About the Course Announcements

Ask a Question

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Jnit 4 - Week	1:	
Course outline		Ass
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Data Attachment		As per ou
Week 0 Assignment 0	\	1) The t
Week 1:		is TF (a)
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Lecture 02 :     Properties of Pure		(d)
Substances  Lecture 03 :		o a b c
Properties of Pure Substances (contd.)		O d
Lecture 04 :     Introduction to     Property Tables		No, the a Score: 0 Accepted
Lecture 05 :     Properties of Pure     Substances: Example     problems (contd.)		2) The 10 kl
Quiz : Assignment 1		is TI
O Feedback for Week 1		(a) (b)
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		with have Assu
		(a)
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		No, the a Score: 0
		Accepted b
		5) Cor

## is TRUE? (d) None of the above $\bigcirc$ a b

Accepted Answers:

is TRUE?

a

b

○ c

Score: 0

No, the answer is incorrect.

Assume  $g = 10 \text{ m/s}^2$ .

(a) 60 grams

(b) 102 grams

(c) 162 grams

(d) 222 grams

No, the answer is incorrect.

Common data for Question 5 to 7:

T [°C]

50

The value of  $P_1$  (in kPa) is

(a) 7.384

(b) 12.350

(c) 19.941

(d) 101.3

No, the answer is incorrect.

The value of  $x_1$  is

Accepted Answers:

(a) 0.24

(b) 0.34

(c) 0.44

(d) 0.64

No, the answer is incorrect.

(a) Compressed liquid

(b) Superheated vapour

(d) Cannot be determined

(c) Saturated liquid-vapour mixture

8) Common data for Question 8 and 9:

The value of  $v_2$  (in m<sup>3</sup>/kg) is

Determine the missing properties for water.

T [°C]

250

P [kPa]

400

 $v \, [\mathrm{m}^3/\mathrm{kg}]$ 

 $v_2$ 

 $v \, [\mathrm{m}^3/\mathrm{kg}]$ 

 $v_3$ 

 $v \, [\mathrm{m}^3/\mathrm{kg}]$ 

 $v_4$ 

x

0.75

P [kPa]

1000

T [°C]

 $T_4$ 

P [kPa]

500

Phase description

Phase 3

Phase description

Phase 2

Accepted Answers:

7) Phase 1 is

a

( b

○ c

d

Score: 0

b

○ c

 $\bigcirc$  d

Score: 0

No, the answer is incorrect.

(a) 0.595

(b) 0.773

(c) 1.01

(d) 1.24

No, the answer is incorrect.

(a) Compressed liquid

(b) Superheated vapour

(d) Cannot be determined

(c) Saturated liquid-vapour mixture

Common data for Question 10 and 11:

140

Determine the missing properties for water.

The value of  $v_3$  (in m<sup>3</sup>/kg) is

(a) 0.509

(b) 0.306

(c) 0.217

(d) 0.001

No, the answer is incorrect.

(a) Superheated vapour

(c) Compressed liquid

No, the answer is incorrect.

(a) 151.86

(b) 179.91

(c) 198.32

(d) 212.42

No, the answer is incorrect.

The value of  $v_4$  (in m<sup>3</sup>/kg) is

Accepted Answers:

(a) 0.0011

(b) 0.0978

(c) 0.1461

(d) 0.1933

No, the answer is incorrect.

Accepted Answers:

 $\circ$  a

b

○ c

d

Score: 0

Accepted Answers:

(d) Cannot be determined

(b) Saturated liquid-vapour mixture

Common data for Question 12 to 13:

Determine the missing properties for water.

The value of  $T_4$  (in °C) is

Accepted Answers:

11) Phase 3 is

Accepted Answers:

Phase 2 is

 $\circ$  a

b

C

 $\bigcirc$  d

Score: 0

○ a

( b

 $\bigcirc$  d

10)

 $\bigcirc$  a

( b

○ c

d

Score: 0

○ a

( b

0 c

d

Score: 0

 $\circ$  a

b

○ c

 $\bigcirc$  d

12)

No, the answer is incorrect.

Accepted Answers:

а

Accepted Answers:

Determine the missing properties for water.

P [kPa]

 $P_1$ 

Accepted Answers:

Accepted Answers:

(d) None of the above

No, the answer is incorrect.

(1) 100°C, 90 kPa (2) 100°C, 110 kPa (3) 100°C, 1 m³/kg

Accepted Answers:

Mentor

1 point

Assignment 1 The due date for submitting this assignment has passed. As per our records you have not submitted this assignment.

The temperature of water is increased from -10°C to 110°C keeping the pressure constant at 0.5 kPa. Which among the following statements regarding the phase of water in this process (a) Ice sublimes directly from the solid to the vapour phase.

Due on 2019-08-14, 23:59 IST.

(b) Ice melts first into a liquid and then subsequently evaporates.

d

(a) Ice sublimes directly from the solid to the vapour phase.

Determine the phase of water for each of the following states.

(c) Ice melts first into a liquid and then subsequently evaporates.

2) The temperature of water is increased from -10°C to 50°C keeping the pressure constant at

(b) Ice melts into a liquid and remains as a liquid till the temperature reaches 50°C.

(a) (1) Superheated vapour, (2) Subcooled liquid, (3) Saturated liquid-vapour mixture

(c) (1) Subcooled liquid, (2) Superheated vapour, (3) Saturated liquid-vapour mixture

A pressure cooker has the lid screwed on tight. A small opening with  $A=6 \text{ mm}^2$  is covered

with a petcock that can be lifted to let steam escape. How much mass should the petcock

(b) (1) Subcooled liquid, (2) Superheated vapour, (3) Superheated vapour

(d) (1) Superheated vapour, (2) Superheated vapour (3) Subcooled liquid

have to allow boiling at 130°C with an outside atmosphere at 100 kPa?

 $v~[\mathrm{m}^3/\mathrm{kg}]$ 

7.72

x (if applicable)

 $x_1$ 

Phase description

Phase 1

10 kPa. Which among the following statements regarding the phase of water in this process

No, the answer is incorrect. Score: 0

( c

(c) Ice melts into a liquid and remains as a liquid till the temperature reaches 110°C.

	-	-			-							-										-	-

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Progress