavior usually share a relatively low thermal conductivity.

Structural searches are performed by the USPEX code,^{30,31} in combination with structural relaxations and total energy calculations within the generalized gradient approximation^{32–34} and G₀W₀-BSE optical absorption³⁵ as implemented in the VASP code.^{36–38} Thermal conductivity is predicted by the ShengBTE code,^{39,40} where the fully iterative solution of the Boltzmann transport equation is adopted. Phonon dispersions are calculated by the PHONONPY code.^{41–43} The computational details and calculated data are provided in

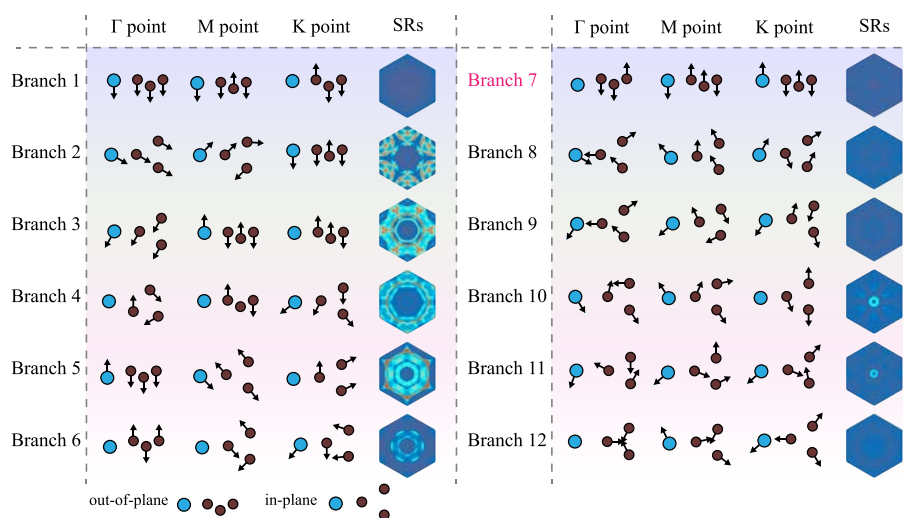


Figure 3. Phonon vibrational modes at the Γ , M, and K points and scattering-rate (SR) distributions in the first Brillouin zone for each phonon branch in 3-9 BC₃ monolayer.

section I in the [Supporting Information](#). The lattice thermal conductivity is defined as⁴⁰

$$\kappa = \sum_{\lambda} C_{\lambda} v_{\lambda}^2 \tau_{\lambda} \quad (1)$$

where $C_{\lambda} = n_{\lambda}^0(n_{\lambda}^0 + 1)\hbar^2\omega_{\lambda}^2/Vk_{\text{B}}T^2$ is the heat capacity per mode; $v_{\lambda} = \partial\omega/\partial\mathbf{q}$ is the group velocity; and the phonon relaxation time τ_{λ} is inverse to the scattering rates, which are associated with the anharmonic IFCs $\Phi_{\lambda\lambda'\lambda''}^{\pm}$ as follows

$$W_{\lambda\lambda'\lambda''}^{\pm} = \frac{\hbar\pi}{4N\omega_{\lambda}\omega_{\lambda'}\omega_{\lambda''}} \left\{ \frac{n_{\lambda'}^0 - n_{\lambda''}^0}{n_{\lambda'}^0 + n_{\lambda''}^0 + 1} \right\} \times |\Phi_{\lambda\lambda'\lambda''}^{\pm}|^2 \delta(\omega_{\lambda} \pm \omega_{\lambda'} - \omega_{\lambda''}) \quad (2)$$

where ω_{λ} is the phonon frequency.

