



Passive tracer options, chemical mechanisms and KPP in WRF-Chem 3.6

Ravan Ahmadov

ravan.ahmadov@noaa.gov

Cooperative Institute for Research in Environmental Sciences, CU Boulder

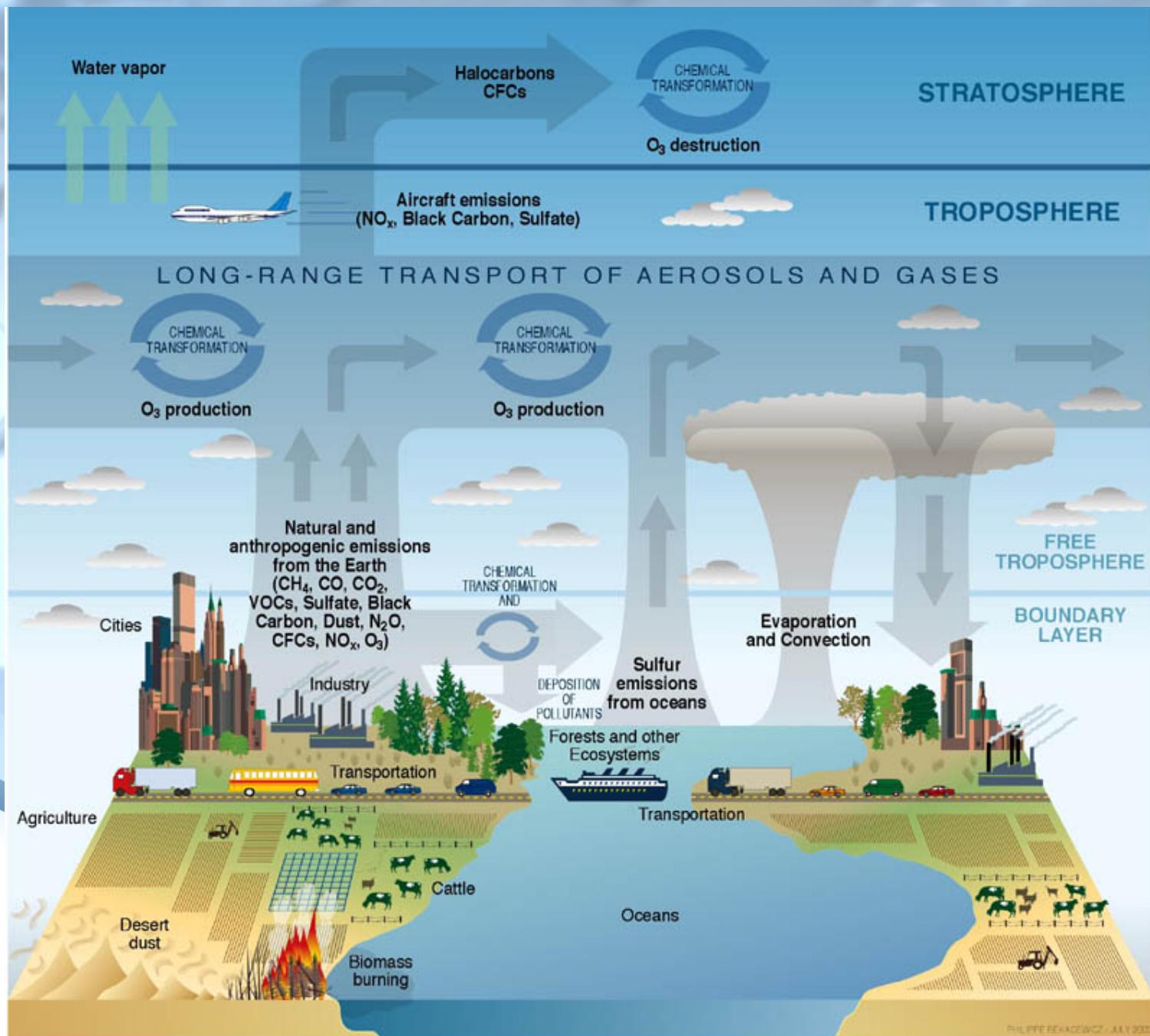
Earth System Research Laboratory, NOAA, Boulder

Acknowledgements: M. Salzmann, S.-W. Kim, S. McKeen, G. Grell, S. Peckham

**WRF-Chem tutorial, ICIMOD
Kathmandu, June 1-5, 2014**

Outline

- **Passive tracer options in the WRF-Chem V3.6 model**
- **Chemical mechanisms in WRF-Chem**
- **Kinetic PreProcessor (KPP) and WRF-Chem KPP Coupler (WKC)**
- **Adding chemical mechanisms to the WRF-Chem model using KPP**
- **Additional notes**



