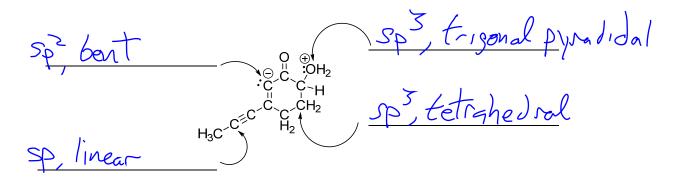
1. Assign formal charges where necessary for the following molecule. All lone pairs are already drawn. (2 points)

2. Provide the **hybridization** and **molecular** geometry for the indicated atoms. (4 points)



3. Identify the functional group in each molecule below. (2 points)

4. Draw a bond line representation from the following Lewis structure including all lone pairs. (2 points)