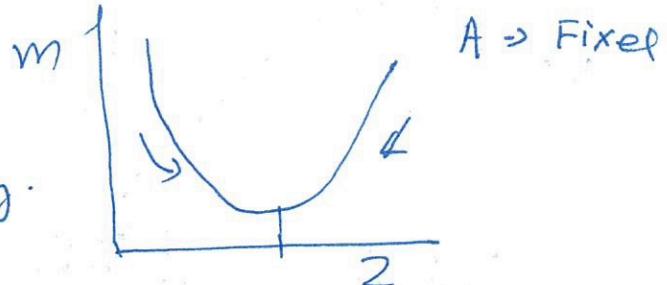
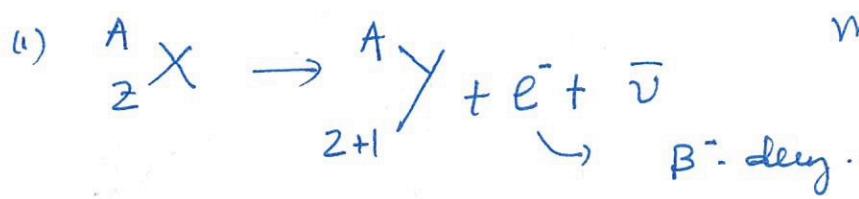
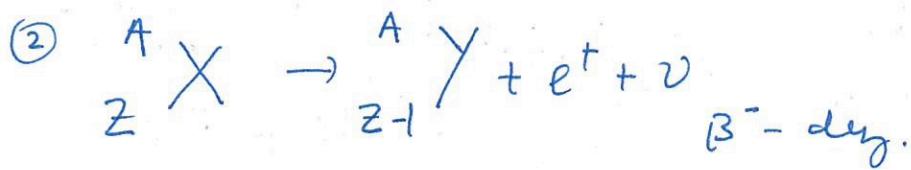


β^- -decay

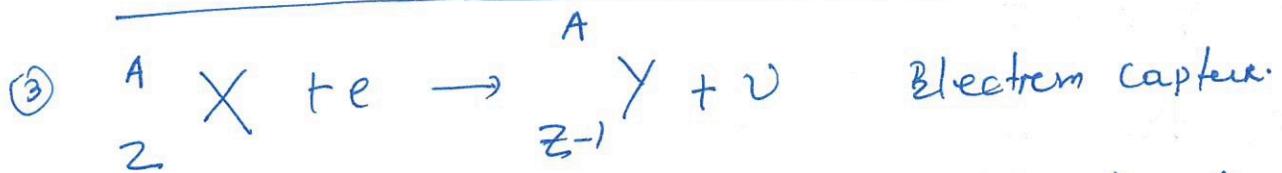
3/10/24



$A \rightarrow$ Fixed



start with



calculate Q value. calculate without ν

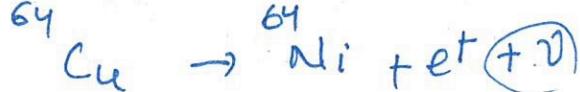
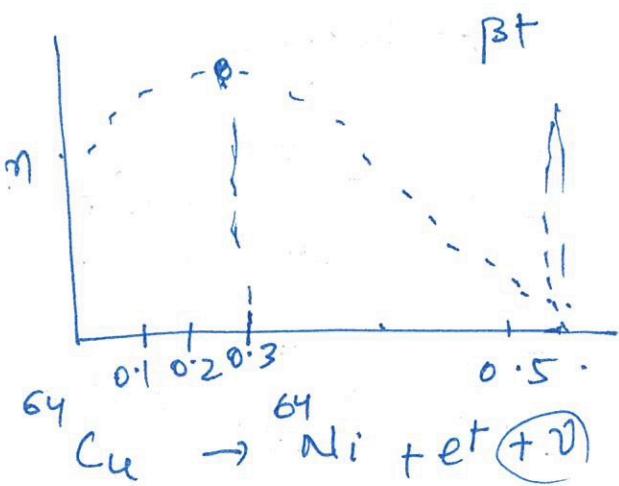
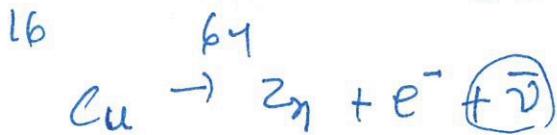
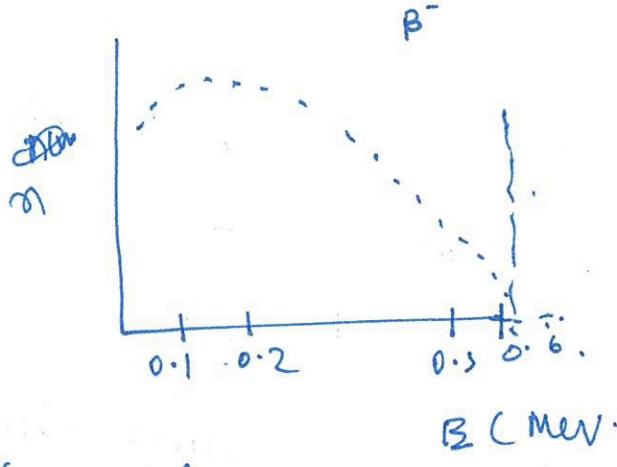
$$Q = [m(X) - m(Y) - m(e) - m(\bar{\nu})]^c^2$$

$$= \left\{ m^a(X) - Zm_e - \left[m^a_{Z+1}(Y) - (Z+1)m_e \right] - m(e) - m(\bar{\nu}) \right\}$$

$$Q = \left[m^a(X) - m^a_{Z+1}(Y) - m(\bar{\nu}) \right]$$

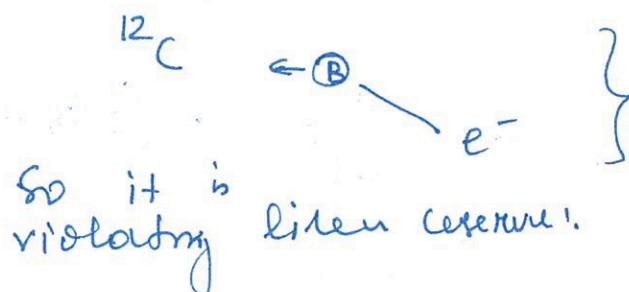
$$\begin{aligned} (2) \frac{\beta^+}{Q} &= m(X) - m_{Z+1}(Y) - m(e) - m(\bar{\nu}) \\ &= m^a(X) - Zm_e - \left(m^a_{Z+1}(Y) - (Z+1)m_e \right) - m(e) - m(\bar{\nu}) \\ &= m^a(X) - \cancel{Zm^a_e} - \boxed{2m(e)} - m(\bar{\nu}) \end{aligned}$$

