

# DFT Approach into the Physical Properties of $M\text{Te}_3$ ( $M=\text{Hf}$ , $\text{Zr}$ ): A Comprehensive Study

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## INTRODUCTION

- ❖ Transition-metal trichalcogenide compounds,  $M\text{Te}_3$  ( $M=\text{Hf}$ ,  $\text{Zr}$ ) have recently piqued interest due to its mechanical, chemical and dynamical stability and also their prospective applications in various electronic and optoelectronic devices.
- ❖ The appearance of superconductivity with  $\sim T_c = 2$  K of  $\text{HfTe}_3$  has been confirmed experimentally.
- ❖ These compounds have a great prospect as a coating materials due to its high value of reflectivity.

## Theoretical Methodology

- DFT calculations are performed using the Cambridge Serial Total Energy Package (CASTEP).
- The generalized gradient approximation (GGA) is used for electron exchange correlation function with the Perdew-Burke-Ernzerhof (PBE) type to perform geometry optimization. Elastic constants were calculated by the 'stress-strain' method using CASTEP program.
- The bulk modulus,  $B$ , shear modulus  $G$  and Young modulus,  $Y$  were obtained from the calculated elastic constants  $C_{ij}$ .

## Structural Properties

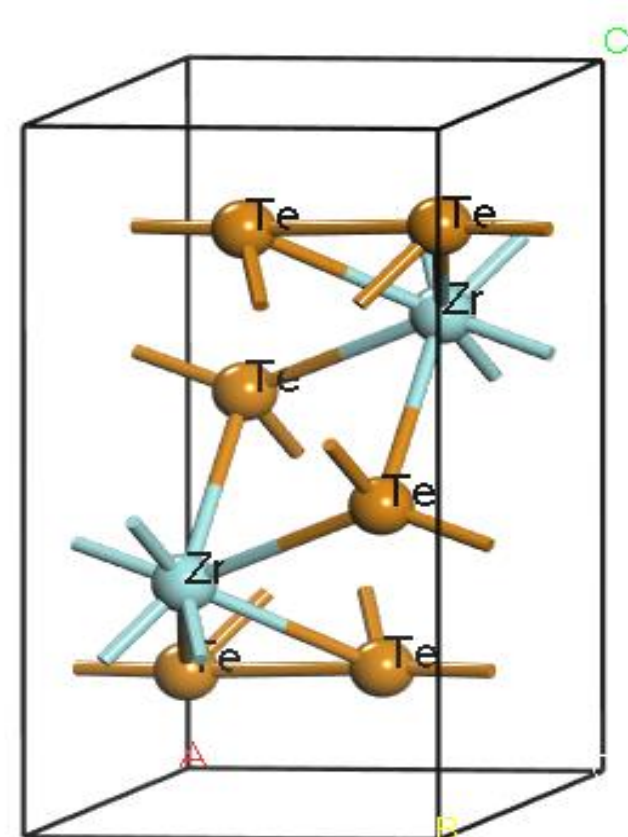


Fig. 1: Optimized crystal structure of  $M\text{Te}_3$  ( $M=\text{Hf}$ ,  $\text{Zr}$ )

Optimized parameters	a	b	c	V	Ref.
$\text{HfTe}_3$	5.883	3.988	10.649	247.882	This
	5.8845	3.9026	10.0551	228.70	Ref. 1
$\text{ZrTe}_3$	5.89480	3.926400	10.1040	231.62421	This
	5.1339	3.6363	9.0032	.....	Ref. 2

Table 1: Optimized lattice parameters,  $a$ ,  $b$  and  $c$  ( $\text{\AA}$ ), unit cell volume  $V$  ( $\text{\AA}^3$ ) of  $M\text{Te}_3$  ( $M=\text{Hf}$  and  $\text{Zr}$ ) compounds

- The obtained structural parameters, such as lattice parameters, angles, and volume, agree with the previous study.
- The crystal system is monoclinic with space group  $P_{21}/m$  (No. 11)

