

Predicting anti-HIV-1 activity of 6-arylbenzonitriles: Computational approach using supraaugmented eccentric connectivity topochemical indices

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Abstract

Highly discriminating adjacency-cum-distance based topochemical indices termed as supraaugmented eccentric connectivity topochemical indices for quantitative structure–activity and structure–property relationships (QSAR/QSPR) have been conceptualized in the present study. These indices were found to exhibit high sensitivity towards the presence and relative position of heteroatom(s), exceptionally high discriminating power and negligible degeneracy for all possible structures of five vertices containing one heteroatom. Utility of these indices was investigated for development of models for prediction of anti-human immunodeficiency virus (HIV)-1 activity using a data set comprising 81 differently substituted 6-arylbenzonitriles. The values of the supraaugmented eccentric connectivity topochemical indices of all the analogues comprising the data set were computed using an in-house computer program. The resultant data was analyzed and suitable models were developed after identification of the active ranges. Subsequently, a biological activity was assigned to each analogue using these models which was then compared with the reported anti-HIV-1 activity. The accuracy of prediction was found to be ~81% for all the three topochemical models. High sensitivity towards presence and relative position of heteroatom(s), exceptionally high discriminating power amalgamated with low degeneracy offer proposed topochemical indices vast potential for isomer discrimination, similarity/dissimilarity, drug design, quantitative structure–activity/structure–property relationships, lead optimization and combinatorial library design.

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1. Introduction

The most critical step in drug discovery continues to be the identification and optimization of lead compounds in a rapid and cost effective way. The *computer-aided drug discovery* (CADD) approach is complementary to the real world of synthesis and screening. It “involves all computer-assisted techniques used to discover, design and optimize compounds with desired structure and properties” [1]. Formulation of quantitative relations among changes in the structural features and physicochemical properties/biological activities is an

interesting task of a computational chemist in view of the potential application of the derived relations in the diagnostic and mechanistic interpretation and prediction of properties/activities [2]. Structure–activity relationships (SARs) are such models, which attempt to relate certain structural aspects of molecules to their physicochemical/biological/toxicological properties [3]. The graph-theoretical approach to quantitative structure–property and structure–activity relationships (QSPR/QSAR) is based on mathematical representation of the molecular structure. The molecular descriptors derived therefrom are commonly named topological indices (TIs) [4]. Topological descriptors have gained considerable popularity as these can be derived from molecular structures using low computational resources [2]. TIs have the advantage that, unlike other molecular descriptors, they can be computed

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rapidly for any known or unknown chemical structure [5]. The application of TIs to the design and selection of novel active compounds is probably one of the most active areas of research on the application of such descriptors to biological problems [1]. The use of these descriptors covers most areas of the drug development process: lead discovery and lead optimization. They also include virtual screening, drug design, combinatorial library design, QSPR/QSAR, structure–pharmacokinetics, structure–toxicity relationships, and so forth [6]. The topological and topochemical descriptors are collectively referred to as topological descriptors. Topostructural descriptors encode information strictly on the adjacency and connectedness of atoms within a molecule whereas topochemical descriptors encode information relating to both molecular topology and the chemical nature of atoms and bonds within a molecule [7]. These indices are derived from matrices, like distance matrix and/or adjacency matrix, which represent a molecular graph. When the distance or adjacency matrix is weighted corresponding to the heteroatom(s) like N, O, Cl, etc., in a molecule, the matrix may be termed as chemical distance or chemical adjacency matrix, respectively. Indices or descriptors derived from such matrices are known as topochemical indices or topochemical descriptors.

Despite advances made in the therapeutic management, human immunodeficiency virus (HIV) infection has remained an intractable problem, and complete eradication of the virus an unrealized goal [8]. The human immunodeficiency virus is the cause of the acquired immunodeficiency syndrome (AIDS). Various compounds have been reported by De Clercq to inhibit the replication of causative retrovirus called HIV-1, *in vitro* [9]. In 1987, zidovudine (azidothymidine, AZT), a nucleoside RT inhibitor (NRTI), was approved in the USA as the first chemotherapeutic agent against HIV/AIDS [10,11]. A new diarylpyrimidine (DAPY) non-nucleoside reverse transcriptase inhibitor (NNRTI) was found to be suitable for high compliance oral treatment of HIV-1 infection [12]. However, resistance to anti-HIV compounds develops rapidly, sometimes within a few days of initiating treatment [13,14]. Acyclic nucleoside phosphonates (such as cidofovir, adefovir and tenofovir) bring a new dimension to the therapy of viral infections, as they offer a broader spectrum of activity, a longer duration of antiviral action and a lower risk of resistance development compared with available treatments [15]. Successful applications of multi-drug cocktails using inhibitors of HIV-1 protease and reverse transcriptase have been hailed as milestone in the treatment of AIDS [16]. Anti-HIV therapy, today, is in need of new drugs, which are less toxic, active against the drug resistant mutants selected by current therapies, or addressed towards novel targets in the viral replicative cycle [17,18].

In the present study, the relationship between adjacency-cum-distance based topochemical indices termed as supraugmented eccentric connectivity topochemical indices and anti-HIV-1 activity of 6-arylbenzonitriles has been investigated and suitable models developed for prediction of anti-HIV-1 activity.

2. Methodology

2.1. Calculation of topological indices

Three adjacency-cum-distance based topochemical indices termed as supraugmented eccentric connectivity topochemical indices, i.e. supraugmented eccentric connectivity topochemical index-1 ($^{SAc}\xi_1^c$), supraugmented eccentric connectivity topochemical index-2 ($^{SAc}\xi_2^c$) and supraugmented eccentric connectivity topochemical index-3 ($^{SAc}\xi_3^c$) have been proposed in the present study.

