# Physics 415 - Lecture 35: Black-Body Radiation

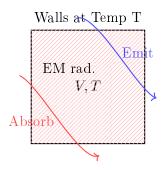
#### April 18, 2025

# **Summary: Photon Statistics**

- Applies to bosonic particles whose total particle number is not conserved (e.g., photons, phonons).
- Equivalent to BE statistics with chemical potential  $\mu = 0$ .
- Mean occupation number:  $\overline{n}_r = \frac{1}{e^{\beta \epsilon_r} 1}$  (Planck distribution).
- Canonical partition function (unrestricted sum):  $Z = \prod_r \frac{1}{1 e^{-\beta \epsilon_r}}$ .
- Helmholtz free energy:  $F = -T \ln Z = T \sum_r \ln(1 e^{-\beta \epsilon_r})$ .

# **Black-Body Radiation**

"Black-body radiation" = electromagnetic (EM) radiation in thermal equilibrium within a cavity (volume V) whose walls are maintained at temperature T.



The walls continually absorb & emit radiation. In equilibrium, the properties of the EM radiation depend only on T.

Wave-particle duality of QM  $\implies$  EM radiation can be described as waves (classical EM) or particles ("photons").

- Photons are bosons.
- Photon number is not conserved (can be absorbed/emitted by walls).
- Interactions between photons are negligible (because Maxwell's equations are linear).

Therefore, we may treat the equilibrium EM radiation as an ideal gas of photons obeying photon statistics ( $\mu = 0$ ).

#### Single-Particle States for Photons

Need to specify the states r and energies  $\epsilon_r$ . From Maxwell's equations in vacuum, the electric field  $\vec{E}$  satisfies the wave equation:

$$\nabla^2 \vec{E} = \frac{1}{c^2} \frac{\partial [}{\partial 2}] \vec{E} t^2$$

Plane wave solutions  $\vec{E} = \text{Re}\{\vec{E}_0 e^{i(\vec{k}\cdot\vec{x}-\omega t)}\}$  require the dispersion relation  $\omega = c|\vec{k}| = ck$ . Also,  $\nabla \cdot \vec{E} = 0 \implies \vec{k} \cdot \vec{E}_0 = 0$ , so  $\vec{E}$  is transverse to  $\vec{k}$ . For each  $\vec{k}$ , there are two linearly independent modes of oscillation (polarization states) perpendicular to  $\vec{k}$ .

In the particle description (QM):

- Photon energy:  $\epsilon = \hbar \omega = \hbar ck$ .
- Photon momentum:  $\vec{p} = \hbar \vec{k}$ .
- Energy-momentum relation:  $\epsilon = c|\vec{p}| = cp$ . (Massless particle).

A single-photon state r is specified by its wave-vector  $\vec{k}$  (or momentum  $\vec{p}$ ) and its polarization state s (two possibilities, s = 1, 2). The energy  $\epsilon_k = \hbar c k$  is independent of polarization.

#### Photon Gas Thermodynamics

The mean number of photons in a state  $(\vec{k}, s)$  is given by the Planck distribution  $(\mu = 0)$ :

$$\overline{n}_{\vec{k},s} = \frac{1}{e^{\beta \epsilon_k} - 1} = \frac{1}{e^{\beta \hbar ck} - 1}$$

Counting States: Use Periodic Boundary Conditions (PBC) on a large box  $V = L_x L_y L_z$ . Allowed wave-vectors:  $k_i = 2\pi n_i/L_i$ . Sum over states  $\sum_r$  becomes sum over  $(\vec{k}, s)$ . Spin degeneracy g = 2 for polarization.

$$\sum_{r} \to \sum_{\vec{k} = 1, 2} \to gV \int \frac{d^3k}{(2\pi)^3} = 2V \int \frac{d^3k}{(2\pi)^3}$$

Convert integral to frequency  $\omega = ck$ . Use spherical coordinates in k-space  $(d^3k = 4\pi k^2 dk)$ .

$$\sum_{r} \to 2V \int_{0}^{\infty} \frac{4\pi k^{2} dk}{(2\pi)^{3}} = \frac{V}{\pi^{2}} \int_{0}^{\infty} k^{2} dk$$

Since  $k = \omega/c$ ,  $dk = d\omega/c$ :

$$\sum_{r} \to \frac{V}{\pi^2} \int_0^\infty \left(\frac{\omega}{c}\right)^2 \left(\frac{d\omega}{c}\right) = \int_0^\infty d\omega \left(\frac{V\omega^2}{\pi^2 c^3}\right)$$

We define the density of modes (states) per unit frequency range  $g(\omega)$ :

$$g(\omega)d\omega = \frac{V\omega^2}{\pi^2c^3}d\omega$$

$$g(\omega) = \frac{V\omega^2}{\pi^2 c^3}$$

### Planck's Formula

Use this result to investigate the mean energy of the photon gas.

$$E = \sum_{r} \overline{n}_{r} \epsilon_{r} = \int_{0}^{\infty} d\omega \, g(\omega) \overline{n}(\omega) \epsilon(\omega)$$

where  $\overline{n}(\omega) = 1/(e^{\beta\hbar\omega} - 1)$  and  $\epsilon(\omega) = \hbar\omega$ .

$$E = \int_0^\infty d\omega \left(\frac{V\omega^2}{\pi^2 c^3}\right) \left(\frac{1}{e^{\beta\hbar\omega} - 1}\right) (\hbar\omega)$$

$$E = \frac{V\hbar}{\pi^2 c^3} \int_0^\infty d\omega \frac{\omega^3}{e^{\beta\hbar\omega} - 1}$$

Write this as  $E = \int_0^\infty d\omega (\frac{dE}{d\omega})$ . The spectral energy density (mean energy per unit volume per unit frequency range) is  $u(\omega) = \frac{1}{V} \frac{dE}{d\omega}$ .

