



Class overview today - April 4, 2016

- **Part I - Natural diffusion**
 - Introduction to the diffusion process
 - Mathematical description of diffusion
 - Hillslope diffusive processes
- **Part II - Solving the diffusion equation**
 - Components of the diffusion equation
 - General form of the diffusion equation
 - Solution of the diffusion equation for hillslope diffusion



Introduction to Quantitative Geology

Lecture 5

Natural diffusion:

Hillslope sediment transport, Earth's thermal field

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4.4.2016



Goals of this lecture

- Introduce the **diffusion process**
- Present some examples of **hillslope diffusive processes**
(heave/creep, solifluction, rain splash)



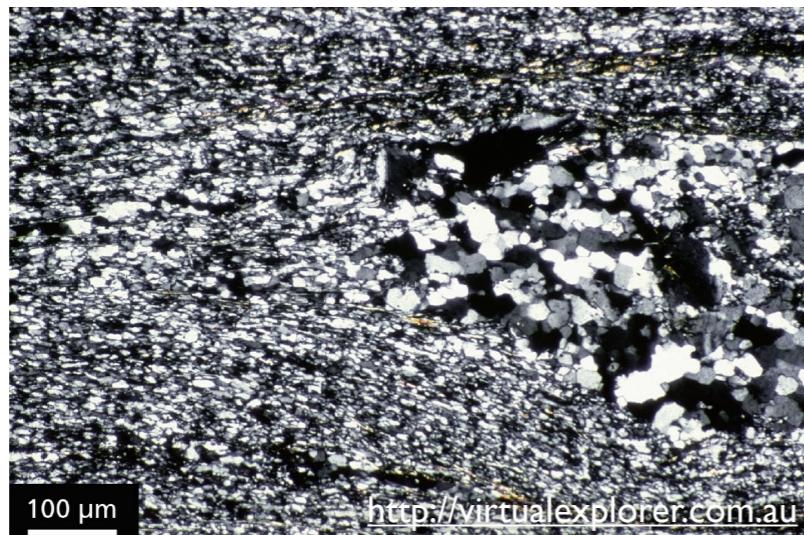
Natural diffusion

- **What are some examples of diffusion processes in nature?**



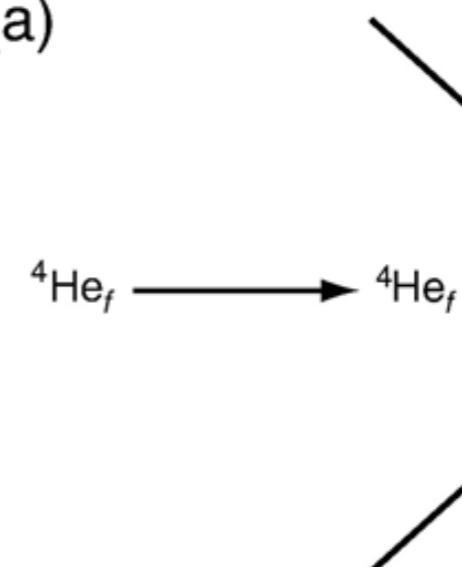
Diffusion as a geological process

Grain boundary
sliding



${}^4\text{He}$ diffusion in apatite

(a)

