

Page No.	
Date	

Name :- Hemil Chovatiya

E. No :- T0120IT219

Branch :- Information Technology

Division :- IT 3

Institute :- PIET

1) Gallium Arsenide (GaAs) - Infrared.

$$E_g (\text{Band gap}) = 1.3 \text{ V} \quad \lambda = 780 \text{ nm to } 1 \text{ mm.}$$

$$h = 6.6 \times 10^{-34} \text{ J s}$$

$$c = 3 \times 10^8 \text{ m/s.}$$

$$E_g = \frac{hc}{\lambda}$$

$$0.013 \times 1.6 \times 10^{-19} = \frac{6.6 \times 10^{-34}}{\lambda} \times 3 \times 10^8$$

$$\lambda = 955 \text{ nm}$$

2) Gallium Arsenide Phosphide (GaAsP):-

$$E_g = 1.881 \quad \lambda = 590 \text{ to } 960 \text{ nm.}$$

- Red to Infrared; Orange

$$E_g = \frac{hc}{\lambda}$$

$$0.01881 \times 1.6 \times 10^{-19} = \frac{6.6 \times 10^{-34}}{\lambda} \times 3 \times 10^8$$

$$\lambda = 660 \text{ nm}$$

3) Aluminum Gallium Arsenide Phosphide.
(AlGaAsP)

$$E_g = 1.82 \text{ V} \quad \lambda = 560 \text{ nm to } 700 \text{ nm}$$

(High Brightness red, orange-red,
orange and yellow)

$$E_g = \frac{hc}{\lambda}$$

$$0.01825 \times 1.6 \times 10^{-19} = \frac{6.6 \times 10^{-34}}{\lambda} \times 3 \times 10^8$$

$$\lambda = 680 \text{ nm}$$

4) Gallium phosphide (Ga-P) $E_g = 2.25 \text{ V}$ red, yellow and green.

$$E_g = \frac{hc}{\lambda} \quad 520 - 700 \text{ nm}$$

$$E_g = \frac{hc}{\lambda}$$

$$0.0225 \times 10^2 \times 1.6 \times 10^{-19} = \frac{6.6 \times 10^{-34} \times 3 \times 10^8}{\lambda}$$

$$\lambda = 550 \text{ nm}$$

5) Aluminium gallium phosphide (Al-Ga-P)

$$E_g = 2.175 \text{ V} \quad \text{green} \quad 520 - 570 \text{ nm}$$

$$E_g = \frac{hc}{\lambda}$$

$$0.02175 \times 10^2 \times 1.6 \times 10^{-19} = \frac{6.6 \times 10^{-34} \times 3 \times 10^8}{\lambda}$$

$$\lambda = 570 \text{ nm}$$

6) Gallium Nitride (Ga-N) green, emerald green.

$$E_g = 2.25 \quad 520 - 560 \text{ nm}$$

$$E_g = \frac{hc}{\lambda}$$

$$0.0225 \times 10^2 = \frac{6.6 \times 10^{-34} \times 3 \times 10^8}{\lambda \times 1.6 \times 10^{-19}}$$

$$\lambda = 550 \text{ nm}$$

7) Gallium Indium Nitride (Ga-In-N)

near Ultraviolet, blush green and blue, 100-550nm

$$E_g = 2.75 \text{ V}$$

$$0.027 \times 10^2 \times 1.6 \times 10^{-19} = \frac{6.6 \times 10^{-34} \times 3 \times 10^8}{\lambda}$$

$$\lambda = 450 \text{ nm}$$

Page No.	
Date	

8) Silicon Carbide :-

Range of $\lambda = 500 \text{ } \text{Å} < \lambda < 570$.

Band gap = 2.2 eV

$$\lambda = \frac{19.87 \times 10^{-26}}{2.2 \times 1.6 \times 10^{-19}}$$

$$\lambda = 564 \text{ nm}$$