$$\begin{aligned} (3) Q &= m(X) + m(e) - m_{Z-1}(Y) \\ &= m^a(X) - Zm_e + m_e - \left(m^a_{Z-1}(Y) + (Z-1)m_e \right) \\ &= m^a(X) - 2(m_e) + m_e - m(Y) - 2m_e - m_e - m(\bar{\nu}) \\ &= m^a(X) - m(Y) - m(\bar{\nu}) \end{aligned}$$



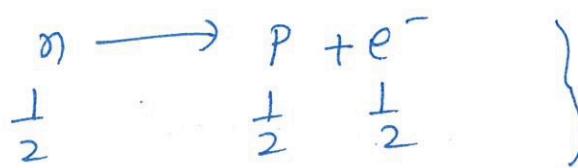
Explain the missing energy.

① Energy.



Back-to-Back due to conservation of linear momenta.

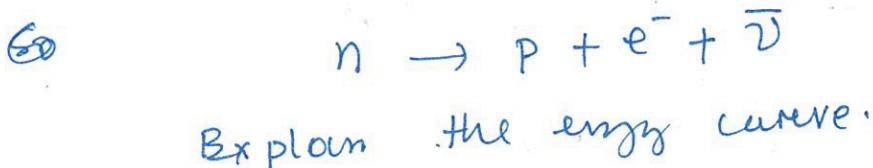
③



violating the conservation of spin.

• Pauli
B. Fermi

1931
1934



$$\text{missing energy} = B_B - E_B$$

Discovered - 1956

Neutron.

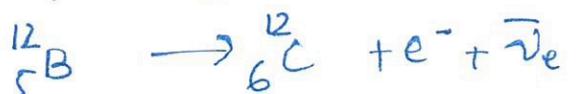
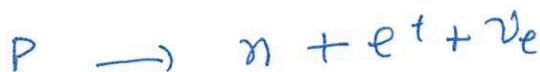
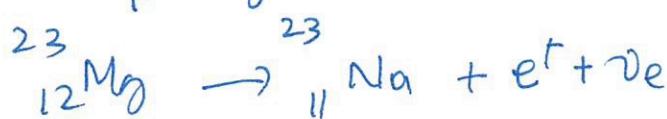


neutrino

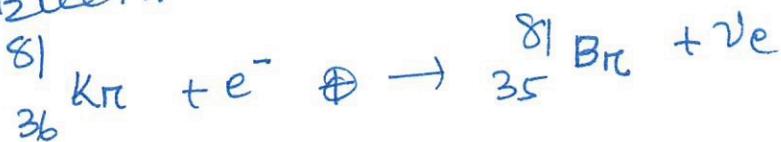


neutral little one.

(2)

① -ve β -decay:Inverse decinverse β -decay.② +ve β -decay:

③ Electron Capt.



Rate:

$$T_{\text{D}} = 0 \rightarrow \infty$$

$$\frac{dN_1}{dt} = -\lambda_1 N_1$$

$$\frac{dN_2}{dt} = -\lambda_2 N_2 + \lambda_1 N_1$$

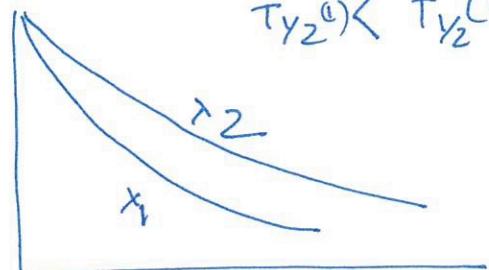
$$\frac{dN_2}{dt} = -\lambda_2 N_2 + \lambda_1 N_0 e^{-\lambda_1 t} e^{\lambda_2 t}$$

$$\Rightarrow e^{\lambda_2 t} \frac{dN_2}{dt} + e^{\lambda_2 t} \lambda_2 N_2 = \lambda_1 N_0 e^{(\lambda_2 - \lambda_1)t}$$

$$\Rightarrow \frac{d(e^{\lambda_2 t} N_2)}{dt} = -\lambda_1 N_0 e^{(\lambda_2 - \lambda_1)t}$$

$$N_2 = \left(\frac{\lambda_1}{\lambda_2 - \lambda_1} \right) N_0 [e^{-\lambda_1 t} - e^{-\lambda_2 t}]$$

$$\lambda_1 > \lambda_2 \\ T_{Y_2}^{(1)} < T_{Y_2}^{(2)}$$



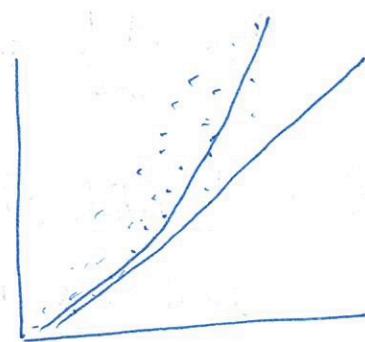
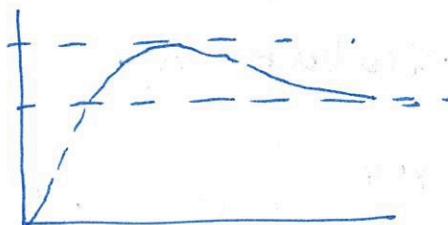
$$e^{\lambda_2 t} N_2 = \frac{\lambda_1}{\lambda_2 - \lambda_1} N_0 e^{(\lambda_2 - \lambda_1) t} + C$$

$$C = N_2 - \frac{\lambda_1}{\lambda_2 - \lambda_1} N_0$$

N_2

Nuclear Decay α -decay (1)

$$BE = a_v A - a_s A^{2/3} - a_c \frac{Z(Z-1)}{A^{1/3}} - Q_{sys} \Delta \frac{(N-Z)^2}{A} + \text{Pairing term}$$



if BE is not sufficient, nucleus \Rightarrow breaks up :

if the nucleus breaks in comparable fragment are called nuclear fission

otherwise it's nuclear decay α, β, γ



If $Q > 0$ no breaking up

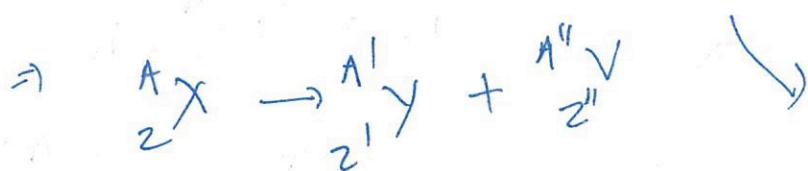
$Q < 0$ it will break.