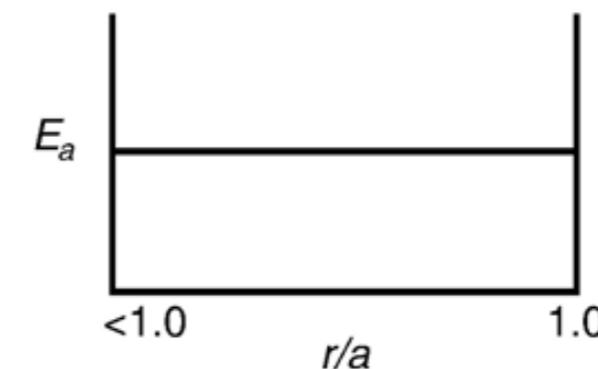


Shuster et al., 2006

Rain splash



Hillslope erosion



Thermochronology

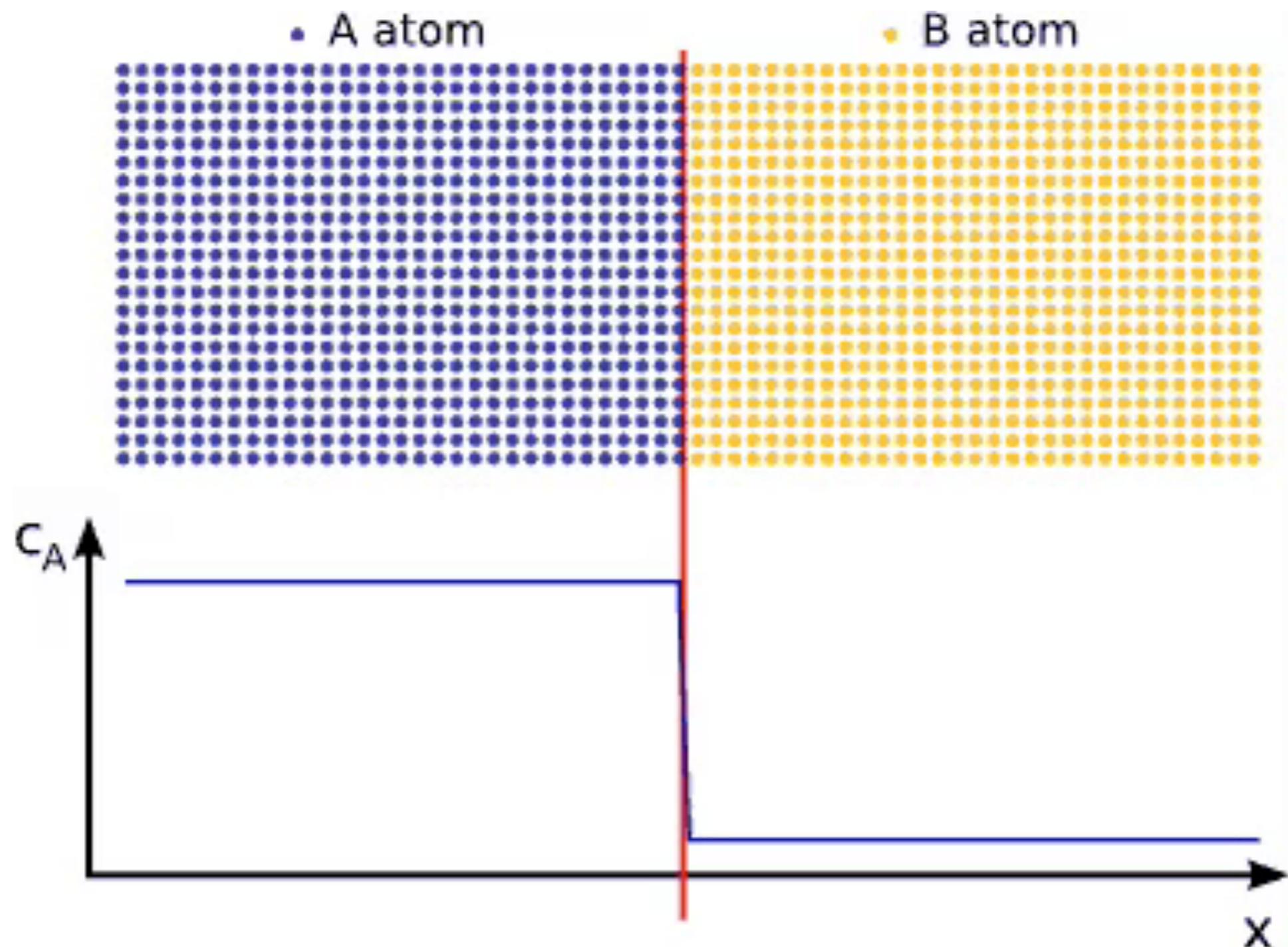


General concepts of diffusion

- **Diffusion** is a process resulting in mass transport or mixing as a result of the random motion of diffusing particles

