

1. A neutral atom with atomic number 6 will have 6 electrons and 6 protons.
2. The maximum number of electrons that can exist in an given atom shell n is calculated by the formula

$$N = 2n^2$$

This means that the maximum number of electronics in the third shell of an atom is

$$2 \cdot 3^2 = 2 \cdot 9 = 18$$

3. Materials are categorized into three groups called isolators, semiconductors, and conductors. Isolators have the largest band gap between the conduction band and valence band. Semiconductors will have a smaller band gap than isolators. Conductors have an overlap between the conduction band and the valence band.
4. There are several types of atoms that have four valence electrons, for example silicon and germanium. These types of atoms are classified as semiconductors.
5. A single atom in a silicon crystal forms four covalent bounds.