Question 2: Enzyme kinetics

Answers to 8.1:

First, 3 basic reactions:

$$\begin{cases}
E + S & \stackrel{k_1}{\longrightarrow} ES \\
ES & \stackrel{k_2}{\longrightarrow} E + S \\
ES & \stackrel{k_3}{\longrightarrow} E + P
\end{cases}$$

Let [E]. [S]. [ES]. [P] be concentration

$$0 \frac{d[E]}{dt} = -k_1[E][S] + k_2[ES] + k_3[ES]$$

$$\frac{d[ES]}{dt} = K_1[E][S] - K_2[ES] - K_3[ES]$$

$$\Theta \frac{dlPJ}{dt} = K_3 LESJ$$