2.2. Supraugmented eccentric connectivity topochemical index-1

The supraugmented eccentric connectivity topochemical index-1, denoted by $^{SAc}\xi_1^c$, is defined as the summation of the quotients of the product of adjacent vertex chemical degrees and squared chemical eccentricity of the concerned vertex, for all vertices in the hydrogen suppressed molecular graph.

It is expressed as;

$$^{SAc}\xi_1^c = \sum_{i=1}^n \left(\frac{M_{ic}}{E_{ic}^2} \right) \quad (1)$$

where M_{ic} is the product of chemical degrees of all vertices (v_j), adjacent to vertex i , E_{ic} is the chemical eccentricity, and n is the number of vertices in graph G .

2.3. Supraugmented eccentric connectivity topochemical index-2

The supraugmented eccentric connectivity topochemical index-2, denoted by $^{SAc}\xi_2^c$, can be defined as the summation of the quotients of the product of adjacent vertex chemical degrees and cubic chemical eccentricity of the concerned vertex, for all vertices in the hydrogen suppressed molecular graph.

It can be expressed as;

$$^{SAc}\xi_2^c = \sum_{i=1}^n \left(\frac{M_{ic}}{E_{ic}^3} \right) \quad (2)$$

where M_{ic} is the product of chemical degrees of all vertices (v_j), adjacent to vertex i , E_{ic} is the chemical eccentricity, and n is the number of vertices in graph G .

2.4. Supraugmented eccentric connectivity topochemical index-3

The supraugmented eccentric connectivity topochemical index-3, denoted by $^{SAc}\xi_3^c$, is defined as the summation of the quotients of the product of adjacent vertex chemical degrees and fourth power of chemical eccentricity of the concerned vertex, for all vertices in the hydrogen suppressed molecular

graph. It is expressed as;

$$^{SAc}\xi_3^c = \sum_{i=1}^n \left(\frac{M_{ic}}{E_{ic}^4} \right) \quad (3)$$

where M_{ic} is the product of chemical degrees of all vertices (v_j), adjacent to vertex i , E_{ic} is the chemical eccentricity, and n is the number of vertices in graph G .

Superaugmented eccentric connectivity topochemical indices can be easily calculated from the chemical distance matrix (D_c) and augmentative chemical adjacency matrix (A_c^g) obtained by modifying chemical adjacency matrix (A_c). The derivation of matrices for calculation of supraugmented eccentric connectivity topochemical index-1 ($^{SAc}\xi_1^c$), supraugmented eccentric connectivity topochemical index-2 ($^{SAc}\xi_2^c$) and supraugmented eccentric connectivity topochemical index-3 ($^{SAc}\xi_3^c$) for three isomers of eight membered molecule (heptylamine) has been exemplified in Fig. 1. The sensitivity of the proposed topochemical descriptors towards presence and relative position of heteroatom(s) for all three, four and five membered isomers containing one heteroatom has been illustrated in Table 1. Discriminating power and degeneracy of the supraugmented eccentric connectivity topochemical indices were investigated using all possible structures with three, four and five vertices containing one heteroatom (Table 2).

2.5. Model development

A data set [19] comprising 81 analogues of 6-arylbenzotriazoles was selected for the present investigation. The basic structure for these analogues is depicted in Fig. 2 and various substituents are enlisted in Table 3.

The values of all the three supraugmented eccentric connectivity topochemical indices were computed for each analogue using an in-house computer program based on the method explained in Fig. 1. For the selection and evaluation of range specific features, exclusive activity ranges were determined from the frequency distribution of response level and subsequently identifying the active range by analyzing the resultant data by maximization of the moving average with respect to the active compounds ($<35\%$ = inactive, $35\text{--}65\%$ = transitional, $>65\%$ = active) [20]. Subsequently, each analogue was assigned a biological activity, which was then compared with the reported [19] anti-HIV-1 activity. The anti-HIV-1 activity was reported quantitatively as IC_{50} (μM) values in different concentrations. The analogues possessing IC_{50} values of $\leq 1.0 \mu M$ were considered to be active and analogues possessing IC_{50} values of $> 1.0 \mu M$ were considered to be inactive for the purpose of present study. The percentage degree of prediction of a particular range and overall degree of prediction of each model were also determined. The results are summarized in Tables 1–4 and Figs. 1–3.

3. Results and discussion

The main hypothesis in the QSPR and QSAR approach is that information about all properties (physical, chemical, and

biological) of a chemical substance can be derived from its molecular structure. An inspection of the published QSPR and QSAR models shows that molecular graph descriptors/TIs are used with success in modeling various properties and demonstrates that they are valuable descriptors of chemical structure [21]. In recent years, TIs have been reported and utilized for chemical documentation, isomer discrimination, study of molecular complexity, chirality, similarity/dissimilarity, QSAR/QSPR, drug design and database selection, lead optimization, rational combinatorial library design and for deriving multilinear regression models [1,22]. High discriminating power and absence of degeneracy are two properties of an ideal topological index, which the researchers in theoretical chemistry are striving to achieve.