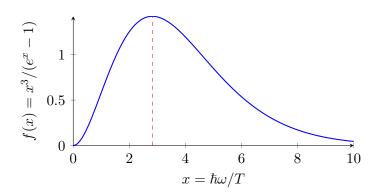
$$u(\omega) = \frac{\hbar}{\pi^2 c^3} \frac{\omega^3}{e^{\beta \hbar \omega} - 1}$$

This is Planck's formula for black-body radiation.

Plot  $u(\omega)$  vs  $\omega$ . It is useful to use a dimensionless variable  $x = \beta \hbar \omega = \hbar \omega / T$ .  $\omega = Tx/\hbar$ ,  $d\omega = Tdx/\hbar$ .

$$u(\omega)d\omega = \frac{\hbar}{\pi^2 c^3} \frac{(Tx/\hbar)^3}{e^x - 1} \left(\frac{T}{\hbar} dx\right) = \frac{T^4}{\pi^2 (\hbar c)^3} \frac{x^3}{e^x - 1} dx$$

Let  $f(x) = \frac{x^3}{e^x - 1}$ .



The function f(x) peaks at  $x_{max} \approx 2.821$ . This means the peak frequency  $\omega_{max}$  in the spectrum satisfies  $\hbar \omega_{max}/T \approx 2.821$ .

$$\hbar\omega_{max} \approx 2.821T$$
 or  $\omega_{max} \approx \frac{2.821T}{\hbar}$ 

As T increases, the maximum of the spectral distribution shifts to higher frequencies, proportional to T. This is **Wien's displacement law**.

Low Frequency Limit ( $\hbar\omega \ll T$ , or  $x \ll 1$ ):  $e^x \approx 1 + x$ .  $u(\omega) \approx \frac{\hbar}{\pi^2 c^3} \frac{\omega^3}{(1+x)-1} = \frac{\hbar}{\pi^2 c^3} \frac{\omega^3}{x} = \frac{\hbar}{\pi^2 c^3} \frac{\omega^3}{\beta \hbar \omega} = \frac{T}{\pi^2 c^3} \omega^2$ .

$$u(\omega) \approx \frac{T\omega^2}{\pi^2 c^3}$$
 (Rayleigh-Jeans formula)

This classical result follows from equipartition: each EM mode (oscillator) has average energy T.  $T \times (\text{number of modes per frequency per volume})$  gives  $T \times (\omega^2/(\pi^2 c^3))$ . Note: The temperature

T drops out of the average energy per mode  $\bar{\epsilon} = \hbar \omega \bar{n}(\omega) \approx \hbar \omega/(\beta \hbar \omega) = T$ , consistent with equipartition.

The RJ formula predicts  $u(\omega) \propto \omega^2$ , which diverges at high frequencies  $(\omega \to \infty)$ . If integrated, it gives infinite total energy density. This is the "ultraviolet catastrophe" of classical physics. Planck's resolution was to introduce quantized energies  $E = n\hbar\omega$ , leading to the  $e^{\beta\hbar\omega}-1$  denominator which suppresses high frequencies. This was a first success of quantum theory.

### Total Energy Density (Stefan-Boltzmann Law)

Compute the total energy density u = E/V.

$$u = \int_0^\infty u(\omega)d\omega = \frac{T^4}{\pi^2(\hbar c)^3} \int_0^\infty dx \frac{x^3}{e^x - 1}$$

The definite integral is  $\int_0^\infty \frac{x^3}{e^x - 1} dx = \Gamma(4)\zeta(4) = 3!(\pi^4/90) = 6\pi^4/90 = \pi^4/15$ .

$$u = \frac{T^4}{\pi^2 (\hbar c)^3} \frac{\pi^4}{15} = \frac{\pi^2}{15(\hbar c)^3} T^4$$

Energy density  $u \propto T^4$ . This is the **Stefan-Boltzmann Law**.  $u = \sigma_E T^4$ , where  $\sigma_E = \frac{\pi^2}{15(\hbar c)^3}$ . (If using T in Kelvin,  $\sigma_E = \frac{\pi^2 k_B^4}{15\hbar^3 c^3}$ ). Qualitatively: Each thermally relevant photon mode has energy  $\sim T$ . The relevant modes

Qualitatively: Each thermally relevant photon mode has energy  $\sim T$ . The relevant modes are those with  $\hbar\omega \sim T$ , or  $\hbar ck \sim T$ , so  $k \lesssim k_{max} \sim T/(\hbar c)$ . The number of modes per volume up to  $k_{max}$  is  $\sim k_{max}^3 \sim (T/(\hbar c))^3 \propto T^3$ . Total energy density  $u \sim (\# \text{ modes } / \text{ vol}) \times (\text{avg energy per mode}) \propto T^3 \times T = T^4$ .