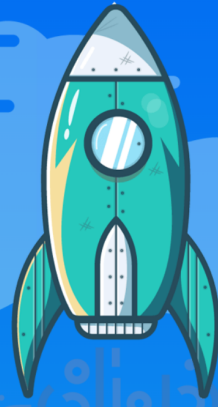


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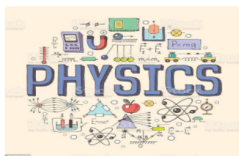


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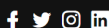
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The English set of multiple-choice questions that have 5 questions.

Time: 3 Minutes

[Strat Exam](#)

Chemistry

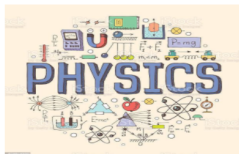


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Time: 3 Minutes

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Physics



The Physics set of multiple-choice questions that have 5 questions

Time: 3 Minutes

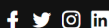
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Arabic



The Arabic set of multiple-choice questions that have 5 questions.

Time: 3 Minutes

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Math Exam

Time left : 2 Minutes**Q1. $(5*5+2)/3$**

- ☐ 12
☒ 11
☐ 10
☐ 9

Q2. What is the next number in this sequence: 2, 4, 8, 16, 32 ...

- ☐ 80
☒ 90
☐ 54
☐ 64

Q3. $(10+2-3)*9/9-1*(10/5)$

- ☒ 7
☐ 6
☐ 5
☐ 4

Q4. $(2*3)+2*6/3-10$

- ☐ 4
☐ 3
☐ 1
☒ 0

Q5. $(10+2-3)*9/9-1*(10/2)$

- ☐ 0
☐ 1
☐ 2
☒ No of the above answers

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Math Exam

Your Result is: 60%

Q1. $(5*5+2)/3$

- ☐ 12
☒ 11
☐ 10
☐ 9

Q2. What is the next number in this sequence: 2, 4, 8, 16, 32 ...

- ☐ 80
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- ☒ 7
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Q4. $(2*3)+2*6/3-10$

- ☐ 4
☐ 3
☐ 1
☒ 0

Q5. $(10+2-3)*9/9-1*(10/2)$

- ☐ 0
☐ 1
☐ 2
☒ No of the above answers

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Congrats You Pass

Your Result is: 60%

OK



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Chemistry Exam

Your Result is: 40%

Q1. The correct formula for aluminum nitrate is

- ☒ **Al(NO₃)₃**
- ☐ Al(NO₂)₃
- ☐ Al₃(NO₃)
- ☐ Al₃N₂

Q2. A substance releases heat when it changes from

- ☒ solid to liquid
- ☐ liquid to gas
- ☐ **liquid to solid**
- ☐ solid to gas

Q3. Given the balanced equation: $2\text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2\text{H}_2\text{O}(\text{l})$ How many grams of H₂O are formed if 9.00 mol H₂(g) reacts completely with an excess of O₂(g)? The molar mass of H₂O is 18.0g/mol.

- ☒ 81.0g
- ☐ **162g**
- ☐ 36.0g
- ☐ 18.0g

Q4. Which element has exactly five

- ☒ **P**
- ☐ Ge
- ☐ Se
- ☐ Ba

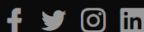
Q5. Which element is a metal?

- ☒ Se (atomic number = 34)
- ☐ **Co (atomic number = 27)**
- ☐ C (atomic number = 6)
- ☐ Br (atomic number = 35)

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Unfortunately You Fail

Your Result is: 40%

[OK](#)


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Math Exam

Your Result is: 60%

Q1. $(5*5+2)/3$

- ☐ 12
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☐ 3
☐ 1
☒ 0

Q5. $(10+2-3)*9/9-1*(10/2)$

- ☐ 0
☐ 1
☐ 2
☒ No of the above answers

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Ibrahim
ibrahim@gmail.com

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#	Exam	Result/100	Status	Date
1	Math	100	Pass	2022-01-04
2	Math	60	Pass	2022-01-05
3	Math	100	Pass	2022-01-05
4	Math	100	Pass	2022-01-05
5	Science	80	Pass	2022-01-05
6	Science	100	Pass	2022-01-05
7	English	80	Pass	2022-01-05
8	English	60	Pass	2022-01-05
9	Chemistry	20	Fail	2022-01-05
10	Physics	20	Fail	2022-01-05
11	Arabic	40	Fail	2022-01-05



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