



Structure of disordered materials



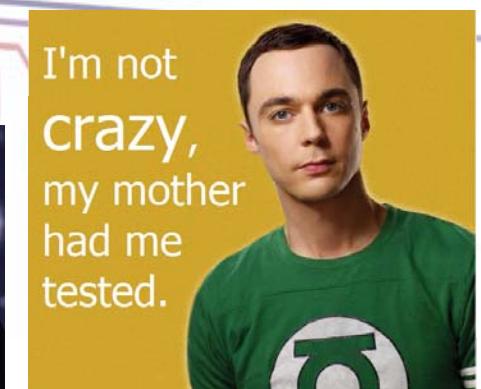
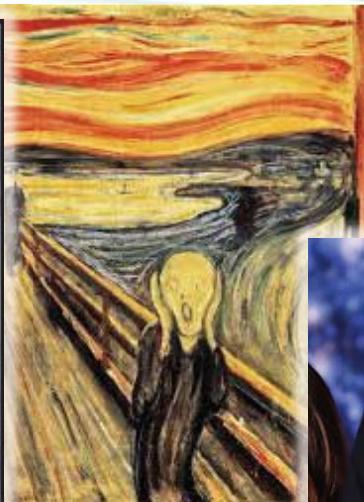
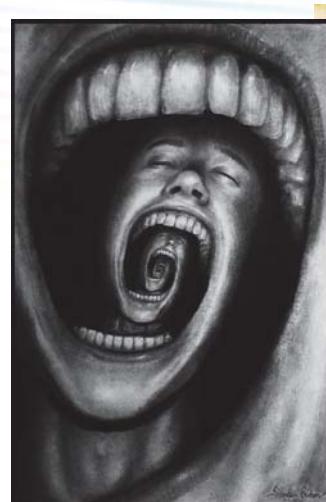
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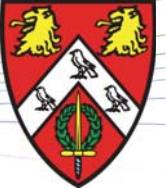
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16th Oxford School on Neutron Scattering
2 - 13 September 2019
St. Anne's College, University of Oxford



Disorder



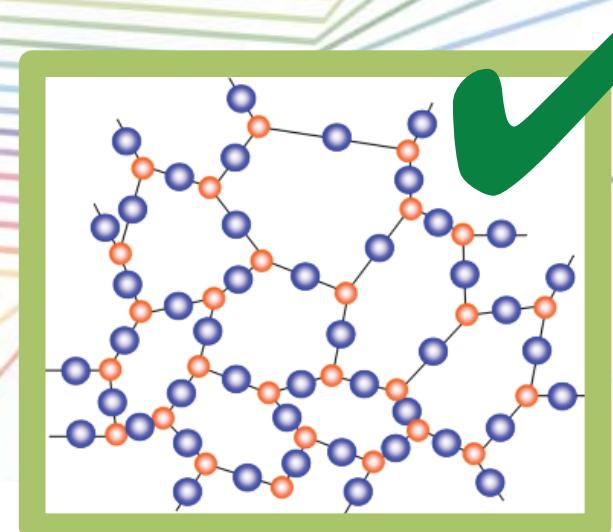
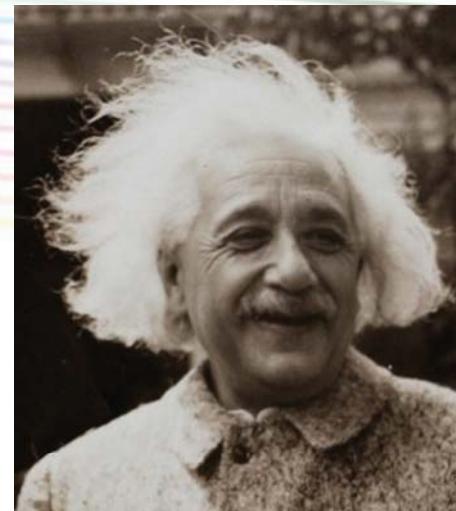


Disordered things



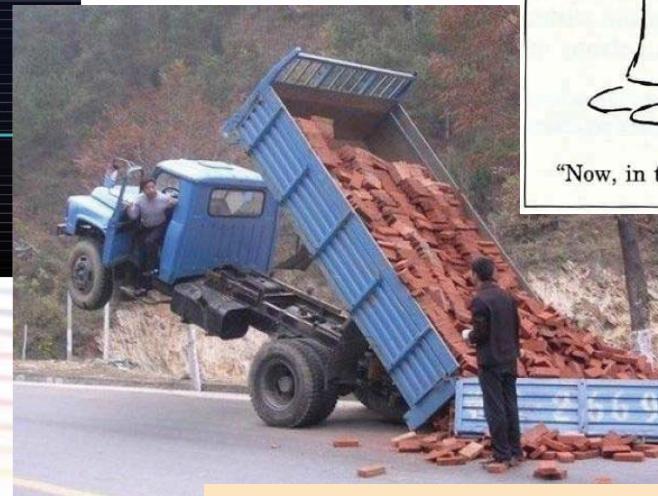
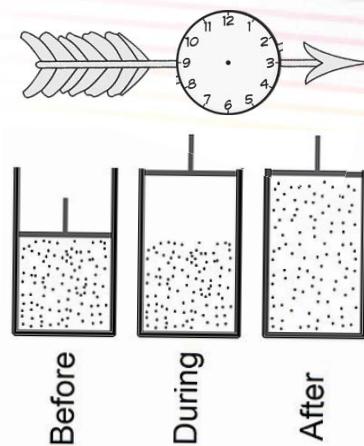
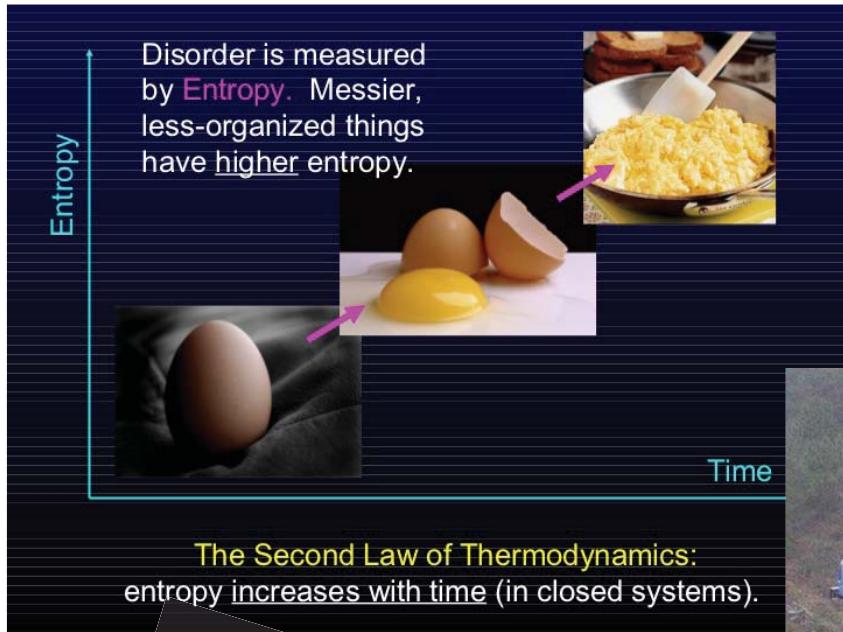
"If a cluttered desk is a sign of a cluttered mind, of what, then, is an empty desk a sign?"

— Albert Einstein



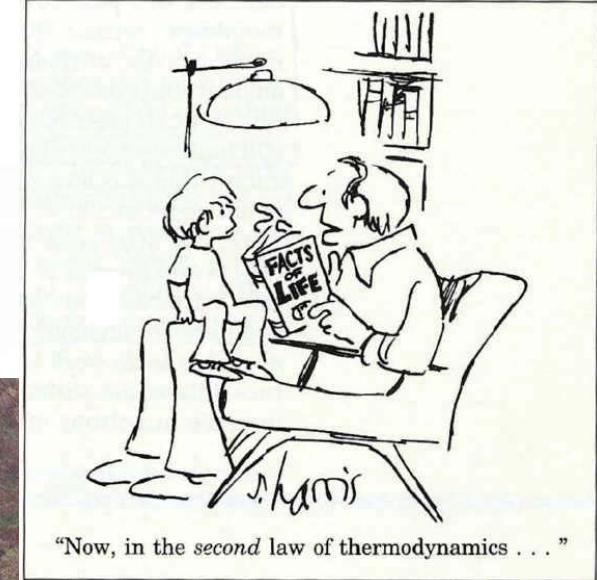


Disorder in Physics

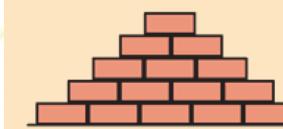


Entropy

$$\frac{dS}{dt} > 0$$



If you tossed bricks off a truck, which kind of pile of bricks would you more likely produce?



Disorder is more probable than order.





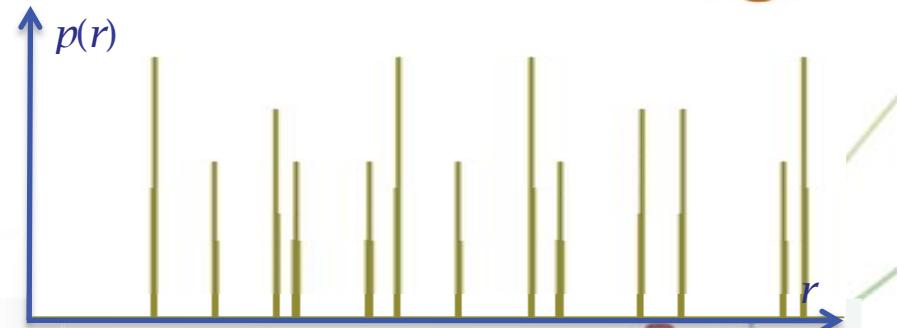
Order vs disorder



Disorder is fun

Creative clutter
is better than
idle neatness.
-Anonymous

Order is boring





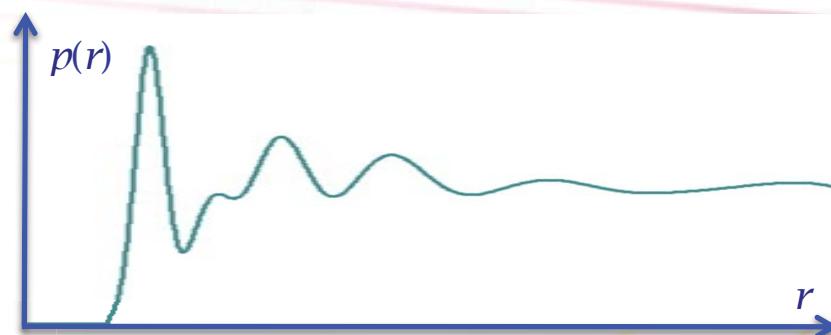
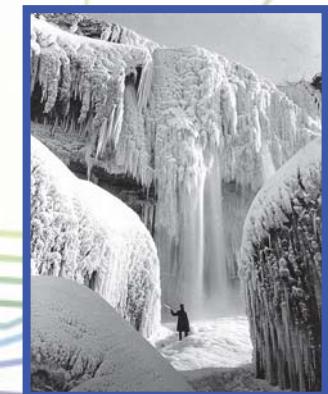
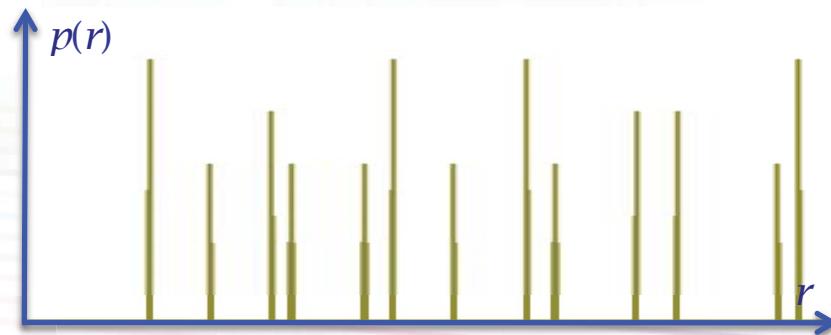
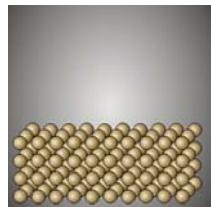
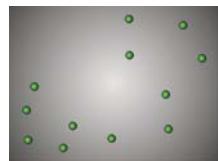
Pair distribution function

©Jan van Haasteren



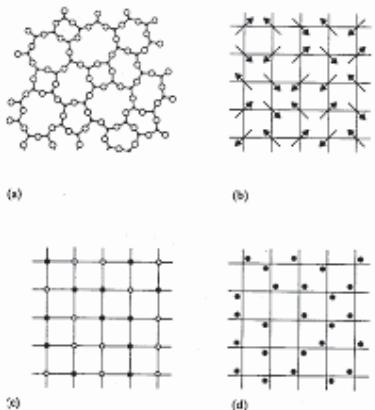


Structure in real space

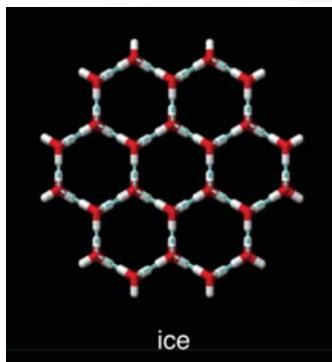




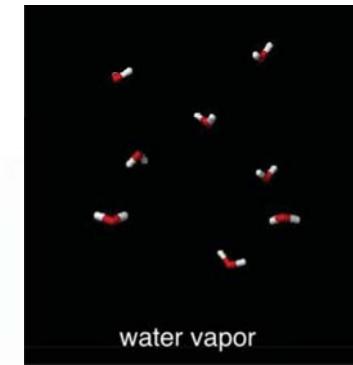
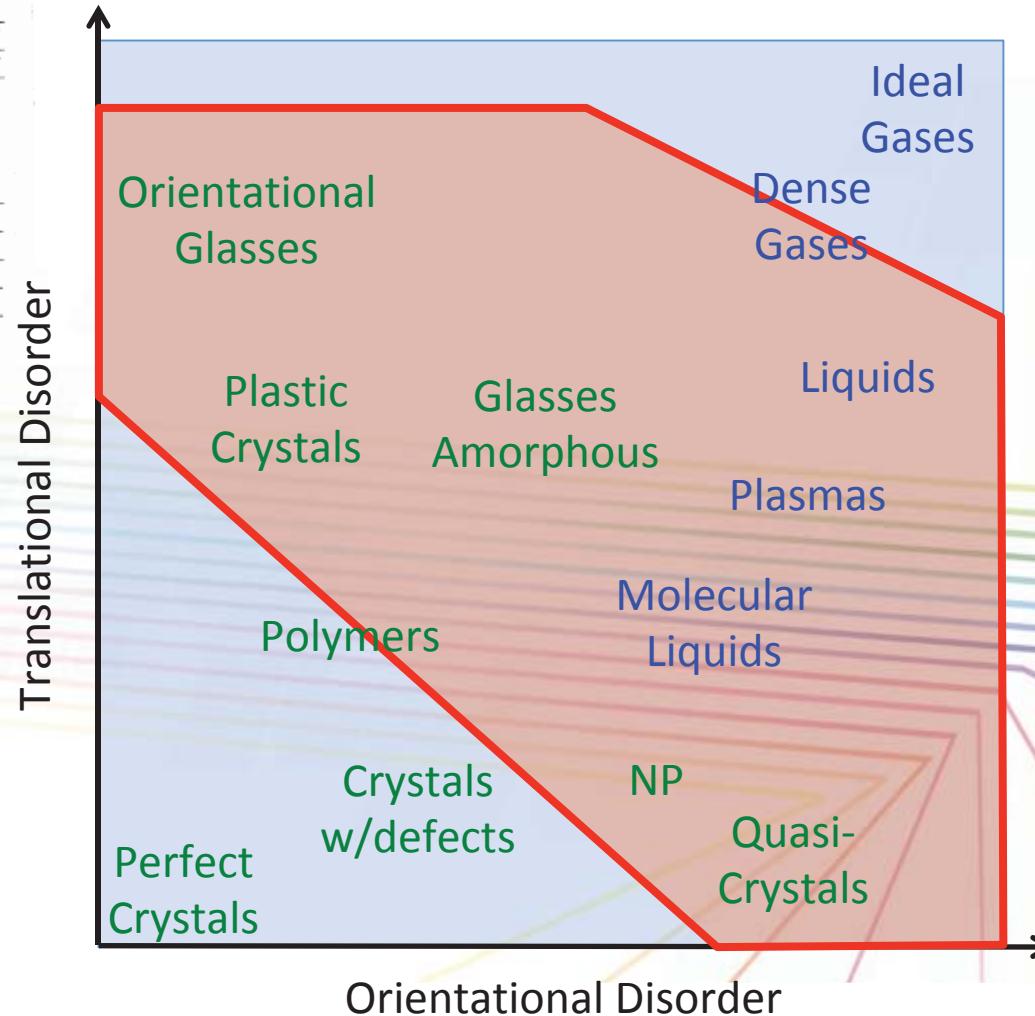
Disorder in solids and fluids



