

# Practice Problems

# Practice Problems

1. Determine the Miller indices of the plane in the figure below.

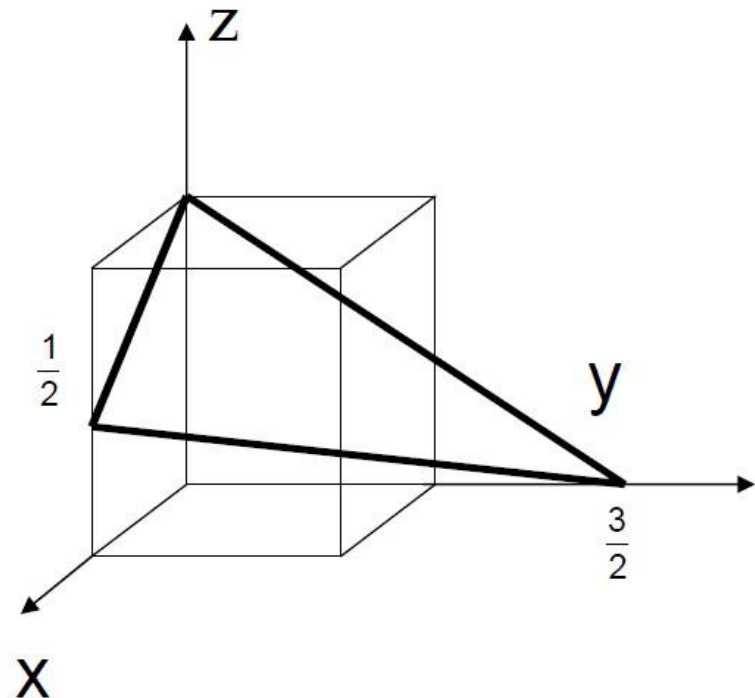
a) (623)

b) (431)

c) (341)

d) (432)

e) (346)



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Answer:

e)(346)

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2. Consider a metal with an FCC structure and an atomic weight of 92.9. When monochromatic x-radiation having a wavelength of 0.1028 nm is focused on the crystal, the angle of diffraction ( $2\theta$ ) for the (311) set of planes in this metal occurs at 71.2 degrees (for the first order reflection  $n=1$ ).
  - a. Calculate the interplanar spacing for this set of planes.
  - b. Calculate the lattice parameter for this metal.
  - c. Calculate the density of the metal (units of  $\text{g/cm}^3$ )

# Practice Problems

Answer:

$$a) \quad d = \frac{\lambda}{2\sin\theta} = \frac{0.1028}{2\sin(71.2^\circ)} = 0.0883nm$$

$$b) \quad a = d\sqrt{h^2 + k^2 + l^2} = 0.0883\sqrt{3^2 + 1^2 + 1^2} = 0.293nm$$

$$c) \quad \rho = \frac{nA}{a^3 N_A} = \frac{4.92.9}{6.02 \times 10^{23}} (0.293 \times 10^{-7})^3 \cdot = 24.53g / cm^3$$

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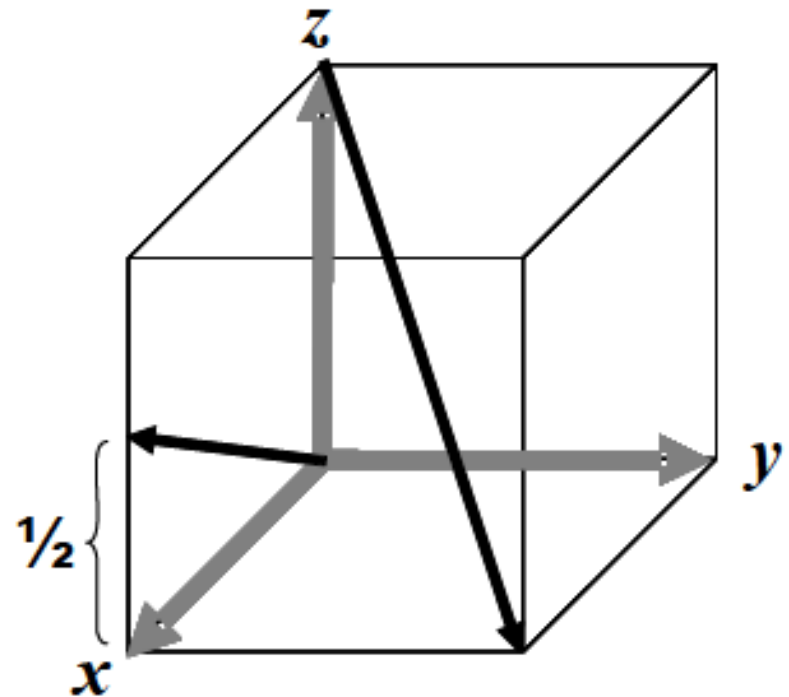
3. Determine the Miller indices of the two directions shown in the figure below:

a)  $[101]$  and  $[\bar{1}12]$

b)  $[211]$  and  $[21\bar{1}]$

c)  $[201]$  and  $[11\bar{1}]$

d)  $[002]$  and  $[\bar{2}11]$



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Answer:

For the first direction, use  $(0,0,0)$  as the point of origin:

$x=1, y=0, z= \frac{1}{2}$ . Clear the fractions to get  $[201]$

For the second direction, use  $(0,0,1)$  as the point of origin:

$x= 1, y=1, z=-1$  or  $[11\bar{1}]$

So the answer is **c)**

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4. Refer to the table below and determine which **two** elements are likely to have the greatest mutual solubility.

	<b>Nb</b>	<b>Fe</b>	<b>Cu</b>	<b>Ni</b>
<b>Crystal structure</b>	<b>BCC</b>	<b>BCC</b>	<b>FCC</b>	<b>FCC</b>
<b>Lattice Parameter</b>	<b>0.330 nm</b>	<b>0.286 nm</b>	<b>0.361</b>	<b>0.352</b>
<b>Electronegativity</b>	<b>1.61</b>	<b>1.83</b>	<b>1.90</b>	<b>1.91</b>

- (a) Nb-Fe
- (b) Nb-Ni
- (c) Fe-Cu
- (d) Cu-Ni



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## Answer:

There are several factors that improve mutual solubility which include:

- Similar size
- Same crystal structure
- Similar electronegativity

For this reason, Cu-Ni should have the best mutual solubility. So the answer is **d**).

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5. How many directions belong to the  $\langle 100 \rangle$  family?

a) 2

b) 4

c) 6

d) 12

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## Answer

We have three principal axes  $[100]$ ,  $[010]$ ,  $[001]$  as well as their reverse directions. So the total is 6 or **c**).