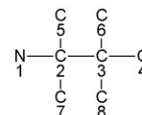
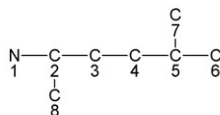
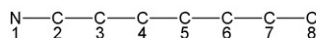
In the present study, three novel supraugmented eccentric connectivity topochemical indices, denoted by $^{SAc}\xi^c$ have been conceptualized. These indices can be easily calculated from chemical distance matrix (D_c) and augmentative chemical adjacency matrix (A_c^g). The consideration of both eccentricity and degree of the vertices results in significant changes in the topochemical indices value with a minor change in the branching of molecules.

As observed from Fig. 1, the value of supraugmented eccentric connectivity topochemical index-1 changes by about five times (from 0.995 to 4.473), the value of supraugmented eccentric connectivity topochemical index-2 changes by more than eight times (from 0.211 to 1.750) and the value of supraugmented eccentric connectivity topochemical index-3 changes by more than 16 times (from 0.046 to 0.713) following branching of eight membered molecule. These supraugmented eccentric connectivity topochemical indices were found to be more sensitive towards branching using three isomers of eight membered molecule (heptylamine).

First-generation topological indexes were integer numbers obtained by simple (“bookkeeping”) operations from local vertex invariants, which were integer numbers, involving only one vertex at a time; second-generation TI’s were real numbers obtained via sophisticated (“structural”) operations from integer local vertex invariants, involving more than one vertex at a time whereas third-generation TI’s are real numbers based on real-number local vertex invariants, having extremely low or no degeneracy [23]. The successful application of topological descriptors is somewhat limited owing to discriminating power of less than two digits only. The highest discriminating power was reported to be 295.4 in respect of augmented eccentric connectivity index [24] for all possible structures containing five vertices. Authors suggest that the topological indices having discriminating power ≥ 100 for structures containing only five vertices {with or without heteroatom(s)} may be treated as fourth-generation topological descriptors.

The values of supraugmented eccentric connectivity topochemical indices were computed for all the possible structure of three, four and five vertices using in-house computer program. The values and the structures have been presented in Table 1 and their comparison is presented in Table 2. Superaugmented eccentric connectivity topochemical indices have revealed exceptionally high discriminating power.

Arbitrary vertex numbering

Chemical Distance Matrices (D_c)

	1	2	3	4	5	6	7	8	E_{ic}
1	0	1	2	3	4	5	6	7	7
2	1.167	0	1	2	3	4	5	6	6
3	2.167	1	0	1	2	3	4	5	5
4	3.167	2	1	0	1	2	3	4	4
5	4.167	3	2	1	0	1	2	3	4.167
6	5.167	4	3	2	1	0	1	2	5.167
7	6.167	5	4	3	2	1	0	1	6.167
8	7.167	6	5	4	3	2	1	0	7.167

	1	2	3	4	5	6	7	8	E_{ic}
1	0	1	2	3	4	5	5	2	5
2	1.167	0	1	2	3	4	4	1	4
3	2.167	1	0	1	2	3	3	2	3
4	3.167	2	1	0	1	2	2	3	3.167
5	4.167	3	2	1	0	1	1	4	4.167
6	5.167	4	3	2	1	0	2	5	5.167
7	5.167	4	3	2	1	2	0	5	5.167
8	2.167	1	2	3	4	5	5	0	5

	1	2	3	4	5	6	7	8	E_{ic}
1	0	1	2	3	2	3	2	3	3
2	1.167	0	1	2	1	2	1	2	2
3	2.167	1	0	1	2	1	2	1	2.167
4	3.167	2	1	0	3	2	3	2	3.167
5	2.167	1	2	3	0	3	2	3	3
6	3.167	2	1	2	3	0	3	2	3.167
7	2.167	1	2	3	2	3	0	3	3
8	3.167	2	1	2	3	2	3	0	3.167

Chemical Adjacency Matrices (A_c)

	1	2	3	4	5	6	7	8	V_{ic}
1	0	1	0	0	0	0	0	0	1
2	1.167	0	1	0	0	0	0	0	2.167
3	0	1	0	1	0	0	0	0	2
4	0	0	1	0	1	0	0	0	2
5	0	0	0	1	0	1	0	0	2
6	0	0	0	0	1	0	1	0	2
7	0	0	0	0	0	1	0	1	2
8	0	0	0	0	0	0	1	0	1

	1	2	3	4	5	6	7	8	V_{ic}
1	0	1	0	0	0	0	0	0	1
2	1.167	0	1	0	0	0	0	1	3.167
3	0	1	0	1	0	0	0	0	2
4	0	0	1	0	1	0	0	0	2
5	0	0	0	1	0	1	1	0	3
6	0	0	0	0	1	0	0	0	1
7	0	0	0	0	1	0	0	0	1
8	0	1	0	0	0	0	0	0	1

	1	2	3	4	5	6	7	8	V_{ic}
1	0	1	0	0	0	0	0	0	1
2	1.167	0	1	0	1	0	1	0	4.167
3	0	1	0	1	0	1	0	1	4
4	0	0	1	0	0	0	0	0	1
5	0	1	0	0	0	0	0	0	1
6	0	0	1	0	0	0	0	0	1
7	0	1	0	0	0	0	0	0	1
8	0	0	1	0	0	0	0	0	1

Augmentative Chemical Adjacency Matrices (A_c)

	1	2	3	4	5	6	7	8	M_{ic}
1	0	2.167	0	0	0	0	0	0	2.167
2	1	0	2	0	0	0	0	0	2
3	0	2.167	0	2	0	0	0	0	4.334
4	0	0	2	0	2	0	0	0	4
5	0	0	0	2	0	2	0	0	4
6	0	0	0	0	2	0	2	0	4
7	0	0	0	0	0	2	0	1	2
8	0	0	0	0	0	0	2	0	2