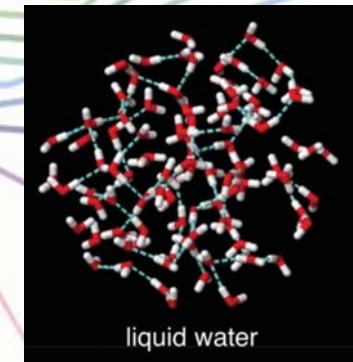
Solids



Total Scattering Techniques



Fluids





Amorphous solids and glasses

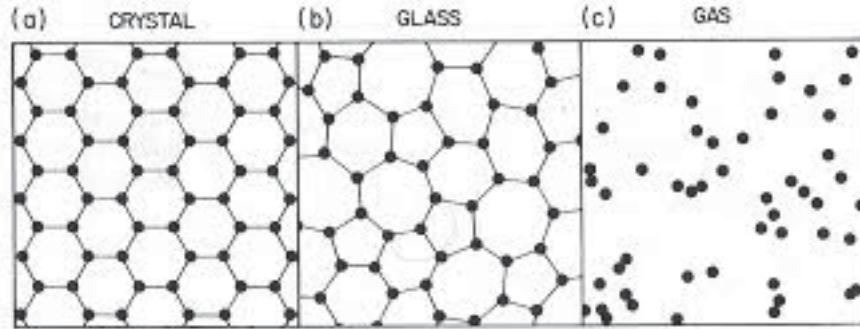
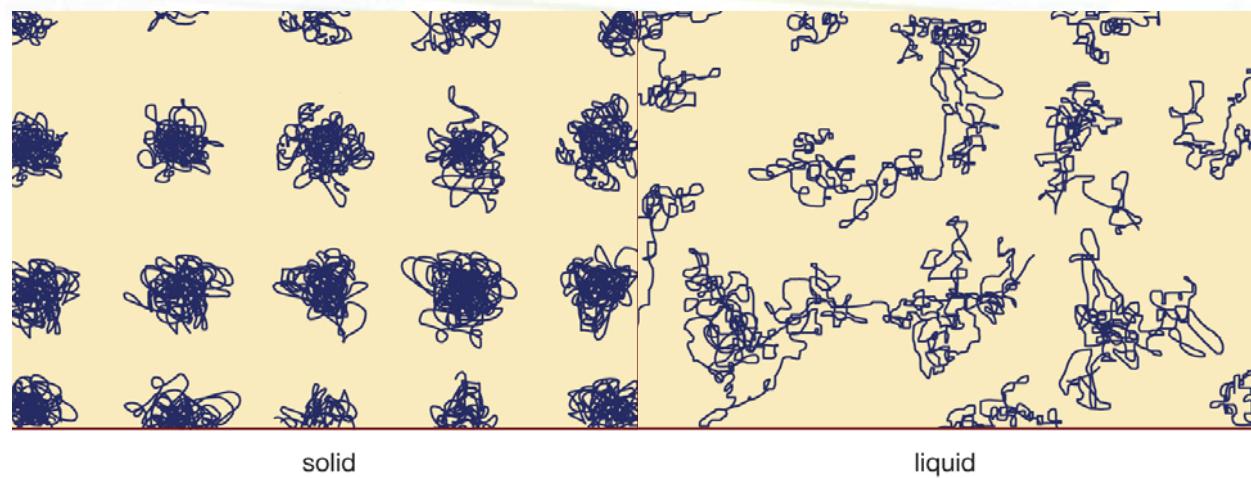
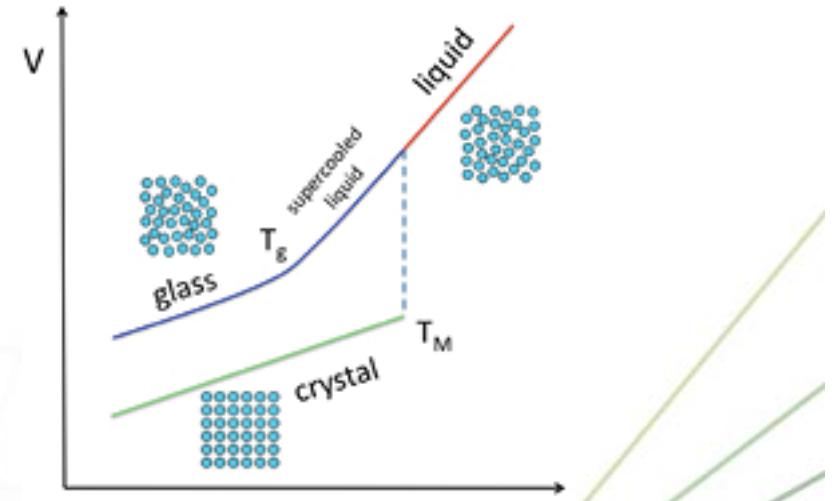


Figure 1.6 Schematic sketches of the atomic arrangements in (a) a crystalline solid, (b) an amorphous solid, and (c) a gas.

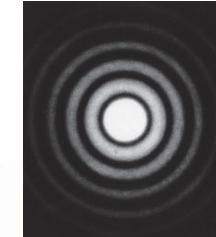
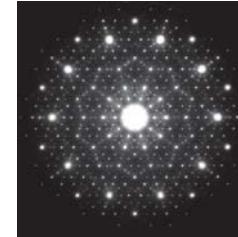
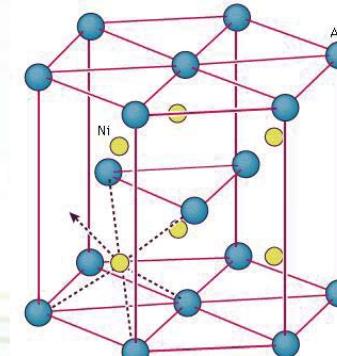




Structure by diffraction



Spatial distribution of atoms or molecules in the system

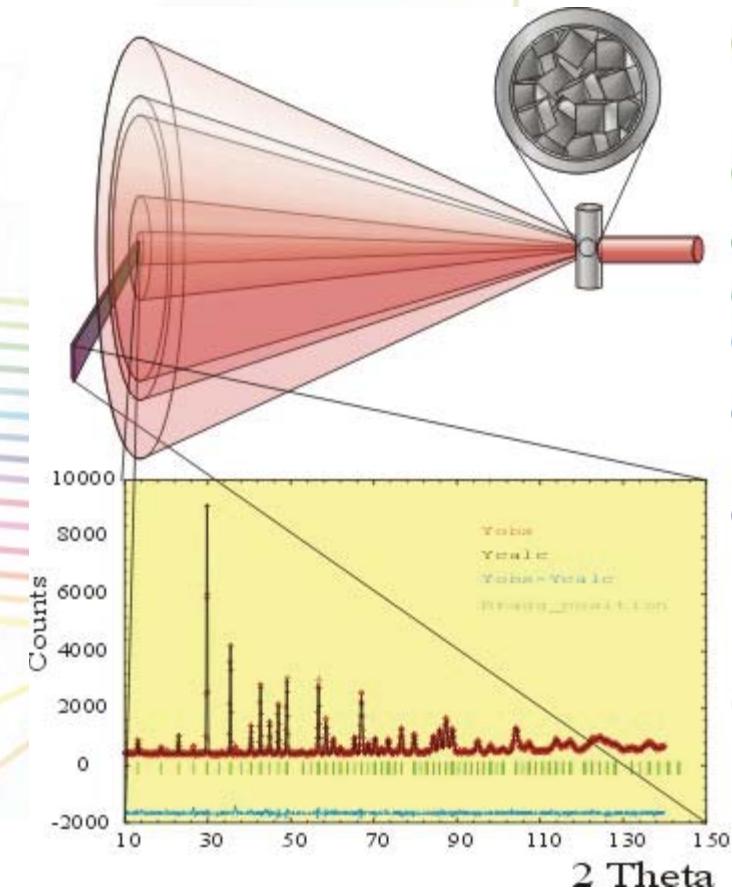


Crystalline solids

- { Equilibrium positions
- Well defined Bragg peaks

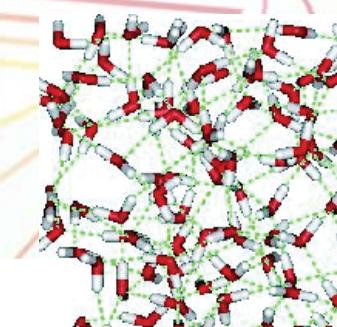
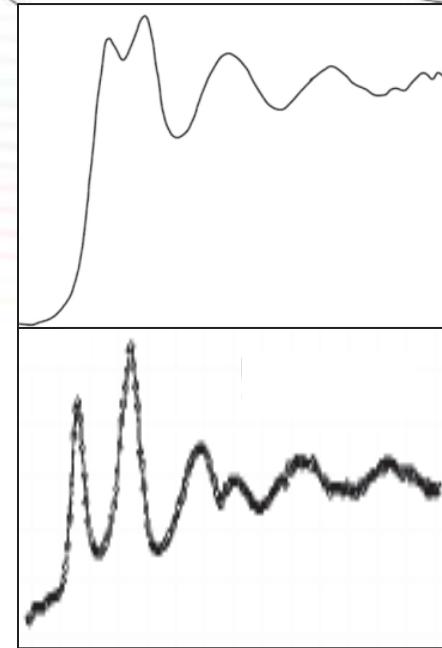
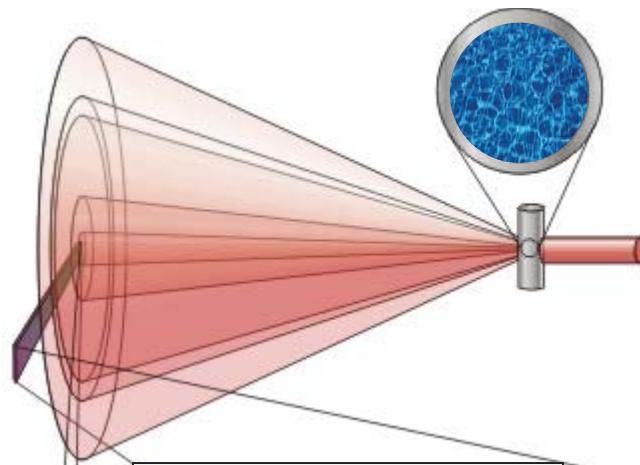
Disordered systems

- { Distribution of equilibrium positions
- No Bragg peaks





Disordered systems





Structure factor for a gas

Exercise

- A- Determine the static structure factor for a monoatomic gas of punctual non-interacting particles.
- B- Determine the static structure factor for a gas where the particles have finite size and consequently an effective repulsion distance. Determine the maximum possible density of such a gas.
- C- Study the low- Q limit of the structure factor. Should the structure factor be null at the origin?

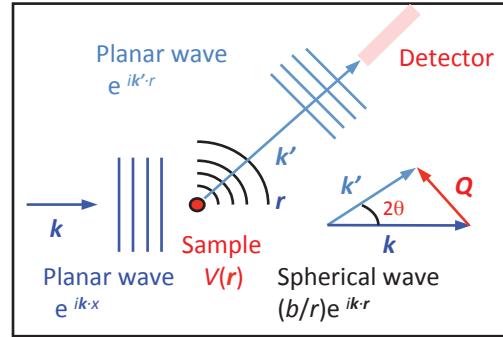




Neutron scattering



Experiment



Microscopic properties

Intensity
 $I(2\theta, \omega)$

Scattering Cross Section
 $d^2\sigma/d\Omega d\omega$

Dynamical SF
 $S(\vec{Q}, \omega)$

$$I(2\theta, \omega) = C \Phi_0 \frac{d^2\sigma}{d\sigma d\omega}(2\theta, \omega) \epsilon(k')$$

Beam

Sample

Detector

$$\frac{d^2\sigma}{d\sigma d\omega}(2\theta, \omega) = N \frac{k'}{k} \frac{\sigma}{4\pi} S(\vec{Q}, \omega)$$

Interaction System



Microscopic properties



Dynamical Structure Factor $S(\mathbf{Q}, \omega)$

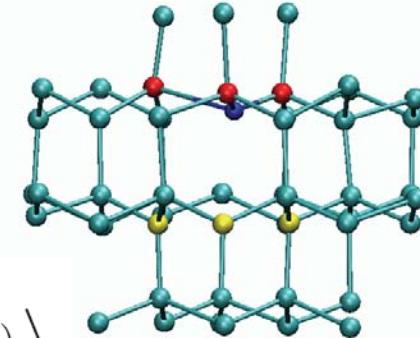
Microscopic Configuration

$$\{\mathbf{r}_1(t), \mathbf{r}_2(t), \dots, \mathbf{r}_N(t)\}$$

$$S(\vec{Q}, \omega) = \frac{1}{2\pi} \int_{-\infty}^{+\infty} dt e^{-i\omega t} \frac{1}{N} \sum_{i,j}^N \left\langle e^{-i\vec{Q} \cdot \vec{r}_i(0)} e^{-i\vec{Q} \cdot \vec{r}_j(t)} \right\rangle$$

$$\delta(\vec{r}) = \frac{1}{(2\pi)^3} \int_{\text{all } \vec{Q}} e^{i\vec{Q} \cdot \vec{r}} d\vec{Q}$$

$$\rho(\vec{r}, t) = \sum_{i=1}^N \delta(\vec{r} - \vec{r}_i(t))$$



Microscopic particle density

$$S(\vec{Q}, \omega) = \frac{1}{2\pi} \iint d\vec{r} dt e^{i(\vec{Q} \cdot \vec{r} - \omega t)} G(\vec{r}, t)$$

van Hove's correlation function

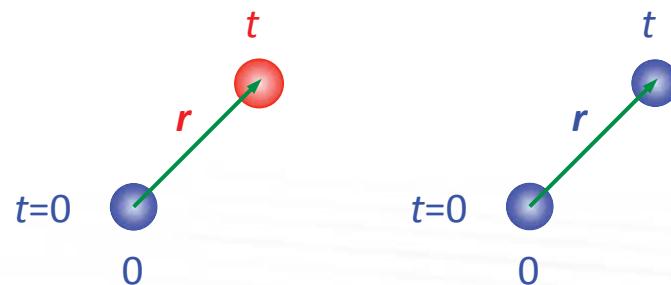
Probability density of having a given atom somewhere,
e.g. at $(0, 0)$, and any atom at (\vec{r}, t)

$$G(\vec{r}, t) = \frac{1}{N} \int d\vec{r}' \left\langle \rho(\vec{r}', 0) \rho(\vec{r} + \vec{r}', t) \right\rangle$$

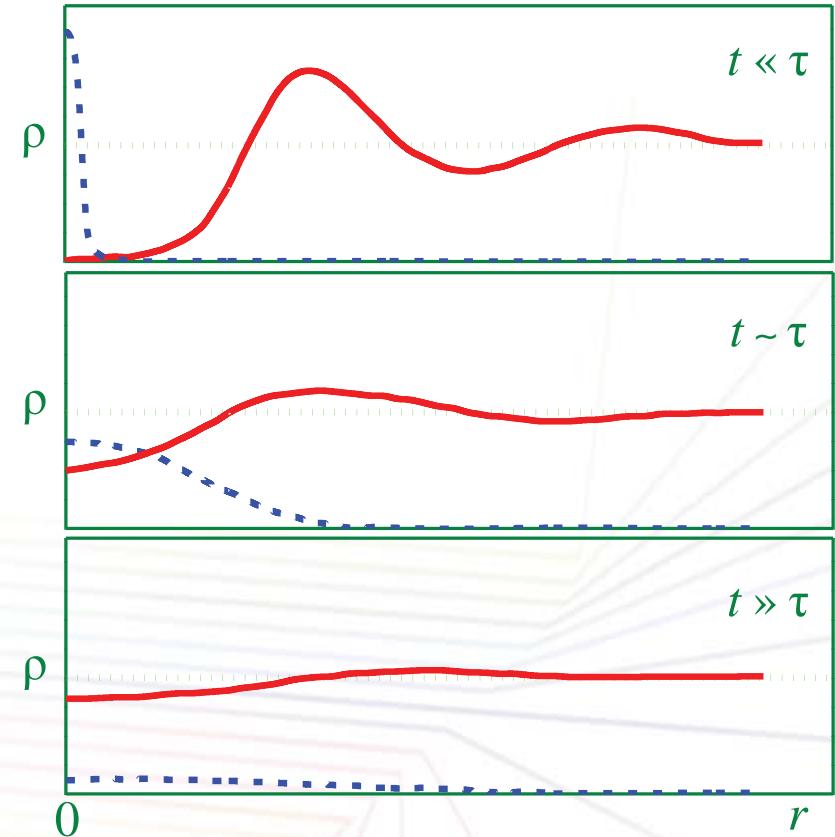


Correlation functions

$$G(\mathbf{r}, t) = G_d(\mathbf{r}, t) + G_s(\mathbf{r}, t)$$



Normalisation	N	$N-1$	1
Large r	ρ	ρ	0
Large t	ρ	ρ	0



$$\frac{d^2\sigma}{d\Omega d\omega}(2\theta, \omega) = N \frac{k'}{k} \left\{ |\langle b \rangle|^2 S_d(\vec{Q}, \omega) + \langle |b|^2 \rangle S_s(\vec{Q}, \omega) \right\}$$



Coherent and incoherent xs



Exercise

Starting with the total scattering cross section and considering the self and distinct contributions, derive the expression showing explicitly the coherent and incoherent contributions.





