

ASESMA 2023: Hands-on session -- Day-2

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Topics of the session:

1. The basics: benzene and graphene

- Single molecule calculations: benzene
- Basics of post-processing: plotting wavefunctions
- Periodic systems: graphene

2. DFT bread and butter: convergence tests and lattice optimization

- Basic convergence tests on fcc Si: energy cutoff and k-grid
- Optimizing the lattice constant:
- Extracting physical information: bulk modulus

To get the latest version of the exercises, in the ASESMA 2023 folder execute: ``git pull``

About Quantum ESPRESSO

More info about Quantum ESPRESSO can be found in:

- <https://www.quantum-espresso.org/>
- Quantum ESPRESSO (QE) documentation:
 - on-line manuals ([pw.x code](#)) and ([pp.x code](#)) for input file description
- Useful resources:
 - [Input generator](#) from materials cloud
 - Pseudopotentials [SSSP library](#) and [GBRV library](#)

Exercise 1: Benze

How to calculate and plot molecular orbitals of benzene
(actually $\text{sign}(\psi(r)) \cdot |\psi(r)|^2$)

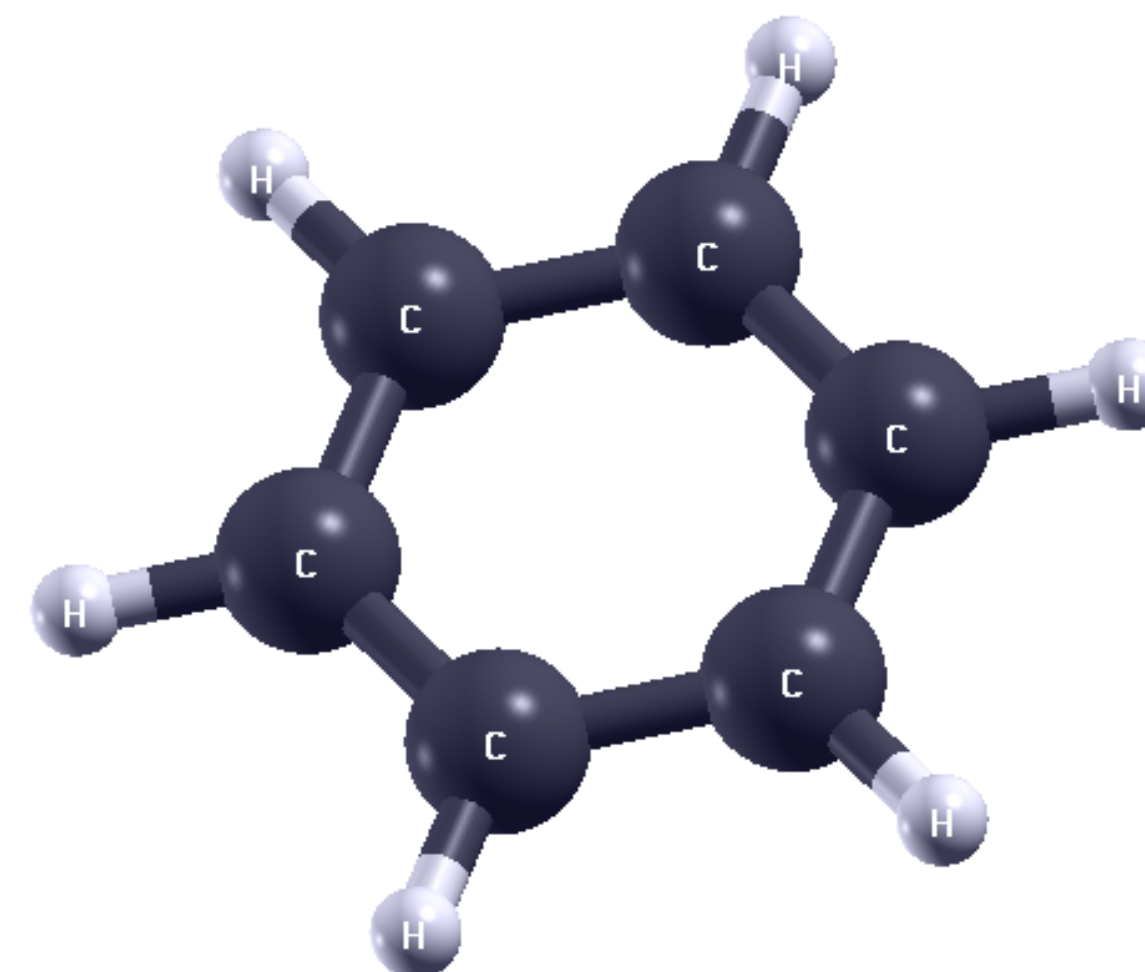
Step 0:

View the Benzene Molecule

```
xcrysden --pwi pw.benzene.scf.in
```

Move the mouse around to take a look at the molecule.

The ".pwi" format stands for quantum espresso pw.x input file. For a pw.x output file, that would be ".pwo".