	1	2	3	4	5	6	7	8	M_{ic}
1	0	3.167	0	0	0	0	0	0	3.167
2	1	0	2	0	0	0	0	1	2
3	0	3.167	0	2	0	0	0	0	6.334
4	0	0	2	0	3	0	0	0	6
5	0	0	0	2	0	1	1	0	2
6	0	0	0	0	3	0	0	0	3
7	0	0	0	0	3	0	0	0	3
8	0	3.167	0	0	0	0	0	0	3.167

	1	2	3	4	5	6	7	8	M_{ic}
1	0	4.167	0	0	0	0	0	0	4.167
2	1	0	4	0	1	0	1	0	4
3	0	4.167	0	1	0	1	0	1	4.167
4	0	0	4	0	0	0	0	0	4
5	0	4.167	0	0	0	0	0	0	4.167
6	0	0	4	0	0	0	0	0	4
7	0	4.167	0	0	0	0	0	0	4.167
8	0	0	4	0	0	0	0	0	4

Superaugmented Eccentric Connectivity Topochemical Index -1

$${}^{SAc}\xi_1^c = \sum_{i=1}^n \left(\frac{M_{ic}}{E_{ic}^2} \right)$$

$$= [2.167/49+2/36+4.334/25+4/16+4/17.4+4/26.7+2/38+2/51] = [3.167/25+2/16+6.334/9+6/10.03+2/17.4+3/26.7+3/26.7+3.167/25] = [4.167/9+4/4+4.167/4.7+4/10.03+4.167/9+4/10.03+4.167/9+4/10.03]$$

$$= 0.995 \quad = 2.020 \quad = 4.473$$

Superaugmented Eccentric Connectivity Topochemical Index -2

$${}^{SAc}\xi_2^c = \sum_{i=1}^n \left(\frac{M_{ic}}{E_{ic}^3} \right)$$

$$= [2.167/343+2/216+4.334/125+4/64+4/72.4+4/137.9+2/234.5+2/368.1] = [3.167/125+2/64+6.334/27+6/31.8+2/72.4+3/137.9+3/137.9+3.167/125] = [4.167/27+4/8+4.167/10.2+4/31.8+4.167/27+4/31.8+4.167/27+4/31.8]$$

$$= 0.211 \quad = 0.577 \quad = 1.750$$

Superaugmented Eccentric Connectivity Topochemical Index -3

$${}^{SAc}\xi_3^c = \sum_{i=1}^n \left(\frac{M_{ic}}{E_{ic}^4} \right)$$

$$= [2.167/2401+2/1296+4.334/625+4/256+4/301.5+4/712.8+2/1446.4+2/2638.5] = [3.167/625+2/256+6.334/81+6/100.6+2/301.5+3/712.8+3/712.8+3.167/625] = [4.167/81+4/16+4.167/22.1+4/100.6+4.167/81+4/100.6+4.167/81+4/100.6]$$

$$= 0.046 \quad = 0.171 \quad = 0.713$$

Fig. 1. Calculation of values of supraaugmented eccentric connectivity topochemical index-1 (${}^{SAc}\xi_1^c$), supraaugmented eccentric connectivity topochemical index-2 (${}^{SAc}\xi_2^c$) and supraaugmented eccentric connectivity topochemical index-3 (${}^{SAc}\xi_3^c$) for three isomers of eight membered molecule (heptylamine).

Table 1

Index values of for all possible structure with three, four and five vertices containing one heteroatom

Compound number	Structure	$S_{Ac} \xi_1^c$	$S_{Ac} \xi_2^c$	$S_{Ac} \xi_3^c$
1		2.214	1.755	1.543
2		1.737	1.113	0.773
3		12.123	11.061	10.150
4		1.402	0.606	0.270
5		1.474	0.644	0.290
6		2.875	1.648	1.024
7		3.506	2.474	1.997
8		4.174	2.048	1.006
9		8.888	7.053	6.206
10		7.156	4.688	3.305
11		6.427	4.158	2.929
12		29.967	26.111	23.655
13		23.247	18.181	14.742
14		98.047	88.562	80.434
15		2.078	0.859	0.367
16		2.050	0.857	0.372
17		2.181	0.921	0.405
18		2.191	0.921	0.402
19		1.595	0.624	0.256
20		1.579	0.610	0.247
21		1.822	0.773	0.348
22		4.438	2.379	1.366
23		20.620	14.716	11.188
24		4.723	2.225	1.049
25		20.466	13.410	9.484
26		22.374	14.845	10.570
27		18.727	12.166	8.620
28		4.974	2.298	1.079

Table 1 (Continued)

Compound number	Structure	$S_{Ac} \xi_1^c$	$S_{Ac} \xi_2^c$	$S_{Ac} \xi_3^c$
29		4.821	2.230	1.051
30		4.448	1.988	0.903
31		4.693	2.107	0.963
32		11.461	5.647	2.785
33		10.779	5.225	2.537
34		45.738	32.509	24.641
35		49.571	35.468	26.931
36		44.826	28.566	19.920
37		53.225	34.464	23.973
38		12.318	5.608	2.585
39		146.750	114.733	92.863
40		136.677	106.796	86.690
41		104.383	75.504	57.865
42		377.743	311.139	260.574
43		5.523	2.518	1.162
44		6.378	3.072	1.497
45		6.025	2.768	1.288
46		4.090	1.710	0.740
47		4.033	1.685	0.727
48		4.695	2.044	0.923
49		4.507	1.946	0.868
50		13.050	6.202	2.984
51		11.592	5.260	2.414
52		14.364	6.947	3.395
53		24.323	11.668	5.606