Coherent and incoherent scattering

$$\frac{d^2\sigma}{d\Omega d\omega}(2\theta, \omega) = N \frac{k'}{k} \left\{ \frac{\sigma_{coh}}{4\pi} S(\vec{Q}, \omega) + \left(\frac{\sigma_{inc}}{4\pi} \right) S_s(\vec{Q}, \omega) \right\}$$

coherent

incoherent

$$\sigma_{coh} = 4\pi |\langle b \rangle|^2 \quad \text{squared mean}$$

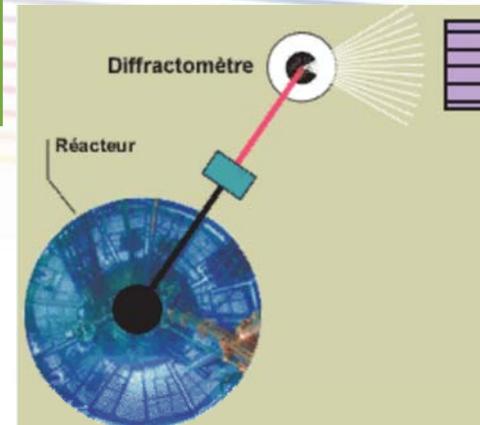
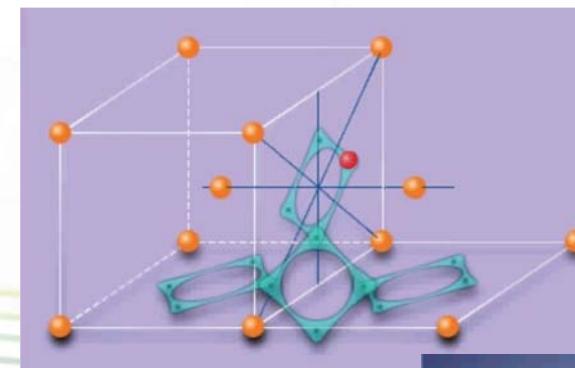
$$\sigma_{inc} = 4\pi \left(\langle |b|^2 \rangle - |\langle b \rangle|^2 \right) \text{variance}$$

Scattering cross section

$$\sigma = \sigma_{coh} + \sigma_{inc}$$

Structural information
Collective dynamics

Diffraction
(no energy discrimination)



Diffusive
Motions
Internal
Dynamics



Static structure factor

Exercises

Find the relationship between the van Hove time-dependent correlation function $G(\mathbf{r}, t)$ and the static structure factor $S(\mathbf{Q})$.



Propose an expression for the static correlation function using the macroscopic density of the material and the real space pair distribution function $g(\mathbf{r})$.

Using this expression, find the relationship between the static structure factor $S(\mathbf{Q})$ and $g(\mathbf{r})$.





Static structure factor



$$S(\vec{Q}) = \int_{-\infty}^{+\infty} d\omega S(\vec{Q}, \omega) = \int d\vec{r} e^{i\vec{Q}\cdot\vec{r}} G(\vec{r}, 0)$$

Static approximation

$S(Q)-1$ &
 $g(r)-1$
become a
FT pair

$$S(\vec{Q}) - 1 = \rho \int_V d\vec{r} [g(\vec{r}) - 1] e^{i\vec{Q}\cdot\vec{r}}$$
$$\rho [g(\vec{r}) - 1] = \frac{1}{(2\pi)^3} \int d\vec{Q} [S(\vec{Q}) - 1] e^{-i\vec{Q}\cdot\vec{r}}$$

Definition

$$F(\vec{Q}) = S(\vec{Q}) - 1$$

$$G(\vec{r}) = 4\pi\rho r [g(\vec{r}) - 1]$$

$$F(\vec{Q}) = \int_V d\vec{r} \frac{G(\vec{r})}{4\pi r} e^{i\vec{Q}\cdot\vec{r}}$$

$$\frac{G(\vec{r})}{4\pi r} = \frac{1}{(2\pi)^3} \int d\vec{Q} F(\vec{Q}) e^{-i\vec{Q}\cdot\vec{r}}$$

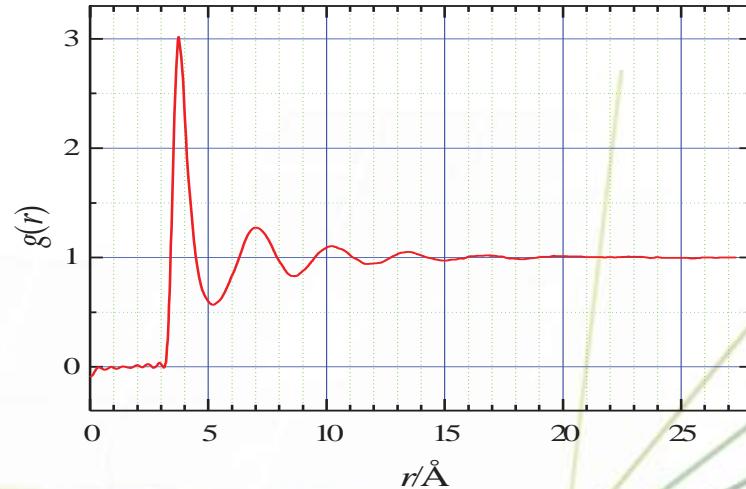
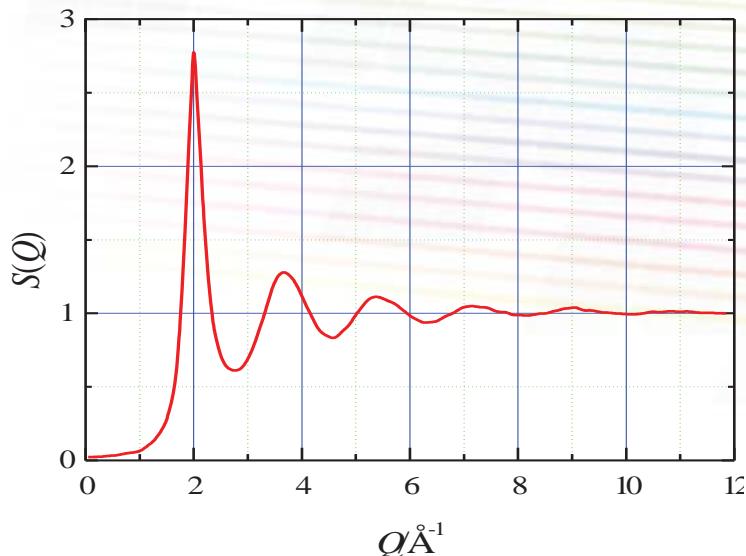


Isotropic case



$$Q F(Q) = \int_0^\infty G(r) \sin(Qr) dr$$

$$G(r) = \frac{2}{\pi} \int_0^\infty Q F(Q) \sin(Qr) dQ$$



$$S(Q) - 1 = \frac{4\pi\rho}{Q} \int_0^\infty r [g(r) - 1] \sin(Qr) dr$$

$$g(r) - 1 = \frac{1}{2\pi^2\rho r} \int_0^\infty Q [S(Q) - 1] \sin(Qr) dQ$$



Isotropic structure factor



Exercises

Consider now the isotropic case and find the relationship between the isotropic structure factor $S(Q)$ and the isotropic pair distribution function $g(r)$.



- A- Consider a well-ordered system with a single (sharp) peak in the real space correlation function and calculate the corresponding structure factor.
- B- Discuss the changes you could expect in the structure factor if you had a distribution of distances around a single mean-value.
- C- Discuss now what you should observe if you had multiple distances, each one with a distribution around the mean value.



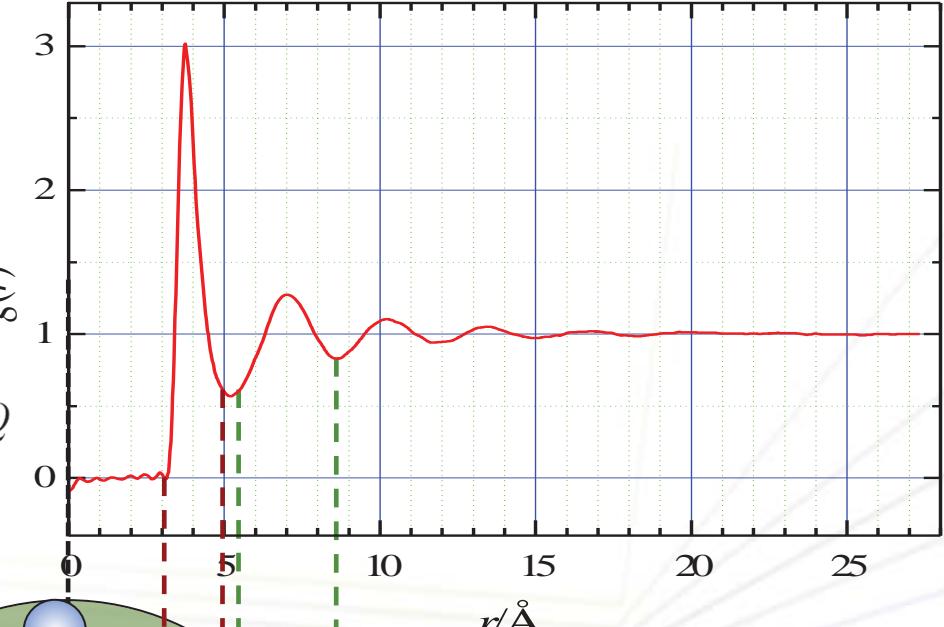
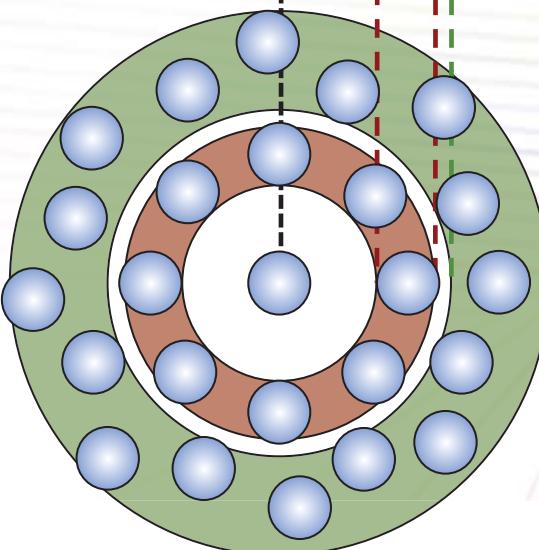
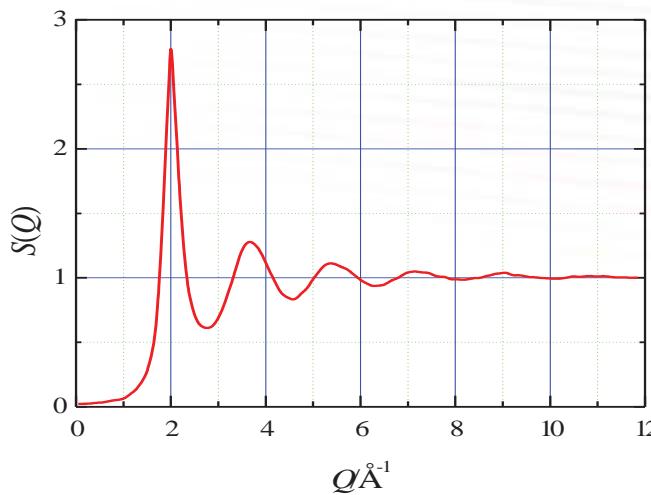


Pair distribution function



$$S(Q) - 1 = \frac{4\pi\rho}{Q} \int_0^\infty r [g(r) - 1] \sin(Qr) dr$$

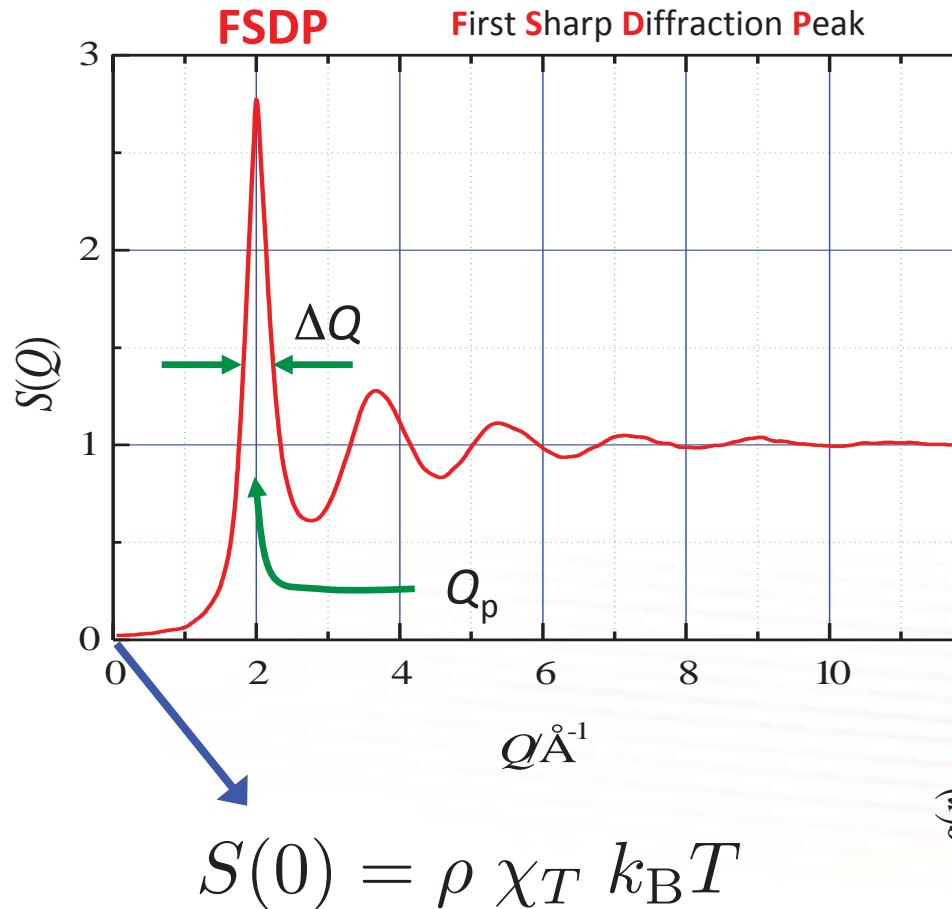
$$g(r) - 1 = \frac{1}{2\pi^2\rho r} \int_0^\infty Q [S(Q) - 1] \sin(Qr) dQ$$



2D analogy



First Sharp Diffraction Peak

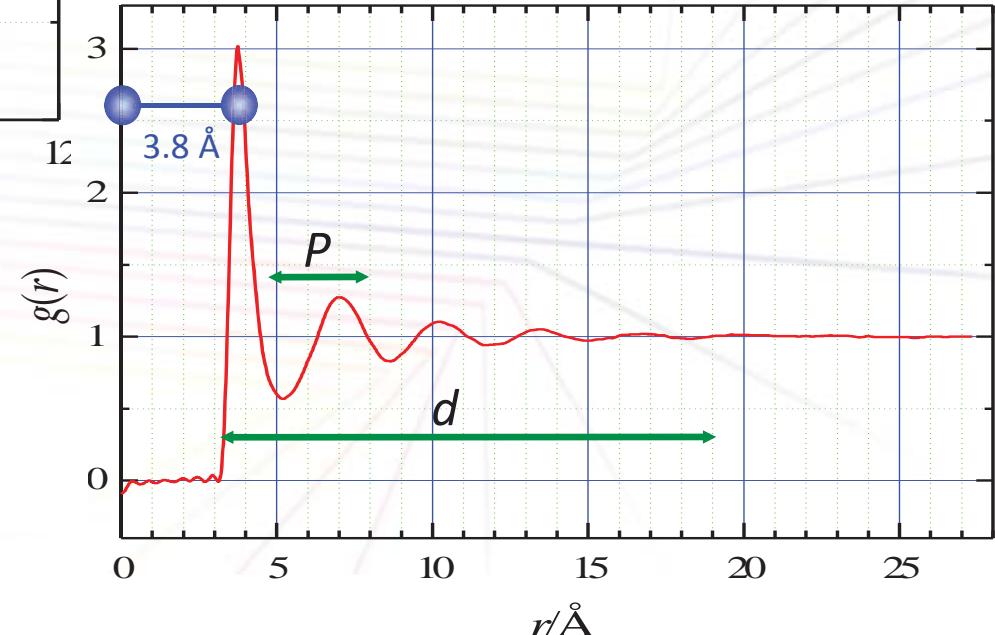


Liquid Ar @ 85K

J.L. Yarnell *et al.* (1973) PRA 7, 2130

$$S(\infty) = 1$$

Fourier Transformation

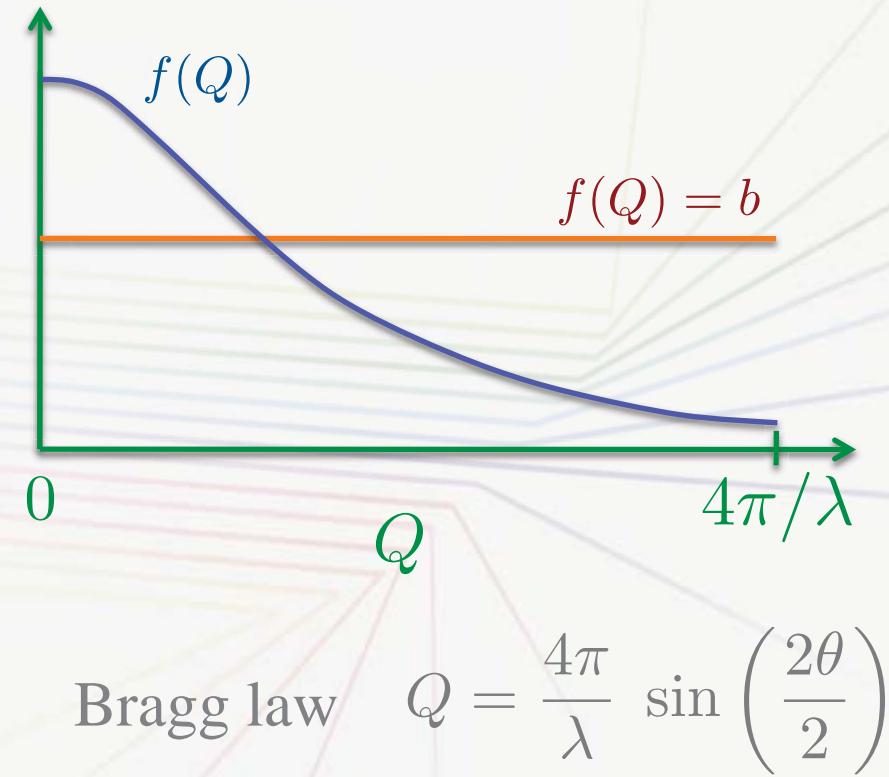
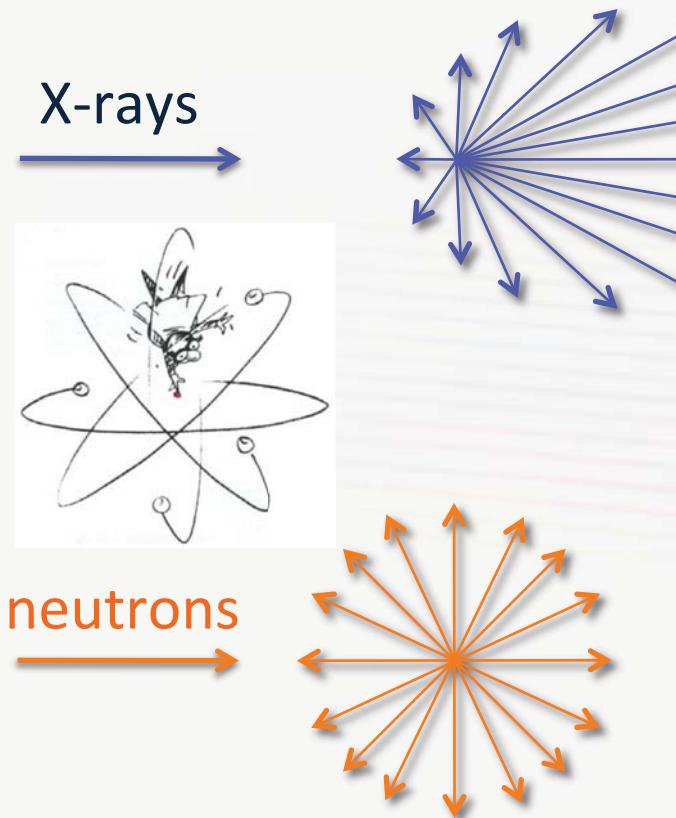




