

Experiment - 4

* Aim:-

Determination of plank's Constant.

* Apparatus Required:-

0-10V power supply, a one way key, a rheostat, digital millimeter, digital voltmeter and 1k resistor and different known wavelength of LED.

* Theory:-

- 1) Planck's Constant (h), a physical constant was introduced by german physicist named max planck in 1900. The significance of Planck's Constant is that quanta (small packets of energy).
Can be determined by frequency of radiation and Planck's constant.
- 2) It describes the behavior of particles and waves at atomic level as well as the particle nature of light.
- 3) An LED is a two terminal semiconductor light source. In the unbiased condition a potential barrier is developed across the p-n junction of the LED. reduced at a particular voltage the height of potential barrier become very low and the LED starts glowing, in the forward

biased condition electron crossing the junction are excited, and when they return to their normal state, energy is emitted. This particular voltage is called the knee voltage.

The light energy emitted during forward biasing is given:-

$$E = \frac{hc}{\lambda} \longrightarrow (1)$$

where, c = velocity of light
 h = plank's constant
 λ = wave length of light.

If v is the forward voltage applied across the LED it begins to emit light (the knee voltage), the energy given to electrons crossing the junction is.

$$E = ev \longrightarrow (2)$$

Equating (1) & (2) we get

The knee voltage V can be measured for LED with different values of λ (wave length of light).

$$V = \frac{hc}{e} \left(\frac{1}{\lambda} \right) \longrightarrow (3)$$

Planck's Constant = $\frac{hc}{e}$

Using the known value $\frac{e}{c} = 5.35 \times 10^{-28} \frac{C}{m}$

Alternatively, we can write Eqn 3 as.

$$h = \frac{e}{c} \lambda V$$

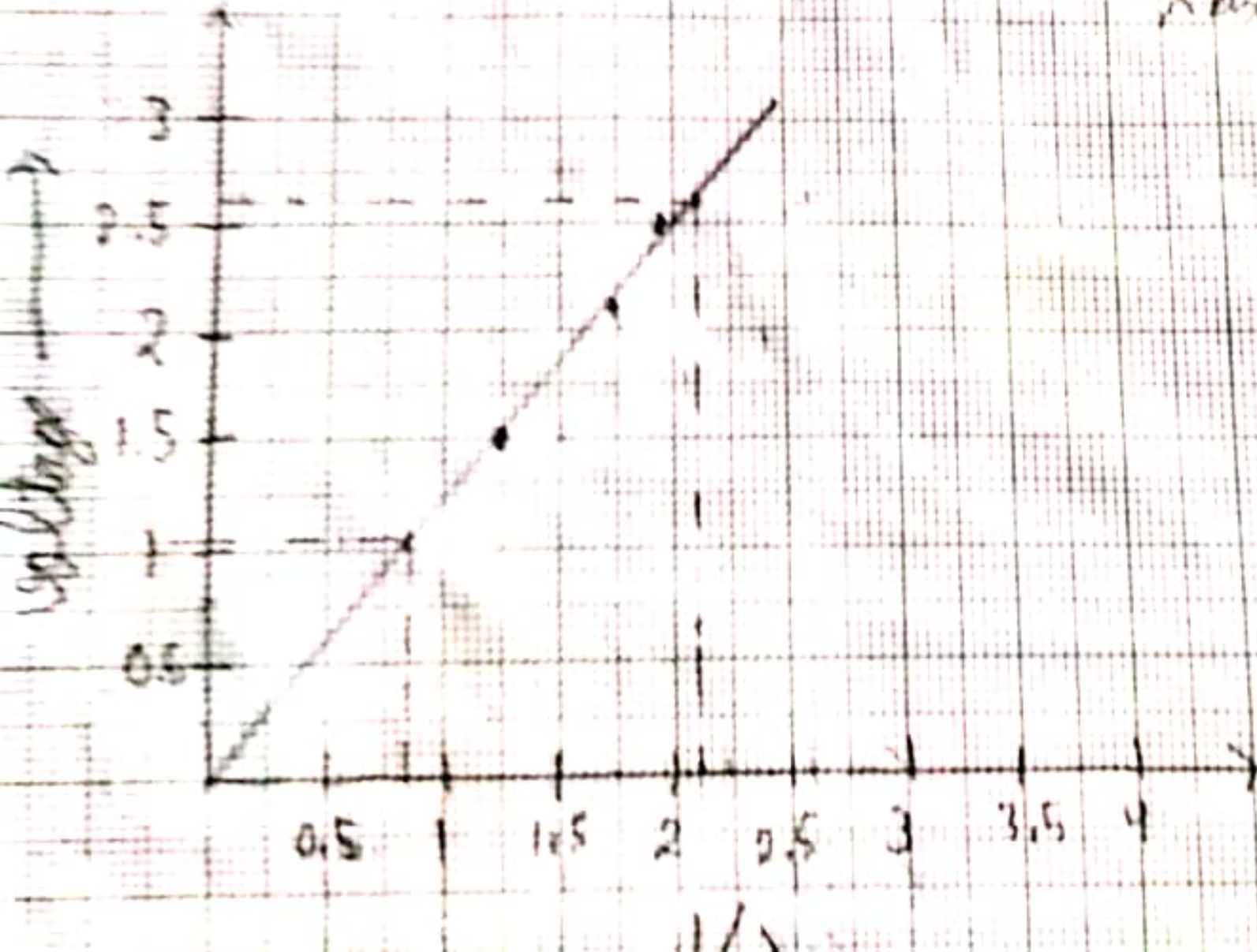
* Observation Table 1:-

	Colour	wave length	knee voltage	$\lambda \times V$
①	Red LED	650×10^{-9}	1.59	1035.5
②	yellow LED	570×10^{-9}	2.17	1236.9
③	Infrared LED	1250×10^{-9}	1.12	1018.08
④	Green LED	510×10^{-9}	2.43	1239.3
⑤	Blue LED	476×10^{-9}	2.61	1246.2

