Name:

1. Assign formal charges where necessary for the following molecule. All lone pairs are already drawn. (2 points)

2. Provide the **hybridization** and **molecular** geometry for the indicated atoms. (4 points)

3. Identify the functional group in each molecule below. (2 points)

$$\begin{array}{c} \mathsf{O} \\ \mathrel{\parallel} \\ \mathsf{C} \\ \mathsf{C} \\ \mathsf{C} \\ \mathsf{H}_2 \\ \mathsf{N} \end{array}$$

4. Draw a bond line representation from the following Lewis structure including all lone pairs. (2 points)