The passive tracer chemistry options in WRF-Chem3.6

- Passive tracer options (chem_opt=13-15): no chemistry, transport only, anthropogenic or any other assigned emissions can be used
- Passive tracer options for greenhouse gases (chem_opt=16, 17): no chemistry, transport only, anthropogenic emissions and biospheric fluxes of CO₂ and CH₄
- You can run the chemistry options as passive tracer simulations by setting gaschem_onoff=0 in namelist.input
- In passive tracer type of simulation all the tracers are advected (chem_adv_opt in namelist.input), vertically mixed (vertmix_onoff) and also mixed by cumulus parameterization (chem_conv_tr)
- Passive tracer transport simulations are very useful to evaluate the transport and mixing of chemicals, emissions and for other tasks; Computationally very efficient!
- 4 In WRF-Chem the meteorological and chemical variables share the same

The gas phase chemistry mechanisms in WRF-Chem3.6

- Regional Acid Deposition Model, 2nd generation (**RADM2**)
- Regional Atmospheric Chemistry Mechanism (**RACM**)
- RACM - Mainz Isoprene Mechanism (**RACM-MIM**)
- RACM - Earth System Research Laboratory (**RACM-ESRL**), *updated RACM-MIM*
- Carbon-Bond Mechanism version Z (**CBMZ**)
- Model of Ozone and Related Chemical Tracers (**MOZART**)
- Statewide Air Pollution Research Center (**SAPRC99**)
- Common Representative Intermediates Mechanism (**CRIMech**)

different implementations, coupled to different aerosol schemes and aqueous chemistry, suitable for different applications ranging from regional air quality to global atmospheric chemistry simulations (no halogen chemistry yet!)

Few notes about the chemistry options in WRF-Chem

- Computationally efficient options using the RADM scheme (e.g. chem_opt=1, 2, 101): **for air quality modeling, not all biogenic VOCs are included**
- Options based on the RACM schemes (e.g. chem_opt=102-105): **more updated version of RADM, more biogenic VOCs are included, updated isoprene chemistry, reaction rates**
- Carbon Bond Mechanisms (e.g. chem_opt=110, 120): **Widely used in different air quality models (CAMx, CMAQ,...) for ozone modeling**
- MOZART (chem_opt= 111): **used in some of global atmospheric chemistry models (NCAR, GFDL)**

Few notes about the chemistry options in WRF-Chem

- Options based on SAPRC99: (e.g. chem_opt=195,198): There are different versions of this chemistry mechanism that are available from UC Riverside: SAPRC07, a compact version, a version for toxics etc.
- Options based on the CRIMech schemes (e.g. chem_opt=600, 601): Very detailed chemistry mechanism based on Master Chemical Mechanism; It is suitable for detailed chemistry analysis; If you have anthropogenic emissions and measurements to run and evaluate this scheme; Also, it's computationally very expensive!

Table 1. RACM Mechanism Species List

No.	Species	Definition	Carbon Number	Molecular Weight
<i>Stable Inorganic Compounds</i>				
Oxidants				
1	O ₃	ozone		48
2	H ₂ O ₂	hydrogen peroxide		34
Nitrogenous compounds				
3	NO	nitric oxide		30
4	NO ₂	nitrogen dioxide		46
5	NO ₃	nitrogen trioxide		62
6	N ₂ O ₅	dinitrogen pentoxide		108
7	HONO	nitrous acid		47
8	HNO ₃	nitric acid		63
9	HNO ₄	pernitric acid		79
Sulfur compounds				
10	SO ₂	sulfur dioxide		64
11	SULF	sulfuric acid		98
Carbon oxides				
12	CO	carbon monoxide	1	28
13	CO ₂	carbon dioxide	1	44
<i>Abundant Stable Species</i>				
14	N ₂	nitrogen		28
15	O ₂	oxygen		32
16	H ₂ O	water		18
17	H ₂	hydrogen		2
<i>Inorganic Short-Lived Intermediates</i>				
18	O ^{3P}	ground state oxygen atom, O(^{3P})		16
19	O ^{1D}	excited state oxygen atom, O(^{1D})		16
Odd hydrogen				
20	HO	hydroxy radical		17
21	HO ₂	hydroperoxy radical		33