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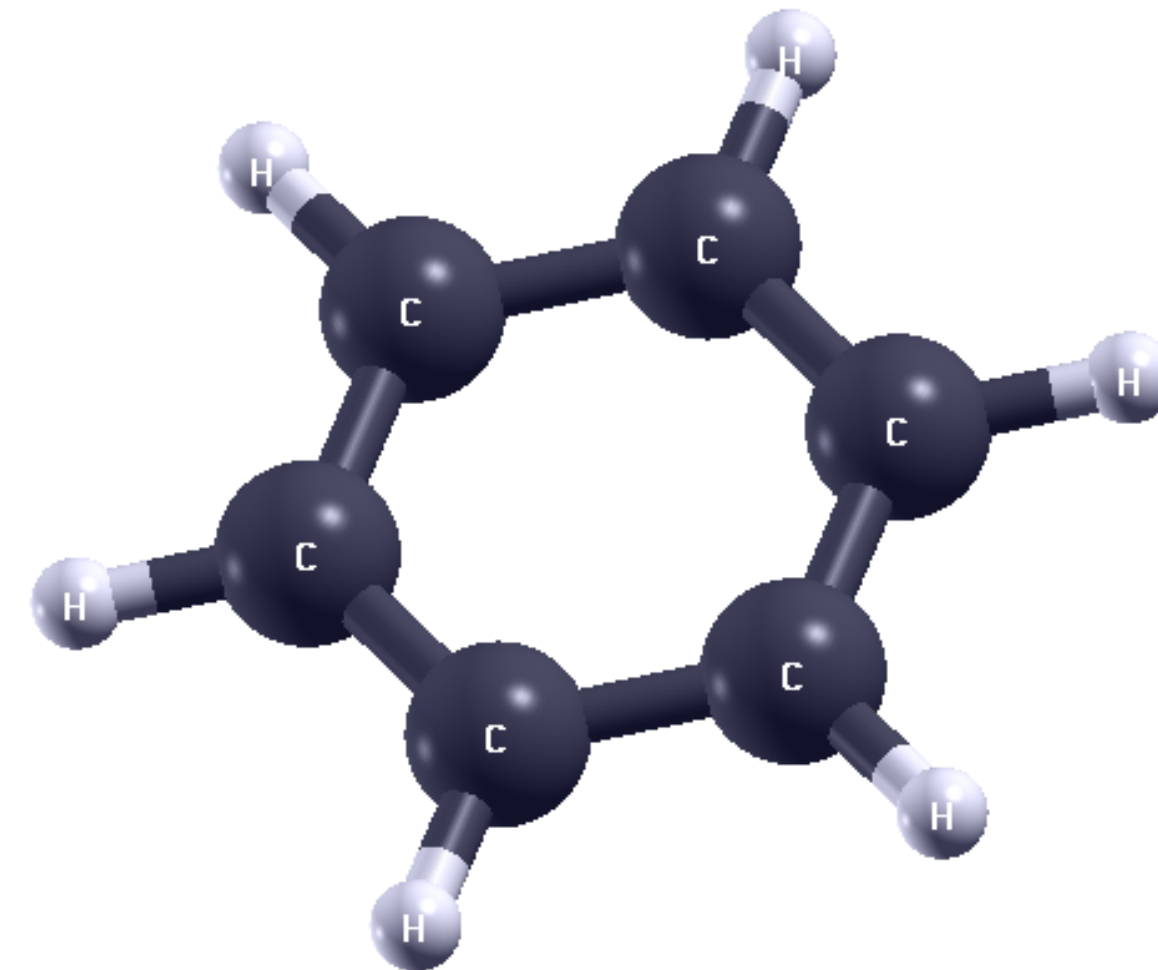
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Step 1:

Perform the SCF Calculation

```
pw.x -in pw.benzene.scf.in \  
> pw.benzene.scf.out
```

The pw.x code will now perform a DFT self-consistent field (SCF) calculation.



Step 2:

Postprocess the wavefunction data

```
pp.x -in pp.benzene.psi2.in > pp.benzene.psi2.out
```

The resulting wavefunction amplitudes $\text{sign}(\psi(r)) \cdot |\psi(r)|^2$ are written to files *psi2.benzene_K001_B0*.xsf*

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Step 3:

Plot a single molecular orbital

```
xcrysden --xsf psi2.benzene_K001_B006.xsf
```

Make a fancy display of the molecular orbital (follow the instructions of the tutor). The README.md contains the info necessary to save the current state view. You can try and save that as *MO-state.xcrysden* and then use the same view with another orbital:

```
xcrysden --xsf psi2.benzene_K001_B005.xsf --script MO-state.xcrysden
```


Step 4:

Plot All Molecular Orbitals

Run in the terminal:

```
bash plot-psi2.sh
```

The shell script is trying to plot all wavefunctions you computed, be patient (this will take a while).