i.e. system prefers to lower down the ground state energy.

$$\Rightarrow Q = [m[X] - m[Y] - m[V]]$$

$$= [-BB(X) + BB(Y) + BB(V)]$$

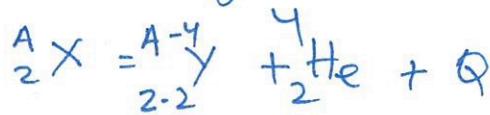
$$m_A + m_Z - m_{A'} - m_{Z'} - m_{A''} - m_{Z''}$$



$$\left. \begin{array}{l} A = A' + A'' \\ Z = Z' + Z'' \end{array} \right\}$$

$$Q = BB(Y) + BB(V) - BE(X)$$

α -decay:



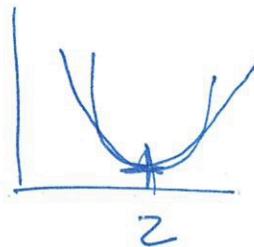
(Q)

$$Q = [BE(Y) + BE(\text{He}) - BE(X)]$$

just use the formula & calculate Q .

From β -stability curve gives you

which Z value, nucleus is more stable.

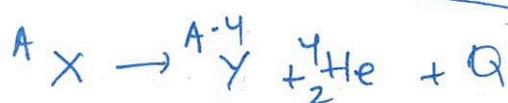


i.p.

$$\boxed{Z = \frac{A/2}{1 + 0.0078 A^{2/3}}}$$

if do calculation $A > 150$, Q been +ve - i.p. we should not have any nuclei, but we see upto 208 why:

150 — 208
stable } exist.



use conservation of linear momentum

$$\left. \begin{aligned} |\vec{P}_Y| &= |\vec{P}_d| = |\vec{P}| \\ p_y &= p_d = P \end{aligned} \right\} \text{same magnitude of momentum must be equal.}$$

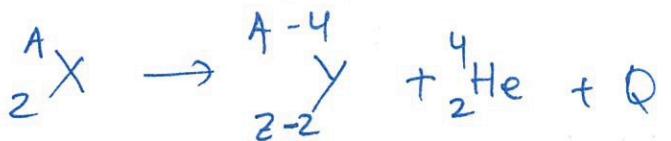
$$KE = \frac{p_d^2}{2m_d} + \frac{p_y^2}{2m_y} = Q$$

$$\Rightarrow \frac{p^2}{2} \left[\frac{m_d + m_y}{m_d m_y} \right] = Q \Rightarrow \frac{p^2}{2} = \frac{Q m_d m_y}{m_d + m_y}$$

$$\Rightarrow KE = \frac{p^2}{2m_d} = \left(\frac{Q m_d m_y}{m_d + m_y} \right) \frac{1}{m_d} = Q \left(\frac{m_y}{m_d + m_y} \right) = Q \left(\frac{A-4}{A} \right) = Q \left(1 - \frac{4}{A} \right)$$

$$K_Y = \frac{p^2}{2m_y} = Q \left(\frac{m_d}{m_d + m_y} \right) = Q \frac{4}{A+4} = Q \frac{4}{A}$$

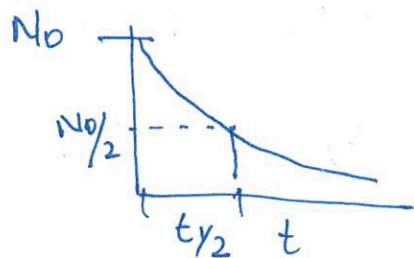
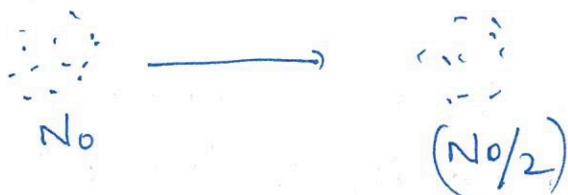
Q-value



$A > 150$, Q is positive.

