Paul Gladen - Aug 29, 2024, 12:29 PM CDT

Assignment #3 - Recrystallization Worksheet	
Description I worked in a group with	Complete observations while in lab. Due at 5:00 pm the day following your lab section.
The work for this assignment is in	My notebook
Grade	10 / 10
Graded on Aug 29, 2024, 12:	29 PM CDT

Ilana Berlin - Aug 27, 2024, 3:13 PM CDT

Purpose:

Learn process of recrystallization by dissolving and recrystallizing phenyl benzoate, benzil, and trans-Cinnamic acid

Observe the range of melting points of recrystallized phenyl benzoate, benzil, and trans-Cinnamic acid

Boiling points of solvents for parts 1-2:

95% ethanol <u>78</u> °C hexane <u>69</u> °C water <u>100</u> °C

1. Phenyl benzoate

Describe the original appearance of the compound:

opaque white powder, slightly glittery

Mass of the compound before recrystallization <u>196</u> mg		
dissolved into clear colorless liquid		
Approximate amount of solvent:0.73_mL 95% ethanol		
Describe the appearance of the compound after recrystallization:		
opaque white crystalline chunk formed at bottom of tube		
Mass of the compound after recrystallization117 mg		
Percent recovery <u>59.7</u> %		
Literature mp <u>70 °</u> C Observed mp <u>58.4-61.3 °</u> C		
2. Benzil		
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Describe the appearance of the compound after recrystallization :

long crystalline fibers formed throughout solution in tube
Mass of the compound after recrystallization 196 mg
Percent recovery99.0%
Literature mp <u>94-96</u> °C Observed mp <u>88.2-95.3</u> °C
3. trans-Cinnamic Acid
Describe the original appearance of the compound :
dull white powder
Mass of the compound before recrystallization 304 mg
mass of the sompound solore rearystanization mg
Approximate amount of solvent: 0.65mL of 95% ethanol
Dissolved into clear colorless liquid
Describe the appearance of the compound after recrystallization :
large opaque white crystalline flakes
Mass of the compound after recrystallization 288 mg
Percent recovery94.7%
reitent recovery <u>34.7</u> /6
Literature mp132-135°C