Table 1 (Continued)

Compound number	Structure	$S_{Ac}^{ec} \xi_1^c$	$S_{Ac}^{ec} \xi_2^c$	$S_{Ac}^{ec} \xi_3^c$
54		26.654	13.103	6.469
55		27.100	13.468	6.696
56		8.416	5.085	3.362
57		9.298	5.716	3.800
58		11.731	5.682	2.759
59		11.162	5.390	2.607
60		11.539	5.666	2.785
61		60.692	47.533	38.607

The discriminating power may be defined as the ratio of highest to lowest value for all possible structures of same number of vertices. The ratio of highest to lowest value for all possible structure of five vertices containing a heteroatom in case of supraugmented eccentric connectivity topochemical index-1 is 239, supraugmented eccentric connectivity topochemical index-2 is 510 and supraugmented eccentric connectivity topochemical index-3 is 1055, respectively. Exceptionally high discriminating power of the proposed new generation indices makes it more sensitive towards any change in molecular structure. Extreme sensitivity towards branching as well as discriminating power of all the three proposed indices is clearly evident from the respective index values of all the possible

Table 2

Comparison of discriminating power and degeneracy of supraugmented eccentric connectivity topochemical indices using all possible structures having three, four and five vertices containing one heteroatom

	$S_{Ac}^{ec} \xi_1^c$	$S_{Ac}^{ec} \xi_2^c$	$S_{Ac}^{ec} \xi_3^c$
For three vertices			
Minimum value	1.737	1.113	0.773
Maximum value	11.061	10.150	9.369
Ratio	1:6.368	1:9.119	1:12.12
Degeneracy ^a	0/3	0/3	0/3
For four vertices			
Minimum value	1.402	0.610	0.270
Maximum value	98.047	88.562	80.434
Ratio	1:70	1:146	1:298
Degeneracy		0/11	0/11
For five vertices			
Minimum value	1.579	0.610	0.247
Maximum value	377.743	311.139	260.574
Ratio	1:239	1:510	1:1055
Degeneracy	0/47	1/47	1/47

^a Degeneracy = Number of compounds having same values/total number of compounds with same number of vertices.

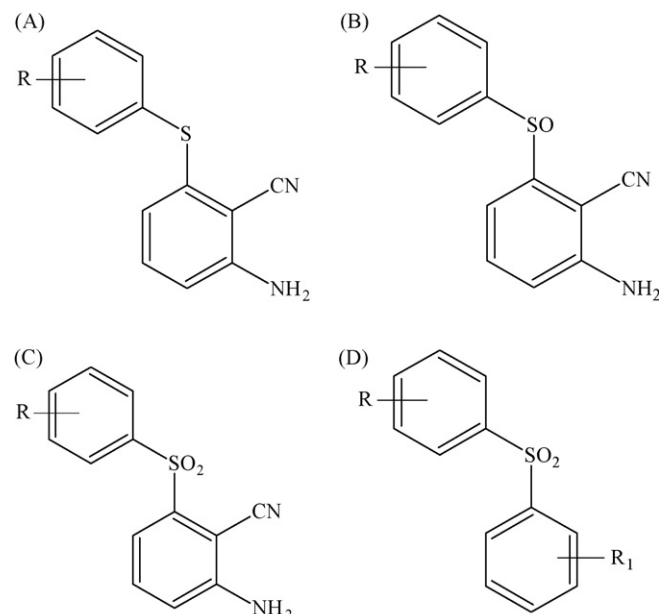


Fig. 2. Basic structures of 6-arylbenzonitriles.

structures with five vertices containing one heteroatom (Table 1).

Degeneracy is the measure of ability of an index to differentiate between the relative positions of atom in a molecule. The supraugmented eccentric connectivity topochemical index-1 did not exhibit any degeneracy for all possible structures with three, four and five vertices with a heteroatom, whereas the supraugmented eccentric connectivity index-2 and supraugmented eccentric connectivity topochemical index-3 has very low degeneracy of one in case of all possible structures with five vertices containing one heteroatom (Table 2). Extremely low degeneracy indicates the enhanced capability of these indices to differentiate and demonstrate slight variations in the molecular structure. This means that the likeliness of different structures to have same value is extremely remote.



Relationship of supraugmented eccentric connectivity topochemical indices with anti-HIV-1 activity of 6-arylbenzonitriles was studied and suitable models were developed for prediction of anti-HIV-1 activity. Though all the analogues in the datasets possess varying degree of biological activity but only those analogues having IC_{50} values of $\leq 1.0 \mu M$ were considered to be active for the purpose of present study. The methodology used in the present studies aims at the development of suitable models for providing lead molecules through exploitation of the active ranges in the proposed models based on topochemical indices. Proposed models are unique and differ widely from conventional QSAR models. Both system of modeling have their advantages and limitations. In the modeling used in this study, the system adopted has distinct advantage of identification of narrow active ranges, which may be erroneously skipped during routine regression analysis in conventional QSAR. Since the ultimate goal of modeling is to provide lead structures, therefore, active range(s) of the proposed model(s) can play vital role in providing lead structures [25].