Table 2b. The RACM Mechanism

Reaction No.	Reaction	$A, \text{cm}^3 \text{s}^{-1}$	$E/R, \text{K}$	k^*
<i>Inorganic Reactions</i>				
(R24)	$\text{O}^3\text{P} + \text{O}_2 \rightarrow \text{O}_3$	Table 2f		1.50×10^{-14}
(R25)	$\text{O}^3\text{P} + \text{O}_3 \rightarrow 2 \text{O}_2$	8.00×10^{-12}	2060	7.96×10^{-15}
(R26)	$\text{O}^1\text{D} + \text{N}_2 \rightarrow \text{O}^3\text{P} + \text{N}_2$	1.80×10^{-11}	-110	2.60×10^{-11}
(R27)	$\text{O}^1\text{D} + \text{O}_2 \rightarrow \text{O}^3\text{P} + \text{O}_2$	3.20×10^{-11}	-70	4.05×10^{-11}
(R28)	$\text{O}^1\text{D} + \text{H}_2\text{O} \rightarrow \text{HO} + \text{HO}$	2.20×10^{-10}		2.20×10^{-10}
(R29)	$\text{O}_3 + \text{HO} \rightarrow \text{HO}_2 + \text{O}_2$	1.60×10^{-12}	940	6.83×10^{-14}
(R30)	$\text{O}_3 + \text{HO}_2 \rightarrow \text{HO} + 2 \text{O}_2$	1.10×10^{-14}	500	2.05×10^{-15}
(R31)	$\text{HO} + \text{HO}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$	4.80×10^{-11}	-250	1.11×10^{-10}
(R32)	$\text{H}_2\text{O}_2 + \text{HO} \rightarrow \text{HO}_2 + \text{H}_2\text{O}$	2.90×10^{-12}	160	1.70×10^{-12}
(R33)	$\text{HO}_2 + \text{HO}_2 \rightarrow \text{H}_2\text{O}_2 + \text{O}_2$	Table 2f		2.92×10^{-12}
(R34)	$\text{HO}_2 + \text{HO}_2 + \text{H}_2\text{O} \rightarrow \text{H}_2\text{O}_2 + \text{O}_2 + \text{H}_2\text{O}$	Table 2f		6.58×10^{-30}
(R35)	$\text{O}^3\text{P} + \text{NO} \rightarrow \text{NO}_2$	Table 2d		1.66×10^{-12}
(R36)	$\text{O}^3\text{P} + \text{NO}_2 \rightarrow \text{NO} + \text{O}_2$	6.50×10^{-12}	-120	9.72×10^{-12}
(R37)	$\text{O}^3\text{P} + \text{NO}_2 \rightarrow \text{NO}_3$	Table 2d		1.58×10^{-12}
(R38)	$\text{HO} + \text{NO} \rightarrow \text{HONO}$	Table 2d		4.87×10^{-12}
(R39)	$\text{HO} + \text{NO}_2 \rightarrow \text{HNO}_3$	Table 2d		1.15×10^{-11}
(R40)	$\text{HO} + \text{NO}_3 \rightarrow \text{NO}_2 + \text{HO}_2$	2.20×10^{-11}		2.20×10^{-11}
(R41)	$\text{HO}_2 + \text{NO} \rightarrow \text{NO}_2 + \text{HO}$	3.70×10^{-12}	-250	8.56×10^{-12}
(R42)	$\text{HO}_2 + \text{NO}_2 \rightarrow \text{HNO}_4$	Table 2d		1.39×10^{-12}
(R43)	$\text{HNO}_4 \rightarrow \text{HO}_2 + \text{NO}_2$	Table 2e		8.62×10^{-2}
(R44)	$\text{HO}_2 + \text{NO}_3 \rightarrow 0.3 \text{ HNO}_3 + 0.7 \text{ NO}_2 + 0.7 \text{ HO} + \text{O}_2$	3.50×10^{-12}		3.50×10^{-12}
(R45)	$\text{HO} + \text{HONO} \rightarrow \text{NO}_2 + \text{H}_2\text{O}$	1.80×10^{-11}	390	4.86×10^{-12}
(R46)	$\text{HO} + \text{HNO}_3 \rightarrow \text{NO}_3 + \text{H}_2\text{O}$	Table 2f		1.47×10^{-13}
(R47)	$\text{HO} + \text{HNO}_4 \rightarrow \text{NO}_2 + \text{O}_2 + \text{H}_2\text{O}$	1.30×10^{-12}	-380	4.65×10^{-12}
(R48)	$\text{O}_3 + \text{NO} \rightarrow \text{NO}_2 + \text{O}_2$	2.00×10^{-12}	1400	1.82×10^{-14}
(R49)	$\text{O}_3 + \text{NO}_2 \rightarrow \text{NO}_3 + \text{O}_2$	1.20×10^{-13}	2450	3.23×10^{-17}
(R50)	$\text{NO} + \text{NO} + \text{O}_2 \rightarrow \text{NO}_2 + \text{NO}_2$	3.30×10^{-39}	-530	1.95×10^{-38}
(R51)	$\text{NO}_3 + \text{NO} \rightarrow \text{NO}_2 + \text{NO}_2$	1.50×10^{-11}	-170	2.65×10^{-11}
(R52)	$\text{NO}_3 + \text{NO}_2 \rightarrow \text{NO} + \text{NO}_2 + \text{O}_2$	4.50×10^{-14}	1260	6.56×10^{-16}
(R53)	$\text{NO}_3 + \text{NO}_2 \rightarrow \text{N}_2\text{O}_5$	Table 2d		1.27×10^{-12}
(R54)	$\text{N}_2\text{O}_5 \rightarrow \text{NO}_2 + \text{NO}_3$	Table 2e		4.36×10^{-2}
(R55)	$\text{NO}_3 + \text{NO}_3 \rightarrow \text{NO}_2 + \text{NO}_2 + \text{O}_2$	8.50×10^{-13}	2450	2.29×10^{-16}
(R56)	$\text{HO} + \text{H}_2 \rightarrow \text{H}_2\text{O} + \text{HO}_2$	5.50×10^{-12}	2000	6.69×10^{-15}
(R57)	$\text{HO} + \text{SO}_2 \rightarrow \text{SULF} + \text{HO}_2$	Table 2d		8.89×10^{-13}
(R58)	$\text{CO} + \text{HO} \rightarrow \text{HO}_2 + \text{CO}_2$	Table 2f		2.40×10^{-13}