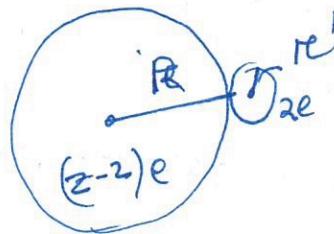
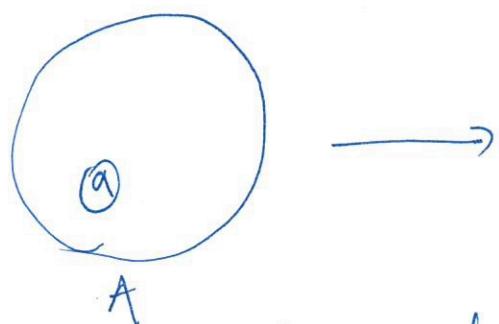
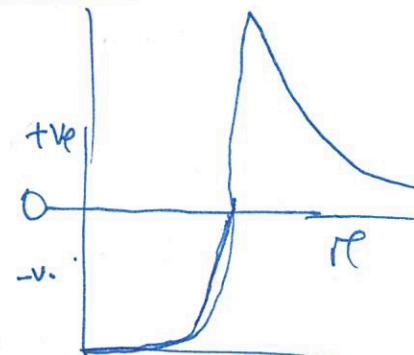
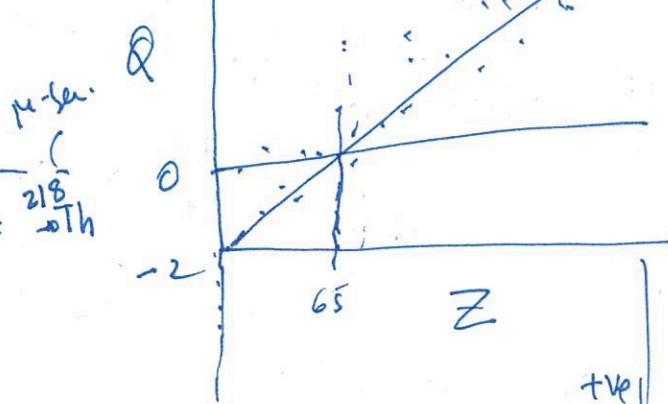
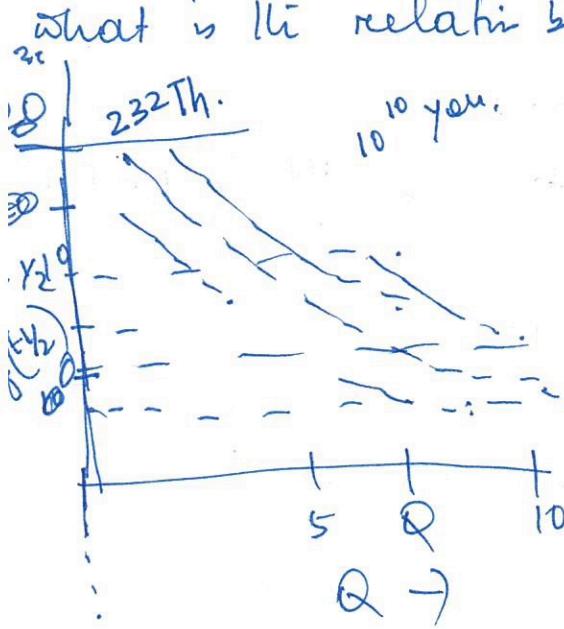
This is unusual
possible to have
 α -decay $A > 150$

But we also know that ${}^{208}\text{Pb}$ is very stable.



Q is observable & $t_{1/2}$ also experimentally observable.

What is the relation between Q & $t_{1/2}$?



$$V(r) = \frac{(Z-2)e \cdot Ze}{4\pi \epsilon_0 r}$$

If within nuclei $r < R$
↳ nuclear force dominant.

$V(r) =$ Nuclear attractive potential
= Coulomb repulsion potential

$r < R$
 $r > R$

Two

K_E is found.

$\propto \text{den.}$

$$\frac{dN}{dt} =$$

$t=0 \quad N_0 \quad \rightarrow t$
 $t: \quad N = N_0 e^{-\lambda t}$

at what time $N \rightarrow N/2$

$$\frac{N_0}{2} = N_0 e^{-\lambda t/2} \Rightarrow \frac{1}{2} = e^{-\lambda t/2} \Rightarrow \ln(2) = \lambda t/2$$

$$t_{1/2} = \frac{\ln 2}{\lambda}$$

$$\lambda = \frac{0.693}{t_{1/2}}$$

$\lambda \rightarrow$ is decay const.

\hookrightarrow probability of decay per atom unit time
of exponential smaller.

$$\frac{dN}{dt} \propto N$$

$$\Rightarrow \frac{dN}{dt} = -\lambda N \quad \text{under needn. at tm. t.}$$

$\frac{dN}{dt}$ is Rat.

$$\frac{dN}{N} = -\lambda dt$$

$$\int \frac{dN}{N} = -\lambda \int dt$$

use boundary cond.

$$\Rightarrow \ln N = -\lambda t + C$$

$$t=0 \quad N = N_0$$

$$\ln N_0 = C$$

$$\text{Thm.} \quad \ln N = -\lambda t + \ln N_0$$

$$\Rightarrow \ln \left(\frac{N}{N_0} \right) = -\lambda t$$

$$\Rightarrow \frac{N}{N_0} = e^{-\lambda t}$$

$\rightarrow t$

$$\Rightarrow N(t) = N_0 e^{-\lambda t}$$

what is the height of this potential from zero?
Coulomb Barrier height.

$$V_C = \frac{q(z-2)e^2}{4\pi\epsilon_0 r}$$

$$Z = 82, N = 208$$

$$r = R + r_e'$$

$$R = 1.18 \times A^{1/3}$$

$$= 1.18 \times (208)^{1/3}$$

$$\rightarrow 6.062$$

$$\Rightarrow r_c = \frac{2 \times 80}{9.2} \left(\frac{e^2}{4\pi\epsilon_0} \right)$$

$$R \approx 7.27 \text{ fm.}$$

$$r_e' = 1.504 \text{ fm.}$$

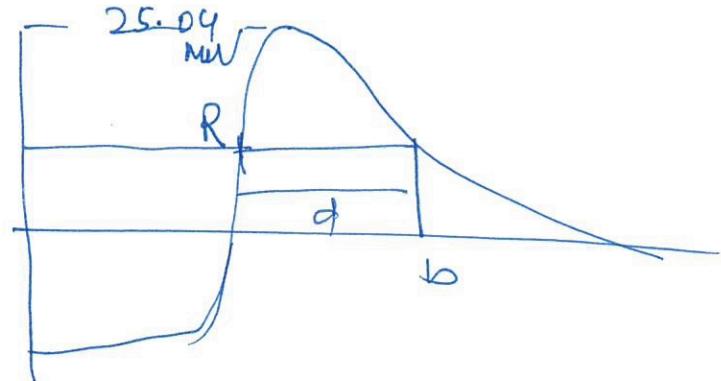
$$= \frac{2 \times 80}{9.2 \text{ fm}} \times 1.44 \text{ MeV fm} \Rightarrow 9.2 \text{ fm.}$$

$$= 25.04 \text{ MeV } \} \quad \text{But } Q \text{ is } \approx 4-5 \text{ MeV.}$$

probability.

