

Crystallography

PHY312

Insulators (high bandgap)



Rocksalt

$E_g = 3.1 \text{ eV}$



Gypsum

$E_g = 5\text{-}7 \text{ eV}$



Diamond

$E_g = 5.47 \text{ eV}$



Calcite Magnesite

$E_g = 6 \text{ eV}$

$E_g = 5.4 \text{ eV}$

Insulators (lower bandgap)



$$E_g = 2.3 \text{ eV}$$

Metals



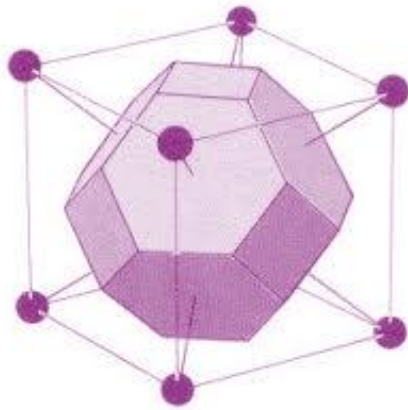
Semiconductors



GaP, 2.26 eV


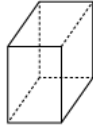
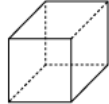
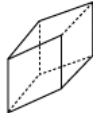
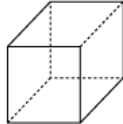
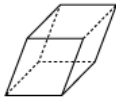
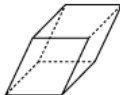


Silicon, 1.1 eV

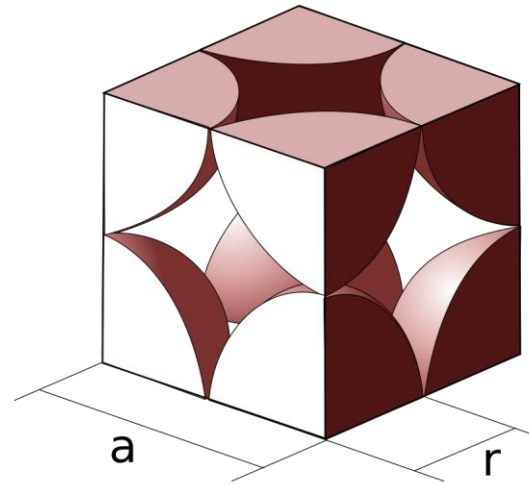
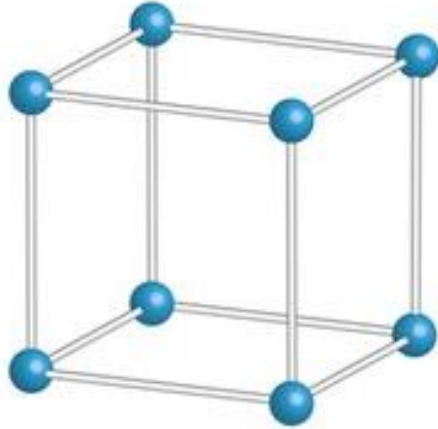


Wigner seitz cell (3D)

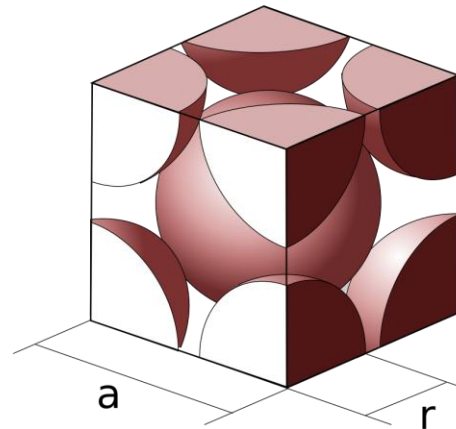
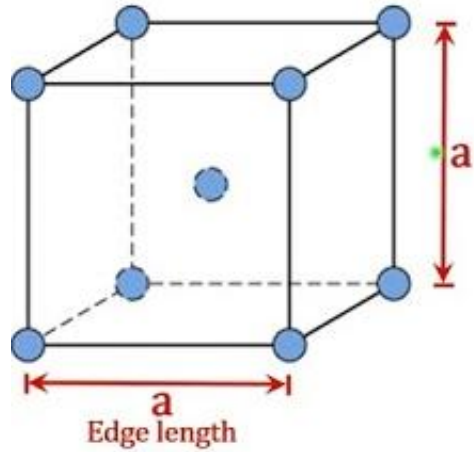
Crystal System Table