What chemistry option should I use for my project?

- What are the chemical species do I need to simulate by WRF-Chem? Ozone, carbon monoxide, benzene or secondary organic aerosols...
- How important are the biogenic hydrocarbons? (e.g. *isoprene chemistry*)
- What speciation information is available in my emission data?
- What chemical measurements do I have for evaluation of the model output?
- What is my computational power? Other constraints?

Some references (*check also the user's guide*)

- Stockwell, W. R., et al. (1997), A new mechanism for regional atmospheric chemistry modeling, *J. Geophys. Res.-Atmos.*, 102(D22), 25847-25879.
- Stockwell, W. R., et al. (1990), The 2nd generation regional acid deposition model chemical mechanism for regional air-quality modeling, *J. Geophys. Res.-Atmos.*, 95(D10), 16343-16367.
- Horowitz, L. W., et al. (2003), A global simulation of tropospheric ozone and related tracers: Description and evaluation of MOZART, version 2, *J. Geophys. Res.*, 108(D24), 4784, doi:10.1029/2002JD002853.
- Gross, A., and W. R. Stockwell (2003), Comparison of the EMEP, RADM2 and RACM mechanisms, *Journal of Atmospheric Chemistry*, 44(2), 151-170.
- Geiger, H., et al. (2003), The tropospheric degradation of isoprene: an updated module for the regional atmospheric chemistry mechanism, *Atmos. Environ.*, 37(11), 1503-1519.
- Luecken, D. J., et al. (2008), Effects of using the CB05 vs. SAPRC99 vs. CB4 chemical mechanism on model predictions: Ozone and gas-phase photochemical precursor concentrations, *Atmos. Environ.*, 42(23), 5805-5820.
- Cai et al. (2011), Photochemical Modeling in California with Two Chemical Mechanisms: Model Intercomparison and Response to Emission Reductions, *J. Air & Waste Manage. Assoc.*
- Jenkin et al., (2008), A Common Representative Intermediates (CRI) mechanism for VOC degradation. Part 1: Gas phase mechanism development

KPP in WRF-Chem

Kinetic PreProcessor (KPP) reads chemical reactions and rate constants from ASCII input files and automatically generates code for chemistry integration using the Rosenbrok solver

No KPP for aerosols!

Advantages:

- less time consuming than manual coding**
- less error prone**
- numerically efficient**
- flexibility in updating mechanism with additional species and equations**
- suitable for adjoint code development**