Neutrons vs X-rays



$$\left(\frac{d^2\sigma}{d\Omega dE'} \right) = \frac{k'}{k} \frac{1}{2\pi\hbar} \sum_{jj'} f_j(\vec{Q}) f_{j'}(\vec{Q}) \int_{-\infty}^{\infty} \langle e^{-i\vec{Q}\cdot\vec{r}_{j'}(0)} e^{i\vec{Q}\cdot\vec{r}_j(t)} \rangle e^{-i\omega t} dt$$





Form factor



Exercise

The Fermi's pseudo-potential works well for the neutron-nucleus interaction yielding to a constant scattering length.

Explain qualitatively the effect that a non-punctual potential has on the structure factor.

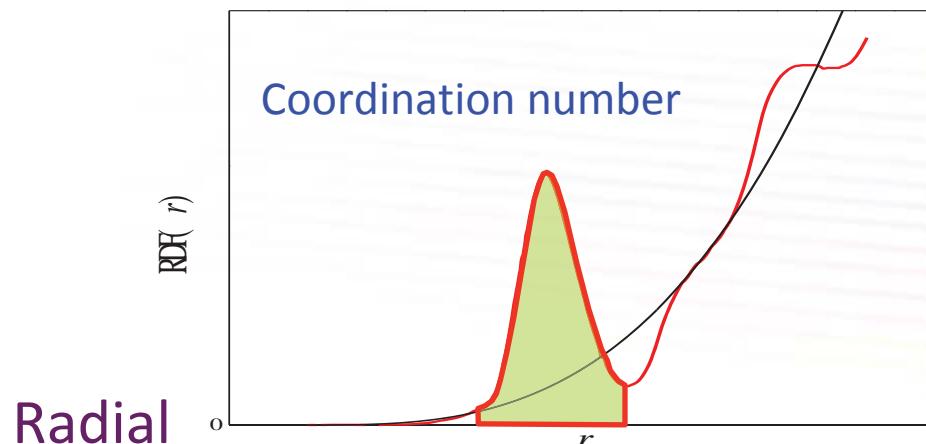
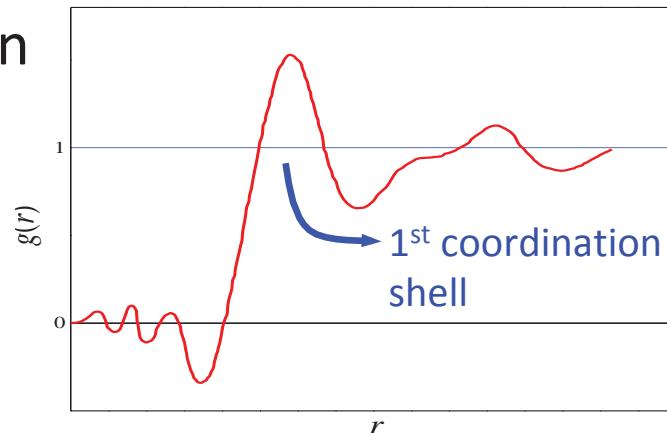




Related functions



Pair distribution function
 $g(r)$



Radial distribution function $RDF(r)$

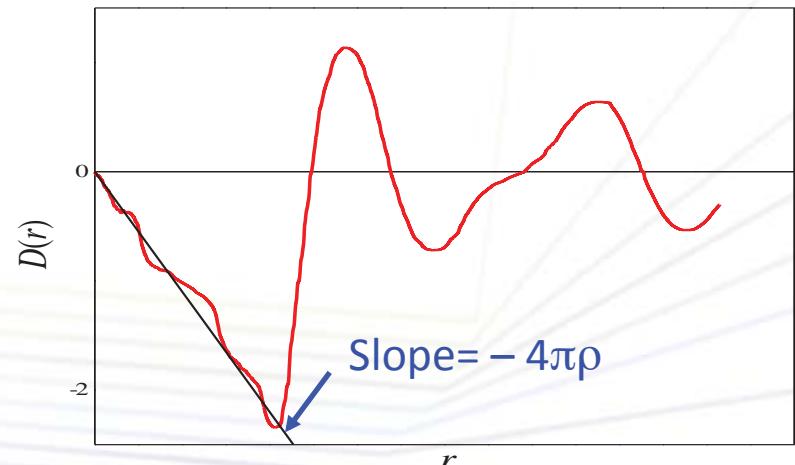
$$RDF(r) = 4\pi r^2 \rho g(r)$$



$$T(r) = RDF(r)/r = 4\pi r \rho g(r) = 4\pi r \rho + G(r)$$

Pair correlation function $G(r)$ or density function $D(r)$

$$G(r) = D(r) = 4\pi r \rho [g(r) - 1]$$



Remember!

$$g(r) - 1 \propto FT \{ S(Q) - 1 \} / \rho$$



Correlation functions

Exercises

Consider a system with a FSDP at 2.1 \AA^{-1} with a FWHM of 0.62 \AA^{-1} .

What could you say about the characteristics of the corresponding pair distribution function?

How many coordination shells could you observe in such a system?



Taking into account the repulsion region in the interaction potential, show that the low- r region of density function is proportional to the density.





Multiatom systems



System of n chemical species



$$\bar{b}^2 \underbrace{[S(Q) - 1]}_{F(Q)} = \sum_{\alpha=1}^n \sum_{\beta=1}^n c_{\alpha} c_{\beta} b_{\alpha} b_{\beta} [S_{\alpha\beta}(Q) - 1]$$

$n(n+1)/2$ independent partial $S_{\alpha\beta}(Q)$

Change b_{α} by

- Isotopic substitution
- X-ray experiments
- Anomalous diffraction

$$\bar{b}^2 = \sum_{\alpha=1}^n \sum_{\beta=1}^n c_{\alpha} c_{\beta} b_{\alpha} b_{\beta}$$

$$\mathbf{F}_{\text{exp}}(Q) = \mathbf{A} \mathbf{F}_{\text{p}}(Q)$$

NDIS: $|\mathbf{A}| < 0.1$

Binary system:

Two different species: x, y

Fixed composition: constant c_x, c_y

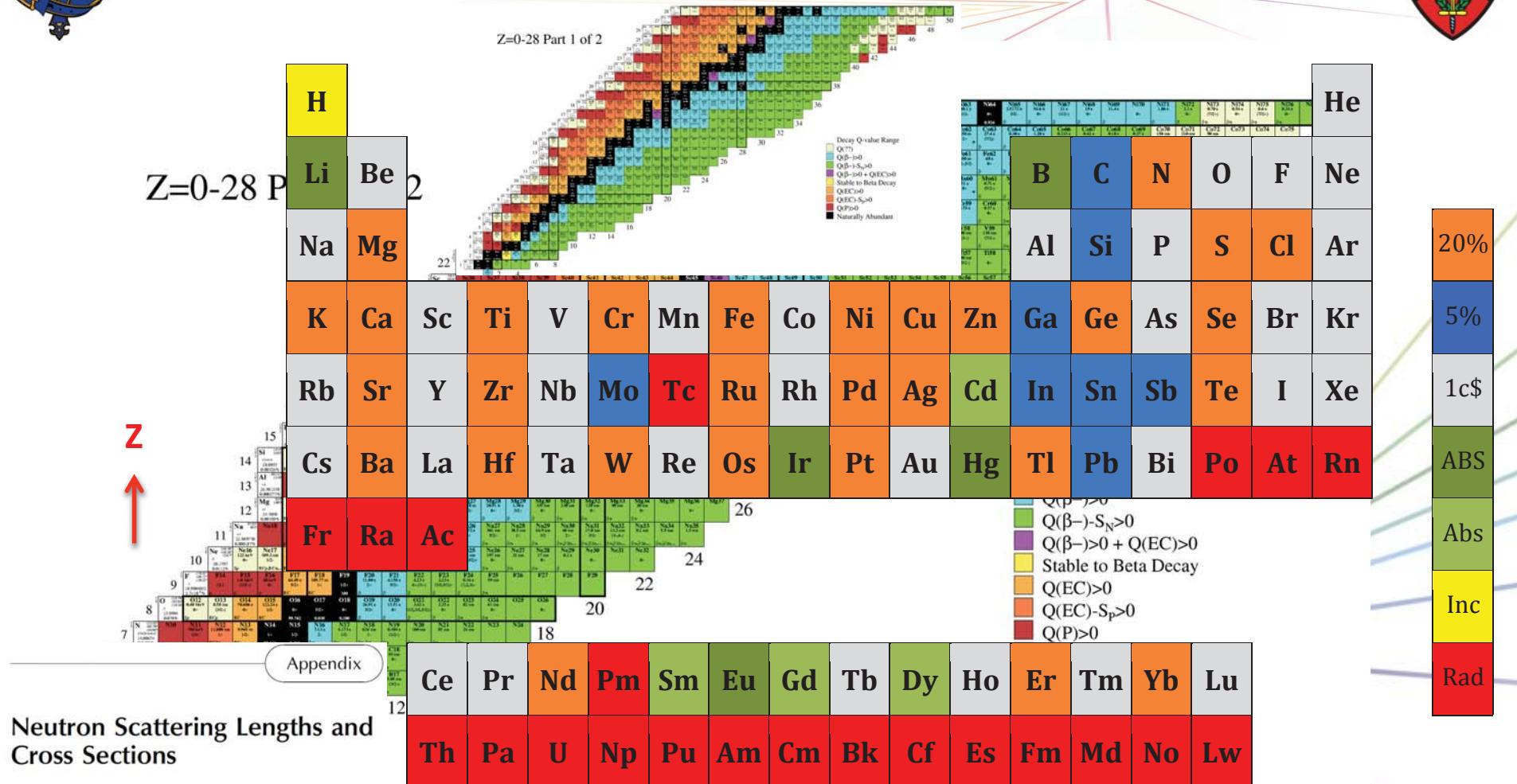
Isotopes with good contrast

$b_{\alpha i}$: scattering length of isotope i of species α

$$\bar{b}^2 \begin{pmatrix} F_{S1}(Q) \\ F_{S2}(Q) \\ F_{S3}(Q) \end{pmatrix} = \begin{pmatrix} c_X^2 b_{X1}^2 & c_Y^2 b_{Y1}^2 & 2c_X c_Y b_{X1} b_{Y1} \\ c_X^2 b_{X2}^2 & c_Y^2 b_{Y2}^2 & 2c_X c_Y b_{X2} b_{Y2} \\ c_X^2 b_{X3}^2 & c_Y^2 b_{Y3}^2 & 2c_X c_Y b_{X3} b_{Y3} \end{pmatrix} \begin{pmatrix} F_{XX}(Q) \\ F_{YY}(Q) \\ F_{XY}(Q) \end{pmatrix}$$



Isotopic substitution



Javier Dawidowski, José Rolando Granada, Javier Roberto Santisteban, Florencia Cantargi and Luis Alberto Rodríguez Palomino
Comisión Nacional de Energía Atómica, Consejo Nacional de Investigaciones Científicas y Técnicas, Centro Atómico Bariloche and Instituto Balseiro, Bariloche, Río Negro, Argentina

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Periodic table showing elements with isotopes with $> 20\%$ scattering length contrast (orange), 5 - 20 % contrast (blue), mono-isotopic, lack of scattering length contrast or prohibitively expensive isotopes (grey), elements with high absorption coefficients where non-absorbing isotopes are available (green), elements with isotopes to overcome incoherent scattering effects (yellow) and radioactive elements (red).



Weighting factors



Exercise

Consider a binary system for which you have performed three different total scattering experiments with three different isotopic compositions.

- A- Write down the expression of the weighting factors matrix
- B- Evaluate this matrix for the case of the Ag₂Se, if you used the following isotopic compositions:

^{107}Ag -^{nat}Se, ^{109}Ag -⁷⁶Se and ^{nat}Ag-⁷⁶Se.

Assume that for each substitution, you have replaced 100% of the atoms by its corresponding isotopes.

- C- Invert the matrix to obtain the partial structure factors from the three experimental structure factors.





A binary system

Silver chalcogenides



X= S, Te or Se

Network formers

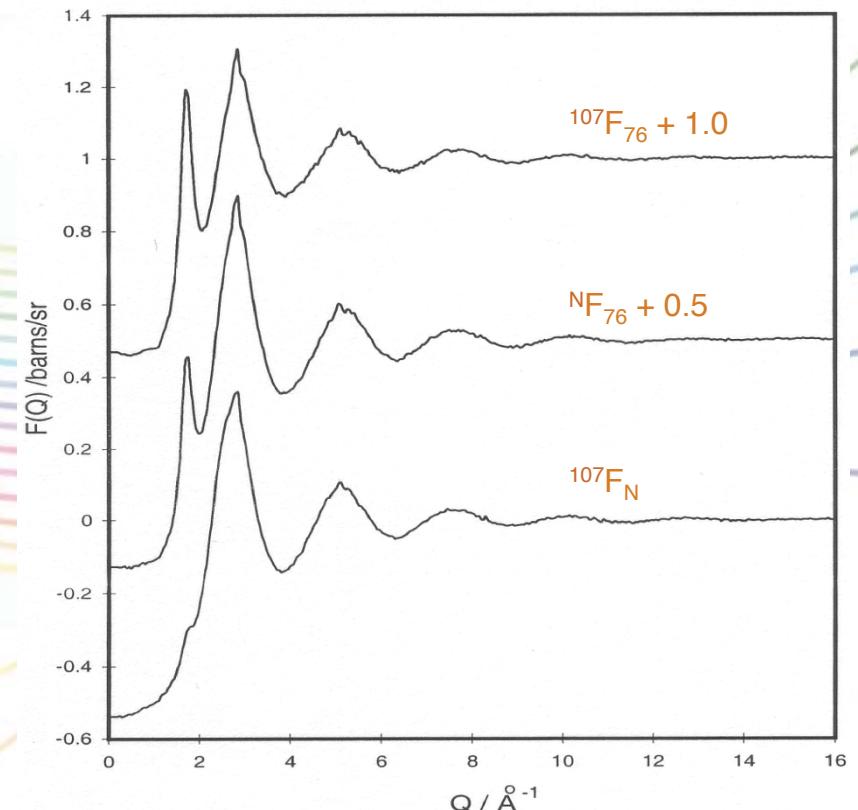


Fast-ion conductor or semiconductor glasses

Samples: $^{107}\text{Ag}_2^{\text{nat}}\text{Se}$, $^{109}\text{Ag}_2^{76}\text{Se}$, $^{\text{nat}}\text{Ag}_2^{76}\text{Se}$

Isotope	b (fm)	σ_a (b)	σ_s (b)
^nAg	5.922	24.6	4.99
^{107}Ag	7.64	14.6	7.44
^{109}Ag	4.19	35.4	2.55
^nSe	7.97	4.55	8.31
^{76}Se	12.2	33.1	18.7

$$\begin{bmatrix} {}^{107}_{\text{nat}} F_{S1}(Q) \\ {}^{109}_{76} F_{S2}(Q) \\ {}^{107}_{\text{nat}} F_{S3}(Q) \end{bmatrix} = \begin{bmatrix} 0.2594 & 0.0706 & 0.2706 \\ 0.0780 & 0.1654 & 0.2272 \\ 0.1559 & 0.1654 & 0.3211 \end{bmatrix} \begin{bmatrix} F_{\text{AgAg}}(Q) \\ F_{\text{SeSe}}(Q) \\ F_{\text{AgSe}}(Q) \end{bmatrix}$$