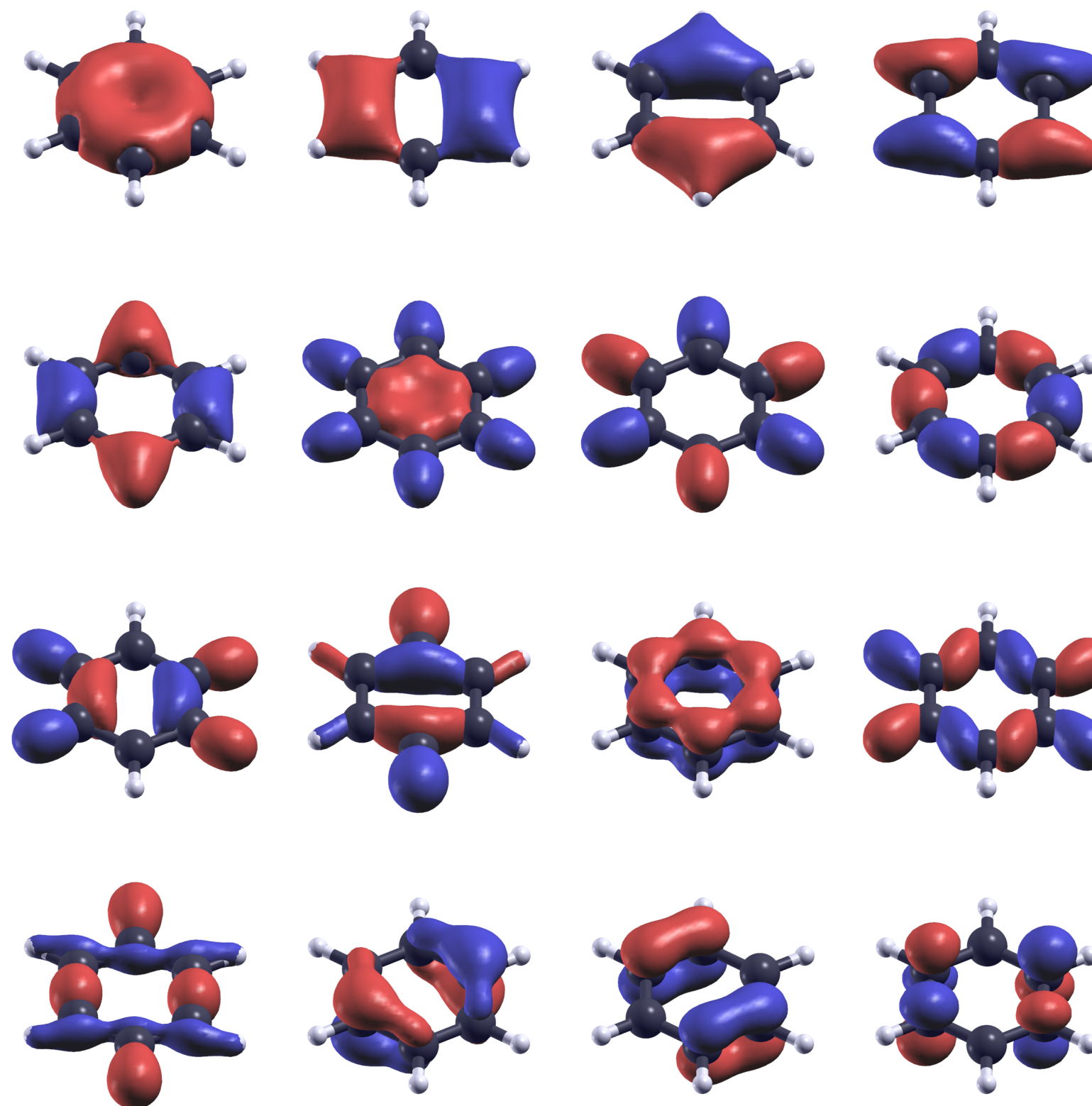
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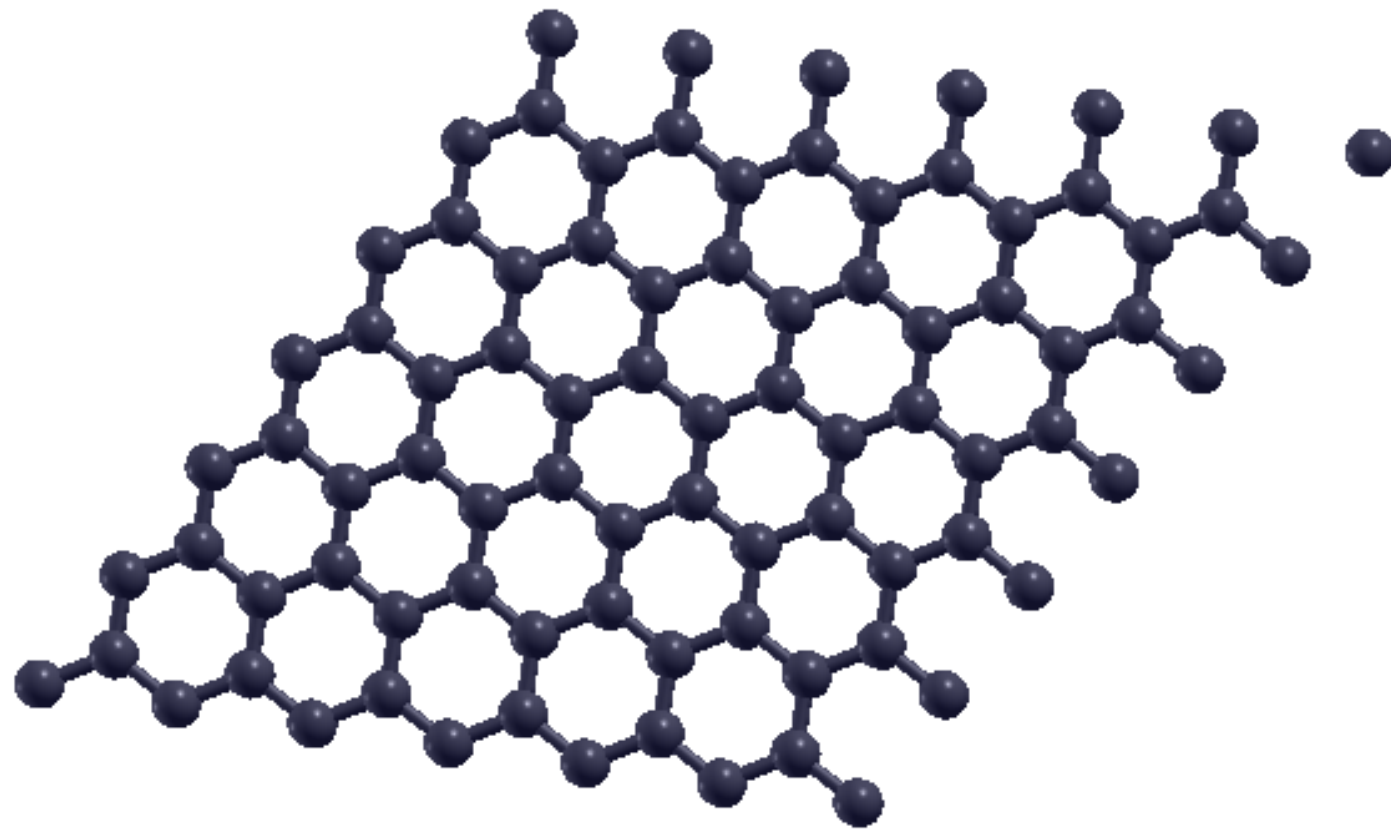