Structural searches identify that the proposed 3-X configurations stabilize as 3-9 BC₃ monolayer in 2D boron carbides (see [Figure 1a](#)). The carbon triangles result in a reduced symmetry ($P\bar{6}m2$), compared to hexagonal BC₃ ($P6/mmm$) (see [Figure S2](#)). The optimized lattice constant of 4.04 Å is considerably larger than 2.59 Å of hexagonal BC₃⁴⁴ and 2.46 Å of graphene.⁴⁵ Thereby the density is smaller in 3-9 BC₃ (0.55 g/cm²) than in hexagonal BC₃ (0.68 g/cm²) and graphene (0.76 g/cm²). Structural details and stability are summarized in section II in the [Supporting Information](#). The lattice thermal conductivity at room temperature is 18.54 W/mK in 3-9 BC₃, which is significantly lower than ~400 W/mK in hexagonal BC₃⁴⁴ and ~3000 W/mK in graphene.⁴⁶

Most interestingly, the 3-9 BC₃ monolayer shows a linear thermal conductivity versus temperature, not the traditional $\kappa \sim 1/T$ law in most materials including 3-9 AlC₃ and graphene (see [Figure 1b](#)). The expected huge phonon gap and splitting between longitudinal optical (LO) and transverse optical (TO) modes are observable in 3-9 AlC₃ and 2D GaN,²⁸ but not in 3-9 BC₃ (see [Figure 2a](#) and b); thus, this anomalous behavior awaits to be discussed deeply.

[Figure 1c,d](#) reveals that the phonon transport is dominated by the flexural acoustic (FA) mode at low temperature and by the flexural optical (FO) mode (the seventh branch in BC₃ and the fourth in AlC₃ in detail) at high temperature. According to the reflection-symmetry-based scattering selection,^{19,47} the

eigenvectors of phonon modes under a horizontal reflection operation σ_{h} satisfy

$$e_{\lambda}^x \xrightarrow{\sigma_{\text{h}}} e_{\lambda}^x, \quad e_{\lambda}^y \xrightarrow{\sigma_{\text{h}}} e_{\lambda}^y, \quad e_{\lambda}^z \xrightarrow{\sigma_{\text{h}}} -e_{\lambda}^z \quad (3)$$

Three-phonon scatterings with an odd number (one or three) of out-of-plane modes are prohibited, such as FA + FA \rightarrow FA, FA + TA \rightarrow TA, and FO \rightarrow FA + FA. Thereby the FA and FO modes determine the thermal conductivity. The FA branch routinely obeys the $\kappa \sim 1/T$ trend. The FO contribution first rises and then drops slightly with temperature, and it overwhelms the FA branch after the crossovers. At relatively high temperature, the FO mode contributes above 40% in BC₃ but ~26% in AlC₃. Therefore, it is feasible to redefine the thermal conductivity versus temperature by the FO mode in 3-9 BC₃.

The linear thermal conductivity versus temperature behavior disappears when the Born effective charges (Z^*) and the dielectric constants (ϵ^*) are neglected. In addition, the thermal conductivity is underestimated by 35.28% in BC₃ and by 10.27% in AlC₃ at room temperature. [Figures 2c,d](#) and [S3](#) reveal that the variations in scattering rates by the inclusion of Z^* and ϵ^* mainly appear at the FO mode. Upon neglecting the Z^* and ϵ^* , the scattering channels are greatly activated for the FO modes in 3-9 BC₃ and AlC₃ near the Γ and K points, respectively. The removal of long-range electrostatic Coulomb interactions seriously suppresses the FO contribution, which is no longer dominant in the thermal conductivity (see [figure S4](#)).

To comprehensively reveal this linear behavior, we focus on the following questions: (1) Why does the thermal conductivity from the optical branch first increase and then decrease with temperature? (2) Why does the seventh (the fourth) branch contribute the most to the thermal conductivity among optical modes in 3-9 BC₃ (AlC₃)? (3) Why does the seventh branch in 3-9 BC₃ contribute relatively more than the fourth in 3-9 AlC₃?

The thermal conductivity is determined by the volumetric specific heat capacity (C_V), the group velocity (v_{λ}), and the relaxation time (τ_{λ}). The group velocity is nearly temperature-independent,^{28,48} which is clarified by the lattice constant and the thermal expansion coefficient versus temperature in 3-9

BC₃ (see Figure S5). The detailed analyses are described in section III in the Supporting Information. As temperature increases, the heat capacity first increases and finally converges because of the total activation of optical phonons (see Figure 2e). Our simulations further reveal that the heat capacity displays a very strong second-order (i.e., $\sim T^2$) character in a wide temperature range (see figure S6). On the other hand, the temperature-enhanced scattering suppresses the thermal conductivity as the known $\sim 1/T$ trend according to the Bose distribution $n_{\lambda}^0 = (\exp[(E_{\lambda} - E_F)/k_B T] - 1)^{-1}$ (see Figure 2f). Therefore, it is more likely for the linear thermal conductivity to appear in this temperature range. Meanwhile, it is necessary for the percentage contribution from optical modes to be large enough to reshape the thermal conductivity with temperature. That is why the linear behavior is present in 3-9 BC₃ but not in 3-9 AlC₃ and graphene. The underlying mechanisms are described in section IV in the Supporting Information in more detail.