$$P = e^{-\frac{r_e'}{d}}$$

Q5



$$\hookrightarrow V = \sqrt{\frac{2m}{t^2} (V_0 - E)}$$

$$\hookrightarrow r \cdot d = \int r(r) dr$$

$$\hookrightarrow P = e^{-2G}$$

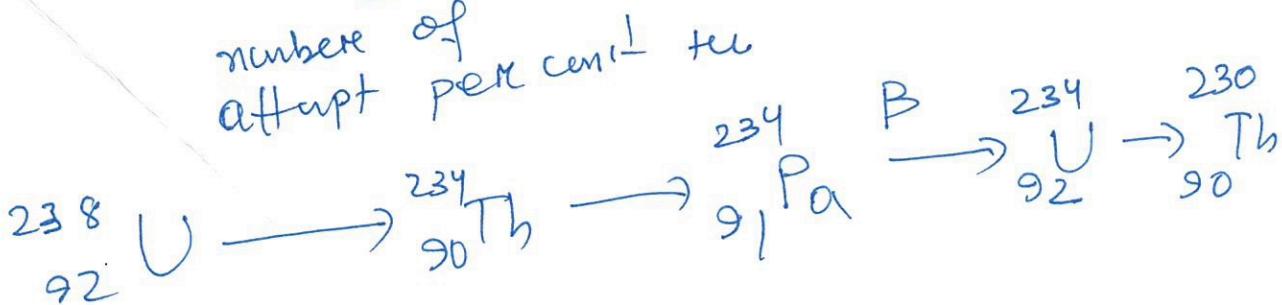
$$t = \frac{0.693}{\lambda}$$

$$G = \frac{q Z e^2}{4 \pi \epsilon_0 t \lambda}$$

speed of
 α -particle.

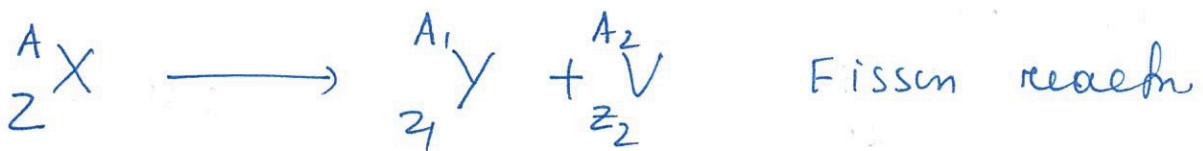
$$\lambda = f P \rightarrow \text{probability of attu}$$

\hookleftarrow
number of
attempt per cent⁻¹ sec



04/10/2024

Nuclear Fission

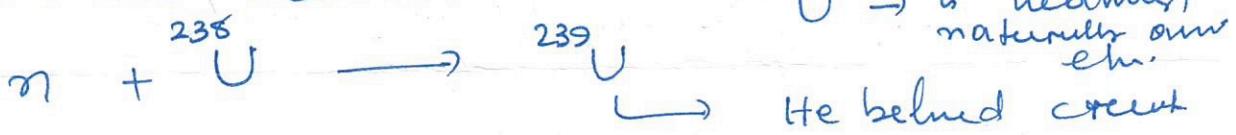


$$\left. \begin{array}{l} A = A_1 + A_2 \\ Z = Z_1 + Z_2 \\ N = N_1 + N_2 \end{array} \right\}$$

Tell the history.

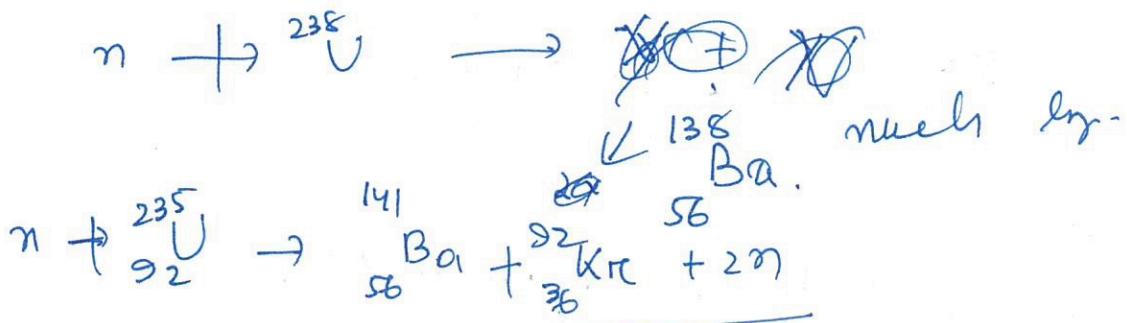
After theory of β -decay by Fermi 1934, it was established that β -decay can take down or up the atomic number by 1. (one).

So Fermi began bombarding n to the nuclei, try to create new heavier nuclei



But he failed crucial radioactive & could not understand what exactly happens

- [1938] Hahn and Strassmann



This was piezlij

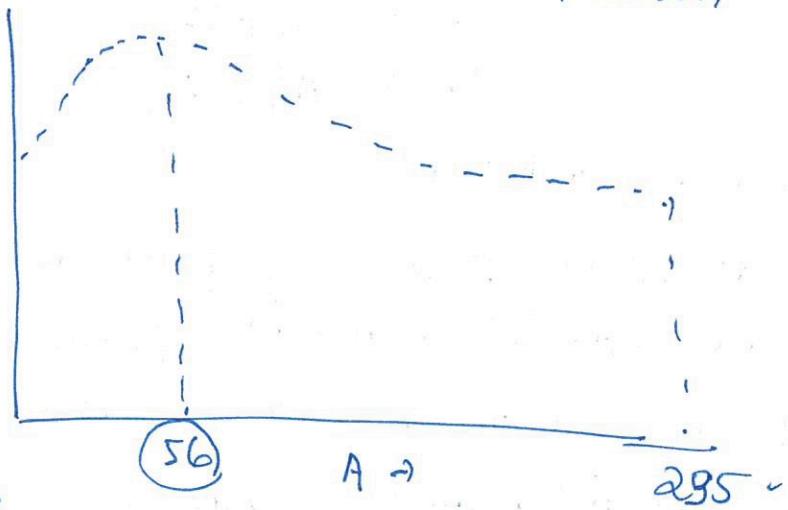
[1939] Lise Meitner & Otto Robert Frisch.
↳ Fleld Nazi Germany.

Try theoretically explain it by observing lot of any
out com.