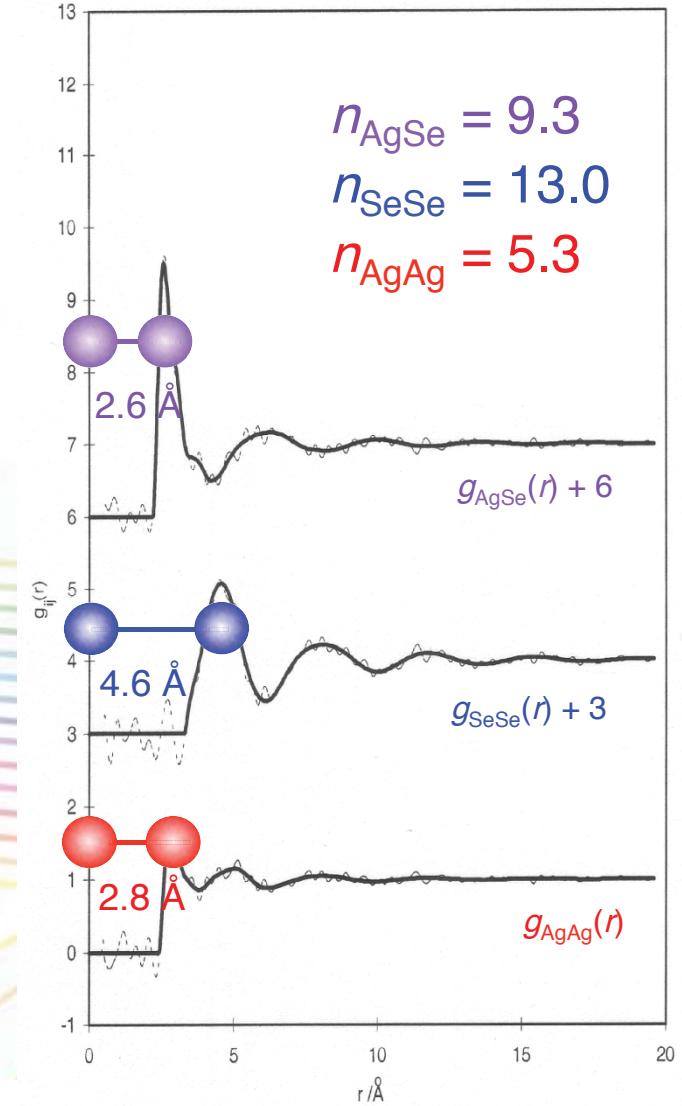
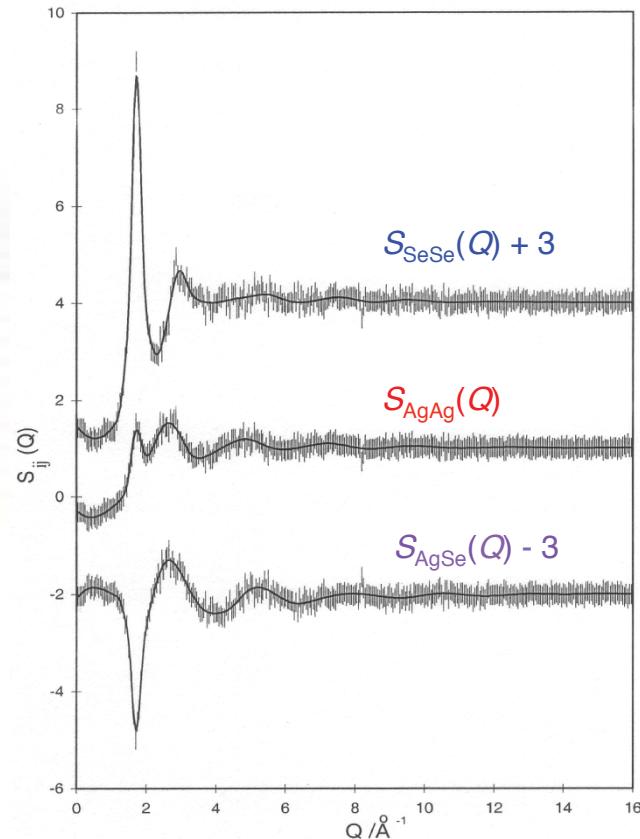




Partial structure factors

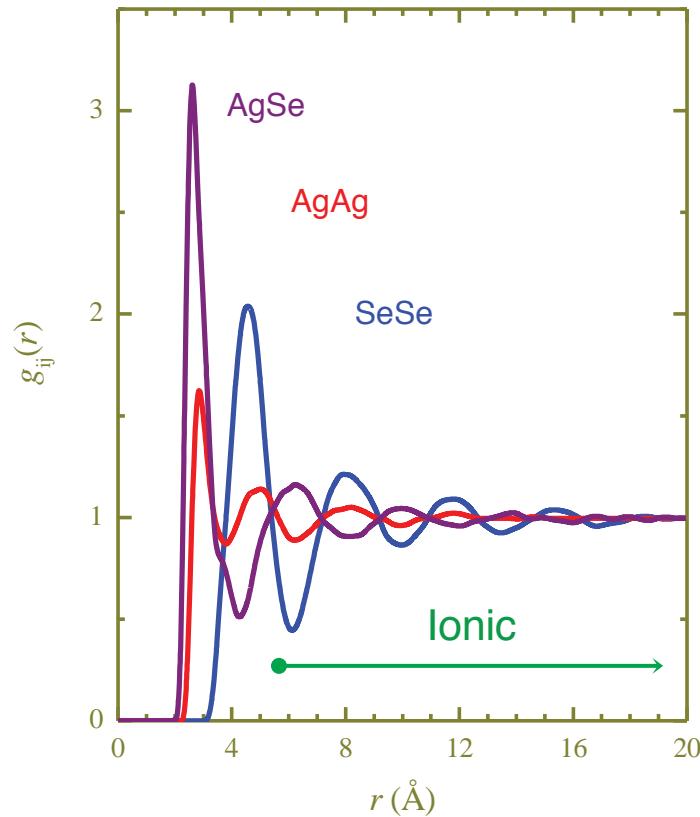


$$\begin{bmatrix} F_{\text{AgAg}}(Q) \\ F_{\text{SeSe}}(Q) \\ F_{\text{AgSe}}(Q) \end{bmatrix} = \begin{bmatrix} 12.17 & 17.31 & -22.50 \\ 8.11 & 32.22 & -29.63 \\ -10.09 & -25.00 & 29.30 \end{bmatrix} \begin{bmatrix} {}^{107}_{\text{nat}} F_{\text{S1}}(Q) \\ {}^{109}_{76} F_{\text{S2}}(Q) \\ {}^{76}_{\text{nat}} F_{\text{S3}}(Q) \end{bmatrix}$$





Ionic behaviour





First difference method

$$\bar{b}^2 F(Q) = \sum_{\alpha=1}^n \sum_{\beta=1}^n c_\alpha c_\beta b_\alpha b_\beta F_{\alpha\beta}(Q) \quad \text{Substitution} \longrightarrow \gamma: \gamma_1, \gamma_2$$

Important! We change scattering lengths but not composition

$$\begin{aligned} \bar{b}^2 F_{\gamma 1}(Q) &= c_\gamma^2 b_{\gamma 1}^2 F_{\gamma\gamma}(Q) + c_\gamma b_{\gamma 1} \sum_{\alpha \neq \gamma}^n c_\alpha b_\alpha F_{\alpha\gamma}(Q) + \sum_{\alpha, \beta \neq \gamma}^n c_\alpha c_\beta b_\alpha b_\beta F_{\alpha\beta}(Q) \\ \bar{b}^2 F_{\gamma 2}(Q) &= c_\gamma^2 b_{\gamma 2}^2 F_{\gamma\gamma}(Q) + c_\gamma b_{\gamma 2} \sum_{\alpha \neq \gamma}^n c_\alpha b_\alpha F_{\alpha\gamma}(Q) + \sum_{\alpha, \beta \neq \gamma}^n c_\alpha c_\beta b_\alpha b_\beta F_{\alpha\beta}(Q) \end{aligned}$$

Correlation function of atom γ with all other components

$$\bar{b}^2 \Delta F_\gamma(Q) = c_\gamma^2 (b_{\gamma 1}^2 - b_{\gamma 2}^2) F_{\gamma\gamma}(Q) + c_\gamma (b_{\gamma 1} - b_{\gamma 2}) \sum_{\alpha \neq \gamma}^n c_\alpha b_\alpha F_{\alpha\gamma}(Q)$$

$$\frac{\bar{b}^2 \Delta F_\gamma(Q)}{c_\gamma^2 (b_{\gamma 1}^2 - b_{\gamma 2}^2)} = F_{\gamma\gamma}(Q) + \frac{\sum_{\alpha \neq \gamma}^n c_\alpha b_\alpha F_{\alpha\gamma}(Q)}{c_\gamma (b_{\gamma 1} + b_{\gamma 2})} \xrightarrow{\text{small}}$$



A ternary system



Li in ND₃

Metal-nonmetal transition at 7 MPM

Class A metals

Conductivity 15000 $\Omega^{-1} \text{ cm}^{-1} \text{ mol}^{-1}$

3 species \Rightarrow 6 different experiments!

First difference method

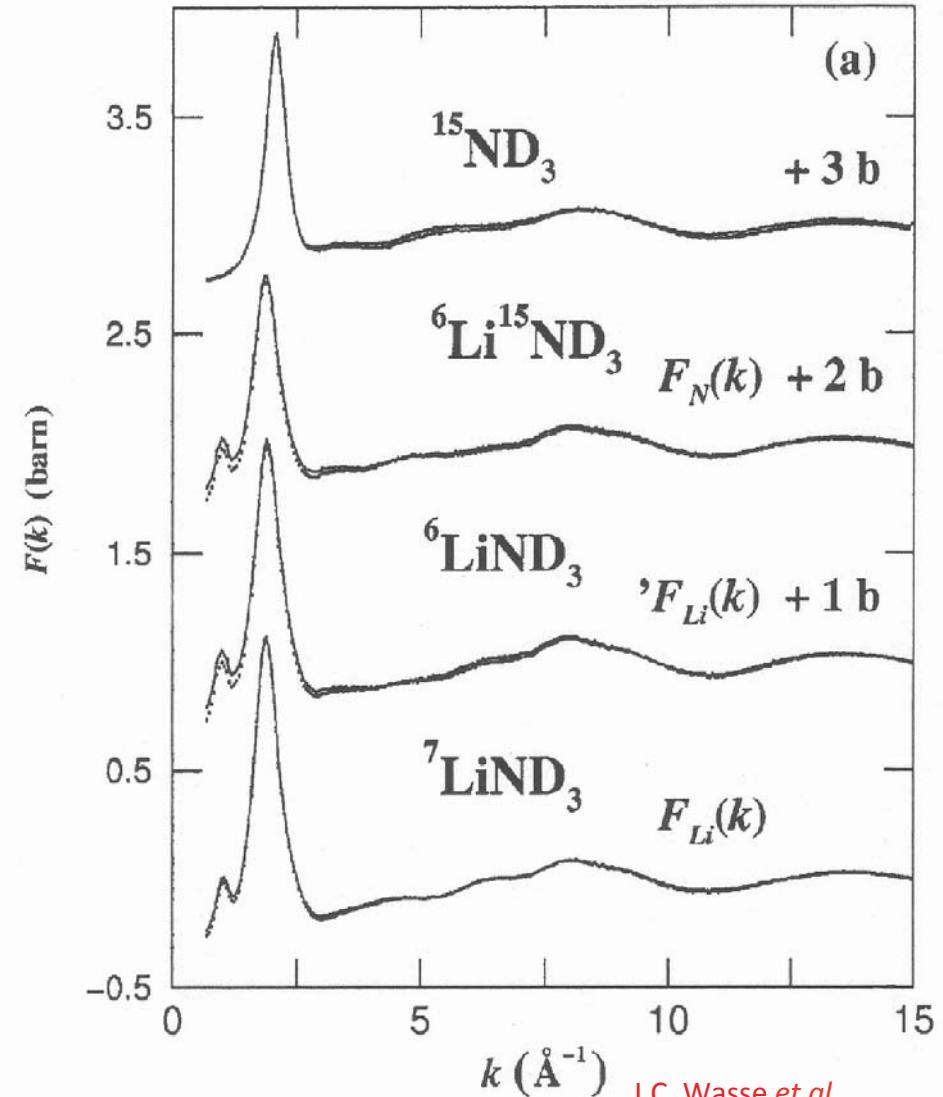
Samples:

⁶Li in ^{nat}ND₃

⁷Li in ^{nat}ND₃

⁶Li in ¹⁵ND₃

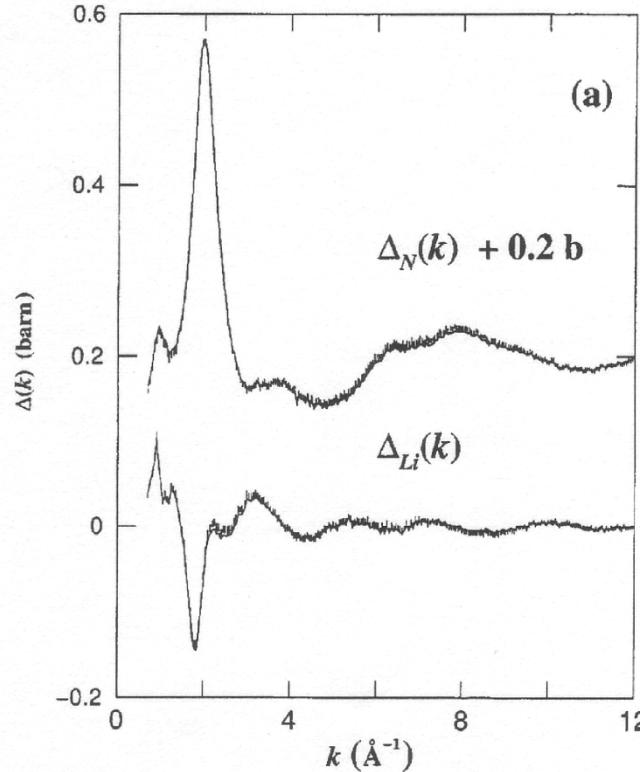
¹⁵ND₃



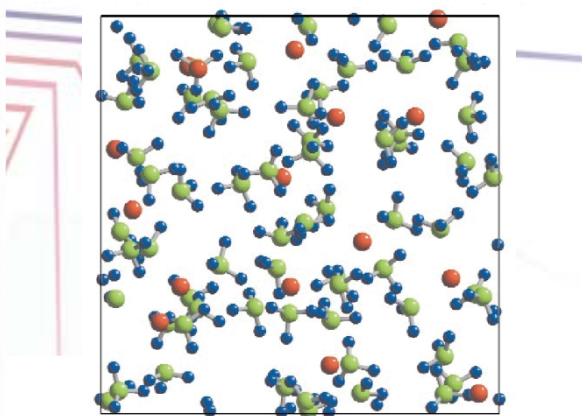
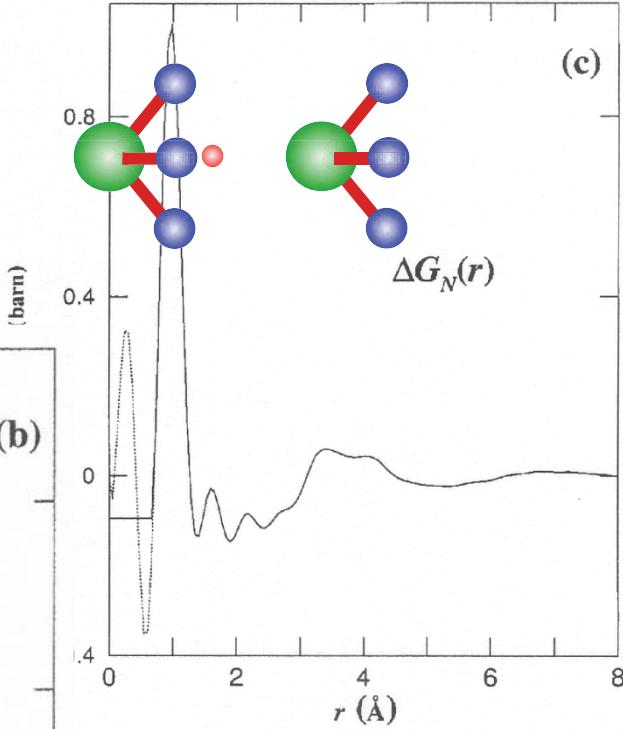
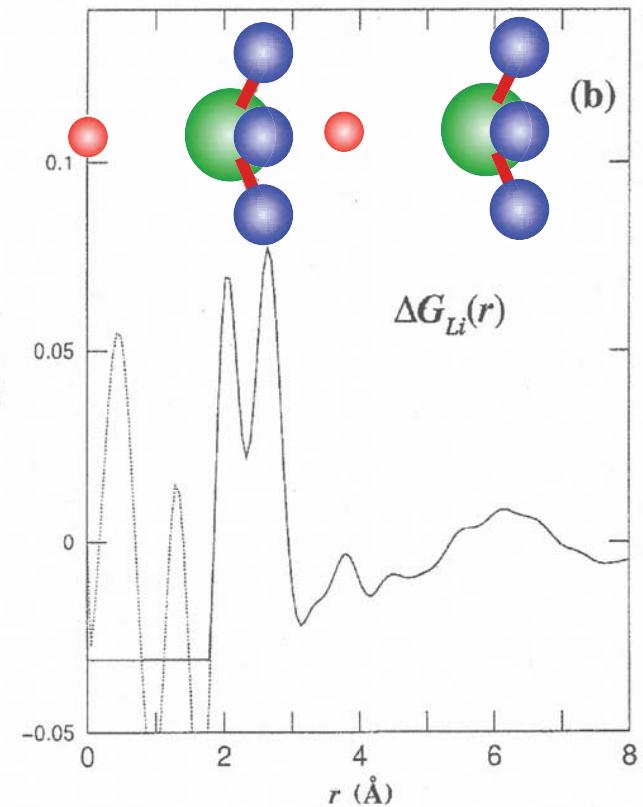
J.C. Wasse *et al.*
PRB 61, 11993 (2000)