System	Axial length	Axial Angle	Unit Cell Geometry
Cubic	$a = b = c$	$\alpha = \beta = \gamma = 90^\circ$	
Tetragonal	$a = b \neq c$	$\alpha = \beta = \gamma = 90^\circ$	
Orthorhombic	$a \neq b \neq c$	$\alpha = \beta = \gamma = 90^\circ$	
Rhombohedral	$a = b = c$	$\alpha = \beta = \gamma \neq 90^\circ$	
Hexagonal	$a = b \neq c$	$\alpha = \beta = 90^\circ, \gamma = 120^\circ$	
Monoclinic	$a \neq b \neq c$	$\alpha = \gamma = 90^\circ, \beta \neq 90^\circ$	
Triclinic	$a \neq b \neq c$	$\alpha \neq \beta \neq \gamma$	

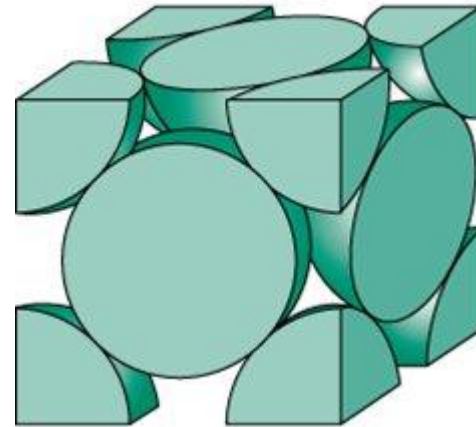
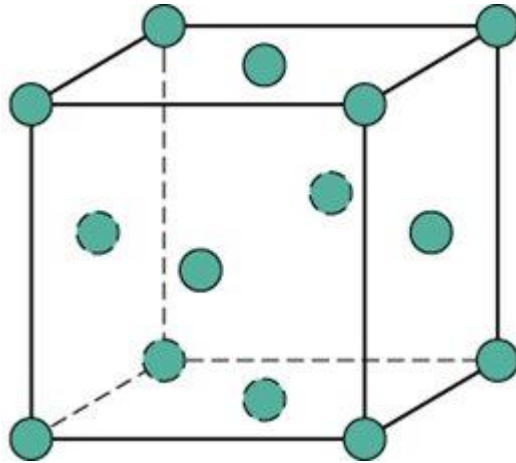
Simple cubic



