

**CENTRAL UNIVERSITY OF HARYANA  
SCHOOL OF ENGINEERING & TECHNOLOGY**

**3<sup>rd</sup> Sessional Exam (July, 2023)**

**Paper Name: Chemistry B. Tech. (CSE) 1<sup>st</sup> Year/II Sem; Paper Code: BT CH 102A, 1 hr, 20 Marks**

*Note: Candidates are required to attempt all the questions.*

1. Prove that  $O_2^{2-}$  is diamagnetic, while  $N_2^{2-}$  is paramagnetic? 5
2. Predict whether the cell reaction is spontaneous or not 5  
$$Fe^{2+}(aq) + Ag^+(aq) \rightarrow Fe^{3+}(aq) + Ag(s)$$

(Given  $E^0_{Fe^{3+}/Fe^{2+}} = 0.77\text{ V}$ ;  $E^0_{Ag^+/Ag} = 0.80\text{ V}$ ;  $F = 96500\text{ C mol}^{-1}$ )
3. A) Calculate pH of 0.005 M  $Ca(OH)_2$  assuming complete dissociation. 5  
B) Calculate the solubility of  $CaF_2$  in solution, when solubility product is  $3.45 \times 10^{-11}$  at room temperature.
4. Discuss the fundamental mode of vibrations in an organic compound. 5

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