Partials... not really



$\Delta G_{Li}(r)$ (barn)



J.C. Wasse et al.
PRB 61, 11993 (2000)



Second difference method



New substitution $\delta: \delta_1, \delta_2$

$$\bar{b}^2 \Delta F_{\gamma\delta 1}(Q) = c_\gamma^2(b_{\gamma 1}^2 - b_{\gamma 2}^2) F_{\gamma\gamma}(Q) + c_\gamma c_\delta(b_{\gamma 1} - b_{\gamma 2}) b_{\delta 1} F_{\gamma\delta}(Q) + c_\gamma(b_{\gamma 1} - b_{\gamma 2}) \sum_{\alpha \neq \gamma, \delta}^n c_\alpha b_\alpha F_{\alpha\gamma}(Q)$$

$$\bar{b}^2 \Delta F_{\gamma\delta 2}(Q) = c_\gamma^2(b_{\gamma 1}^2 - b_{\gamma 2}^2) F_{\gamma\gamma}(Q) + c_\gamma c_\delta(b_{\gamma 1} - b_{\gamma 2}) b_{\delta 2} F_{\gamma\delta}(Q) + c_\gamma(b_{\gamma 1} - b_{\gamma 2}) \sum_{\alpha \neq \gamma, \delta}^n c_\alpha b_\alpha F_{\alpha\gamma}(Q)$$

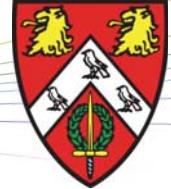
$$\bar{b}^2 \Delta^2 F_{\gamma\delta}(Q) = c_\gamma c_\delta(b_{\gamma 1} - b_{\gamma 2})(b_{\delta 1} - b_{\delta 2}) F_{\gamma\delta}(Q)$$

$$F_{\gamma\delta}(Q) = \frac{\bar{b}^2 \Delta^2 F_{\gamma\delta}(Q)}{c_\gamma c_\delta(b_{\gamma 1} - b_{\gamma 2})(b_{\delta 1} - b_{\delta 2})}$$

Partial structure factor
for pairs γ and δ

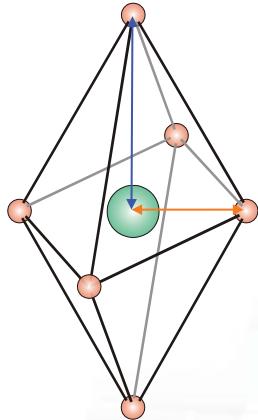


Cu(II) aqua ion



A. Pasquarello *et al.*
Science 291, 856 (2001)

Model Octahedral complex $[\text{Cu}(\text{H}_2\text{O})]^{2+}$
Sixfold coordination



X-ray diffraction
EXAFS
XANES
NDIS } *A priori* assumptions
about structure
Overlap axial Cu-O
and Cu-H

Second difference method

$$\Delta F_H = c_{\text{Cu}}^2 (b_{65}^{-2} - b_{63}^{-2}) F_{\text{CuCu}} + 2 c_{\text{Cu}} (b_{65} - b_{63}) \times (c_{\text{Cl}} b_{\text{Cl}} F_{\text{CuCl}} + c_{\text{O}} b_{\text{O}} F_{\text{CuO}} + c_{\text{H}} b_{\text{H}} F_{\text{CuH}})$$

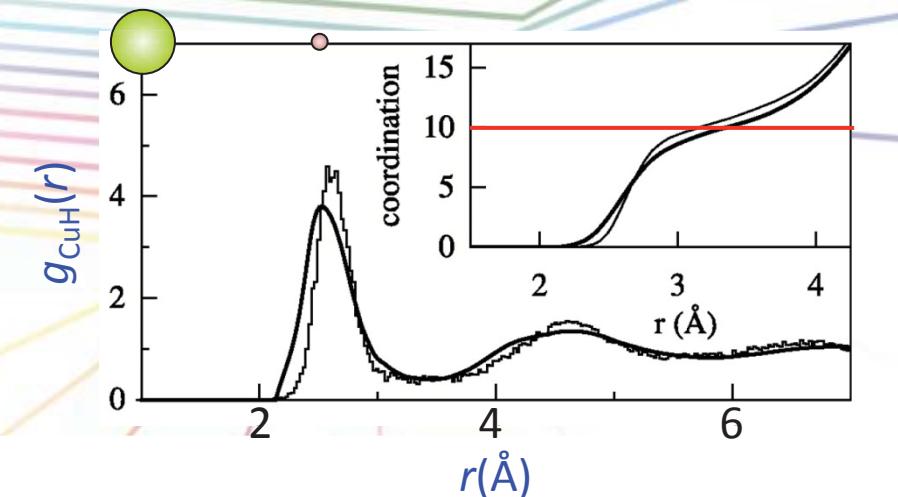
$$\Delta^2 F = 2 c_{\text{Cu}} c_{\text{H}} (b_{65} - b_{63}) (b_{\text{D}} - b_{\text{H}}) F_{\text{CuH}}$$

System:

$\text{Cu}(\text{ClO}_4)_2 + \text{HClO}_4$ in H_2O
→ 10 expts!

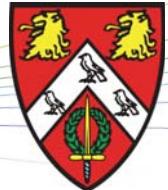
Samples:

$^{65}\text{Cu}(\text{ClO}_4)_2 + \text{HClO}_4$ in H_2O
 $^{63}\text{Cu}(\text{ClO}_4)_2 + \text{HClO}_4$ in H_2O
 $^{65}\text{Cu}(\text{ClO}_4)_2 + \text{DClO}_4$ in D_2O
 $^{63}\text{Cu}(\text{ClO}_4)_2 + \text{DClO}_4$ in D_2O

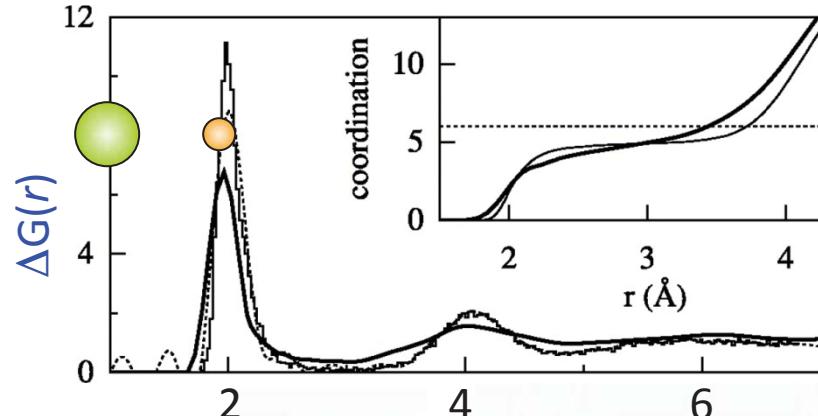




Five-fold coordinated ion



A. Pasquarello *et al.*
Science 291, 856 (2001)

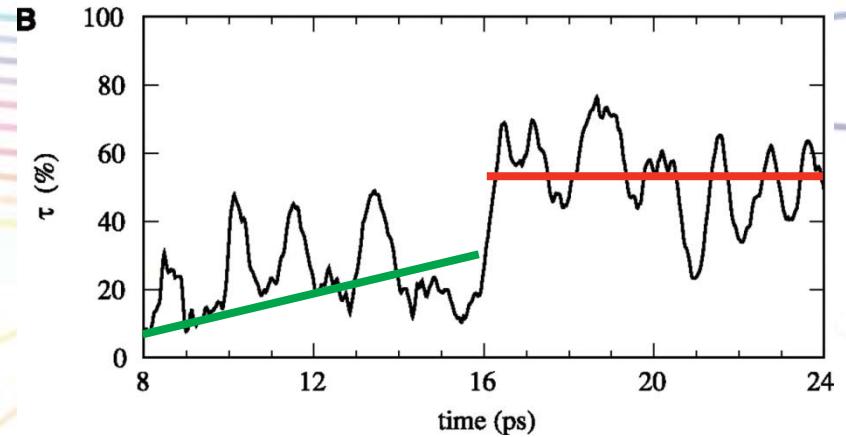
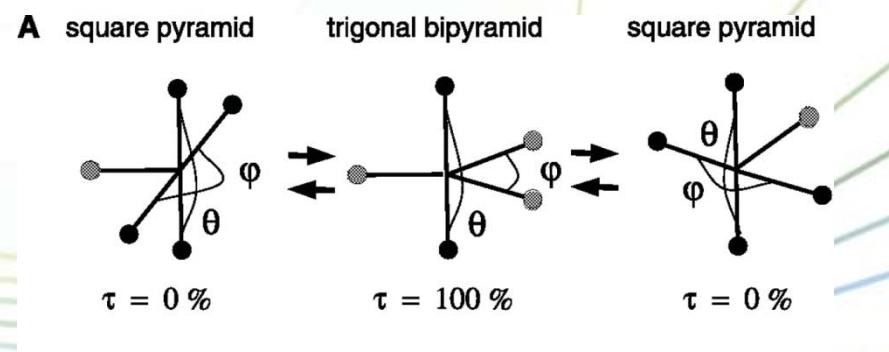


First-principles Molecular
Dynamics Simulation

$$\tau = (\theta - \varphi)/60 \times 100\%$$

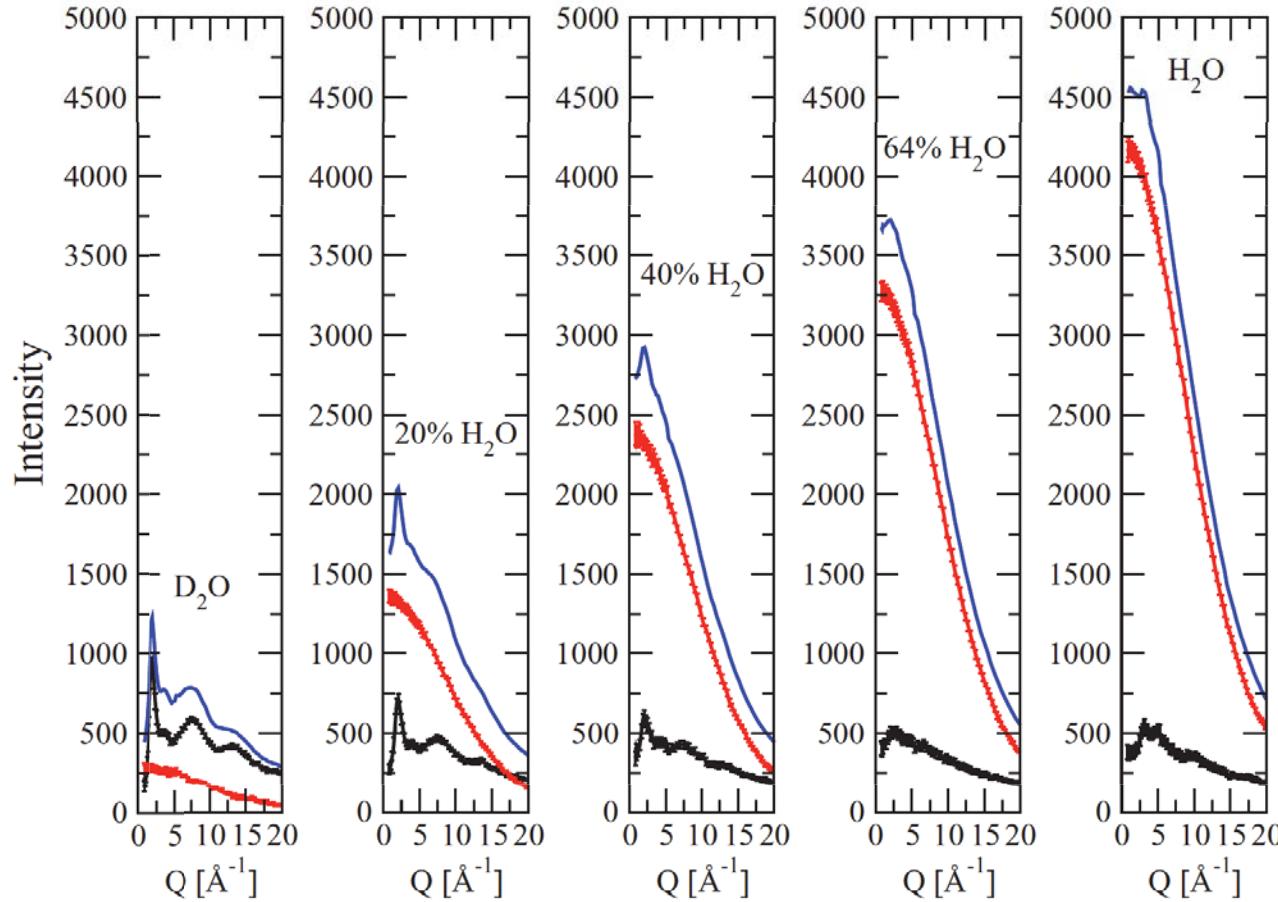
Cu(II) aqua ion is
five-fold coordinated

$$\Delta F = F_{\text{CuO}} + 0.044 F_{\text{CuCu}} + 0.102 F_{\text{CuCl}}$$





H incoherence problem



Isotopes

σ_c	σ_i	σ_s	σ_a
44.89 (4)	0	44.89 (4)	0
1.7568 (10)	80.26(6)	82.02 (6)	0.3326 (7)
1.7589 (11)	79.91(4)	81.67 (4)	0.3326 (7)
5.597 (10)	2.04(3)	7.64(3)	0.000519 (7)
2.89(3)	0.14(4)	3.03(5)	<6E-06

Appendix

Neutron Scattering Lengths and Cross Sections

Javier Dawidowski, José Rolando Granada, Javier Roberto Santisteban, Florencia Cantargi and Luis Alberto Rodríguez Palomino
Comisión Nacional de Energía Atómica, Consejo Nacional de Investigaciones Científicas y Técnicas, Centro Atómico Bariloche and Instituto Balseiro, Bariloche, Río Negro, Argentina

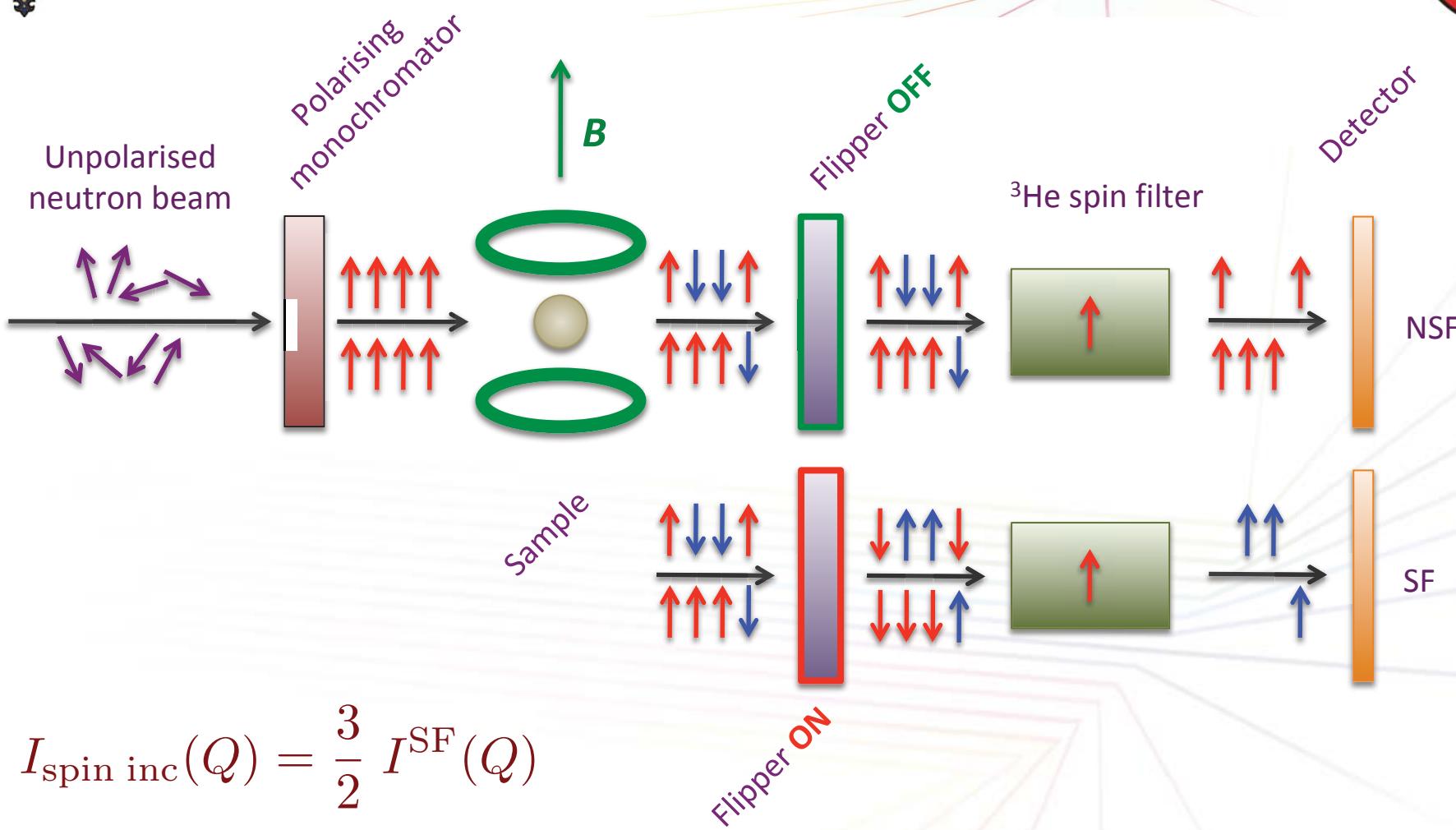
Experimental Methods in the Physical Sciences, Vol. 44.
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Neutron diffraction of hydrogenous materials: Measuring incoherent and coherent intensities separately