## Electronic Properties

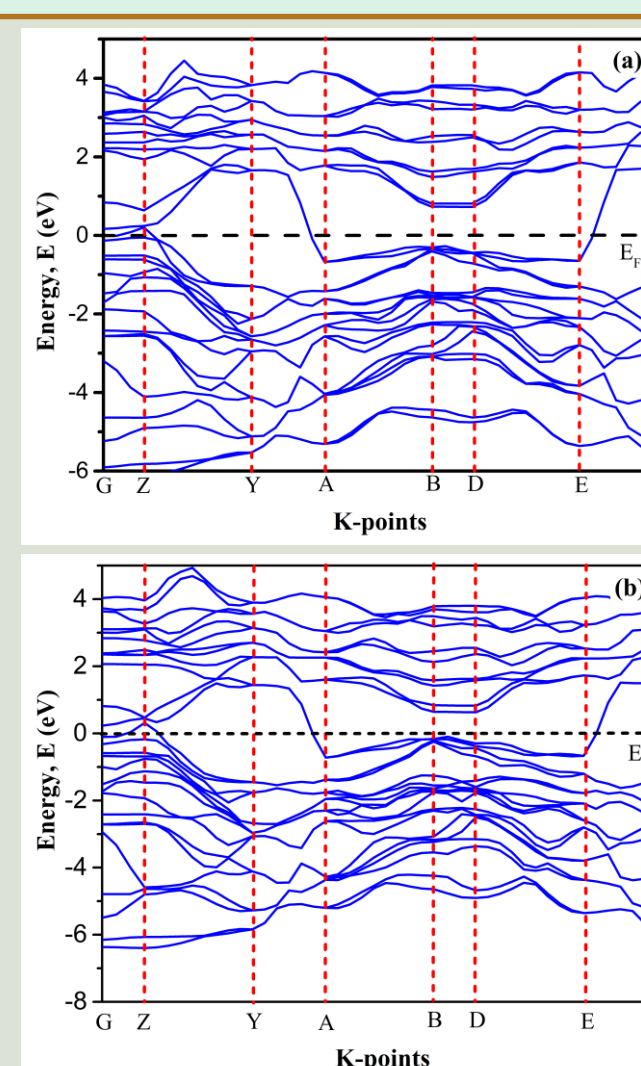


Fig 2. Calculated electronic band structure of (a)  $\text{HfTe}_3$  (b)  $\text{ZrTe}_3$  along the high symmetry directions in the Brillouin zone at ambient conditions.