Table 3

Relationship of supraugmented eccentric connectivity topochemical indices with anti-HIV-1 activity of 6-arylbenzonitriles

Compound number	R	R_1	$S_{\xi_1}^{Ac\epsilon c}$	$S_{\xi_2}^{Ac\epsilon c}$	$S_{\xi_3}^{Ac\epsilon c}$	Anti-HIV-1 activity predicted reported				Reported IC_{50} (μM) values [19]
						$S_{\xi_1}^{Ac\epsilon c}$	$S_{\xi_2}^{Ac\epsilon c}$	$S_{\xi_3}^{Ac\epsilon c}$	Reported	
A ₁	H	—	3.145	0.546	0.104	—	—	—	—	14.6
A ₂	2-OCH ₃	—	3.325	0.541	0.097	—	—	—	—	4.3
A ₃	3-OCH ₃	—	2.552	0.353	0.052	—	—	—	—	6.0
A ₄	4-OCH ₃	—	2.074	0.256	0.033	—	—	—	—	16.0
A ₅	2-CH ₃	—	3.295	0.566	0.107	—	—	—	—	6.1
A ₆	3-CH ₃	—	3.273	0.561	0.106	—	—	—	—	115
A ₇	4-CH ₃	—	2.564	0.376	0.059	—	—	—	—	4.1
A ₈	2-Cl	—	2.904	0.427	0.067	—	—	—	—	7.4
A ₉	3-Cl	—	2.38	0.306	0.041	—	—	—	—	32.0
A ₁₀	4-Cl	—	1.928	0.227	0.028	—	—	—	—	30.0
A ₁₁	2-Br	—	2.069	0.23	0.026	—	—	—	—	5.1
A ₁₂	3-Br	—	1.894	0.201	0.022	—	—	—	—	9.8
A ₁₃	3-F	—	2.946	0.461	0.079	—	—	—	—	6.4
A ₁₄	2-CN	—	3.441	0.58	0.108	—	—	—	—	1.73
A ₁₅	3-CN	—	2.601	0.369	0.056	—	—	—	—	43.8
A ₁₆	4-CN	—	2.108	0.265	0.035	—	—	—	—	12.8
A ₁₇	3-CF ₃	—	2.712	0.355	0.049	—	—	—	—	31.5
A ₁₈	5-NH ₂	—	3.238	0.553	0.105	—	—	—	+	>0.8
A ₁₉	2,5-Cl ₂	—	2.772	0.359	0.048	—	—	—	+	0.43
A ₂₀	3,5-(CH ₃) ₂	—	3.411	0.577	0.108	—	—	—	—	62% @ 5 μM
A ₂₁	3,5-Cl ₂	—	2.807	0.356	0.047	—	—	—	—	1.76
A ₂₂	3-Cl, 5-CH ₃	—	2.538	0.324	0.043	—	—	—	—	2.0
A ₂₃	3-OCH ₃ , 5-CH ₃	—	2.693	0.369	0.054	—	—	—	—	5.1
A ₂₄	3-OCH ₃ , 5-CF ₃	—	2.968	0.384	0.052	—	—	—	—	93.5
B ₁	2-OCH ₃	—	5.68	1.048	0.208	—	+	±	—	4.8
B ₂	3-OCH ₃	—	4.12	0.631	0.102	—	—	—	—	16.0
B ₃	4-OCH ₃	—	3.233	0.432	0.06	—	—	—	—	>200
B ₄	2-CH ₃	—	5.809	1.13	0.237	—	±	—	—	93.0
B ₅	3-CH ₃	—	5.73	1.117	0.235	—	±	±	—	29.23
B ₆	4-CH ₃	—	4.323	0.709	0.123	—	—	—	—	49.0
B ₇	2-Br	—	3.033	0.345	0.04	—	—	—	—	39.2
B ₈	3-Br	—	2.428	0.256	0.028	—	—	—	+	0.08
B ₉	4-Br	—	1.893	0.183	0.018	—	—	—	—	20.21
B ₁₀	2-CN	—	5.946	1.143	0.238	—	±	—	—	3.9
B ₁₁	3-CN	—	4.259	0.672	0.112	—	—	—	—	14.2
B ₁₂	4-CN	—	3.322	0.453	0.064	—	—	—	—	>200
B ₁₃	3-CF ₃	—	4.157	0.6	0.091	—	—	—	—	40.0
B ₁₄	3,5-(CH ₃) ₂	—	5.868	1.133	0.236	—	±	±	+	0.34
B ₁₅	2,5-Cl ₂	—	4.23	0.59	0.085	—	—	—	—	9.85
B ₁₆	3-Cl, 5-CH ₃	—	3.827	0.53	0.077	—	—	—	+	0.32
B ₁₇	3-OCH ₃ , 5-CF ₃	—	4.413	0.629	0.095	—	—	—	—	2.07
C ₁	H	—	11.601	2.503	0.563	—	—	—	—	2.0
C ₂	2-OCH ₃	—	11.314	2.307	0.493	—	—	—	+	0.6
C ₃	3-OCH ₃	—	7.857	1.314	0.227	+	±	±	+	0.9
C ₄	4-OCH ₃	—	5.934	0.849	0.125	—	—	—	—	25
C ₅	2-CH ₃	—	11.865	2.539	0.568	—	—	—	—	2.3
C ₆	3-CH ₃	—	11.729	2.518	0.565	—	—	—	+	0.4
C ₇	4-CH ₃	—	8.549	1.531	0.285	—	—	—	—	9.5
C ₈	2-Cl	—	9.322	1.66	0.307	—	—	—	—	4.1
C ₉	3-Cl	—	6.699	1.009	0.156	+	+	+	+	0.59
C ₁₀	4-Cl	—	5.18	0.685	0.092	—	—	—	—	3.0
C ₁₁	2-Br	—	4.824	0.555	0.065	—	—	—	—	5.0
C ₁₂	3-Br	—	3.628	0.379	0.04	—	—	—	+	0.54
C ₁₃	4-Br	—	2.88	0.274	0.027	—	—	—	—	20.0
C ₁₄	2-F	—	12.004	2.559	0.571	—	—	—	—	3.0
C ₁₅	3-F	—	9.995	1.944	0.394	—	—	—	—	3.0
C ₁₆	2-CN	—	11.994	2.55	0.569	—	—	—	—	5.4
C ₁₇	3-CN	—	8.229	1.42	0.254	—	—	—	—	2.4
C ₁₈	4-CN	—	6.162	0.903	0.136	—	+	+	—	80.0
C ₁₉	3-CF ₃	—	7.584	1.198	0.196	+	±	±	—	3.5
C ₂₀	2,5-Cl ₂	—	7.43	1.111	0.171	+	+	+	+	0.3
C ₂₁	3,5-Cl ₂	—	7.127	1.058	0.162	+	+	+	+	0.07