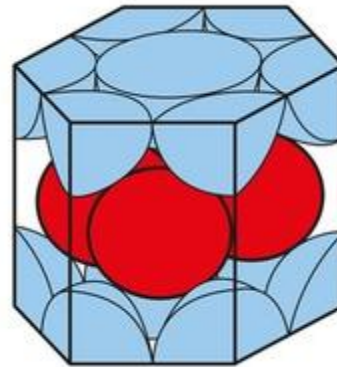
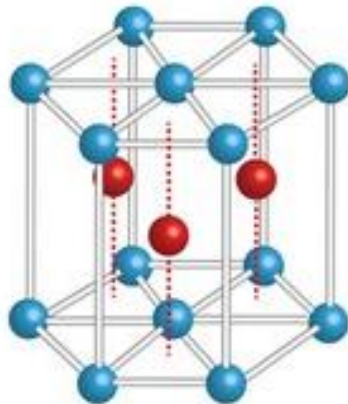
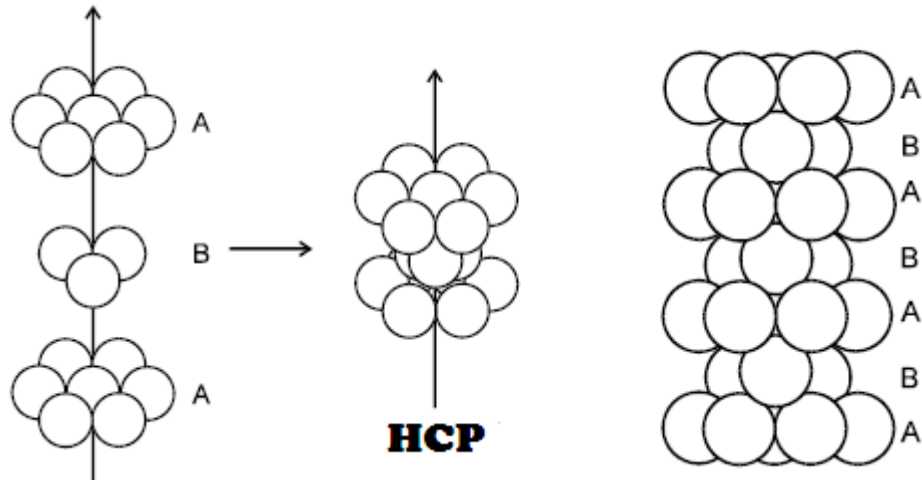
Body centered cubic