Fission
 ↓
 nuclear - split roughly equal } protons
 Nuclear Fission

$^{235}_{92}U$

②



$$BB = \alpha_1 A - \alpha_3 A^{2/3}$$

$$- \alpha_c \frac{2(E^2-1)}{A^{1/2}} - \alpha_{sy} \frac{(A-2Z)^2}{A} \pm \delta(A, Z)$$

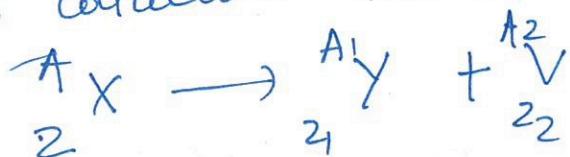
$$\Rightarrow BB = 1328.09 \text{ MeV.}$$

$$(BB/\eta) = 5.6 \text{ MeV}/\eta$$

$$(BB/\eta)_{Ba} = 7.35 \text{ MeV}/\eta$$

$$(BB/\eta)_{K\pi} = (6.88 \text{ MeV/nucleon})$$

Let's calculate the theoretical value.



$$Q = m(X) - m(Y) - m(V)$$

$$\Rightarrow \cancel{m(X)} = BB$$

$$Q = A \cancel{m_n}$$

$$\begin{aligned} & \cancel{N m_n} \\ \left(N m_p + Z m_p - BB(x) \right) &= \left\{ \cancel{N_1 m_n} + Z m_p - BB(y) \right\} \\ &= \left\{ N_2 m_n + Z m_p - BB(y) \right\}. \end{aligned}$$

$$= -BB(x) + BB(y) + BB(v)$$

$$= -5.6 \times A + 7.35 \times A_1 + 6.88 \times A_2$$

$$= -5.6 \times 235 + 7.35 \times 141 + 6.88 \times 92$$

$$= \frac{-1316 + 1036.35 + 632.96}{1669.31}$$

$$= 353.3 \text{ MeV.}$$

What is min A for fission to happen.

$$A_1 = A_2 = A/2 \text{ symmetric f.}$$

$$M = (Z m_p + N m_n - BB)$$

$$BB = Q_V A - Q_S A^{2/3} - Q_C \frac{2(2-1)}{A^{1/3}} - Q_{sys} \frac{(A-2Z)^2}{A} \pm \delta (\text{pair})$$

↑ no chs

$$Q_V \left(\frac{A}{2} \right) - Q_S \left(\frac{A}{2} \right)^{2/3} - Q_C \frac{2(2-1)}{A^{1/3}} \downarrow -$$

$$2 \left(\frac{A}{2} \right)^{2/3} > (A)^{2/3}$$

$\frac{2}{1.5872}$

$$\frac{(A-2Z)^2}{A}$$

$$2 \frac{\left(\frac{A}{2} - \frac{Z}{2} \right)^2}{A/2} = \left(\frac{A-2Z}{A} \right)$$

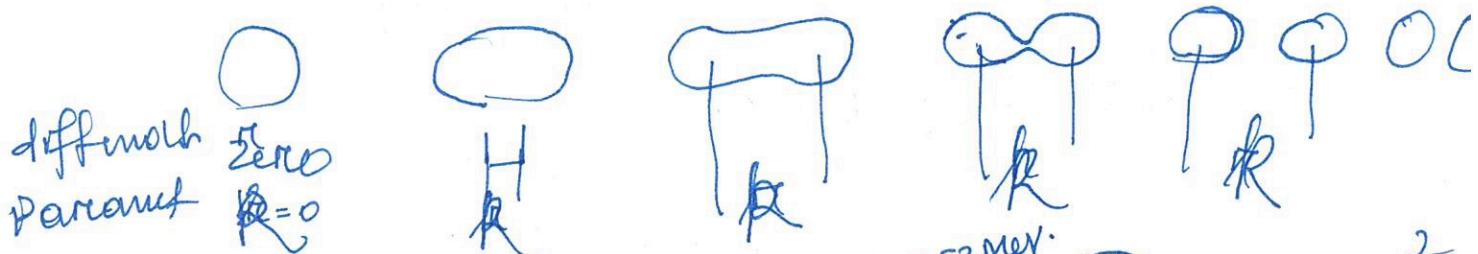
$$mc^2 = \cancel{m_0 c^2} + A$$

$$m = m_0 + A - \left(-\alpha_s A^{2/3} - \alpha_s \frac{z^2}{A^{1/3}} \right)$$

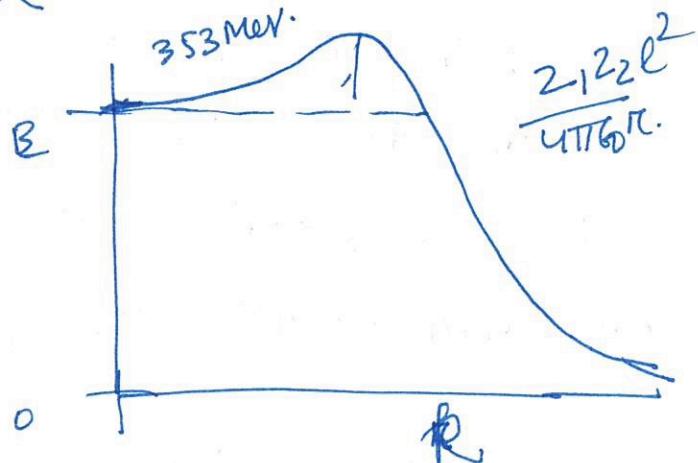
$$\boxed{Q = M - 2M_y}$$

$$= \alpha_s \left[A^{2/3} - 2 \left(\frac{1}{2} \right)^{2/3} \right] \\ \alpha_s \left[\frac{z^2}{A^{1/3}} - 2 \frac{\left(\frac{z}{2} \right)^2}{\left(\frac{1}{2} \right)^{2/3}} \right] \quad \left. \right\} > 0$$