Exercise 2: Graphs

Introduction to periodic systems

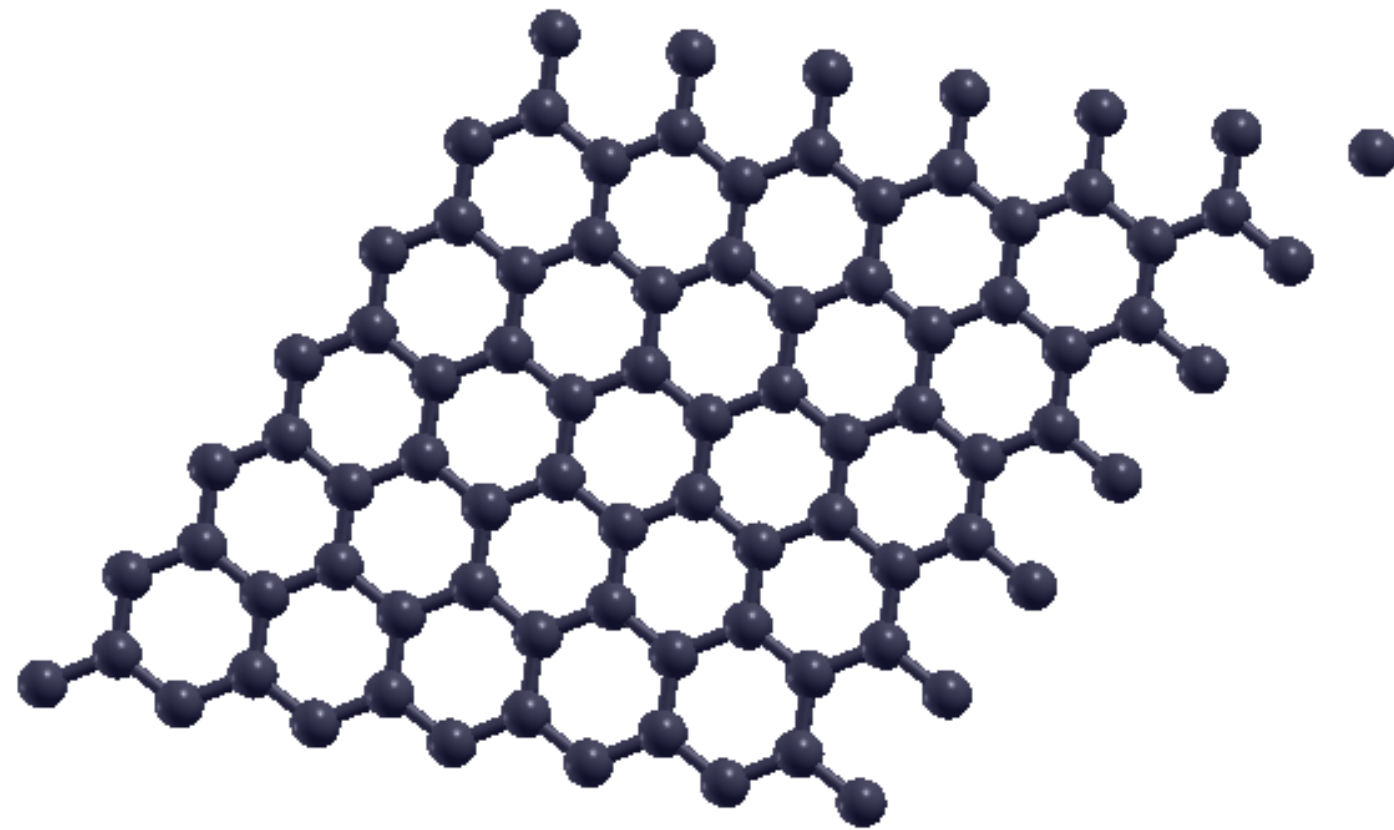
Graphene is a single sheet of carbon atoms