Table 3 (Continued)

Compound number	R	R ₁	S _{Ac} ^{EC} ₁	S _{Ac} ^{EC} ₂	S _{Ac} ^{EC} ₃	Anti-HIV-1 activity predicted reported				Reported IC ₅₀ (μM) values [19]
						S _{Ac} ^{EC} ₁	S _{Ac} ^{EC} ₂	S _{Ac} ^{EC} ₃	Reported	
C ₂₂	3,5-(CH ₃) ₂	—	11.867	2.534	0.567	—	—	—	+	0.01
C ₂₃	3-Br, 5-CH ₃	—	3.798	0.396	0.042	—	—	—	+	0.02
C ₂₄	3-Cl, 5-CH ₃	—	6.858	1.027	0.158	+	+	+	+	0.03
C ₂₅	3-OCH ₃ , 5-CH ₃	—	7.999	1.33	0.229	+	±	±	+	0.05
C ₂₆	3-OCH ₃ , 5-CF ₃	—	7.84	1.228	0.2	+	±	±	+	0.09
C ₂₇	3-OH, 5-CH ₃	—	11.09	2.272	0.488	—	—	—	+	0.43
C ₂₈	3-OCH ₂ CH ₃ , 5-CH ₃	—	6.207	0.881	0.128	+	+	+	+	0.06
C ₂₉	3-O(CH ₂) ₂ CH ₃ , 5-CH ₃	—	5.065	0.638	0.082	—	—	—	+	0.06
C ₃₀	3-O(CH ₂) ₃ CH ₃ , 5-CH ₃	—	4.219	0.48	0.056	—	—	—	+	0.6
C ₃₁	 =1-Naphthyl	—	9.314	1.627	0.297	—	—	—	—	3.32
C ₃₂	 =2-Naphthyl	—	6.753	1.001	0.153	+	+	+	+	0.05
D ₁	2-NO ₂	—	10.329	2.056	0.426	—	—	—	—	8.6
D ₂	2-NO ₂	2-NO ₂	10.852	2.105	0.43	—	—	—	—	2.2
D ₃	3,5-Cl ₂	2-CN, 3-F	7.026	1.034	0.158	+	+	+	—	42
D ₄	3,5-(CH ₃) ₂	2-CN, 3-F	10.273	1.987	0.401	—	—	—	—	1.5
D ₅	3-OCH ₃ , 5-CH ₃	2-CN, 3-F	7.94	1.314	0.226	+	±	±	—	51
D ₆	3,5-(CH ₃) ₂	2-CN, 3-NH-(CH ₂) ₃ CH ₃	4.623	0.551	0.067	—	—	—	—	>12
D ₇	3,5-(CH ₃) ₂	2-CN, 3-cyclo-hexylamnio	4.797	0.566	0.068	—	—	—	—	>200
D ₈	H	3-NH ₂	11.008	2.423	0.552	—	—	—	—	>50

(+) Active analogue; (—) inactive analogue; (±) analogue in the transitional range where activity could not be specifically assigned.

Retrofit analysis of the data in Tables 3 and 4 reveals the following information with regard to models based upon supraaugmented eccentric connectivity topochemical index-1:

- All the compounds were classified either as active or inactive using the model based upon supraaugmented eccentric connectivity topochemical index-1. The overall accuracy of prediction was found to be 80% with regard to anti-HIV-1 activity.
- The active range had supraaugmented eccentric connectivity topochemical index-1 values of 6.207–7.999. Seventy five percent of the analogues in the active range exhibited anti-HIV-1. The average IC₅₀ value of correctly predicted compounds in the active range was found to be only

0.238 μM. This clearly indicates exceptionally high potency of the active range.

- Two inactive ranges—a lower inactive range with index values of <6.207 and an upper inactive range with index values of >7.999 were observed. Activity of 56 out of 69 compounds in these inactive ranges was predicted correctly.
- The ratio of average IC₅₀ values of active range and lower inactive range for correctly predicted analogues was found to be 1:158.79 and ratio of average IC₅₀ values of active range and upper inactive range for correctly predicted analogues was found to be 1:31.45.