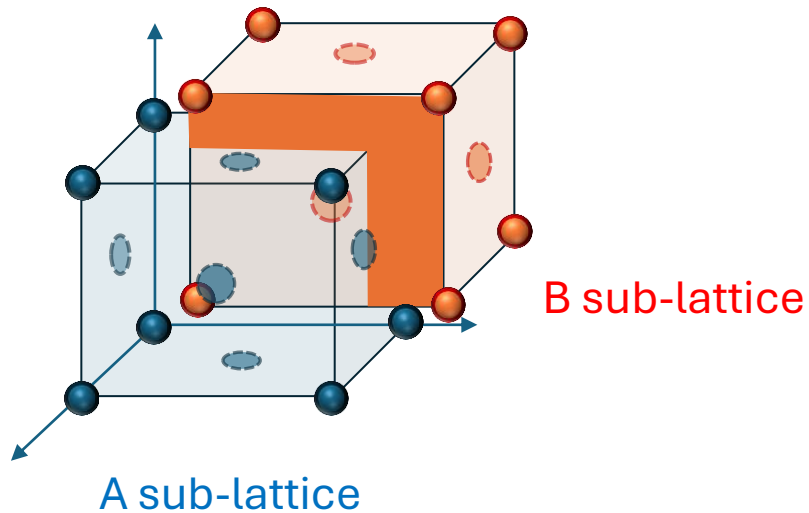
Face centered cubic



Hexagonal closed pack structure



Interpenetrating FCC structure



Co-ordinates

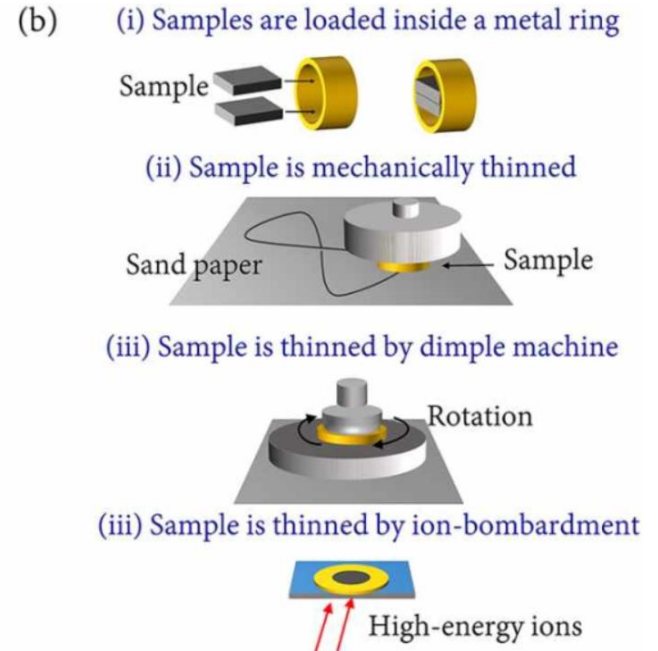
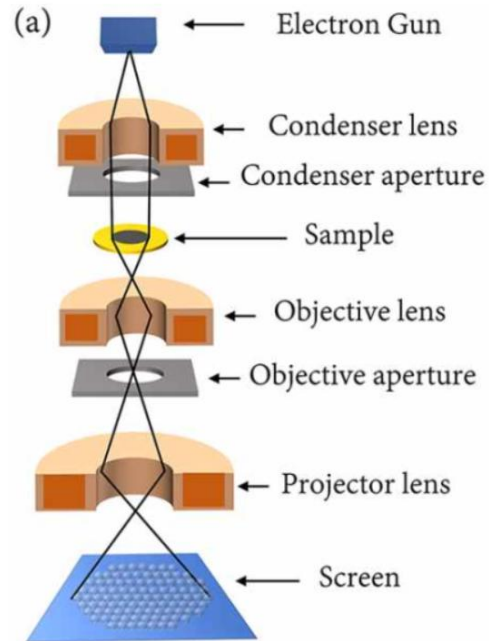