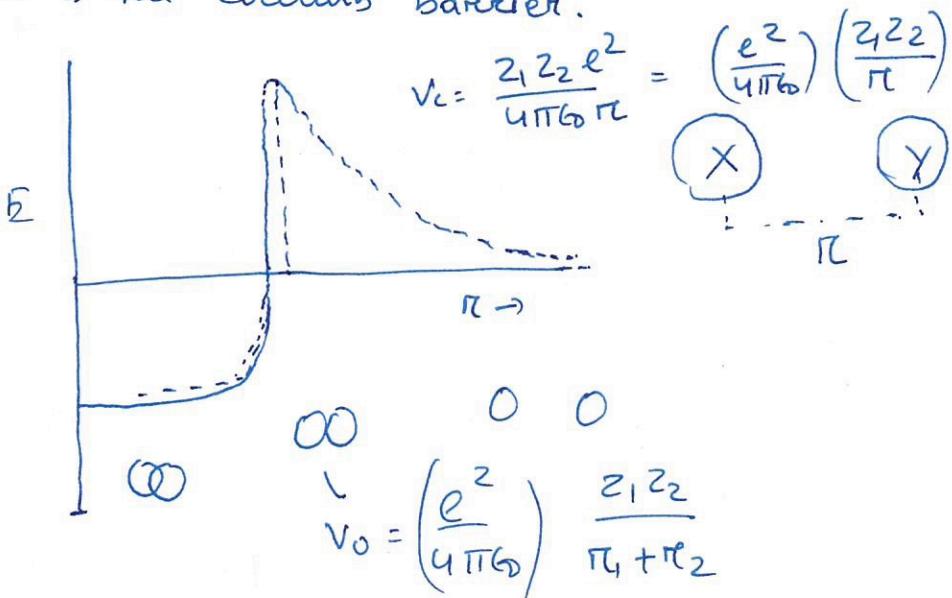
$$A > 150$$



activation Energy
Baron height



How fusion happens - conditions for nuclear fusion.
Actual issue is the coulomb barrier.



$$^{12}_6 C + ^{12}_6 C : V_0 = 1.44 \text{ MeV.fm} \quad \frac{6 \cdot 6}{2 \times 0.8 \times 1.2 \times 3} = 1.44 \times 4.367 = 6.288 \text{ MeV}$$

$$^1 H + ^1 H : V_0 = \frac{1.44}{2 \times 0.8} \text{ MeV} = \frac{1.44}{1.6} \text{ MeV} = 0.9 \text{ MeV}$$

$$^{40}_{20} Ca + ^{40}_{20} Ca : \frac{1.44 \times 20 \times 20}{2 \times 1.8 \times 40 \times 3} = \frac{1.44 \times 20 \times 20}{2 \times 1.18 \times 3.42} = 72.36485 \text{ MeV}$$

Do the discern how to get this
energy to cross the barrier.

→ Kinetic energy \rightarrow accelerator.

\downarrow Take it to high temperature

\downarrow create a fusion

\downarrow There will be random motion

\downarrow

They will have kinetic energy

\downarrow

use this \leftarrow These are thermal kinetic energy
to create \leftarrow reaction.

↓
starts

↑
Thermo-nuclear Fusion

↑

use this \leftarrow To create
reaction.

Room Temperature

$$kT = 0.027 \text{ eV}$$

$$E = kT$$

\downarrow average kinetic energy.

$$\boxed{\langle E \rangle = \frac{3}{2} kT} \Rightarrow T \Rightarrow$$

$$= \frac{0.027 \times 3}{2}$$

$$\frac{3}{2} kT = 8.617 \times 10^{-5} \text{ eV/K} \times T$$

$$900 \text{ KeV} =$$

$$T = \frac{900 \text{ keV}}{8.617 \text{ eV}} \times 10^5 \text{ K} = \frac{900,000}{8.617} \times 10^5 \text{ K} = 1.045 \times 10^{10} \text{ K}$$