Periodic boundary conditions are applied in the plane

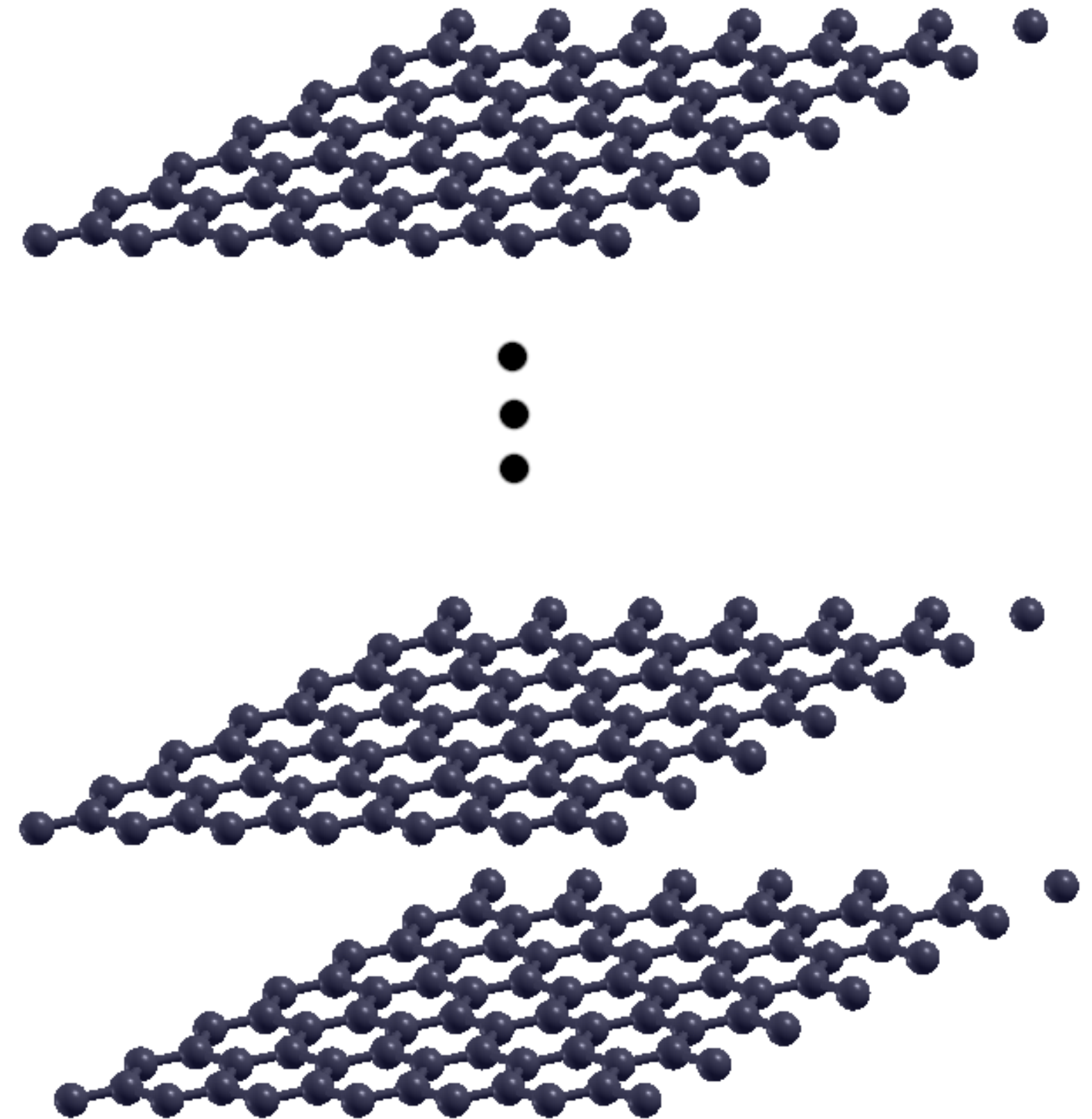


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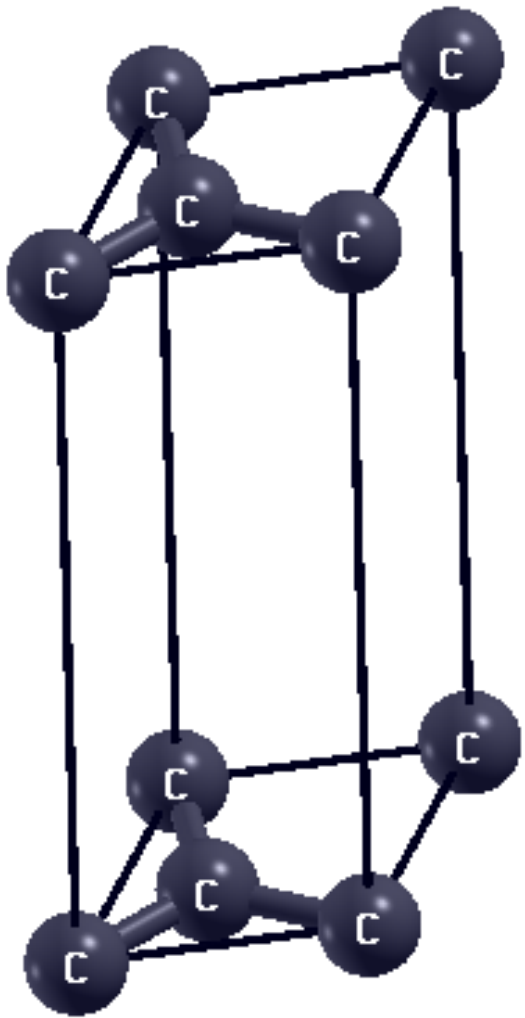


But in quantum espresso every direction is periodic



Computing the electronic structure of graphene

We can describe the graphene sheet with just 2 atoms in the unit cell



Step 1: Perform the SCF Calculation

```
pw.x -in pw.graphene.scf.in \  
> pw.graphene.scf.out
```

Step 2: Make a DOS Plot

```
pw.x -in pw.graphene.nscf.in \  
> pw.graphene.nscf.out
```

Step 3: Calculate DOS Datafile

```
dos.x -in dos.graphene.in \  
> dos.graphene.out
```

The resulting DOS datafile is written to *graphene.dos*

Step 4: Plot the DOS with Gnuplot

```
gnuplot dos.gp
```

Step 5: Perform Bands Calculation

```
pw.x -in pw.graphene.bands.in > pw.graphene.bands.out
```

Step 6: Generate Datafile for Plotting

```
bands.x -in bands.graphene.in > bands.graphene.out
```

The resulting datafile for plotting is written to *graphene.bands.dat.gnu*

Step 7: Plot the bandstructure with Gnuplot

```
gnuplot spaghetti.gp
```

Step 8: Find and Edit Fermi Energy in Gnuplot Files

```
grep Fermi pw.graphene.nscf.out
```

Use the Fermi energy value from the output and edit *dos.gp* and *spaghetti.gp* files accordingly.

