

Chapter 2.2

Question 1

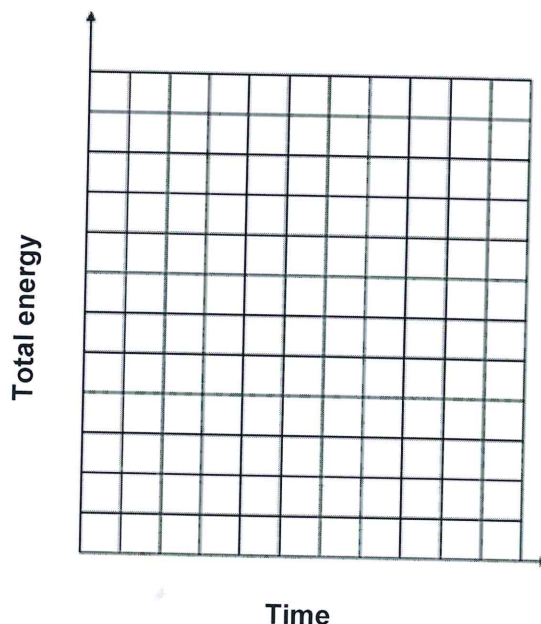
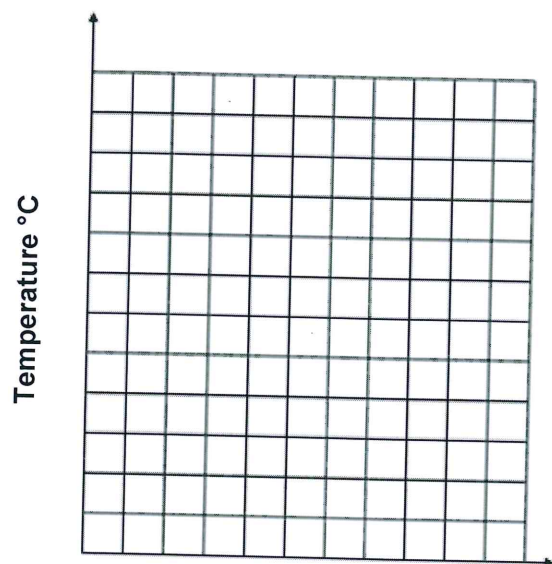
WACE Q

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(5 marks)

An experiment is conducted in which 500 mL of ice at 0.00°C is placed into a beaker. An electric heater with a constant power output is used to heat the beaker.

- (a) Sketch graphs on the axes below showing the temperature change over time, and the energy input over time, as the ice is melted and then brought to a boil. Label melting point and boiling point on the temperature graph. (3 marks)

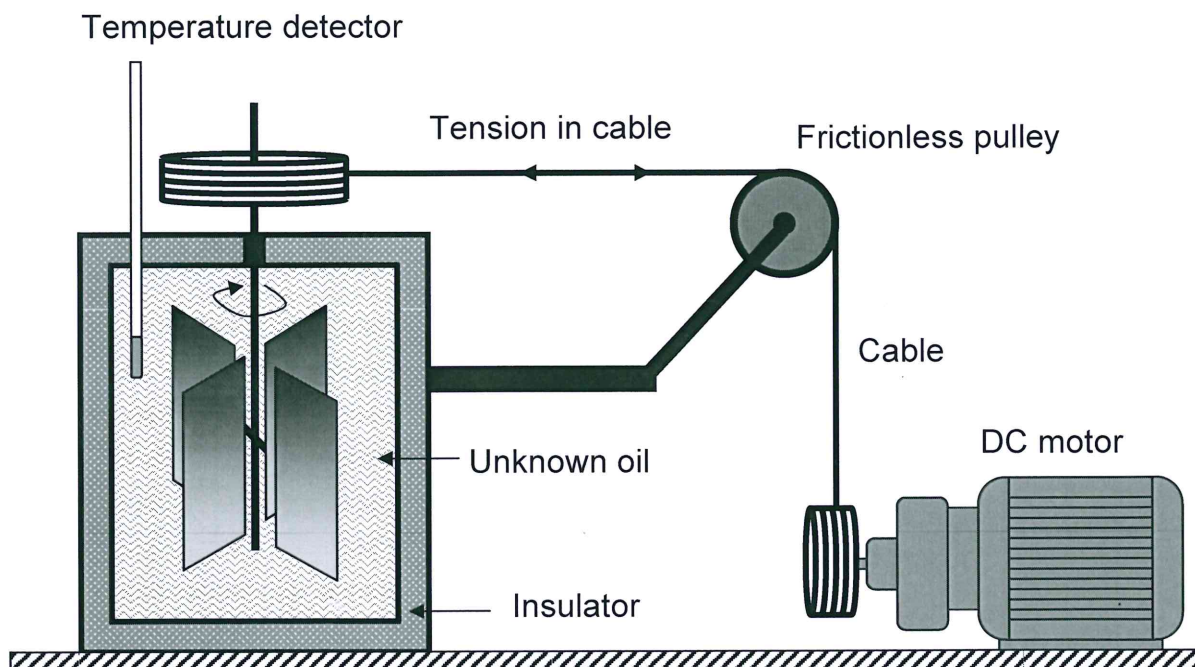


- (b) The experiment was repeated with a small fan placed on the top of the beaker. Explain how this would affect the melting and boiling results. (2 marks)

Question 2

(13 marks)

A modified Joule's heat apparatus, shown in the figure below, can be used to determine the specific heat capacity of an unknown liquid. A 40.0 watt 12.0 volt DC motor provides the necessary energy input to the oil through the motion of the cable and paddle. A student wants to use this apparatus to distinguish whether a liquid is olive oil or generic vegetable oil. 0.500 kg of oil is used in the experiment.



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Question 2 cont

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- (a) Draw a free body diagram showing the forces acting on the frictionless pulley. (2 marks)



- (b) During one trial, the DC motor winds up the cable at a constant velocity for 50 seconds. Calculate the work done by the motor. (2 marks)
- (c) Using previous experimental results, the device was determined to have an 80% energy conversion efficiency from the electric motor to the oil.
- (i) Explain what 'efficiency' means in this context. (2 marks)
- (ii) Using information from parts (b) and (c), calculate the energy input into the oil. (1 mark)
- (iii) What is one possible source of energy loss in the system? (1 mark)
- (d) After several more trials, the results of this experiment are averaged and can be summarised as
'the temperature of this oil is raised 1.7°C from an energy input to the oil of $1.8 \times 10^3 \text{ J}$ '.
- (i) Calculate the specific heat capacity of the liquid using these values. (3 marks)
- (ii) A search on the internet finds the specific heat capacities for olive oil and generic vegetable oil listed as 1.97×10^3 and $1.67 \times 10^3 \text{ J kg}^{-1} \text{ K}^{-1}$, respectively. Explain which oil you think is in the apparatus. (2 marks)