Name:

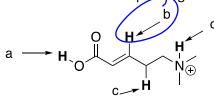
- 1. What is the hybridization and molecular geometry for the nitrogen atom molecule below? (1 point)
 - a. sp², bent

(c. sp³), trigonal planar

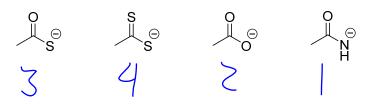
- b. sp³, tetrahedral
- d. sp³, trigonal pyramidal



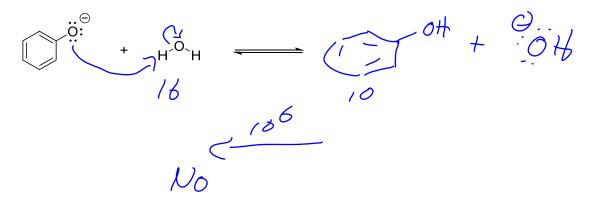
2. Circle the letter corresponding to the **LEAST ACIDIC** proton? (1 point)



3. Rank the following bases from MOST BASIC (1) to LEAST BASIC (4). (2 points)



- 4. a. Provide products and curved arrows for the following acid base reaction.
 - b. Determine whether the acid Is a suitable choice for this protonation. Use pK_a values to justify your answer. (4 points)



5. Provide the IUPAC name for the following molecule. (2 points)