Figure 2g displays that the seventh branch shares the maximum group velocity ~ 32.87 km/s among optical modes in 3-9 BC₃ and hence contributes the most to the thermal conductivity. The large group velocity agrees well with the much dispersive phonon branch and closely correlates with the atom vibration, which arises from the strong long-range Coulomb interaction⁴²

$$\mathbf{F} \equiv -\frac{\partial E}{\partial \mathbf{u}} = -\left(M\omega_0^2 + \frac{4\pi e^2 Z^2}{\Omega \epsilon_{\infty}}\right)\mathbf{u} \quad (4)$$

where the restoring force \mathbf{F} is proportional to the atomic motion displacement \mathbf{u} . The theoretical analyses are described in section V in the Supporting Information in more detail.

Figure 3 reveals that the seventh branch shares the distinctive out-of-plane vibrations. Two C atoms move up, and the other C and B atoms move down; thus, the mass ratio (m_u/m_d) reaches ~ 0.95 at the M and K points. Thereby, the FO mode exhibits the maximum vibrational amplitude by the atom resonance along the out-of-plane direction (listed in Table S2), which greatly enhances the thermal conductivity via enlarging the group velocity. Note that the B atom has a slightly larger amplitude than the C atom because of the slighter mass. Analogically, to realize the atom resonance in 3-9 AlC₃, Al and three C atoms vibrate along the opposite directions (see Figures 4c and S7) and the m_u/m_d reaches 0.75. Similarly, the fourth branch exhibits the maximum amplitude and the maximum group velocity among optical modes (see Tables S2 and S3 and Figure S8). Finally, because of the difference in the atom mass and the m_u/m_d , the maximum amplitude is 0.146 Å in AlC₃, which is much smaller than the 0.208 Å in BC₃. Figure 3 also reveals that the scattering processes mainly occur in the low-frequency range and seldom at the Γ point. For example, the third acoustic branch exhibits the largest scattering rate near the M point and thus greatly lowers the thermal conductivity.

Figure 4a displays that the group velocities are overall comparable in 3-9 BC₃ and AlC₃, with the maximum values at the fourth and seventh branches, respectively. The scattering phase space is clearly larger in AlC₃ than in BC₃, especially near the fourth and seventh branches (see Figures 4b). The thermal transport is thereby slightly weaker in AlC₃ than in BC₃ overall. On the other hand, the fourth-branch contribution to the thermal conductivity is seriously suppressed by the large phase space in 3-9 AlC₃, and hence, the linear behavior is

