

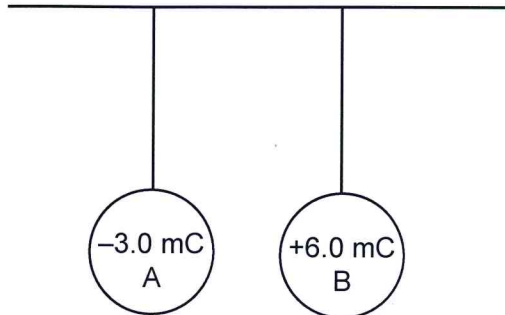
# Chapter 5.1 Exam Q

## Question 1

Year 1

(3 marks)

A student suspended two identical metallic spheres, 'A' and 'B', from an insulated wooden rod, and applied a charge to each, as shown below.



- (a) Did the spheres attract or repel each other? Circle the correct response: (1 mark)
- (b) (i) The student touched the spheres together, and they then moved apart. Determine the overall charge, in coulombs, on the pair of spheres after they were touched together. (1 mark)
- (ii) Determine the charge on each sphere after they had separated. (1 mark)

Charge on A: \_\_\_\_\_ Charge on B: \_\_\_\_\_

# Chapter 5.1 Exam Q

## Question 2

Year 1

(4 marks)

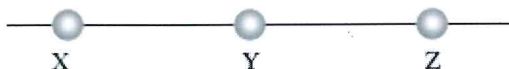
E, F, G and H are very small glass balls. F, G and H are charged, but it is not known what type of charge they possess. In order to test the types of charge on these balls, we charge glass ball E by rubbing it with silk. This removes electrons from the glass ball.

- (a) Once E has been charged, we find E attracts F and F repels G, but G attracts H. Which are two negatively charged balls? (1 mark)

- A E and F
- B F and H
- C E and G
- D F and G

Correct answer: \_\_\_\_\_

Three balls are selected from these four balls and renamed 'X', 'Y' and 'Z'. These three balls are placed in a straight line with a reasonable distance between them. Balls X and Z are fixed in place and Y is kept in equilibrium through electrostatic forces as shown below.



- (b) For Y to stay in a stable position, the charges on the spheres need to be (1 mark)

- A X is positive, Y is positive and Z is negative.
- B X is positive, Y is negative and Z is positive.
- C X is negative, Y is positive and Z is positive.
- D X is negative, Y is negative and Z is positive.

Correct answer: \_\_\_\_\_

- (c) An atom has two electrons removed from it. Which statement is correct? (2 marks)

- A The atom becomes a different isotope.
- B The atom becomes more positive.
- C The atom becomes more negative.
- D The atom is unchanged.

Correct answer: \_\_\_\_\_

Explain your choice.