Retrofit analysis of the data in Tables 3 and 4 reveals the following information with regard to model based

Table 4
Proposed models for the prediction of anti-HIV-1 activity of 6-arylbenzonitriles

Index	Nature of range in the proposed model	Index value	Number of analogues in the range	Number of analogues predicted correctly	Percent accuracy	Overall accuracy of prediction	Average IC ₅₀ (mM) of correctly predicted analogues
S _{Ac} ^{EC} ₁	Lower inactive	<6.207	52	43	82.69	80.25	37.792
	Active	6.207–7.999	12	9	75		0.238
	Upper inactive	>7.999	17	13	76.47		7.486
S _{Ac} ^{EC} ₂	Lower inactive	<0.881	46	38	82.61	80.56	37.214
	Active	0.881–1.111	9	6	66.67		0.183
	Transitional	>1.111 to <1.42	9	N.A.	N.A.		N.A.
	Upper inactive	>1.42	17	14	82.35		6.951
S _{Ac} ^{EC} ₃	Lower inactive	<0.128	46	38	82.61	80.82	37.214
	Active	0.128–0.171	8	6	75		0.183
	Transitional	>0.171 to <0.237	8	N.A.	N.A.		N.A.
	Upper inactive	>0.237	19	15	78.95		12.948

N.A., not applicable.

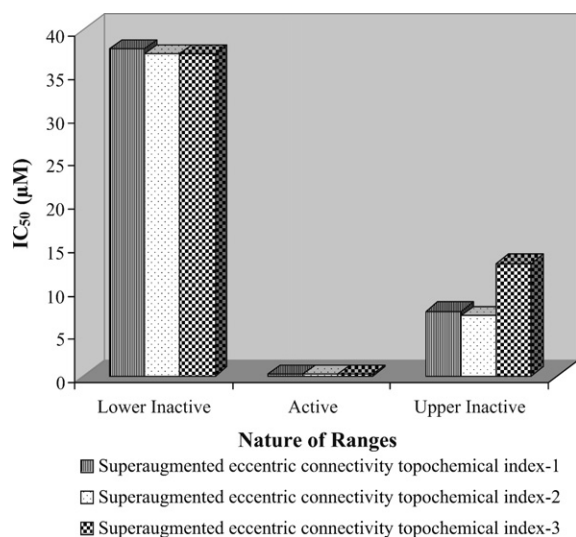


Fig. 3. Average IC_{50} (μM) values of 6-arylbenzonitriles of correctly predicted analogues in various ranges of topochemical models.

upon supraugmented eccentric connectivity topochemical index-2:

- A total of 72 out of 81 compounds were classified either as active or inactive using model based upon supraugmented eccentric connectivity topochemical index-2. The overall accuracy of prediction was found to be $\sim 81\%$ with regard to anti-HIV-1 activity.
- The active range had supraugmented eccentric connectivity topochemical index-2 values of 0.881–1.111. Approximately sixty seven percent of the analogues in the active range exhibited anti-HIV-1 activity. The average IC_{50} value of correctly predicted compounds in the active range was found to be only $0.183 \mu M$. This clearly indicates exceptionally high potency of the active range.
- Two inactive ranges—a lower inactive range with index values of <0.881 and an upper inactive range with index values of >1.42 were observed. Activity of 52 out of 63 compounds in these inactive ranges was predicted correctly.
- A transitional range with index values of >1.111 to <1.42 was observed. The transitional range indicates gradual transition from active to inactive range and vice versa.
- The ratio of average IC_{50} values of active range and lower inactive range for correctly predicted analogues was found to be 1:203.36 and ratio of average IC_{50} values of active range and upper inactive range for correctly predicted analogues was found to be 1:37.98.

Retrofit analysis of the data in Tables 3 and 4 reveals the following information with regard to supraugmented eccentric connectivity topochemical index-3:

- A total of 73 out of 81 compounds were classified either as active or inactive using model based upon supraugmented eccentric connectivity topochemical index-3. The overall

accuracy of prediction was found to be $\sim 81\%$ with regard to anti-HIV activity.

- The active range had supraugmented eccentric connectivity topochemical index-3 values of 0.128–0.171. Seventy five percent of the analogues in the active range exhibited anti-HIV-1 activity. The average IC_{50} value of correctly predicted compounds in the active range was found to be only $0.183 \mu M$. This clearly indicates exceptionally high potency of the active range.
- Two inactive ranges—a lower inactive range with index values of <0.128 and an upper inactive range with index values of >0.237 were observed. Activity of 53 out of 65 compounds in these inactive ranges was predicted correctly.
- A transitional range with index values of >0.171 to <0.237 was observed. The transitional range indicates gradual transition from active to inactive range and vice versa.
- The ratio of average IC_{50} values of active range and lower inactive range for correctly predicted analogues was found to be 1:203 and ratio of average IC_{50} values of active range and upper inactive range for correctly predicted analogues was found to be 1:71.

4. Conclusion

High sensitivity towards presence and relative position of heteroatom(s), exceptionally high discriminating power and negligible degeneracy render all the three conceptualized supraugmented eccentric connectivity topochemical indices extremely beneficial tools for isomer discrimination, similarity/dissimilarity, drug design, quantitative structure–activity/structure–property relationships, lead optimization and combinatorial library design.

Investigations reveal significant correlations of all the three-topochemical indices with anti-HIV-1 activity of 6-arylbenzonitriles. The overall accuracy of prediction for models based on supraugmented eccentric connectivity topochemical indices was $\sim 81\%$. High predictability of the proposed models based upon the topochemical indices offer a vast potential for providing lead structures for the development of potent therapeutic agents with regard to anti-HIV-1 activity.

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