

## Section I / Sig Fig rules and basic knowledge about all science courses/

1. It is for nonzero #s to be significant (1-9)
2. Beginning zeros are not significant
3. In between zeros are significant
4. Ending zeros are significant if there is a dot or line indicates it is important

Example : 700.0 (4 Sig Figs).    780(2 sig figs)    780\_( 3 sig figs)

### Use of scientific notation

To express very large or small quantities

Rule of Addition/ Minus :  $3.01 - 2.00 = 1.01$  ( The difference's decimal place would follow the one(either 3.01 or 2.00) who has least decimal place)

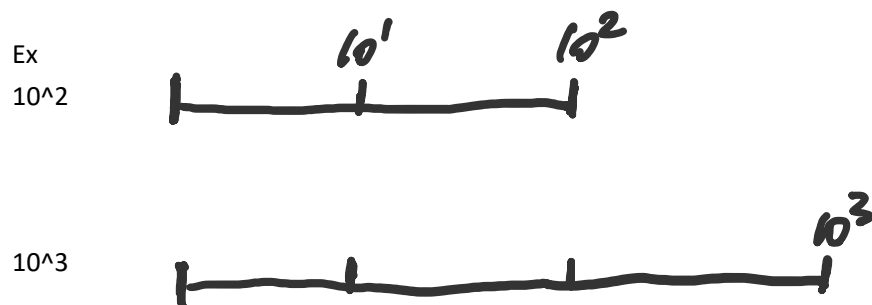
Rules of multiple/division :  $3.00 / 2.0$  ( 2 sig figs) = 1.5 ( so the dividend also has 2 sig figs)

### Prefix conversion

1000 meters =  $1.000 \times 10^3$  meters = 1.000 Km

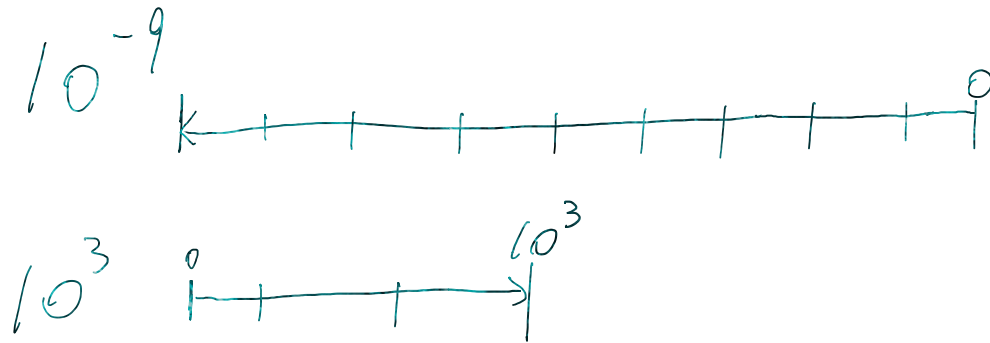
Table 5. SI prefixes			
Factor	Name	Symbol	Factor
$10^{24}$	yotta	Y	$10^{-1}$
$10^{21}$	zetta	Z	$10^{-2}$
$10^{18}$	exa	E	$10^{-3}$
$10^{15}$	peta	P	$10^{-6}$
$10^{12}$	tera	T	$10^{-9}$
$10^9$	giga	G	$10^{-12}$
$10^6$	mega	M	$10^{-15}$
$10^3$	kilo	k	$10^{-18}$
$10^2$	hecto	h	$10^{-21}$
$10^1$	deka	da	$10^{-24}$
			yocto y

To convert it , you can draw a number-line diagram to show the difference



By knowing the difference between two powers , you can convert them easily.

Another example

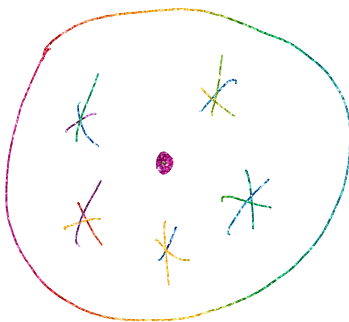


So combine them together, you can find the difference from  $10^{-9}$  to  $10^3$

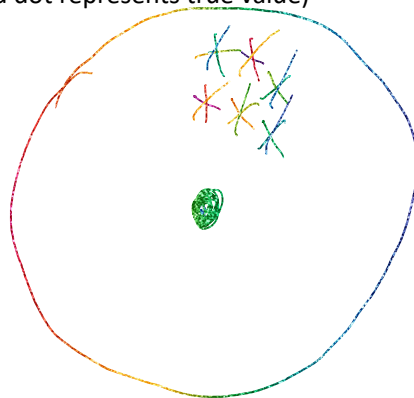


So something like  $1 \cdot 10^3$  meters would be  $1 \cdot 10^{12}$  nanometers

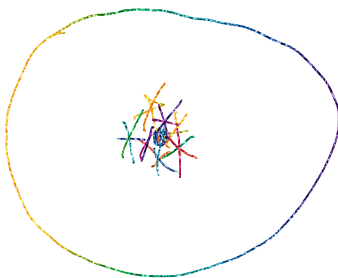
Precision & accuracy ( \* represents the measurements and dot represents true value)



High accuracy / low precision



Low accuracy/high precision



High accuracy / high precision

Precision : the nearness of sequential measurements

Accuracy : The measurements close to true value

Extensive property: amount measured does change before/after the measurements

Intensive property : amount measured does not change before/after the measurements



Side notes: precision is repeating data while accuracy is how much close to the true accurate result

Uncertainty reading: if the cylinder reading is 35.2 mL then add one (guessed) digit to the reading so it becomes 35.22mL.