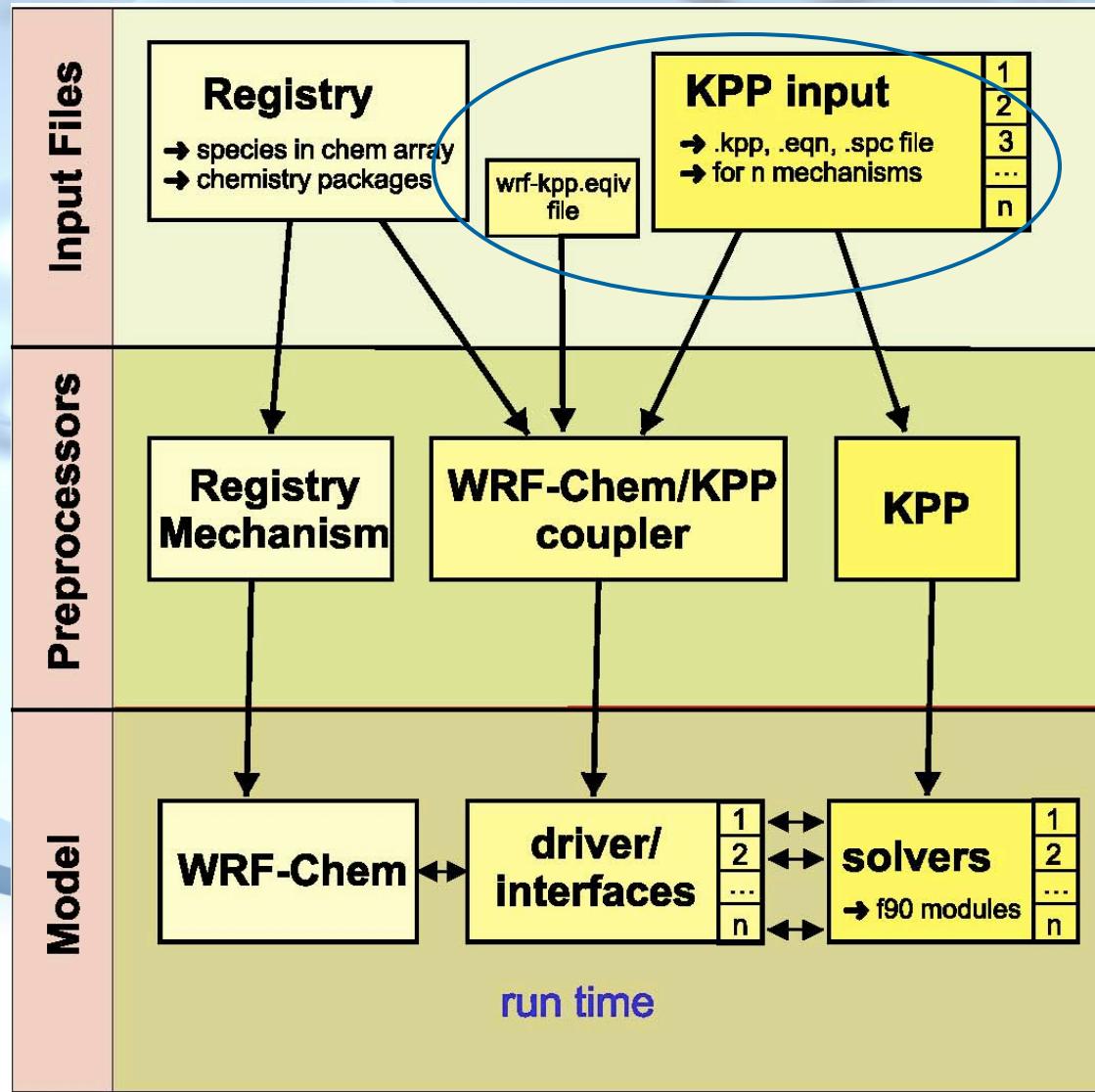
References:

- Damian, V., et al. (2002), The kinetic preprocessor KPP - a software environment for solving chemical kinetics, *Comput. Chem. Eng.*, 26(11), 1567-1579.
- Sandu, A., and R. Sander (2006), Technical note: Simulating chemical systems in Fortran90 and Matlab with the Kinetic PreProcessor KPP-2.1, *Atmos. Chem. Phys.*, 6, 187-195.
- Verwer, J., Spee, E., Blom, J. G., and Hunsdorfer, W. (1999), A second order Rosenbrock method applied to photochemical dispersion problems, *SIAM Journal on Scientific Computing*, 20, 1456–1480.
- www.mpch-mainz.mpg.de/~salzmann/my_home/sub/wkc.html

Implementation of the chemistry mechanisms in WRF-Chem3.6

<i>Chemical mechanisms</i>	<i>Fixed versions</i>	<i>KPP</i>	<i>Coupled to the aerosol schemes</i>
RADM2	Yes	Yes	MADE/SORGAM, GOCART
RACM	None	Yes	MADE/SORGAM, GOCART
RACM-MIM	None	Yes	None
RACM-ESRL	None	Yes	MADE/SORGAM, MADE/SOA_VBS
CB4	None	Yes	None
CBMZ	Yes	Yes	MOSAIC
MOZART	None	Yes	GOCART
SAPRC99	None	Yes	MOSAIC
NMHC9	None	Yes	None
CRIMech	None	Yes	MOSAIC

WKC (WRF-Chem/KPP Coupler) (\$WRF-CHEM/chem/KPP/util/wkc/)



WRF-CHEM3.3 User's guide, 2012

Input files of KPP

*.spc file

Definition of chemical species as variable or fixed value.