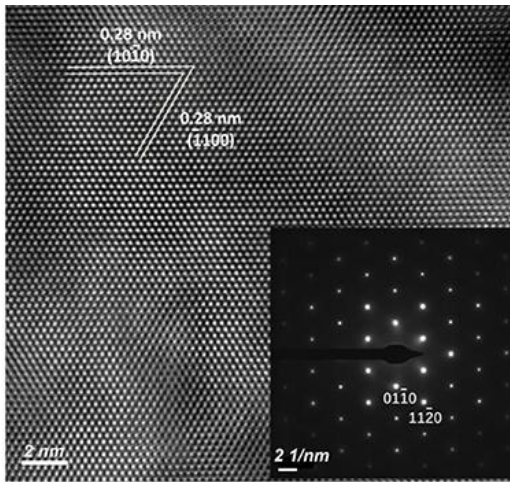
A-sublattice (0 0 0)

B-Sublattice ($\frac{1}{4}$ $\frac{1}{4}$ $\frac{1}{4}$)

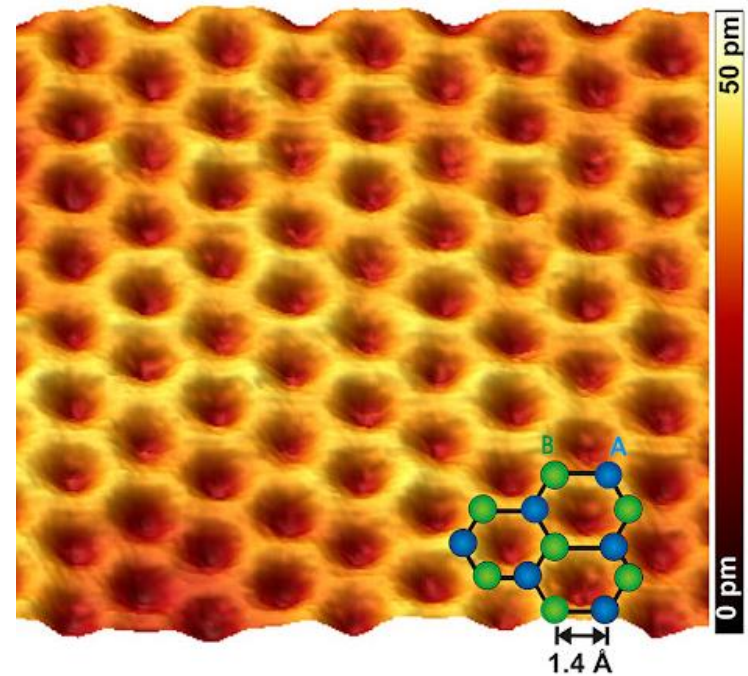
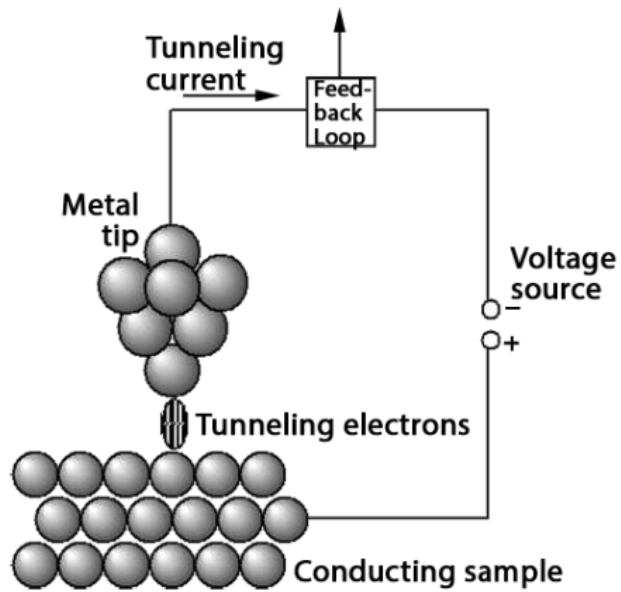
A = B, Diamond structure