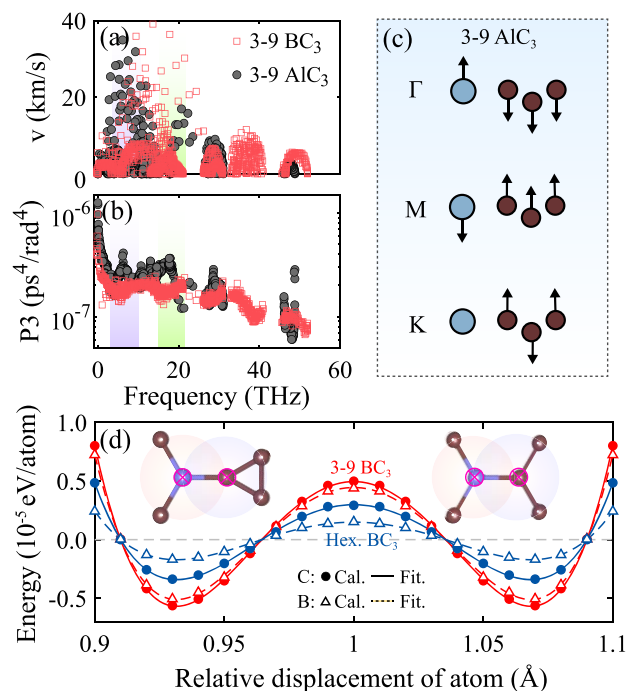


Figure 4. (a) Group velocity and (b) the scattering phase space P_3 in 3-9 BC₃ and AlC₃. The shadow areas indicate the fourth and the seventh branches in 3-9 AlC₃ and BC₃, respectively. (c) The vibrational modes of the fourth branch in 3-9 AlC₃. (d) The regular residuals between the second-order polynomial fittings and the DFT data. The lines represent the higher-order fittings.

absent because of the small percentage. Phonon DOS reveals that B atoms mainly contribute above 20 THz, while Al atoms mainly contribute below 20 THz because of the heavier mass. The down-shift of optical modes, especially the fourth branch, significantly increases the phonon population at low frequency in 3-9 AlC₃, which greatly enlarges the phase space of aa_0 scattering and finally reduces the thermal conductivity.

Our next focus is why the carbon triangles in 3-9 BC₃ seriously suppress the thermal transport, compared to hexagonal BC₃ monolayer, although they consist of the sp^2 covalent bonds with the same chemical stoichiometry. The anharmonicity of covalent bonds is evaluated by the regular residual $O(r_z^3) = E_{\text{cal}}(r_z) - E_{\text{fit}}(r_z)$ between the calculated $E_{\text{cal}}(r_z)$ and the polynomial fitting⁴⁹

$$E_{\text{fit}}(r_z) = a + br_z + cr_z^2 + O(r_z^3) \quad (5)$$

The discussion of the second-order terms is summarized in section VI in the Supporting Information. Two kinds of covalent bonds exist in 3-9 and hexagonal BC₃ monolayers (see the inserted panels in Figure 4d). Figure 4d shows that 3-9 BC₃ possesses a much stronger anharmonicity than hexagonal BC₃, consistent with the more asymmetrical charge distribution in the 3-9 configuration. The charge distribution determines the interatomic interactions (e.g., force constants) and thus reflects the anharmonicity of covalent bonds.⁴⁸ The sp^2 hybridizations are thereby usually irregular in the 3-X structural models because of the asymmetry in the charge distribution by the carbon triangles (see the ELF in Figures 1a and S2b). Furthermore, in 3-9 BC₃, the anharmonicity of C atoms is stronger than that of B atoms because of the more asymmetrical charge distribution around the carbon triangles. The anharmonicity activates the scattering and hence results in

a lower thermal conductivity in 3-9 than in hexagonal BC₃. Analogically, in bulk cases,⁵⁰ T-carbon shares a lower thermal conductivity than C-diamond because of the carbon tetrahedron. Our discussion is also applicable to the ultralow thermal conductivity in chalcogenide materials,⁵¹ where the existence of lone pair cations introduces an asymmetry into the charge distribution and hence enhances the anharmonicity.

In summary, inspired by T-carbon and H-boron, we propose the 3-X structural models to unify the 2D multitriangle materials. Structural searches identify the stability as 3-9 BC₃ monolayer in 2D boron carbides, which exhibits a linear thermal conductivity versus temperature, not the traditional $\kappa \sim 1/T$ law in most materials. To determine what kinds of 2D materials favor such behavior, we summarize the common characteristics by analyzing the absolute and percentage contributions from the optical modes. We note that 2D materials with such behavior usually share a relatively low thermal conductivity and explain why the linear behavior is absent in 3-9 AlC₃ and graphene. We further unveil that the so-called anomalous thermal conductivity versus temperature is usually a linear relationship. Our findings are universal to other 2D materials and are beneficial to promoting nanoelectronic applications in 2D semiconductors.

■ ASSOCIATED CONTENT

SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.jpclett.1c03248>.

Details on the structural parameters, phonon spectra, Bader charges, Born effective charges, ELF, thermal conductivity, phonon vibrational modes, group velocities, and scattering rates (PDF)

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Notes

The authors declare no competing financial interest.

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