László Temleitner, Anne Stunault, Gabriel J. Cuello, and László Pusztai
Phys. Rev. B **92**, 014201 – Published 1 July 2015



Polarised neutrons

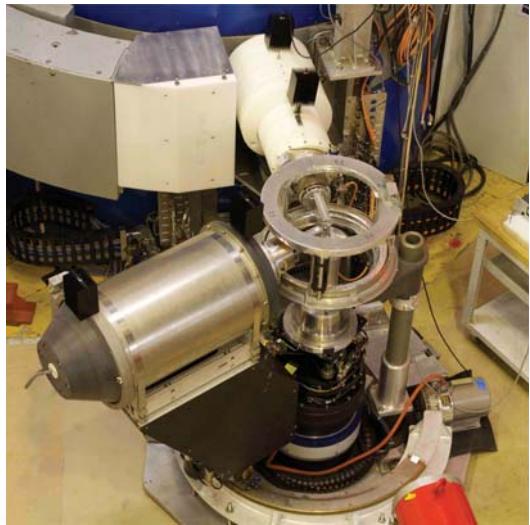


$$I_{\text{spin inc}}(Q) = \frac{3}{2} I^{\text{SF}}(Q)$$

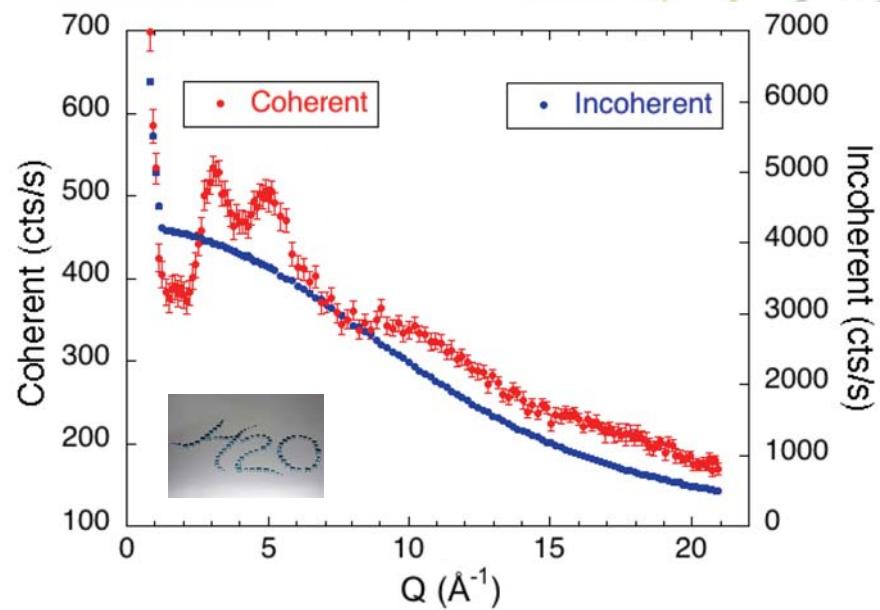
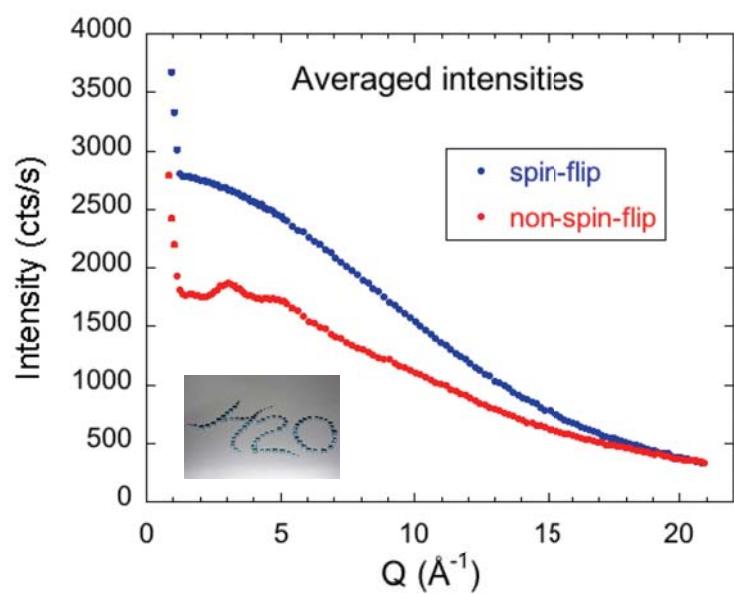
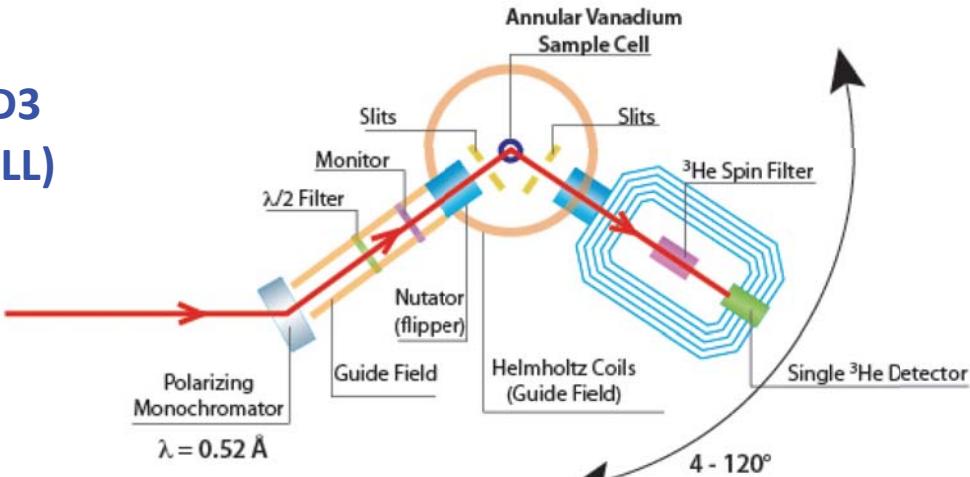
$$I_{\text{coh}}(Q) + I_{\text{isotope inc}}(Q) = I^{\text{NSF}}(Q) - \frac{1}{2} I^{\text{SF}}(Q)$$



The case of water



D3
(ILL)





Instruments

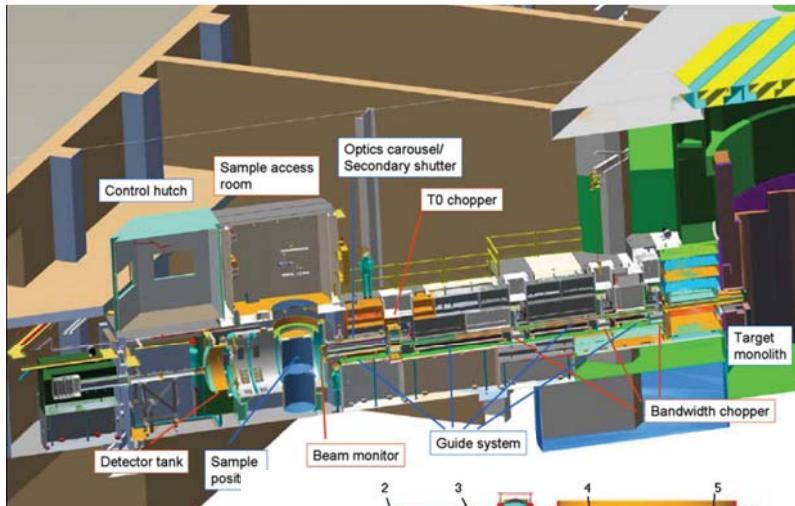
Reactor → 2-axis

Accelerator → TOF

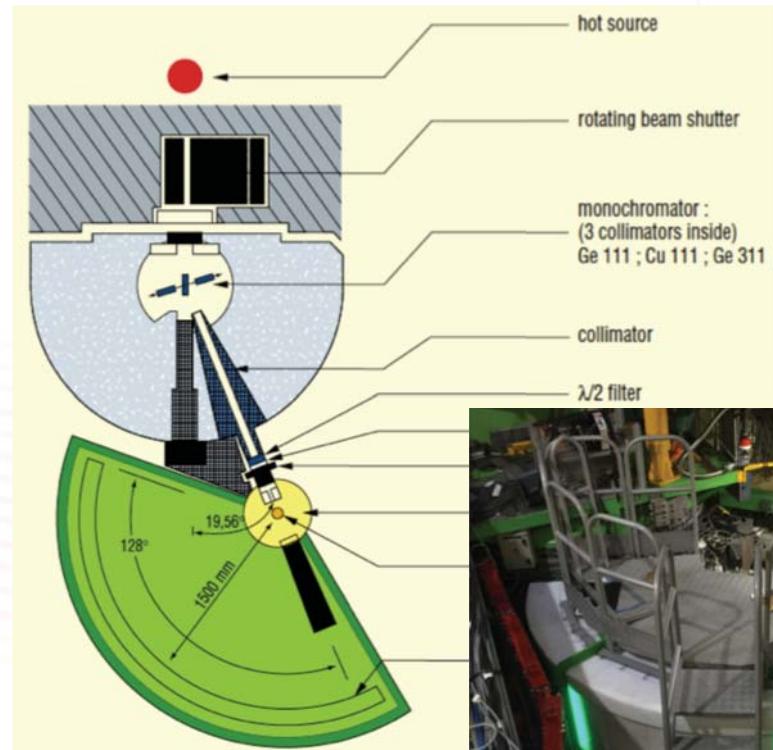
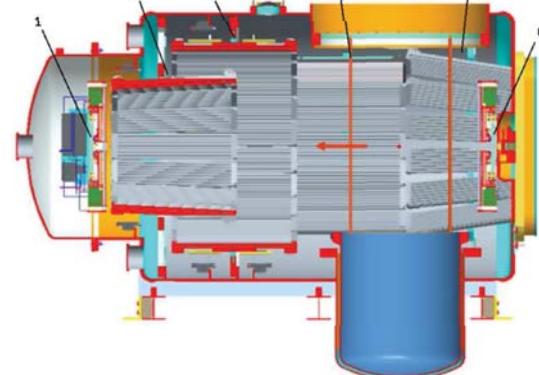
→ Scattering angle

→ Time-of-flight

Elastic scattering → Q



NOMAD
(ORNL)

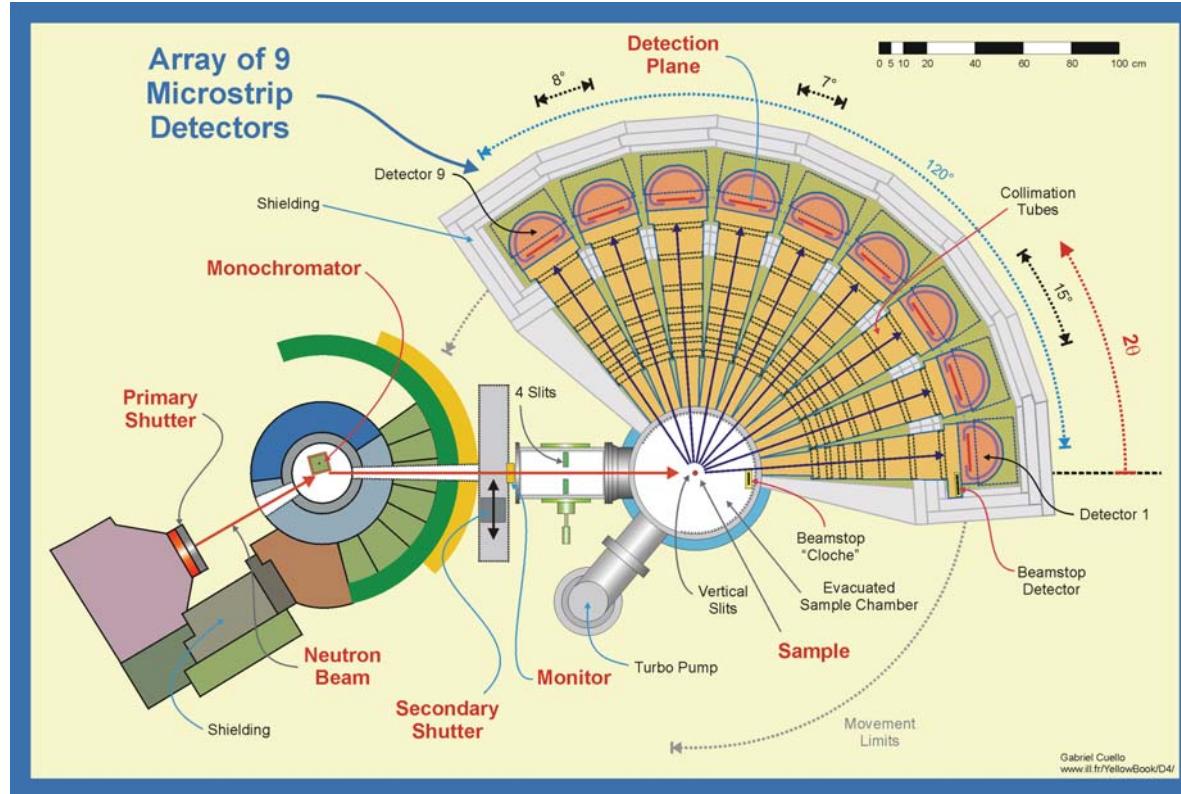
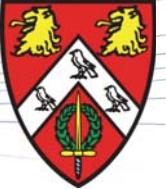


7C2
(LLB)





D4C @ ILL

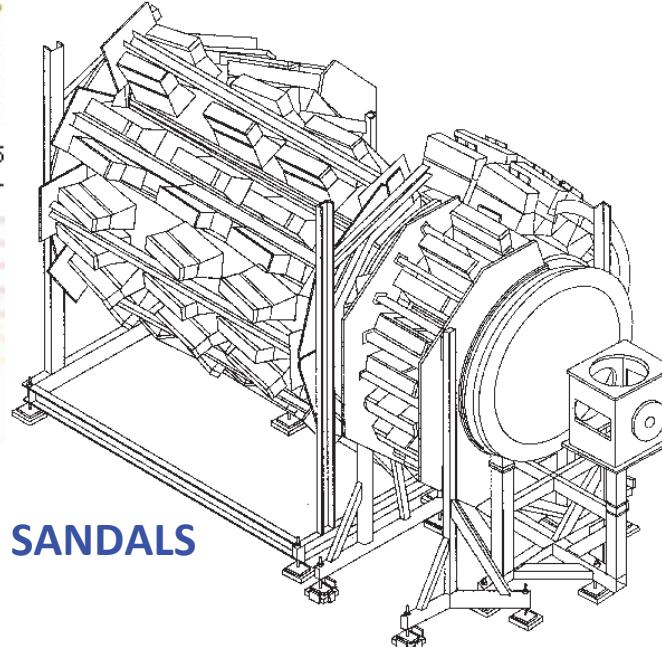
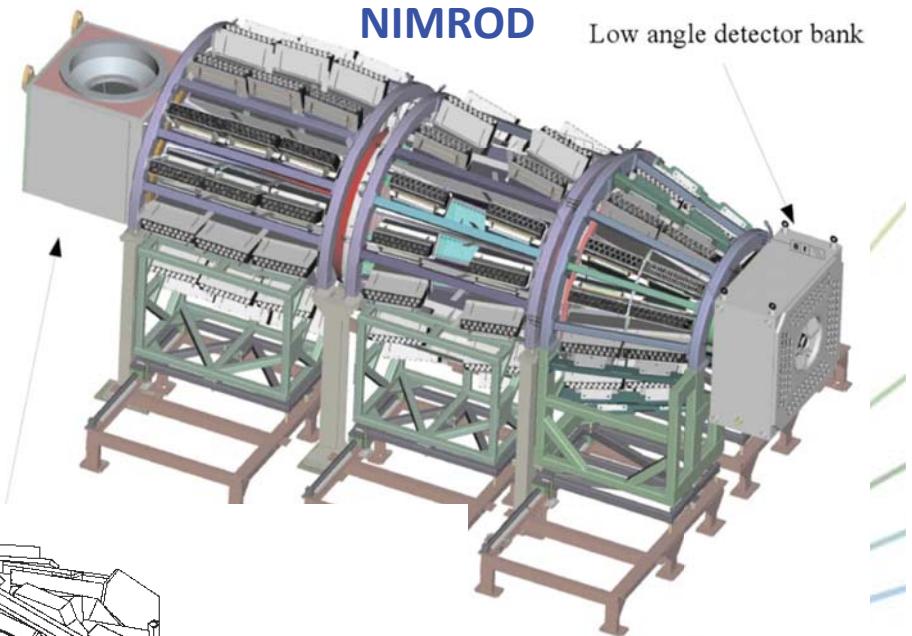
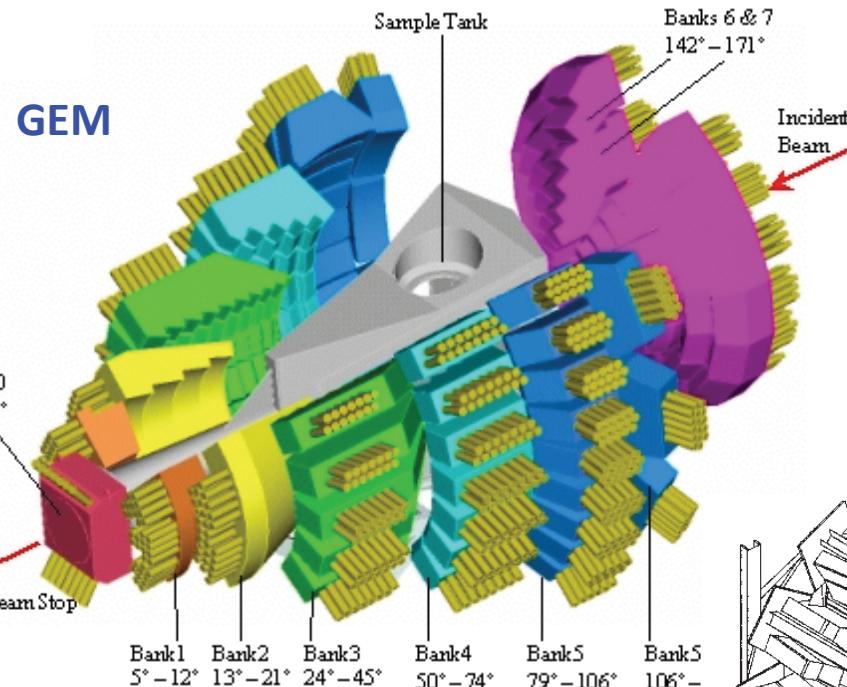