Nuclear fusion. 7/10/24 ①

$${}_{Z_1}^{A_1} X + {}_{Z_2}^{A_2} Y \longrightarrow {}_Z^A V$$

$$Q = m(X) + m(Y) - m(V)$$

$$= Z_1 m_p + Z_2 m_n - BB$$

$$= Z_1 m_p + Z_2 m_n - Z_1 m_p - Z_2 m_n - BB$$

$$= Z_1 m_p + Z_2 m_n - Z_1 m_p - Z_2 m_n - BB$$

$$\Rightarrow A_1 + A_2 = A$$

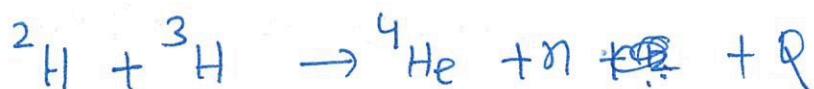
$$Z_1 + Z_2 = Z$$

$$N_1 + N_2 = N$$

$$= -BB^X - BB^Y + BB^V$$

$$\Rightarrow Q = BB^V - BB^X - BB^Y$$

$$= (BB/n)A - (BB/n)A_1 - (BB/n)A_2$$



$${}^2\text{H} : \Delta m = 0.001839$$

$$\Delta B = 0.1714 \text{ MeV}$$

$$(BB/n) = 0.857 \text{ MeV}$$

$$m({}^2\text{H}) = 2.0141024$$

$$m_p = 1.0072764$$

$$m_n = 1.0086654$$

$${}^3\text{H} : \Delta m = 0.008557$$

$$\Delta B = 7.973$$

$$(BB/n) = 2.658 \text{ MeV}$$

$$m({}^3\text{H}) = 3.0160494$$

$${}^4\text{He} : \Delta m = 0.029279$$

$$\Delta B = 27.41 \text{ MeV}$$

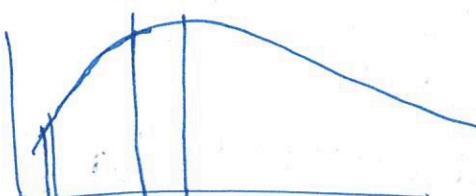
$$(BB/n) = 6.85 \text{ MeV}$$

$$m({}^4\text{He}) = 4.0026034$$

$$Q = 6.85 \times 4 - 0.857 \times 2 - 2.658 \times 3 \text{ MeV}$$

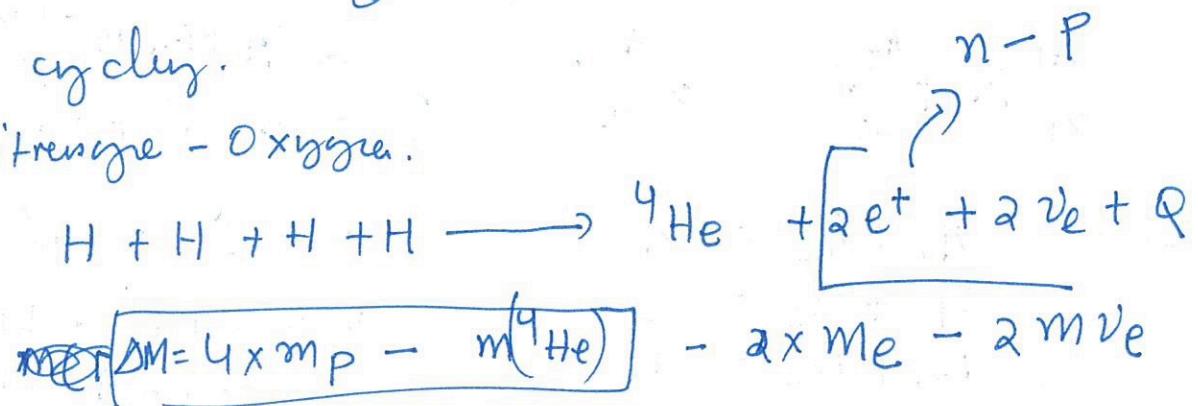
$$= 27.41 - \underbrace{1.714 - 7.974}_{9.688} = 17.722 \text{ MeV}$$

$Q > 0$: There will be release of energy.



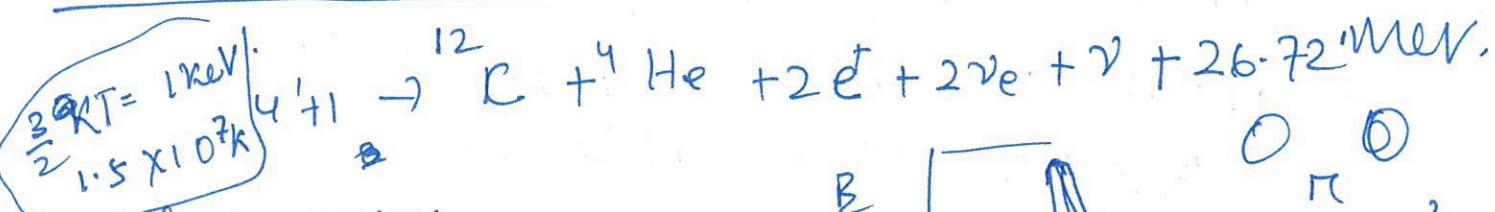
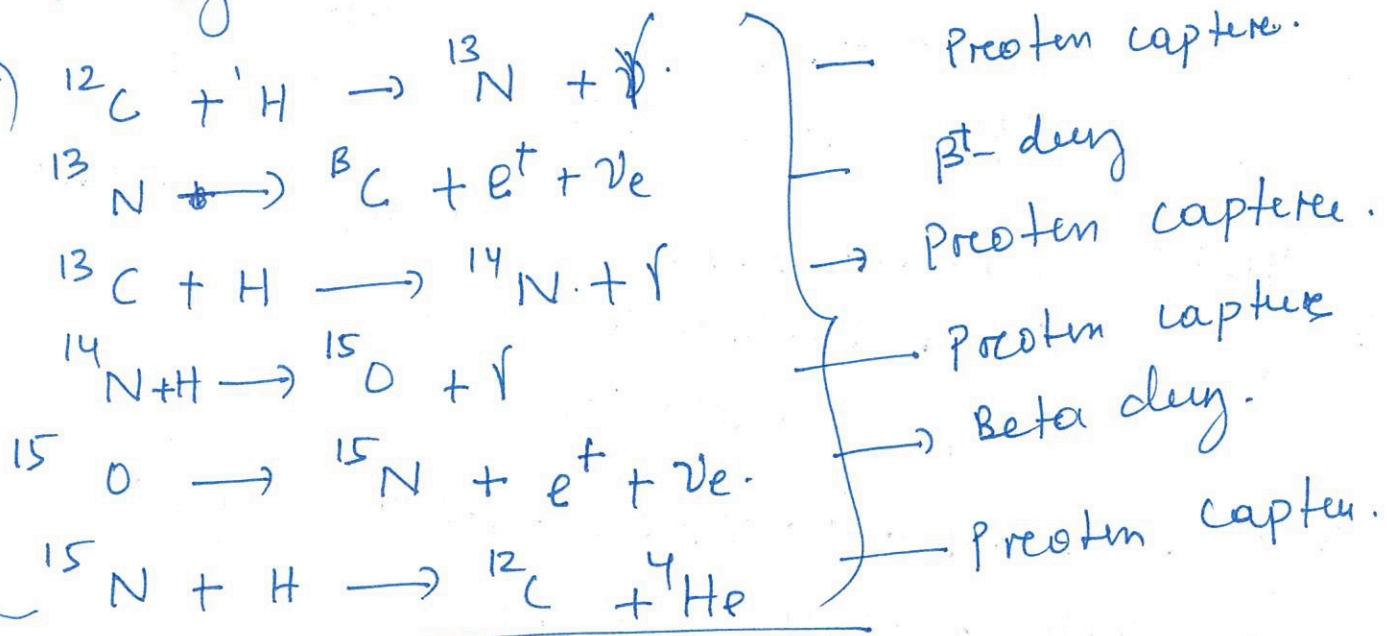
CNO - cycling.

carbon - Nitrogen - Oxygen.



$$\begin{aligned}\Delta M &= (4.031300 - 4.002603) c^2 - 2M_e - 2m\nu_e \\ &= 0.028697 \times 931.5 \text{ MeV} \\ &= 26.72 - \cancel{1.022} \text{ MeV} \\ &= \underline{\underline{25.695 \text{ MeV}}}\end{aligned}$$

interestingly CNO been catalyst for this reaction



Coulomb barrier

$$V_0 = \frac{2e^2}{4\pi\epsilon_0(R_1 + R_2)}$$

900 KeV

$$2.0 \times 1.44 \text{ MeV fm} = \frac{1.44}{(0.8 + 0.8)} = 1 \text{ MeV}$$

0.9 MeV

B



$$\frac{2e^2}{4\pi\epsilon_0 R} = \frac{1.44}{1 \text{ fm}}$$

$$\left(\frac{3kT}{2}\right) = 150 \text{ KeV}$$