*.eqn file

Writing chemical reactions in kpp format

*.kpp file

Model description, computer language, precision, integrator (e.g. Rosenbrock solver)

etc.

*.def file

User defined functions

(also check \$WRF-CHEM/chem/KPP/kpp/kpp-2.1/util/ WRF_conform/ UserRateLaws.f90)

Pre-defined variables in WKC

	KPP equation file	Equation file units	Registry
Photolysis rate	J(Pj_no2)	s ⁻¹	ph_no2
Temperature	TEMP	K	t_phy
Third body concentration	C_M	(molecular moist air) cm ⁻³	Calculated from density
Water vapor concentration	C_H2O	Molecules cm ⁻³	Calculated from qvapor

WRF-CHEM3.6 User's guide, 2013

How to add a new KPP chemistry mechanism to WRF-Chem?

- 1) Add a new mechanism to \$WRF-CHEM/Registry/registry.chem
- 2) Add new variables to the registry if necessary (new species, namelist options etc.)
- 3) Create subdirectories in \$WRF-CHEM/chem/KPP/ “mechanisms” and “inc”
- 4) Create new KPP files with new reactions, rates and species
- 5) Modify some \$WRF-CHEM/chem/*.F (e.g. chem_driver.F) programs in order to include a new chemical mechanism
- 6) Modify \$WRF-CHEM/chem/convert_emiss.F for new species
- 7) Compile a new WRF-CHEM code and run using updated emission files

KPP requires some UNIX tools - flex, yacc, and sed to be installed on your system before compiling the code !

setenv FLEX_LIB_DIR = /usr/local/lib

setenv WRF_KPP =1

Example: Adding a new chemistry mechanism to WRF-Chem for the state of the art SOA parameterization

In order to implement the new SOA mechanism (chem_opt=108 in V3.5) we need to modify the RACM_ESRLSORG gas chemistry:

- Separate MBO from internal alkenes - OLI species
- Add SESQ (biogenic VOC) reactions

Adding the new chemistry package and species to \$WRF-CHEM/Registry/registry.chem

```
state real sesq ikjftb chem 1 - irhusdf=(bdy_interp:dt) "sesq" "SESQ concentration" "ppmv"  
state real mbo ikjftb chem 1 - irhusdf=(bdy_interp:dt) "mbo" "MBO concentration" "ppmv"
```

The new package RACM_SOA_VBS_KPP chem_opt==108

Chem array:

G a s e s :
so2,sulf,no2,no,o3,hno3,h2o2,ald,hcho,op1,op2,paa,ora1,ora2,n2o5,no3,pan,hc3,hc5,hc8,eth,co,ete,
olt,oli,tol,xyl,aco3,tpan,hono,hno4,ket,gly,mgly,dcb,onit,csl,iso,co2,ch4,udd,hket,api,lim,dien,macr,hac
e, ishp,ison,mahp,mpan,nald, **sesq,mbo**, cvasoa1,cvasoa2,cvasoa3,cvasoa4,
cvbsoa1,cvbsoa2,cvbsoa3, cvbsoa4,ho,ho2,

Aerosols:

so4aj,so4ai,nh4aj,nh4ai,no3aj,no3ai, asoa1j,aso1i,aso2j,aso2i,aso3j,aso3i,aso4j,aso4i,
bsoa1j,bsoa1i,bsoa2j,bsoa2i,bsoa3j,bsoa3i,bsoa4j,bsoa4i,
orgpaj,orgpai,ecj,eci,p25j,p25i,antha,seas,soila,nu0,ac0,corn

All species within “chem” array are advected and mixed by WRF-CHEM!

Some species are part of the “misc” array (not advected)

\$WRF-CHEM/Registry/registry.chem:

non-transported radical species for the RACM mechanism

state	real addt	ikj	misc	1	-	r	"addt"	"Radicals" "ppm"
state	real addx	ikj	misc	1	-	r	"adx"	"Radicals" "ppm"
state	real addc	ikj	misc	1	-	r	"adc"	"Radicals" "ppm"
state	real etep	ikj	misc	1	-	r	"etep"	"Radicals" "ppm"
state	real oltp	ikj	misc	1	-	r	"oltp"	"Radicals" "ppm"
state	real olip	ikj	misc	1	-	r	"olip"	"Radicals" "ppm"
state	real cslp	ikj	misc	1	-	r	"cslp"	"Radicals" "ppm"
state	real limp	ikj	misc	1	-	r	"limp"	"Radicals" "ppm"
state	real hc5p	ikj	misc	1	-	r	"hc5p"	"Radicals" "ppm"
state	real hc8p	ikj	misc	1	-	r	"hc8p"	"Radicals" "ppm"
state	real tolp	ikj	misc	1	-	r	"tolp"	"Radicals" "ppm"
state	real xylp	ikj	misc	1	-	r	"xylp"	"Radicals" "ppm"
state	real apip	ikj	misc	1	-	r	"apip"	"Radicals" "ppm"
state	real isop	ikj	misc	1	-	r	"isop"	"Radicals" "ppm"
state	real hc3p	ikj	misc	1	-	r	"hc3p"	"Radicals" "ppm"

Make a new subdirectory in \$WRFV-CHEM/chem/KPP/mechanisms/

The name of this directory should be the same as the package name in the Registry without the “_kpp” suffix.

