1 Degree of freedom of a system with external forces

From MD simulations with different $N(CH_4)$, we compute average kinetic energy E_k of CH_4 as listed in **Table 1**.

Table 1: Average E_k of CH_4 for different $N(CH_4)$

\overline{N}	$\frac{E_k}{N}$ (kJ/mol)
4	2.78
8	3.23
16	3.48
40	3.61
80	3.66
100	3.67
120	3.67
140	3.68
160	3.68

According to the equipartition theorem, energy carried from every degree of freedom (DOF) is equal to $\frac{1}{2}k_{\rm B}T$, and the sum of the energies for all DOF $(N_{\rm DOF} \times \frac{1}{2}k_{\rm B}T)$, is equal to total kinetic energy E_k . From this discussion, we obtain the relation between total kinetic energy E_k and temperature T as **Equation (1)**. To calculate total DOF $(N_{\rm DOF})$ of CH₄ group, we assume it has the form of " $N_{\rm DOF} = aN - b$ " and perform a linear regression with data of E_k and N from **Table 1**, T=298 K, $N_{\rm DOF}$ is equal to 3N - 3 (**Equation (2)**).

$$E_k = \sum_{N=1}^{i} \frac{1}{2} m v_i^2 = \frac{1}{2} k_{\rm B} T \times N_{\rm DOF}$$
 (1)

$$N_{\text{DOF}} = 3.01 \times N - 3.00 \tag{2}$$

This result supports the idea that the calculation of degrees of freedom in LAMMPS is implemented in a way which keeps total linear momentum conserved and center of mass (COM) fixed, despite of the existence of an external force field. This correction of DOF has an effect on the velocities of CH_4 , and, consequently, on the E_k distribution.

The conservation of total linear momentum determines total DOF of CH_4 is equal to 3N-3 instead of 3N, which cause the anomalous relation. Indeed, to use the correct DOF estimation, we modify the source codes of LAMMPS related to the DOF calculations in the NVT simulation.

Lammps-version/src/:

fix_nvt.cpp *call fix_nh.cpp*\
fix_nh.cpp: *call "dof" variable from compute_temp.cpp*\

tdof=temperature->dof

kecurrent=tdof*boltz*t_current

compute.cpp: *set default value of "extra_dof"*\

 $\$ "fix_dof"*\

extra_dof=domain->dimension

fix_dof=0

compute_temp.cpp: *definition of "dof" variable*\

dof=domain->dimension*natoms_temp

dof-=extra_dof+fix_dof

For using Nose-Hoover thermostat, four *.cpp files are the most relevant. fix_nvt.cpp and fix_nh.cpp include all implementations of Nose-Hoover thermostat, and refer to variable dof to compute E_k . compute.cpp and compute_*.cpp include implementations of computation of certain property, e.g. compute_temp.cpp for temperature calculation.

In compute_temp.cpp, we find the definition of $N(CH_4)$ for a 3-dimensional system, dof=3N-extra_dof-fix_dof. In compute.cpp, default value of extra_dof is assigned to be equal to dimension, i.e. 3 for 3-dimensional system.

In conclusion, LAMMPS assigns the total DOF of our system to be (3N-3) instead of 3N, which causes the lower total kinetic energy E_k compared to the reference value, 3.70 kJ/mol (**Equation (3)**), and incorrect E_k distribution compared to the reference canonical distribution (**Equation (4)**).

Canonical distribution at 298 K

$$E_k \text{ per mol} = \frac{3}{2} k_{\rm B} T N_{\rm A} = 3.70 \text{ kJ/mol}$$
 (3)

$$P = 2\left(\frac{E}{\pi}\right)^{\frac{1}{2}} \left(\frac{1}{k_B T}\right)^{\frac{3}{2}} \exp\left\{-\frac{E}{k_B T}\right\} \tag{4}$$

In other words, when we perform a NVT simulation at 298 K using LAMMPS, the target temperature is never achieved, more significant for small numbers of guest molecules, due to loss of energies of 3 DOF, and, in addition, the effective temperature of our simulation is $T_{\rm eff} = \frac{N-1}{N} \times 298$ K. The effective temperature $T_{\rm eff}$ is lower than the imposed one, leading to slower diffusion of CH₄ and smaller diffusion coefficients D_s , which explains the anomalous relation between $N({\rm CH_4})$ and D_s .

To obtain the correct results of D_s at 298 K, we modify compute_temp.cpp, use dof=3N-fix_dof instead of dof=3N-extra_dof-fix_dof. By removing extra_dof variable, LAMMPS is able to calculate the total DOF of CH₄ as 3N.

The modified Lammps is available in 'https://github.com/HengluXu/lammps.git'