Started on Thursday, 2 June 2022, 11:20 AM

State Finished

Completed on Thursday, 2 June 2022, 11:54 AM

Time taken 34 mins 32 secs

Grade 45.00 out of 100.00

Question 1 Correct Mark 5.00 out of 5.00

Ambident nucleophiles are ones which can react with a substrate at either of two nucleophilic sites. Which of the following is not an ambident nucleophile?

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$$\begin{bmatrix} :\text{CN:} \end{bmatrix} & \begin{bmatrix} :\ddot{\text{O}}\text{H} \end{bmatrix} & \begin{bmatrix} :\ddot{\text{O}}\text{CH}_2\text{CH}_2\ddot{\text{S}} : \end{bmatrix}^{2^{-}} \\ \text{I} & \text{II} & \text{III} \\ & \begin{bmatrix} :\ddot{\text{O}}\text{:N::}\ddot{\text{O}} \end{bmatrix}^{-} & \begin{bmatrix} H_{\cdot}\ddot{\text{N}};\ddot{\text{O}}\text{:H}} \end{bmatrix}^{-} \\ \text{IV} & \text{V} \end{bmatrix}$$

Select one:

- a. I
- b. III
- c. IV
- d. II
- e. V

The correct answer is: II

Question 2

Incorrect

Mark 0.00 out of 5.00

Arrange the following amines in order of increasing basicity (least to most) in the gas phase:

NH3 CH3CH2CH2NH2

(CH3CH2CH2)2NH

(CH3CH2CH2)3N

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IV

Select one:

- a. || < || < | < |V
- b. | < |V < || < |||</li>

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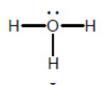
- C. | < || < || < |V
- d. III < I < II < IV
- e. IV < III < II < I

The correct answer is: I < II < III < IV

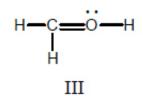
Questic	on 3 Incorrect	Mark 0.00 out of 5.00			
Conside	r the following:				
CH <sub>3</sub> Cl	H <sub>2</sub> CH=CHCH <sub>2</sub> CH <sub>3</sub> I	CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH=CH <sub>2</sub> II			
CH <sub>3</sub> Cl	H=CHCH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub> III	CH <sub>2</sub> =CHCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub> IV			
Which two structures represent the same compound?					
Select on	e:				
	a. II and III				
	b. I and III	× Y	anlış		
	c. II and IV				
	d. I and II				
$\bigcirc$	e. None of these choic	es.			

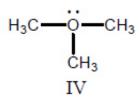
The correct answer is: II and IV

In which structure(s) below does the oxygen have a formal charge of +1?









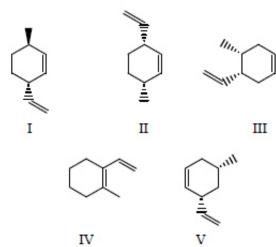
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Select one:

- a. I, III, and IV
- b. I only
- c. II only
- d. I and IV
- e. I and III

The correct answer is: I, III, and IV

Select the structure for cis-3-methyl-6-vinylcyclohexene.



Select one:

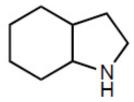
- a. V
- ( b. I
- c. IV
- od. III
- e. II

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The correct answer is: II

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What functional group is present in the following compound?

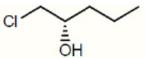


## Select one:

- a. 1∘ alkyl bromide
- b. 2° amine
- c. nitrile
- d. 1∘ amine
- e. 3° amine

The correct answer is: 2° amine

What functional group(s) is/are present in the following compound?



## Select one:

- a. Primary alkyl chloride and primary alcohol
- b. Ether and primary alcohol
- c. Primary alkyl chloride and secondary alcohol



- d. Primary alcohol and secondaryalkyl chloride
- e. Ether and secondary alcohol

The correct answer is: Primary alkyl chloride and secondary alcohol

Question 8

Not answered

Marked out of 5.00

What is the IUPAC name of the following compound?