\$WRF-CHEM/chem/KPP/mechanisms/**racm_soa_vbs**

Copy these files from chem/KPP/mechanisms/racm_esrlsorg to the new subdirectory and rename them except “atoms_red”:

atoms_red

racm_soa_vbs.def

racm_soa_vbs.eqn

racm_soa_vbs.kpp

racm_soa_vbs.spc

racm_soa_vbs_wrfkpp.equiv

racm_soa_vbs.def file

If necessary update equation sets and rate constants etc.

```
#include atoms_red
#include ./racm_soa_vbs.spc
#include ./racm_soa_vbs.eqn

#INLINE F90_RATES
REAL(KIND=dp) FUNCTION k46( TEMP, C_M )
    REAL(KIND=dp), INTENT(IN) :: temp, c_m
    REAL(KIND=dp) :: k0, k2, k3

    k0=2.4E-14_dp * EXP(460._dp/TEMP)
    k2=2.7E-17_dp * EXP(2199._dp/TEMP)
    k3=6.5E-34_dp * EXP(1335._dp/TEMP) * c_m

    k46=k0+k3/(1+k3/k2)

END FUNCTION k46
```

racm_soa_vbs.spc file:

```
#DEFVAR  
  
O3 =IGNORE ;  
  
H2O2 =IGNORE ;  
  
.....  
  
SESQ =IGNORE ;  
  
MBO =IGNORE ;
```

racm_soa_vbs.kpp file:

```
#MODEL racm_soa_vbs  
  
#LANGUAGE Fortran90  
  
#DOUBLE ON  
  
#INTEGRATOR WRF_conform/rosenbrock  
  
.....
```

racm_soa_vbs.eqn file:

```
#EQUATIONS {} ;  
  
{001:J01} NO2+hv=O3P+NO : j(Pj_no2) ;  
  
{002:J02} O3+hv=O1D{+O2} : j(Pj_o31d) ;  
  
{003:J03} O3+hv=O3P{+O2} : j(Pj_o33p)  
  
{004:J04} HONO+hv=HO+NO : j(Pj_hno2) ;  
  
{005:J05} HNO3+hv=HO+NO2 : j(Pj_hno3) ;  
  
.....  
  
chemical reactions:  
  
{024:001} O3P+M{O2}=O3 : (C_M *6.00D-34*(TEMP/300.0)**(-2.4)) ;  
  
{025:002} O3P+O3=M {2O2} : ARR2( 8.00D-12 , 2060.0_dp, TEMP) ;  
  
{026:003} O1D + M = O3P : .78084*ARR2(2.15D-11 , -110.0_dp, TEMP) +  
                           .20946*ARR2( 3.30D-11 , -55.0_dp , TEMP ) ;  
  
{027:004} O1D+H2O=HO+HO : ARR2( 1.63D-10 , -60.0_dp, TEMP ) ;  
  
{028:005} O3+HO=HO2{+O2} : ARR2( 1.70D-12 , 940.0_dp, TEMP ) ;
```

Adding new reactions to racm_soa_vbs.eqn file

{245:222} SESQ+HO=0.36 KET+0.3 HCHO+0.05 ORA1+0.19 OLIP

{246:223} SESQ+O3=0.51 HCHO+0.85 ALD+0.039 ORA1+0.23 KET+0.053 ORA2
+0.63 HO

{247:224} SESQ+NO3=0.9 OLNN+0.10 OLND+0.9 MACR

{248:225} MBO+HO=OLIP

{249:226} MBO+NO3=0.11 OLNN+0.89 OLND

{250:227} MBO+O3=0.02 HCHO+0.99 ALD+0.16 KET+0.30 CO+0.011 H2O2
+0.14 ORA2+0.07 CH4+0.22 HO2+0.63 HO+0.23 MO2
+0.12 KETP+0.06 ETH+0.18 ETPH

TEMP) ;

Reference:

Papiez, M. R., et al. (2009), The impacts of reactive terpene emissions from plants on air quality in Las Vegas, Nevada, *Atmos. Environ.*, 43(27), 4109-4123