Face	d-spacing (Å)	λ (Å)	Flux (10^7 n cm $^{-2}$ s $^{-1}$)	Filter
Si111	1.807	0.7	5.0	Ir
Cu220	1.278	0.5	4.5	Rh
Cu331	0.829	0.35	0.3	Non

$$Q = \frac{4\pi}{\lambda} \sin\left(\frac{2\theta}{2}\right)$$



Instruments @ ISIS





Data reduction



$$I(2\theta, \omega) = C \Phi_0 N \frac{k'}{k} \frac{\sigma}{4\pi} S(\vec{Q}, \omega) \epsilon(k')$$

$$I(2\theta) = C \Phi_0 N \frac{\sigma}{4\pi} \int_{-\infty}^{E_{\max}} d\omega \frac{k'}{k} S(\vec{Q}, \omega) \epsilon(k')$$

Formal aspects

- Elastic scattering (diffraction)
- Stationary beam
- Constant efficiency detector
- One interaction processes (single scattering)

Bragg's law (for Q)
Integration limits ($\pm\infty$)

$$I(2\theta) = C \Phi_0 N / 4\pi (\sigma_{\text{coh}} S(Q) + \sigma_{\text{inc}}) \epsilon(k)$$

Practical aspects

- Monochromatic beam
- No background
- No attenuation
- Single scattering

No beam
No container
No sample
No environment
No detector

No problem!





Experimental corrections

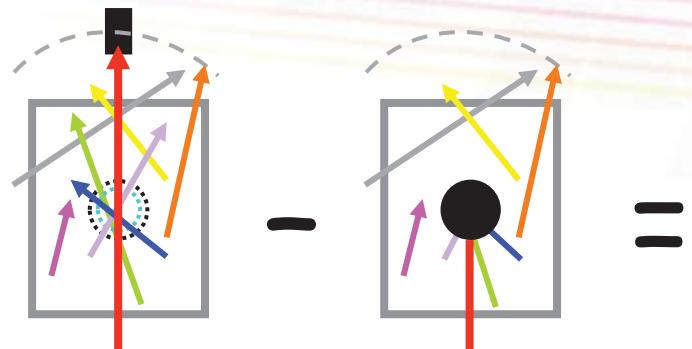


Instrumental effects

- Background
- Detector efficiency
- Detector dead-time
- Instrumental resolution

Sample effects

- Inelasticity
- Attenuation (container)
- Multiple scattering
- Normalisation

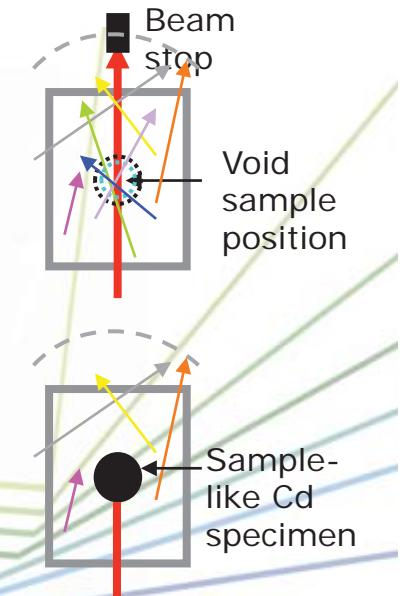


Background noise

Requires two measurements:

- Empty beam
(no sample, no container)

- Sample-like Cd specimen



$$I_b(2\theta) \approx I_{Cd}(2\theta) + T_{SC} [I_E(2\theta) - I_{Cd}(2\theta)]$$

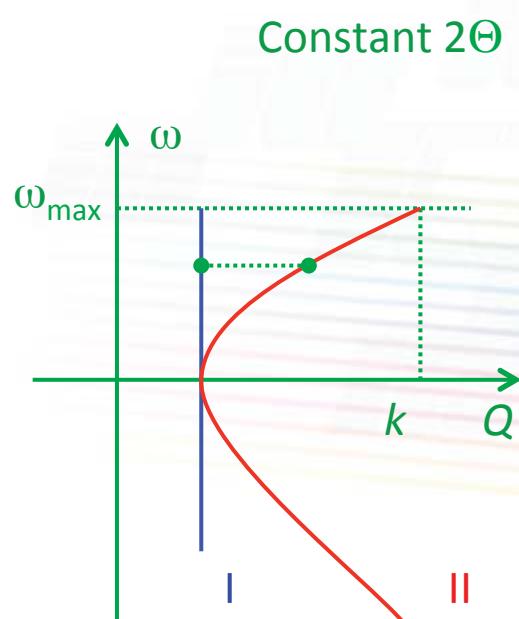
$$T = \exp\{-n \sigma_T(E) d\}$$



Inelasticity effects

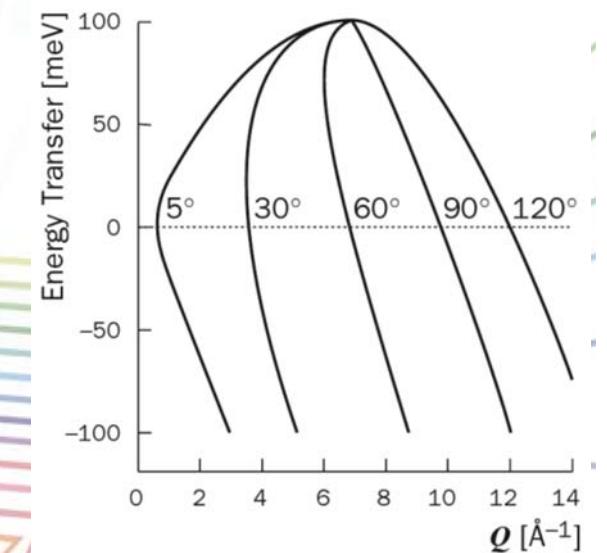


$$I(2\theta) = C \Phi_0 N \frac{\sigma}{4\pi} \int_{-\infty}^{\infty} E_{\max} \frac{k'}{k} S(\vec{Q}, \omega) \epsilon(k') d\omega$$



$$\frac{\hbar^2 Q^2}{2m} = 2E + \hbar\omega - 2\sqrt{E^2 + \hbar\omega E} \cos 2\theta$$

Placzek's correction



These effects are closely associated
to the detector efficiency



Placzek's expansion



Efficiency

- Black detector, $\varepsilon(E) = 1$
- $1/v$ detector, $\varepsilon(E) \propto E^{-1/2}$
- Exponential detector, $\varepsilon(E) = 1 - \exp\{-\alpha (E/E')^{1/2}\}$

- Taylor expansion of $S(Q_i, \omega)$ around $(Q_i, \omega) \longrightarrow S(Q_{||}, \omega)$
- Expansion of $Q_{||}^2 - Q_i^2$, $\varepsilon(k)$ and k'/k in powers of ω/ω_{\max}
- Energy integration

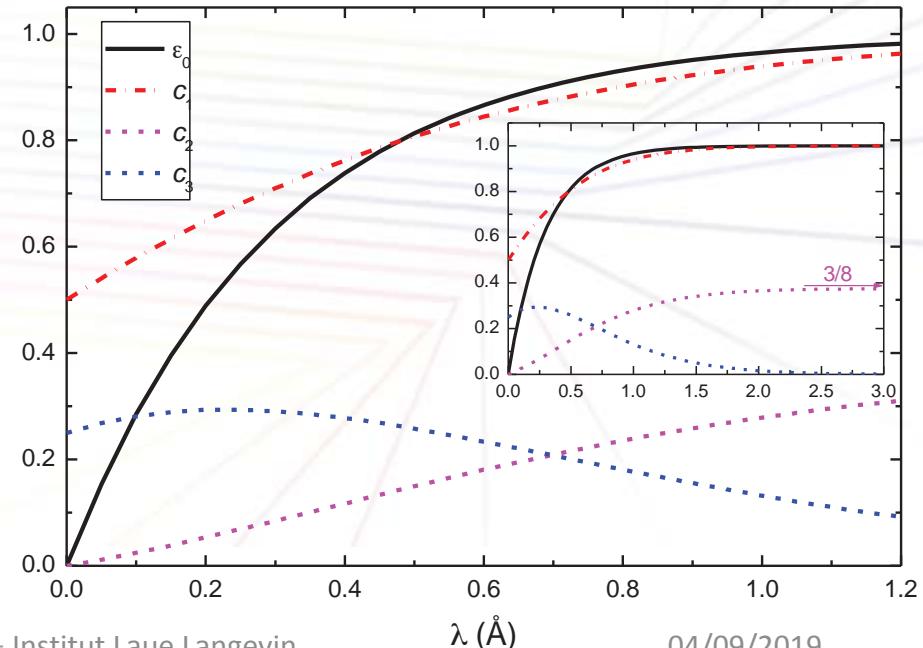
$$S(Q) = \frac{1}{\epsilon_0} \frac{1}{b_{\text{coh}}^2} \left(\frac{d\sigma}{d\Omega} \right)_{\text{corr}} + \left(1 + \frac{b_{\text{inc}}^2}{b_{\text{coh}}^2} \right) \left(C_1 \delta - C_2 \delta^2 + C_3 \delta \gamma - \frac{m}{2M} (\delta + \gamma) \right) - \frac{b_{\text{inc}}^2}{b_{\text{coh}}^2}$$

For an exponential detector

$$C_1 = 1 - \frac{\alpha/2}{e^\alpha - 1}$$

$$C_2 = \frac{3}{8} - \frac{\alpha(\alpha+3)}{8(e^\alpha - 1)}$$

$$C_3 = \frac{\alpha(\alpha+1)}{4(e^\alpha - 1)}$$





Sample related corrections



Attenuation

Multiple Scattering

Sample
+
Container

Minimisation by choosing an
adequate sample geometry

$$I_S^{\text{corr}}(2\theta) = \frac{1}{\alpha_{S,SC}(2\theta)} \left(I_S(2\theta) - I_S^B(2\theta) - \frac{\alpha_{C,SC}(2\theta)}{\alpha_{C,C}(2\theta)} (I_C(2\theta) - I_C^B(2\theta)) \right) - \Delta$$

Paalman & Pings' coefficients

Cylindrical geometry

Blech & Averbach's correction

Complete knowledge of $S(Q,\omega)$
Numerical simulation

Normalisation

Absolute scale
Vanadium diffractogram



Final analysis



$I_{\text{exp}}(2\theta)$

Corrections

$S(Q)$

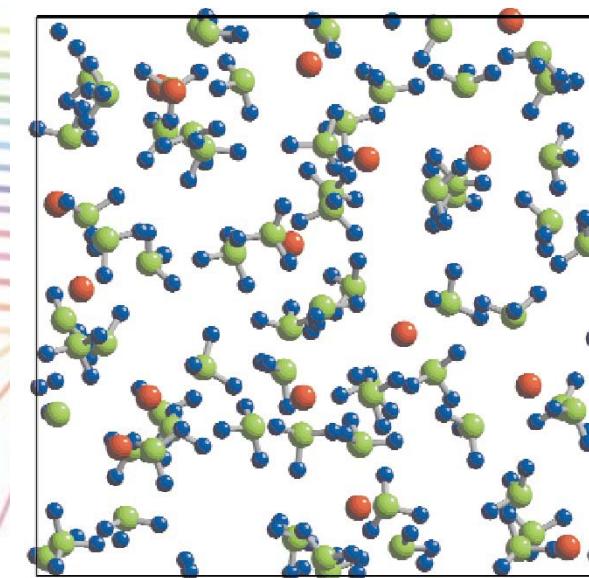
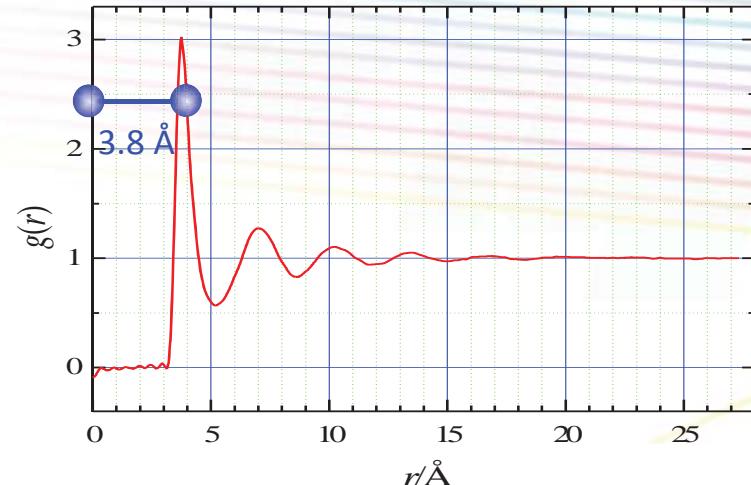
FT

$g(r)$

Truncation effects
Window functions

Interpretation

{ Model
Simulation, e.g. RMC, EPSR, MD, etc.





Further reading



General

D.I. Page, *The structure of liquids by neutron diffraction*, p. 173, Chap. 8 in *Chemical applications of thermal neutron scattering*, B.T.M. Willis (Ed.), Oxford University Press (1973).

J.G. Powles, *The structure of molecular liquids by neutron scattering*, Advances in Physics **22**, 1 (1973).

A.C. Wright, *The structure of amorphous solids by x-ray and neutron diffraction*, Advances in Structure Research by Diffraction Methods **5**, 1 (1974).

P.A. Egelstaff, *Classical fluids*, p. 405, Chap.14 in *Methods of experimental physics*, vol. 23, part B, Academic Press (1987).

P. Chieux, *Introduction to accurate structure factor measurements of disordered materials by neutron scattering*, J. Molec. Struct. **296**, 177 (1993).

G. J. Cuello, *Structure factor determination of amorphous materials by neutron diffraction*, J. Phys.: Condensed Matter **20**, 244109 (2008)

H. E. Fischer, A. C. Barnes, P. S. Salmon, *Neutron and x-ray diffraction studies of liquids and glasses*, Reports on Progress in Physics **69**, 233-299 (2006)

Examples

See the D4 web site (www.ill.fr/YellowBook/D4/pub/) for a list of publications arisen from D4 expts.

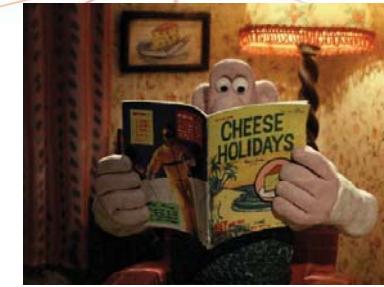
Visit the Disordered Materials Group at ISIS (www.isis.stfc.ac.uk/groups/disordered-materials/)



Further reading

Examples with useful information about data treatment

D.M. North, J.E. Enderby and P.A. Egelstaff,
The structure factor for liquid metals:
I. The application of neutron diffraction techniques,
J. Phys. C **1**, 784 (1968).



H. Bertagnolli, P. Chieux and M.D. Zeidler, *A neutron-diffraction study of liquid acetonitrile: I. $CD_3C^{14}N$,*
Molec. Phys. **32**, 759 (1976).

J.L. Yarnell, M.J. Katz R.G. Wenzel and S.H. Koenig, *Structure factor and radial distribution function for liquid Ar at 85 K*, Phys. Rev. A **7**, 2130 (1973).

Corrections

G. Placzek, Phys. Rev **86**, 377 (1952).

A.K. Soper and P.A. Egelstaff, *Multiple scattering and attenuation of neutrons in concentric cylinders: I. Isotropic first scattering*, Nucl. Instr. Meth. **178**, 415 (1980).

I.A. Blech and B.L. Averbach, *Multiple scattering of neutrons in vanadium and copper*, Phys. Rev. **137**, A1113 (1965).

H.H. Paalman and C.J. Pings, *Numerical evaluation of x-ray absorption factors for cylindrical samples and annular sample cells*, J. Appl. Phys. **33**, 2635 (1962).