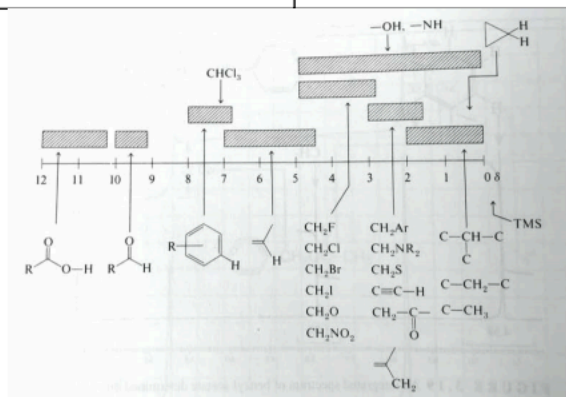


Appendix A

APPROXIMATE CHEMICAL SHIFTS: Protons in Different Molecular Environments

Methyl Protons	ppm	Methylene Protons	ppm	Methine Protons	ppm	Other Protons	ppm
—CH_3	0.9-2.0	>CH_2	1.3-1.9	—CH	1.5-2.2	$\text{C}_6\text{H}_5\text{NH}_2$	3.5-4.5
—O—CH_3	1.4	—O—C(=O)—CH_2	2.2	—O—C(=O)—CH	2.5	$\text{C}_6\text{H}_5\text{H}$	6.5-8.0
—O—C(=O)—CH_3	2.0	—C(=O)—CH_2	2.4	—C(=O)—CH	2.7	—C(=O)—H	10
—C(=O)—CH_3	2.2	$\text{C}_6\text{H}_5\text{CH}_2$	2.7	$\text{C}_6\text{H}_5\text{—CH}$	3.0	—C(=O)—OH	9-12
$\text{C}_6\text{H}_5\text{CH}_3$	2.3	$\text{N}\equiv\text{C—CH}_2$	2.8	—O—CH	3.7	>N—H	0.5-5
X—CH_3	2.7-3.0	—O—CH_2	3.4	X—CH	4.0-4.1	—O—H	0.5-5
—O—CH_3	3.3	X—CH_2	3.4	—C(=O)—O—CH	4.8	>C=CH_2	~4-8
—C(=O)—O—CH_3	3.7-3.9	—C(=O)—O—CH_2	4.1			$\text{CH}_2\text{=CH—O—H}$	>15
$\text{C}_6\text{H}_5\text{O—CH}_3$	3.8	$\text{C}_6\text{H}_5\text{O—CH}_2$	3.9-4.3				



From: Pavia, D. L., Lampman, G. M., Kriz, G. S., & Vyvyan, J. A. (2008). Introduction to Spectroscopy. Cengage Learning. p. 124.