* Observation Table 2:-

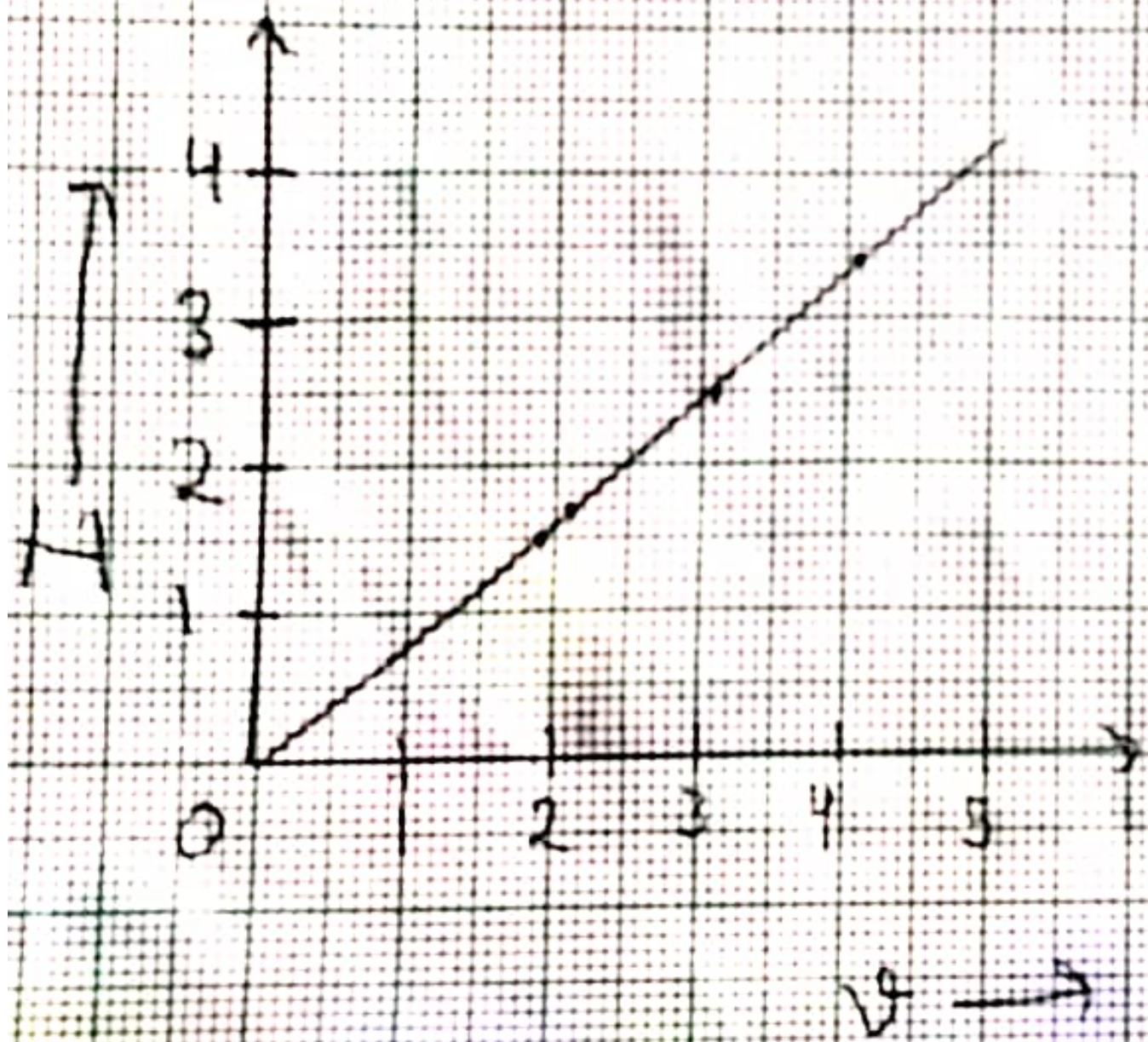
S.No	voltage (V)	Current (I)
1)	1.908	1.592
2)	2.102	1.751
3)	3.100	2.584
4)	4.104	3.420

Y-axis \Rightarrow Force - 0.5 unit
X-axis \Rightarrow Time - 0.5 unit



X-axis) ~~time~~ = 1 unit

Y-axis) ~~time~~ = 1 unit



* Calculations:-

$$\text{Slope} = \frac{y_2 - y_1}{x_2 - x_1} = \frac{2.43 - 2.17}{(1.96 - 1.75) \times 10^6}$$

$$\Rightarrow \frac{0.26}{0.21} \times 10^{-6} = 1.2 \times 10^{-6}$$

$$\text{Slope} = \frac{h\nu}{e}$$

$$\text{Slope} \times \frac{e}{c} = h$$

$$\text{where, } h = 1.2 \times 10^{-6} \times 5.33 \times 10^{-28}$$

$$h = 6.396 \times 10^{-34}$$

$$\approx 6.626 \times 10^{-34}$$

From the graph we get,

$$\frac{1}{\lambda} = 0.8 \times 10^{12}$$

$$\frac{1}{\lambda} = 0.8 \times 10^{12}$$

$$\lambda = \frac{1}{0.8} \times 10^{-12}$$

$$\text{Also, } \lambda = 1.25 \times 10^{-12}$$

$$\lambda = 1250 \text{ nm}$$

wavelength of Infrared LED = 1250 nm

* Results:-

Planck's Constant = 6.626×10^{-34}

wavelength of IR LED = 1250 nm

wavelength of blue LED = 476 nm