Step 9: Replot the DOS and bandstructure

```
gnuplot dos.gp
```

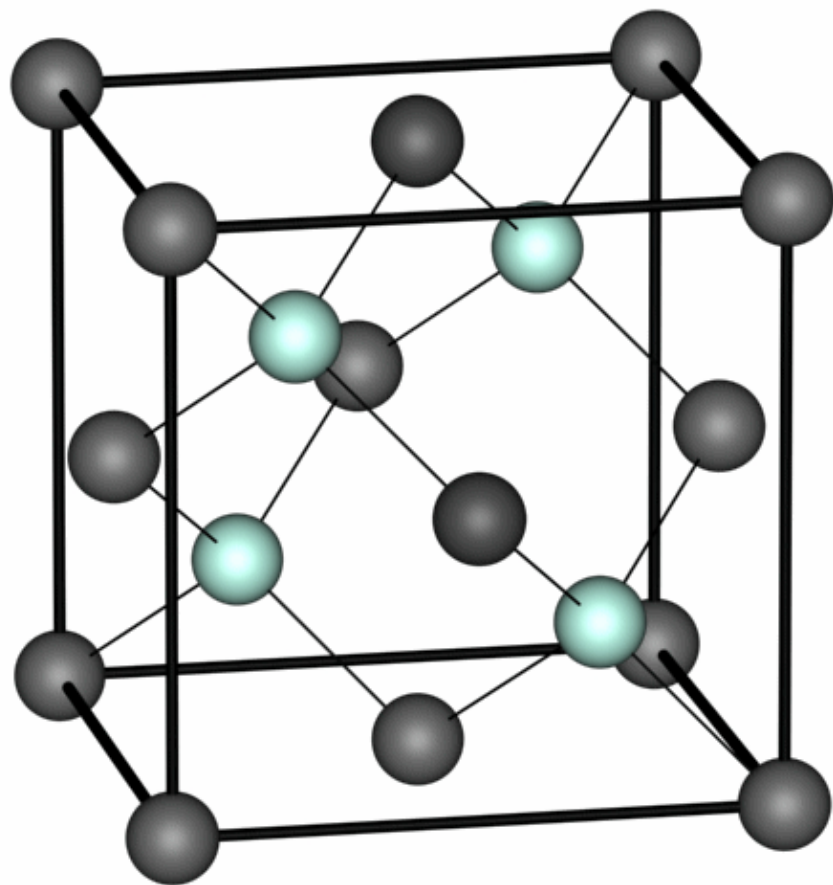
```
gnuplot spaghetti.gp
```

Exercise 3: Silica

Bread and butter of real DFT calculations

Crystal Structure Definition

bulk silicon is a face-centered cubic (FCC) lattice with 2 atoms in the unit cell at positions $[0\ 0\ 0]$ and $[1/4\ 1/4\ 1/4]$ (this is also called *diamond* or *zinc-blend* structure)



```
&CONTROL
calculation='scf',
prefix='silicon',
pseudo_dir='../pseudo/',
outdir='./tmp'
/
&SYSTEM
ibrav = 2,
celldm(1) = 10.2,
nat = 2,
ntyp = 1,
ecutwfc = 28,
/
&ELECTRONS
/
ATOMIC_SPECIES
Si 28.086 Si.pz-vbc.UPF
ATOMIC_POSITIONS
Si 0.00 0.00 0.00
Si 0.25 0.25 0.25
K_POINTS automatic
4 4 4 1 1 1
```

Crystal Structure Definition

- `ibrav=2`: meaning fcc lattice
- Just one:
`celldm(1)=10.2`,
lattice parameter a in Bohr
- `nat=2`: two atoms
- `ntyp=1`: one distinct atomic specie
- Where are the atoms located in the unit cell?
See card ATOMIC POSITIONS: here, in Cartesian axes, in units of a

```
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/
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/
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Si 28.086 Si.pz-vbc.UPF
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Si 0.00 0.00 0.00
Si 0.25 0.25 0.25
K_POINTS automatic
4 4 4 1 1 1
```

Logic of the examples:

Convergence tests for Si bulk consist of the following steps:

1. Converge the basis-set
2. Converge the k-points
3. With converged basis-set and k-points, calculate the lattice parameter of FCC bulk Si
4. With converged basis-set, k-points, and lattice parameter, fit the bulk modulus of FCC Si

Description of folder structure:

- *ex1.ecutwfc/* - Convergence tests for cutoff energy
- *ex2.kpoints/* - Convergence tests for k-points
- *ex3.alat/* - Search of lattice parameter of Si bulk (*alat* = *a* lattice parameter)

Kinetic Energy Cutoff (ecutwfc)

The kinetic energy cutoff ecutwfc (in Ry) determines the size of the Plane-Wave (PW) basis set used to expand wave-functions (i.e. Kohn-Sham orbitals). The default value for the charge density is $\text{ecutrho}=4*\text{ecutwfc}$, which is suitable for norm-conserving pseudopotentials.

A manual test of convergence with respect to the kinetic energy cutoff involves the following tasks (Note: We will not perform this manually!):

1. Change the value of ecutwfc in the pw.si.scf.in input file to different values such as 16, 20, 24, 28, 32 Ry.
2. For each value of ecutwfc, run pw.x and record the final total energy.
3. Store the data in a file, let's say *si.etot_vs_ecut* (each line should contain two values: ecutwfc and total energy).
4. Plot the energies collected in si.etot vs ecut using your preferred plotting program.

Gnuplot:

```
plot "si.etot_vs_ecut" w lp
```

Python:

```
python3 plot_etot_ecut.py
```

Kinetic Energy Cutoff (ecutwfc)

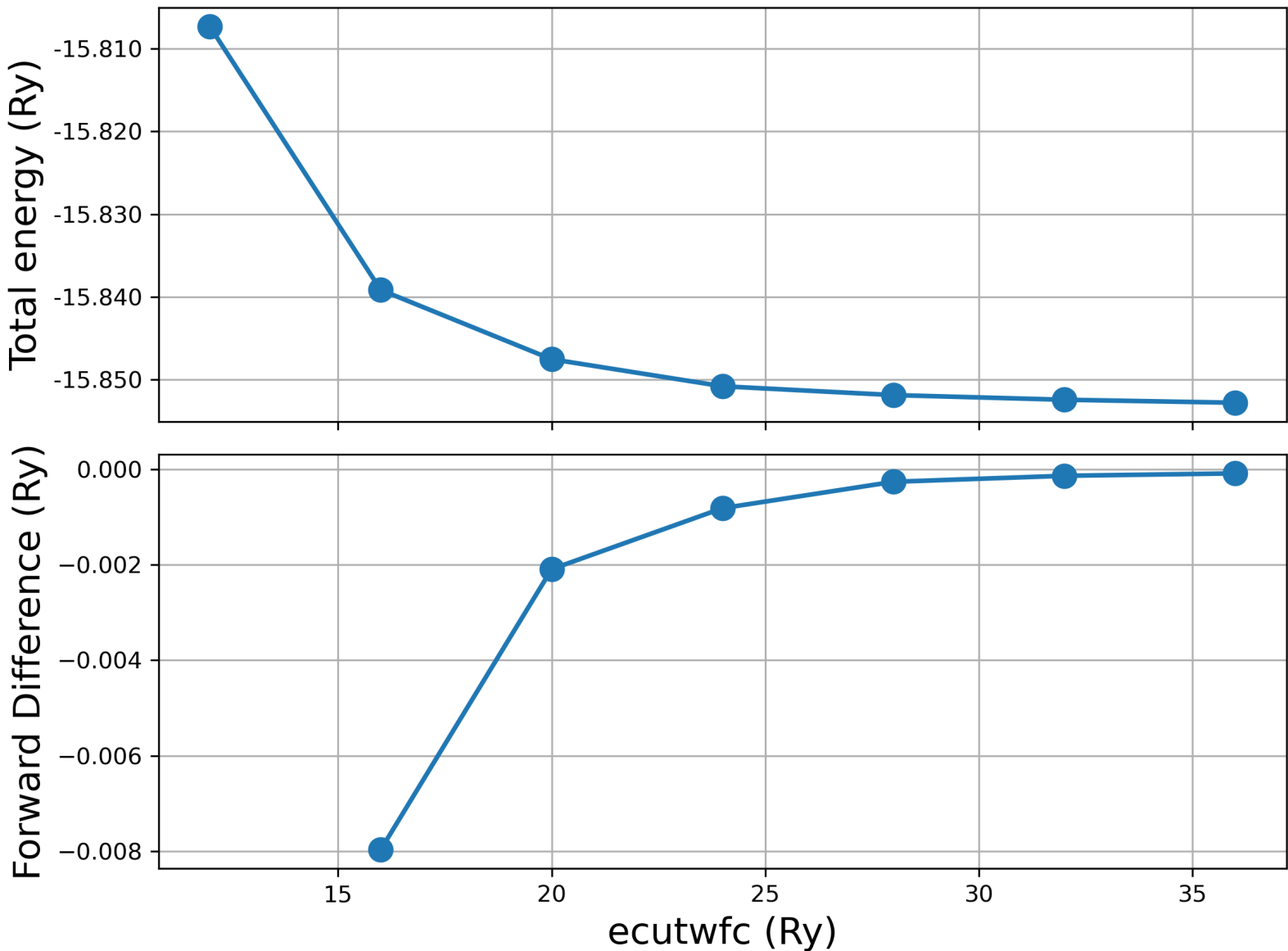
In the end, we want to have something of the type:

12.0000	-15.80731200
16.0000	-15.83916740
20.0000	-15.84754590
24.0000	-15.85081789
28.0000	-15.85188267
32.0000	-15.85244512
36.0000	-15.85280759

We are looking for differences of < mRy

Note:

The energy per-se is meaningless, only energy differences are meaningful!



Convergence with Respect to K-Points

A sufficiently dense grid of k-points is required to accurately represent the periodicity of the system. To test the convergence with respect to k-points, you can modify the K_POINTS card in your input file. Request automatic Monkhorst-Pack grids using the following format:

```
K_POINTS automatic  
nk1 nk2 nk3 k1 k2 k3
```

Gradually increase the values of nk1, nk2, and nk3 while keeping k1, k2, and k3 equal to 1. For example, you can try increasing nk1 = nk2 = nk3 to 2, 4, 6, 8, and so on. Run the pw.x calculation for each set of k-point values.

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Note:

- The first three nk1 nk2 nk3 numbers indicate the number of grid points along crystal axes 1, 2, 3.
- The second three k1 k2 k3 numbers, either 0 or 1, indicate whether the grid starts from 0 or is displaced by half a step along crystal axes 1, 2, 3.
- Convergence is not necessarily monotonic, as there is no variational principle with respect to the number of k-points. Why do you think this is the case? Try to repeat the example with odd values of nk.

Lattice Parameter Determination

In silicon (Si), the equilibrium state is determined solely by the minimum-energy lattice parameter. Due to symmetry, there are no forces on the atoms. You can verify this by setting `tpnfor=.true.` in the namelist `&CONTROL` and checking for the forces reprinted at the end of the calculation.

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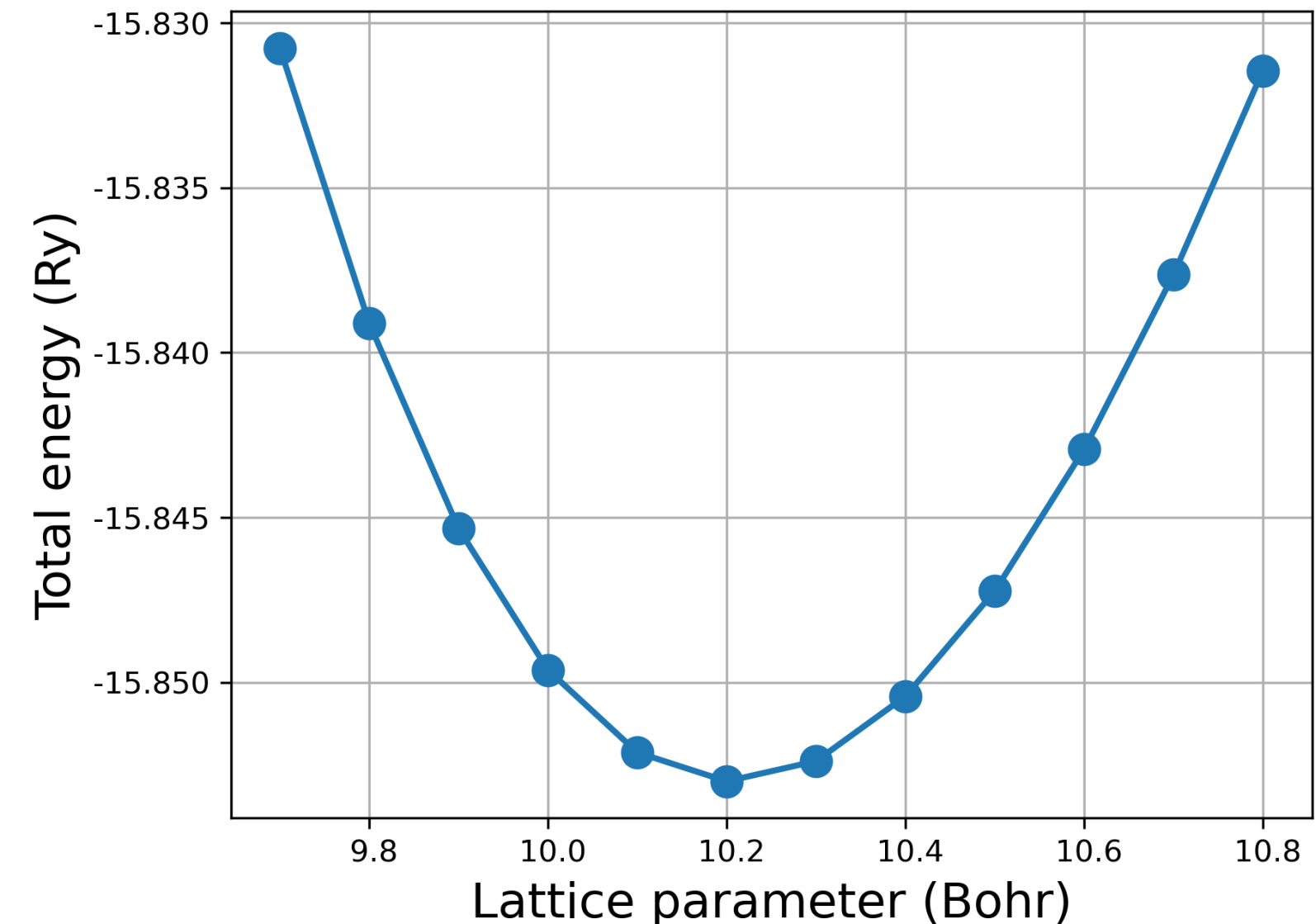
- Choose suitable values for `ecutwfc` (e.g., 36 Ry) and the k-point grid (e.g., 4 4 4 1 1 1).
- Run `pw.x` for values of `celldm(1)` ranging from 9.7 to 10.8 Bohr in steps of 0.1 Bohr

