

Gas-Liq-Solids

Three-laws-Thermo

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Outline

- 1 States of Matter
 - Gas, Liquid, and Solid
- 2 Practical Functions
 - Blocks
 - Figures and Tables
 - Graphs
- 3 Conclusions
 - As for States of Matter

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General Notice

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- formula transformation: molarity & density

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Why do we study kinetic molecular theory?

- Graham's law of effusion?

Understanding from A New Point of View

Here we will discuss the **ideal gas equation** from a new point of view, *i.e.*, **kinetic molecular theory**(KMT).

Understanding from A New Point of View

First we should get aware of the **prerequisite** of KMT.

¹Sun, Ting, *CHEM2100J-FA21-Ch5-6*, pp. 35.

Understanding from A New Point of View

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Recall what has been taught in lectures.

1. A gas is in continuous random motion
2. Gas molecules are infinitesimally small
3. They move in straight lines until collision
4. Gas molecules do not influence one another except during collisions
5. The collisions are elastic

Prerequisites of KMT shown in slides¹

¹Sun, Ting, *CHEM2100J-FA21-Ch5-6*, pp. 35.

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Understanding from A New Point of View

Example

For a model satisfying KMT, suppose there exists N gas molecules in a cubic box with length L . Each molecule has the mass of m , and the speed of u .

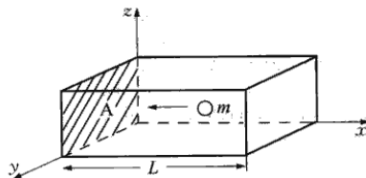


Figure 1.7 An elastic collision of molecule with a wall.

Understanding from A New Point of View

- (1) Calculate the average kinetic energy of each molecule \bar{E}_k .
- (2) If the relationship between average molecule and the temperature is

$$\bar{E}_k = \frac{3}{2}kT$$

where k denotes the Boltzmann constant and satisfies $k = \frac{R}{N_A}$.

Understanding from A New Point of View

Solution

We may assume there are $\frac{N}{3}$ molecules moving along the x -axis direction since there are N molecules and only three directions in total.

Apparently, the momentum of these molecules is mu . For a whole collision process (we can also refer this process as *bouncing*), the distance the molecule travels is $2L$ and the change of momentum is $2mu$.

Understanding from A New Point of View

(Continued)

Hence

Liquid

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This is an example of block.

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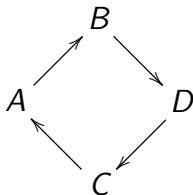
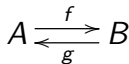
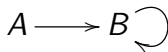
This is another block.

Examples of figures and tables



Figure: An example of figure

Examples of Graphs



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- Smoot Salute!
<http://web.mit.edu/spotlight/smoot-salute>.