A \neq B, Zinc-Blende (ZnS) structure

Transmission electron microscopy

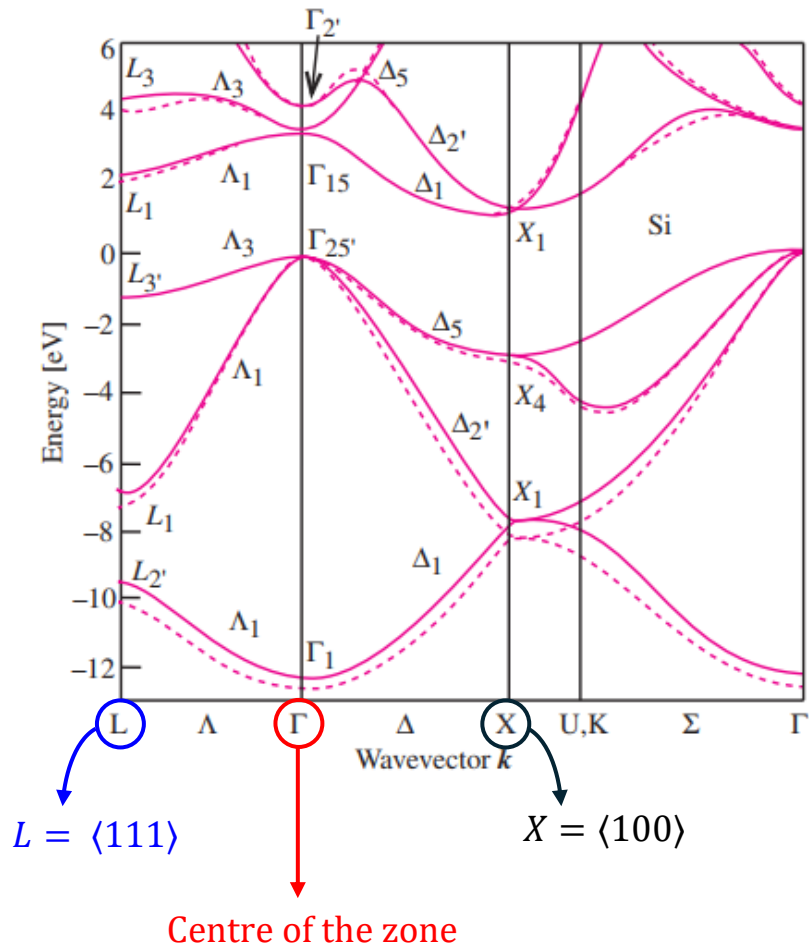


Scanning Tunnelling Microscopy

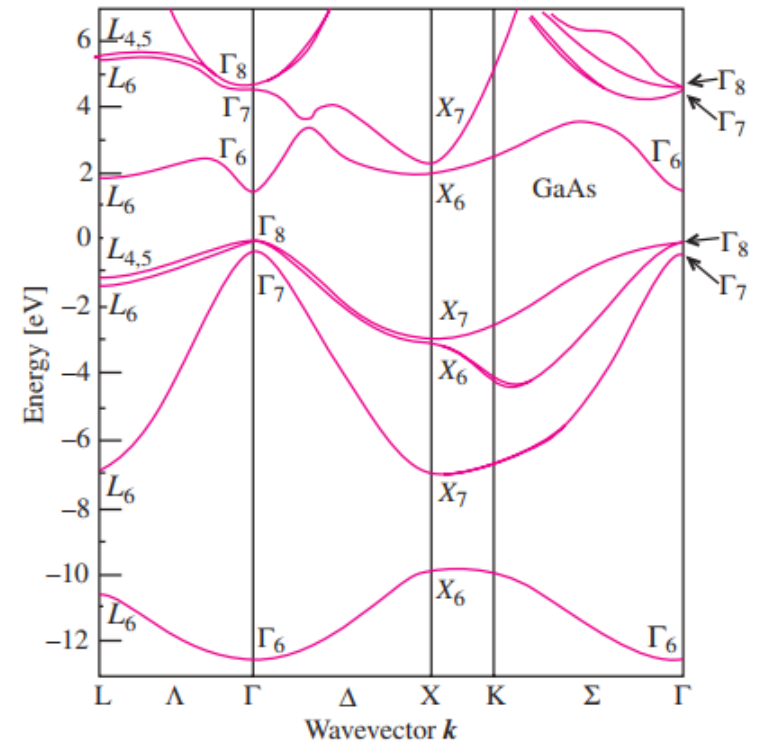


Bandstructure:

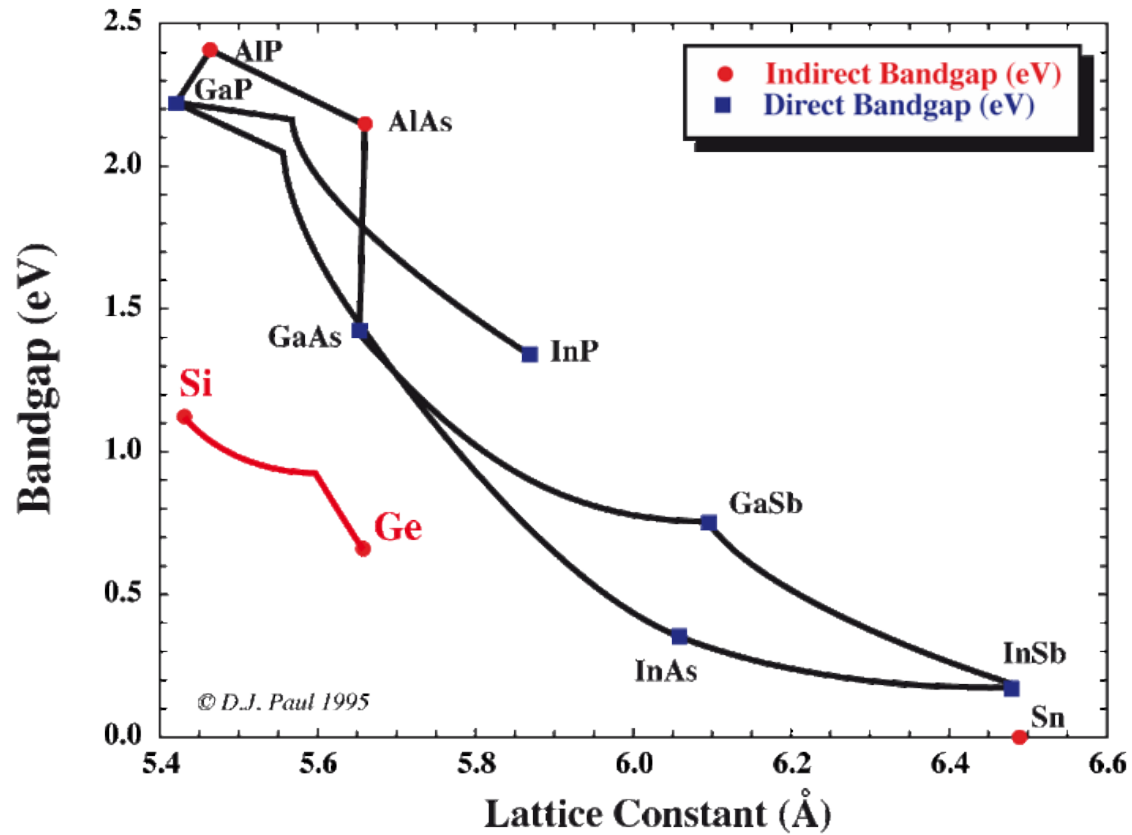
Silicon



GaAs



Bandgap for different semiconductors:



Subcategory metals, nonmetals, and metalloids																					
<div><div></div>Alkali metals</div>																		<div></div> Lanthanides		<div></div> Metalloids	
<div></div> Alkaline earth metals																		<div></div> Actinides		<div></div> Reactive non metals	
<div></div> Transition metals																		<div></div> Post transition metals		<div></div> Noble gases	
<div></div>																		<div></div> Unknown properties			

Subcategory metals, nonmetals, and metalloids

- Alkali metals
- Alkaline earth metals
- Transition metals
- Lanthanides
- Actinides
- Post transition metals
- Metalloids
- Reactive non metals
- Noble gases
- Unknown properties

57 La Lanthanum 138.91 2-8-18-18-9-2	58 Ce Cerium 140.12 2-8-18-19-9-2	59 Pr Praseodymium 140.91 2-8-18-21-8-2	60 Nd Neodymium 144.24 2-8-18-22-8-2	61 Pm Promethium [145] 2-8-18-23-8-2	62 Sm Samarium 150.36 2-8-18-24-8-2	63 Eu Europium 151.96 2-8-18-25-8-2	64 Gd Gadolinium 157.25 2-8-18-25-9-2	65 Tb Terbium 158.93 2-8-18-27-8-2	66 Dy Dysprosium 162.50 2-8-18-28-8-2	67 Ho Holmium 164.93 2-8-18-29-8-2	68 Er Erbium 167.26 2-8-18-30-8-2	69 Tm Thulium 168.93 2-8-18-31-8-2	70 Yb Ytterbium 173.05 2-8-18-32-8-2	71 Lu Lutetium 174.97 2-8-18-32-9-2
89 Ac Actinium [227] 2-8-18-32-18-9-2	90 Th Thorium 232.04 2-8-18-32-18-10-2	91 Pa Protactinium [231.04] 2-8-18-32-20-9-2	92 U Uranium 238.03 2-8-18-32-21-9-2	93 Np Neptunium [237] 2-8-18-32-22-9-2	94 Pu Plutonium [244] 2-8-18-32-24-8-2	95 Am Americium [243] 2-8-18-32-25-8-2	96 Cm Curium [247] 2-8-18-32-25-9-2	97 Bk Berkelium [247] 2-8-18-32-27-8-2	98 Cf Californium [251] 2-8-18-32-28-8-2	99 Es Einsteinium [252] 2-8-18-32-29-8-2	100 Fm Fermium [257] 2-8-18-32-30-8-2	101 Md Mendelevium [258] 2-8-18-32-31-8-2	102 No Nobelium [259] 2-8-18-32-32-8-2	103 Lw Lawrencium [266] 2-8-18-32-32-8-3