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Extracting physical parameters: the Bulk modulus

The bulk modulus is defined as:

$$B = -V \left(\frac{dP}{dV} \right)$$

The Murnaghan equation of state assumes that the bulk modulus (B) is proportional to the pressure (P).

$$B = B_0 + \frac{B'_0}{B_0} \cdot P$$

$$E(V) = E_0 + \left[\frac{1}{B'_0(B'_0 - 1)} \left(\frac{V}{V_0} \right)^{1-B'_0} + \frac{1}{B'_0} \frac{V}{V_0} - \frac{1}{B'_0 - 1} \right]$$

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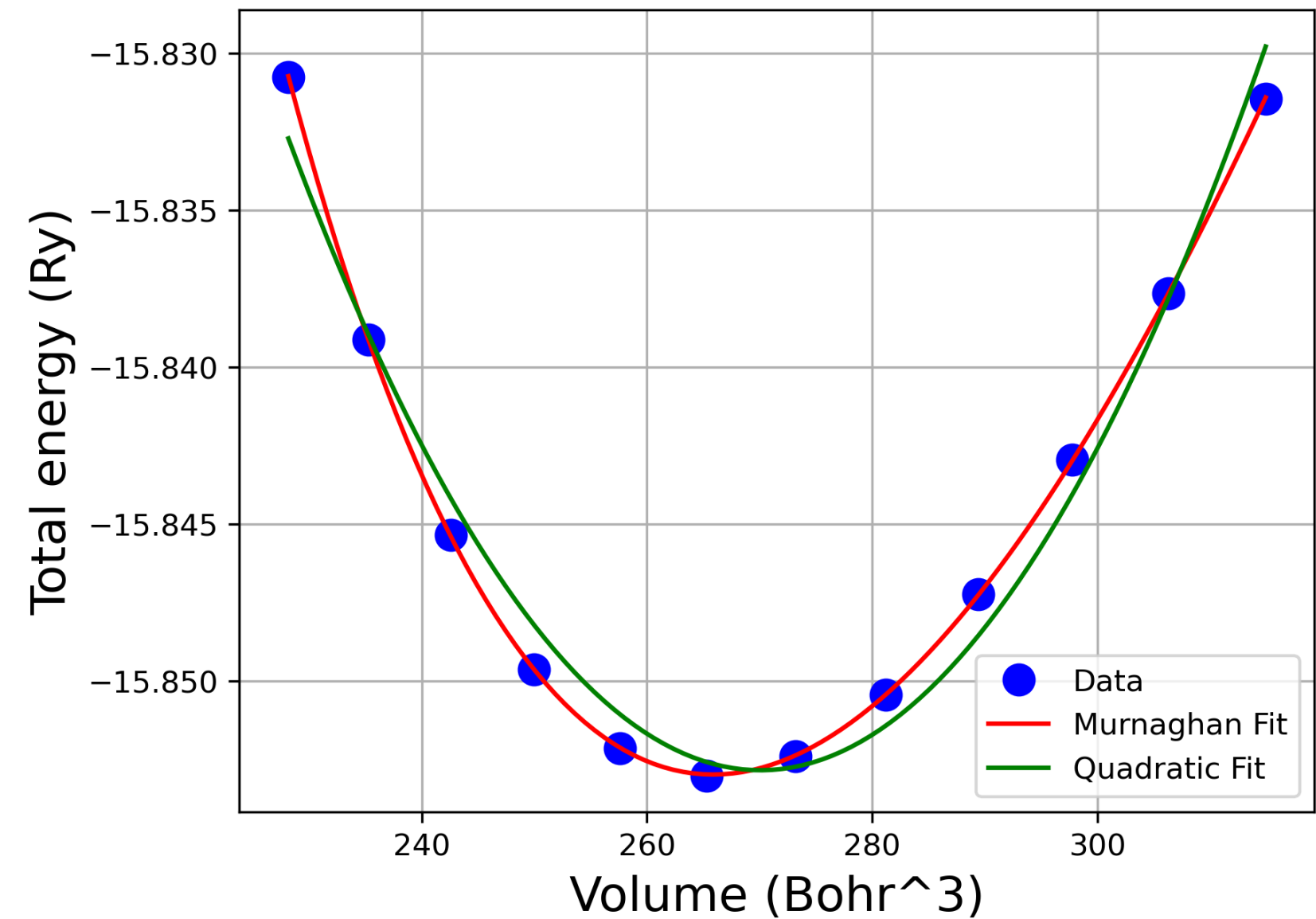
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Thank you ! Questions?

