

Assignment 2

Name: Harikesh Kushwaha

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Imports

```
import pandas as pd
In [1]:
         import matplotlib.pyplot as plt
         import matplotlib as mpl
```

If you are running this file in Google colab, the associated csv file can be downloaded by using the command below:

!wget https://raw.githubusercontent.com/Hari31416/Academic/main/Codes/Assignment2/data In []:

Run the cell above in Google colab and continue with the next cells.

Also, the original excel file can be downloaded by clicking here.

```
df = pd.read csv("data.csv")
In [2]:
In [3]:
```

Out[3]:

```
Distance
                 Energy
       1.000 -31.078234
       0.900 -31.460702
 2
       0.800 -31.710034
       0.850 -31.608172
 4
       0.700 -31.700418
       0.600 -31.199452
       0.550 -30.636954
       0.500 -29.751617
       0.710 -31.718451
       0.720 -31.732030
10
       0.730 -31.741429
11
       0.740 -31.746910
       0.750 -31.748717
12
13
       0.760 -31.747075
       0.770 -31.742194
```

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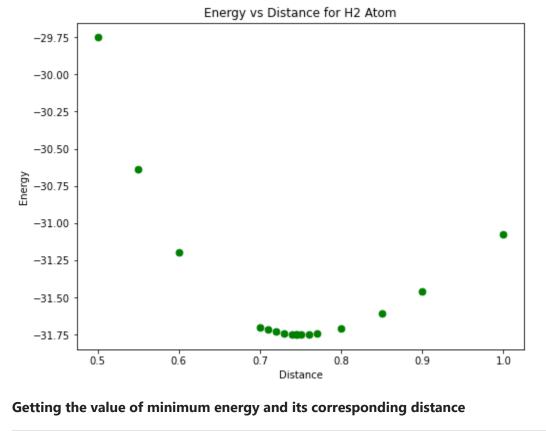
```
#Setting global figure size
In [4]:
         mpl.rcParams['figure.figsize'] = (8, 6)
```

Making a rough plot of the data

0.744 -31.748061

0.746 -31.748420

```
df.plot(x='Distance', y='Energy', kind='scatter', color='green', s = 45)
In [5]:
         plt.title('Energy vs Distance for H2 Atom');
```



#Getting the value of minimum energy and its corresponding distance

```
In [6]:
         min position = df[df['Energy'] == df['Energy'].min()].values
         min energy = float(min position[:,1])
         min energy position = float(min position[:,0])
        The Final Plot
```

```
df.plot(x='Distance', y='Energy', kind='scatter', color='green', s = 45)
In [15]:
         #The point corresponding to minimum energy
         plt.plot(min_energy_position, min_energy, 'bo', markersize=10)
         #Text describing the minimum energy
         plt.text(min_energy_position+0.005, min_energy-0.005,
             f'Minimum Energy = {round(min energy, 3)} eV',
             fontdict={'size': 12, 'color': 'blue', 'weight': 'bold'})
         #Vertical line at the distance corresponding to minimum energy
         plt.vlines(min_energy_position, -32, -29, linestyles='dashed', colors='red')
         #Text describing the minimum distance
         #Modifying the lables
         plt.xlabel('Bond Length $(\AA)$', fontsize=14)
         plt.ylabel('Energy (eV)', fontsize=14)
         #Zooming in on the minimum energy
         plt.ylim([-31.8, -31.6])
         plt.xlim([0.68,0.82])
         #Adding a title
         plt.title('Energy vs Bond Length for H2 Atom (Zoomed in)', fontsize=14);
                       Energy vs Bond Length for H2 Atom (Zoomed in)
           -31.600
```