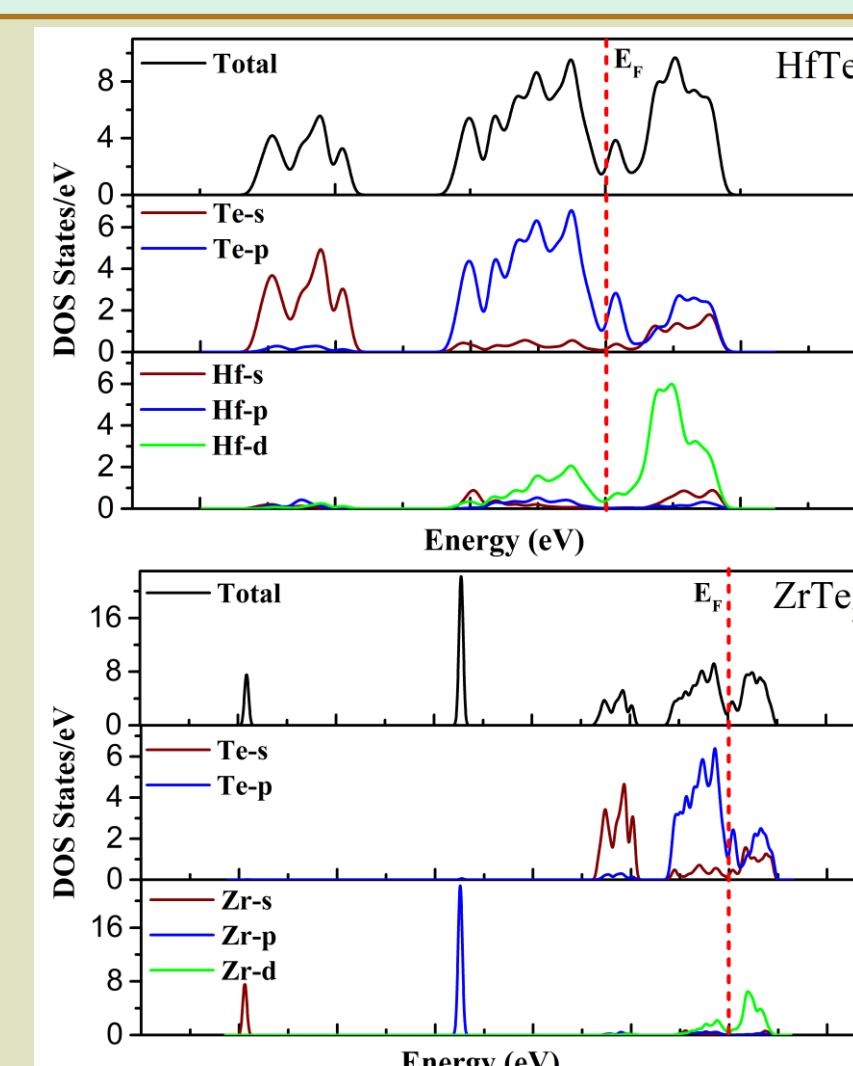


Fig 3. The total and partial energy density of states of (a)  $\text{HfTe}_3$  (b)  $\text{ZrTe}_3$ .

- ✓ Electronic band structure profile demonstrates that the studied compounds are metallic in nature.
- ✓ The Partial density of states (PDOS) confirms that the prime contribution to the conductivity comes from the Te-p orbital electrons.

## Mechanical Properties

Compounds	$C_{11}$	$C_{12}$	$C_{13}$	$C_{23}$	$C_{22}$	$C_{33}$	$C_{44}$	$C_{55}$	$C_{66}$	CP
$\text{HfTe}_3$	94	94	11	18	87	41	28	16	22	66
$\text{ZrTe}_3$	105	16	14	17	82	49	34	9	22	-18

Table 2: The calculated single crystal elastic constants  $C_{ij}$  (GPa) and Cauchy pressure (CP) for  $M\text{Te}_3$  ( $M=\text{Hf}$  and  $\text{Zr}$ ).

Compounds	B	G	Y	G/B	$\nu$
$\text{HfTe}_3$	28	21	50	0.74	0.20
$\text{ZrTe}_3$	35	22	55	0.63	0.23

Table 3: The calculated bulk modulus,  $B$  (GPa), shear modulus,  $G$  (GPa), Young's modulus,  $Y$  (GPa), Pugh ratio  $G/B$ , and Poisson ratio,  $\nu$  for  $M\text{Te}_3$  ( $M=\text{Hf}$  and  $\text{Zr}$ ).

- The title compounds are mechanically stable, which is confirmed from the stiffness constants.
- The value of Pugh's ratio (ratio of shear and bulk modulus) indicates that  $\text{ZrTe}_3$  is less brittle in nature compared to  $\text{HfTe}_3$ .
- The title compounds are predicted to be highly soft and anisotropic in nature.

### References

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## Conclusions

- The obtained structural parameters, such as lattice parameters, angles, and volume, agree with the previous study.
- Electronic band structure demonstrates that there are no band gaps between the valence and conduction bands, making the material metallic.
- The mechanical properties that are technologically important (stiffness constant, elastic moduli, brittle/ductile behavior, Poisson's ratio, anisotropy, and hardness) are thoroughly examined and addressed. When compared to  $\text{HfTe}_3$ , the value of Pugh's ratio (ratio of shear and bulk modulus) indicates that  $\text{ZrTe}_3$  is less brittle in nature. The title compounds are projected to be highly soft and anisotropic in nature.