

# PHY304 - Statistical Mechanics

Spring 2021, IISER Mohali

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## PHY304: Homework 3 Solutions

Due: Friday, February 5, 2021 at 11:00pm.

(Upload your solutions to Moodle as a single .pdf file.)

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1. Consider a system consisting of two particles of mass  $m$ . They interact weakly and are allowed to move in one dimension. Let us denote the position coordinates of particle 1 by  $x_1$  and particle 2 by  $x_2$ . We also denote their momenta by  $p_1$  and  $p_2$ . The boundaries of the one-dimensional space are at  $x = 0$  and  $x = L$ . The total energy of the system is between  $E$  and  $E + \delta E$ . Draw the phase space regions that are accessible to the system. Since the phase space is four dimensional, draw separately the part of the phase space involving  $x_1$  and  $x_2$  and that involving  $p_1$  and  $p_2$ .

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**Solution:**

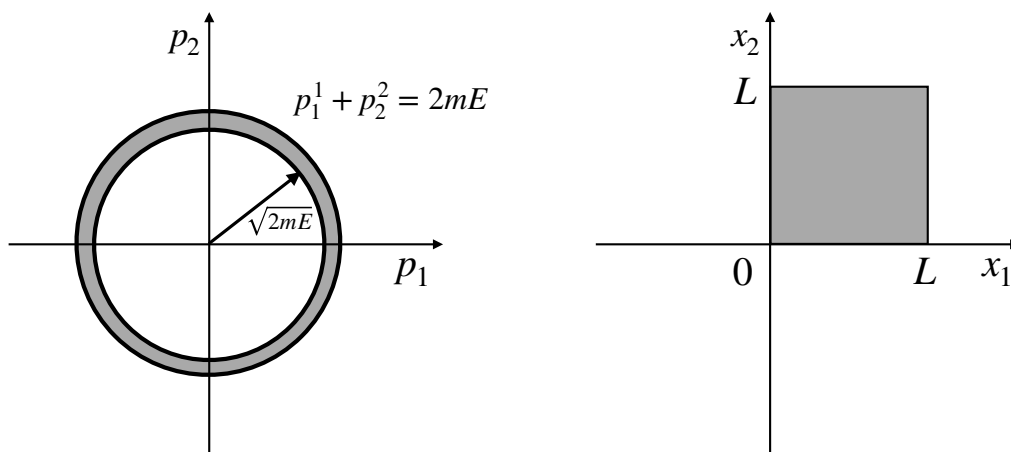


Figure 1: The four-dimensional phase space, shown as two two-dimensional spaces.

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2. Let us consider a system made out of a very large number  $N$  of distinguishable molecules. Let us assume that they are non-moving and non-interacting. Each molecule can have two energy levels: 0 and  $\delta$ , with  $\delta > 0$ . In the large  $N$  limit,  $N \rightarrow \infty$ ,  $E/N$  represents the mean energy per molecule. Compute the mean entropy per molecule,  $S/N$ , as a function of  $E/N$ .

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**Solution:**

When the mean energy per molecule is  $E/N$ , we have  $E/\delta$  molecules occupying the energy  $\delta$ . The number of microstates is then

$$\Omega = \frac{N!}{\left(\frac{E}{\delta}\right)! \left(N - \frac{E}{\delta}\right)!}. \quad (1)$$

The entropy is

$$\begin{aligned} S &= k_B \ln \Omega \\ &= k_B \ln \frac{N!}{\left(\frac{E}{\delta}\right)! \left(N - \frac{E}{\delta}\right)!}. \end{aligned} \quad (2)$$

When  $E/\delta \gg 1$ ,  $N - (E/\delta) \gg 1$ , we have

$$\begin{aligned} \frac{S}{N} &= k_B \left[ \ln N - \frac{E/\delta}{N} \ln(E/\delta) - \left(1 - (E/\delta)/N\right) \ln(N - (E/\delta)) \right] \\ &= k_B \left[ \ln N - \frac{E}{\delta N} \ln \left(\frac{E}{\delta}\right) - \left(1 - \frac{E}{N\delta}\right) \ln \left\{ N \left(1 - \frac{E}{N\delta}\right) \right\} \right] \\ &= k_B \left[ \ln N - \frac{E}{N\delta} \ln \left(\frac{E}{\delta}\right) - \ln N - \ln \left(1 - \frac{E}{N\delta}\right) \right. \\ &\quad \left. + \frac{E}{N\delta} \ln N + \frac{E}{N\delta} \ln \left(1 - \frac{E}{N\delta}\right) \right] \\ &= k_B \left[ -\frac{E}{N\delta} \ln \left(\frac{E}{N\delta}\right) - \ln \left(1 - \frac{E}{N\delta}\right) + \frac{E}{N\delta} \ln \left(1 - \frac{E}{N\delta}\right) \right] \\ &= k_B \left[ -\frac{E}{N\delta} \ln \left(\frac{E}{N\delta}\right) + \ln \frac{1}{1 - E/(N\delta)} - \frac{E}{N\delta} \ln \ln \frac{1}{1 - E/(N\delta)} \right] \\ &= k_B \left[ \frac{E}{N\delta} \ln \frac{N\delta}{E} + \left(1 - \frac{E}{N\delta}\right) \ln \frac{1}{1 - (E/N\delta)} \right]. \end{aligned} \quad (3)$$

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