Update some \$WRF-CHEM/chem/ subroutines

In order to call necessary subroutines for the new chemical mechanism, e.g. SO₂->SO₄ conversion

chem_driver.F:

```
.....  
so2so4_select: SELECT CASE(config_flags%chem_opt)  
CASE (RADM2SORG,RADM2SORG_KPP,RACMSORG_KPP, RACM_SOA_VBS_KPP)  
CALL wrf_debug(15,'gocart so2-so4 conversion')  
call so2so4(chem,p_so2,p_sulf,p_h2o2,p_QC,T_PHY,MOIST,      &  
.....
```

Also the following subroutines in chem/

chemics_init.F

module_input_chem_data.F

mechanism_driver.F

cloudchem_driver.F

module_wetscav_driver.F

aerosol_driver.F

dry_dep_driver.F

emissions_driver.F

module_bioemi_megan2.F

module_add emiss_burn.F

module_ftuv_driver.F

optical_driver.F

module_optical_averaging.F

...

Update chem/convert_emiss.F and generate new anthropogenic and biogenic emissions input (wrfchemi_* and wrfbiochemi* files)

```
.....  
#ifdef DM_PARALLEL  
    IF (wrf_dm_on_monitor()) THEN  
        READ(26,'(12E9.2)') dumc1(ids:ide-1,jds:jde-1)  
    ENDIF  
    DM_BCAST_MACRO(dumc1)  
#else  
    READ(26,'(12E9.2)') dumc1(ids:ide-1,jds:jde-1)  
#endif  
    grid%sebio_sesq(ips:ipe ,jps:jpe ) = dumc1(ips:ipe ,jps:jpe )  
ENDIF  
.....
```

After compiling WRF-CHEM with KPP option

\$WRF-CHEM/chem/KPP/mechanisms/racm_soa_vbs:

Makefile -> ../../util/Makefile_kpp	racm_soa_vbs_Main.f90
Makefile_racm_soa_vbs	racm_soa_vbs_Model.f90
atoms_red	racm_soa_vbs_Monitor.f90
racm_soa_vbs.def	racm_soa_vbs_Parameters.f90
racm_soa_vbs.eqn	racm_soa_vbs_Precision.f90
racm_soa_vbs.kpp	racm_soa_vbs_Rates.f90
racm_soa_vbs.spc	racm_soa_vbs_Update_Rconst.f90
racm_soa_vbs_wrfkpp.equiv	racm_soa_vbs_Util.f90
racm_soa_vbs.map	racm_soa_vbs_mex_Fun.f90
racm_soa_vbs_Function.f90	racm_soa_vbs_mex_Jac_SP.f90
racm_soa_vbs_Global.f90	
racm_soa_vbs_Initialize.f90	
racm_soa_vbs_Integrator.f90	
racm_soa_vbs_Jacobian.f90	
racm_soa_vbs_JacobianSP.f90	
racm_soa_vbs_LinearAlgebra.f90	

WKC generated racm_soa_vbs.map file

Variable species

1 = SULF (n) 32 = HKET (r) 63 = LIMP (r)

2 = CO2 (n) 33 = O3P (r) 64 = HC5P (r)

3 = ORA1 (n) 34 = PHO (r) 65 = HC8P (r)

4 = ORA2 (n) 35 = H2O2 (r) 66 = HCHO (r)

5 = CVASOA1 (n) 36 = ADDT (r) 67 = TOLP (r)

6 = CVASOA2 (r) 37 = ADDX (r) 68 = XYLP (r)

7 = CVASOA3 (r) 38 = ETE (r) 69 = OLIP (r)

8 = CVASOA4 (r) 39 = ADDC (r) 70 = ONIT (r)

9 = CVBSOA1 (n) 40 = PAA (r) 71 = DCB (r)

10 = CVBSOA2 (r) 41 = ISON (r) 72 = XO2 (r)

11 = CVBSOA3 (r) 42 = SESQ (r) 73 = OLI (r)

18 = HC5 (r) 49 = MBO (r) 80 = HC3P (r)

Additional notes

- 1) Always run “clean -a” command after you change any of KPP files
- 2) When you add chemical species for a new chem_opt to registry.chem, place the gases between “so2 ... ho2” and if you add aerosols then place them after “so4aj...”
- 3) Only species within “chem” array (not “misc”!) are used to initialize from previous simulation data when chem_in_opt=1
- 4) Check if the added mechanisms work with pre-existing initial and boundary conditions, emissions, photolysis rates, aerosol modules, dry and wet deposition rates: e.g. check module_dep_simple.F
- 5) You can also simulate some species as passive tracers such as chem_opt=13-15 (prescribed emissions only), e.g. CO
- 6) chem_opt=16,17 (with modeled fluxes and prescribed emissions for CO2 and CH4)



QUESTIONS ?

+O (CH₃)