

# EX\_1

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EX 1.1

EX 1.2

EX 1.3

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During a heating process, the temperature of a system rises by  $10^{\circ}C$ . Express this rise in temperature in  $K$ ,  $^{\circ}F$ , and  $R$

$$\Delta K = \Delta^{\circ}C = 10K$$

$$\Delta R = 1.8\Delta^{\circ}C = 18R$$

$$\Delta^{\circ}F = \Delta R = 18^{\circ}F$$

## EX 1.2

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A vacuum gage connected to a chamber reads 5.8 psi at a location where the atmospheric pressure is 14.5 psi. Determine the absolute pressure in the chamber

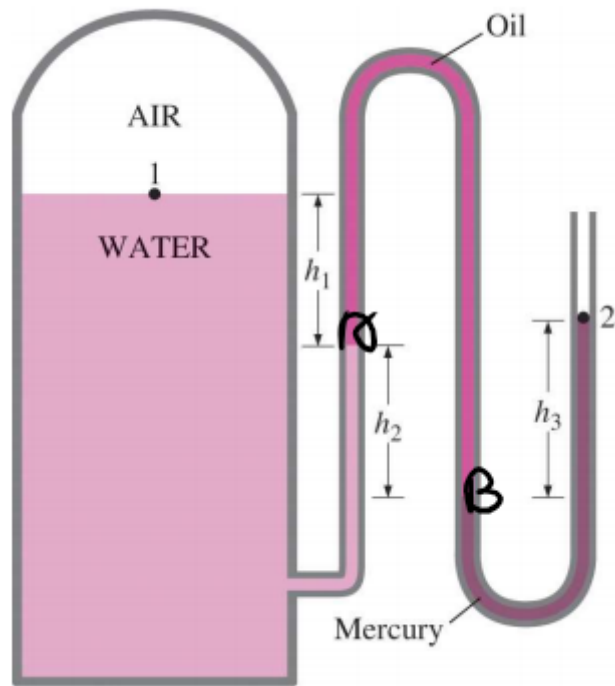
$$\begin{aligned}P_{vac} &= P_{atm} - P_{atm} \\&= 14.5 - 5.8 = 8.7psi\end{aligned}$$

## EX 1.3

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The water in a tank is pressurized by air, and the pressure is measured by a multifluid-manometer as shown in the following figure. The tank is located on a mountain where the atmospheric pressure is  $85.6 \text{ kPa}$ .

Determine the air pressure in the tank if  $h_1 = 0.1 \text{ m}$ ,  $h_2 = 0.2 \text{ m}$ , and  $h_3 = 0.35 \text{ m}$ . Take the densities of water, oil, and mercury to be  $1000 \text{ kg/m}^3$ ,  $850 \text{ kg/m}^3$ , and  $13,600 \text{ kg/m}^3$ , respectively.



$$\begin{cases} P_A = P_1 + \rho_w g h_1 \\ P_B = P_A + \rho_o g h_2 \\ P_B = P_2 + \rho_m g h_3 \end{cases}$$

$$\Rightarrow P_1 = P_2 + \rho_m g h_3 - \rho_w g h_1 - \rho_o g h_2$$

$$= 129.65 kPa$$