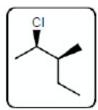
## Select one:

- a. trans-1-chlorocyclohexan-3-ol
- b. trans-4-chlorocyclohexanol
- c. cis-4-chlorocyclohexanol
- d. cis-1-chlorocyclohexan-4-ol
- e. trans-3-chlorocyclohexan-1-ol

The correct answer is: trans-4-chlorocyclohexanol

Quest	ion 9	Correct	Mark 5.00 out of 5.00						
Which molecule would have a dipole moment greater than zero?									
Select c	ne:								
	a. BCl <sub>3</sub>								
	b. H <sub>2</sub> O			✓	Doğru				
$\bigcirc$	c. CCl4								
	d. CO <sub>2</sub>								
	e. BeCl <sub>2</sub>								
The co	rrect answer	is: H2O							

Which staggered Newman projection(s), looking down the C-2? -C-3 bond (C-2 in front and-3 in back), illustrates the following boxed compound?



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Select one:

a. I, II and III

H<sub>3</sub>C

- b. None of these choices.
- c. II and V
- d. I and II
- e. V only

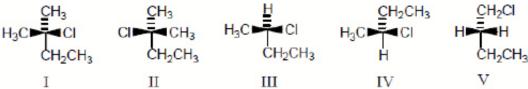
The correct answer is: None of these choices.

Question 11

Correct

Mark 5.00 out of 5.00

(R) -2-Chlorobutane is represented by:



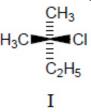
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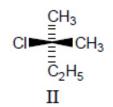
Select one:

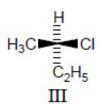
- a. II
- ) b. V
- c. IV
- d. III
- e. I

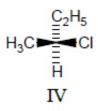
The correct answer is: III

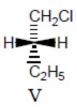
Chiral molecules are represented by:











Select one:

- a. IV alone
- b. I, II, III, IV and V
- c. I and II
- d. I, II, III and IV
- e. III and IV

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The correct answer is: III and  ${\sf IV}$ 

Quest	ion 13	Incorrect	Mark 0.00 out of 5.00		
For th	ne general	ized structure	BrCH2CHClCH2CHClCH2Br there	e exists what number	
of					
stere	oisomers?				
Select	one:				
$\bigcirc$	a. 8				
$\bigcirc$	b. 2				
$\bigcirc$	c. 6				
$\bigcirc$	d. 3				
•	e. 4			🗡 Yanlış	
	orrect answe		Mark 5.00 out of 5.00		
		methylpentan	e are examples of:		
Select		. K.O. O. MO. O. K.G			
		ereomers.			
	b. enantiomers.				
	c. constitutional isomers.				
$\bigcirc$	d. stere	oisomers.			
$\bigcirc$	e. None	of these cho	ces.		
The co	orrect answe	er is: constituti	onal isomers.		

I and II are:

Select one:

- a. constitutional isomers.
- b. identical.
- c. diastereomers.
- d. enantiomers.
- e. not isomeric.

The correct answer is: identical.

Which alkyl halide would be most reactive in an S<sub>N</sub>1 reaction?

Select one:

- \_\_\_\_ a. I
- b. IV
- O c. V
- d. III
- e. II

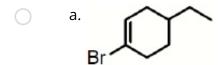
The correct answer is: IV

**Question 17** Marked out of 5.00 Which of the following alkyl bromide isomers would most readily undergo an S<sub>N</sub>2 reaction? Select one: a. 4-bromo-1-butene b. All will react at the same rate. c. 3-bromo-1-butene d. bromocyclobutane e. 1-bromo-1-butene The correct answer is: 4-bromo-1-butene Question 18 Not answered Marked out of 5.00 Which of the following reactions proceeds with inversion of configuration at the carbon bearing the leaving group? Select one: a. E2 b. All of these answer choices are correct. c. S<sub>N</sub>2 d. S<sub>N</sub>1 e. E1 The correct answer is: SN2

Not answered

Which of the following would be most reactive in an S<sub>N</sub>2 reaction?

Select one:



The correct answer is:

The reaction between which combination of substances below cannot be classified as a

Bronsted-Lowry acid-base reaction?

## Select one:

- a. two of these choices
- $\bigcirc$  b.  $H_2SO_4 + CH_3CO_2Na$
- $\bigcirc$  c. BF<sub>3</sub> + NH<sub>3</sub>
- $\bigcirc \quad \text{d. } H_3O^+ + CH_3NH^-$
- e. CH<sub>3</sub>Li + C<sub>2</sub>H<sub>5</sub>OH

The correct answer is:  $BF_3 \ + \ NH_3$