



Quiz # 1

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1. A tall cylindrical vessel with gaseous nitrogen is located in a uniform gravitational field in which the free-fall acceleration is equal to g . The temperature of the nitrogen varies along the height h so that its density is the same throughout the volume. The variation in temperature with respect to the height is given by,

- (a) gM/R
- (b) $-gM/R$
- (c) $2gM/R$
- (d) $-2gM/R$

M is the molar mass of the nitrogen. You can treat the gas as ideal.

2. Which of the following statement/s are wrong ?

- (a) $\Delta Q = 0$ implies $\Delta S = 0$.
- (b) According to the condition for stable equilibrium the entropy should decrease in any small change.
- (c) The third law of thermodynamics implies that $\lim_{T \rightarrow 0} \frac{\partial S}{\partial V} = 0$.
- (d) No engine other than Carnot engine can be 100% efficient.

3. Given $U = U(S, V, N)$, the Maxwell relation $(\partial S / \partial \mu)_{T, V} = (\partial N / \partial T)_{\mu, V}$ follows from the thermodynamic potential/s

- (a) $F = U - TS$
- (b) $\Phi_G = U - TS - \mu N$
- (c) $H = U + PV$
- (d) $G = U + PV - TS$

4. Given $U = U(S, V, N)$ and $F = U - TS$, which of these is/are true if $\partial^2 F / \partial V^2 \geq 0$?

(a) $\kappa_T > 0$

(b) $c_P > 0$

(c) $\kappa_S > 0$

(d) $c_V > 0$

5. The Legendre Transform of $f(x) = \ln(x)$ is given by

(a) $c(m) = -e^m$

(b) $c(m) = \ln(m) - 1$

(c) $c(m) = e^{-m}$

(d) $c(m) = -\ln(m) - 1$