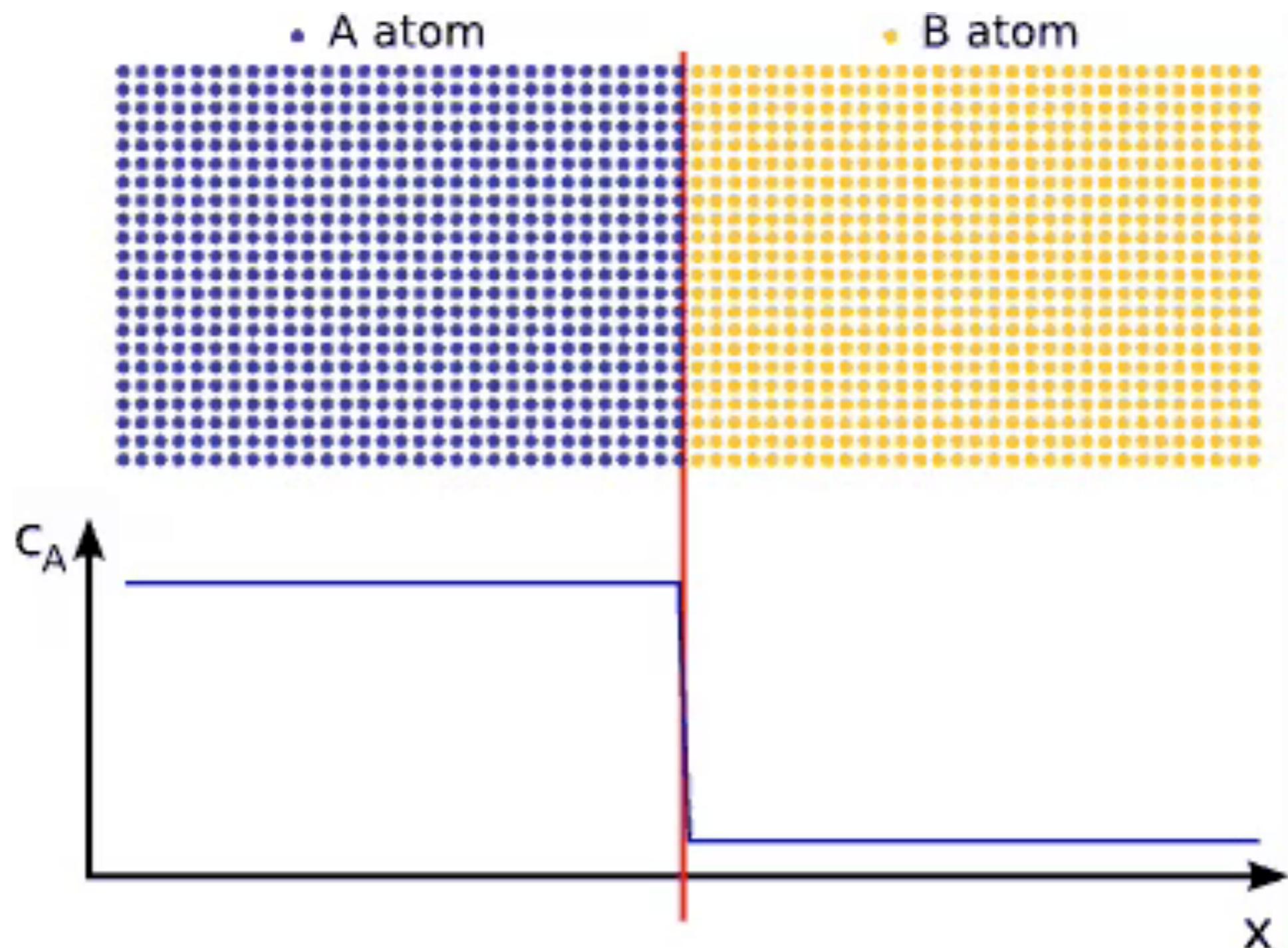


The diffusion process





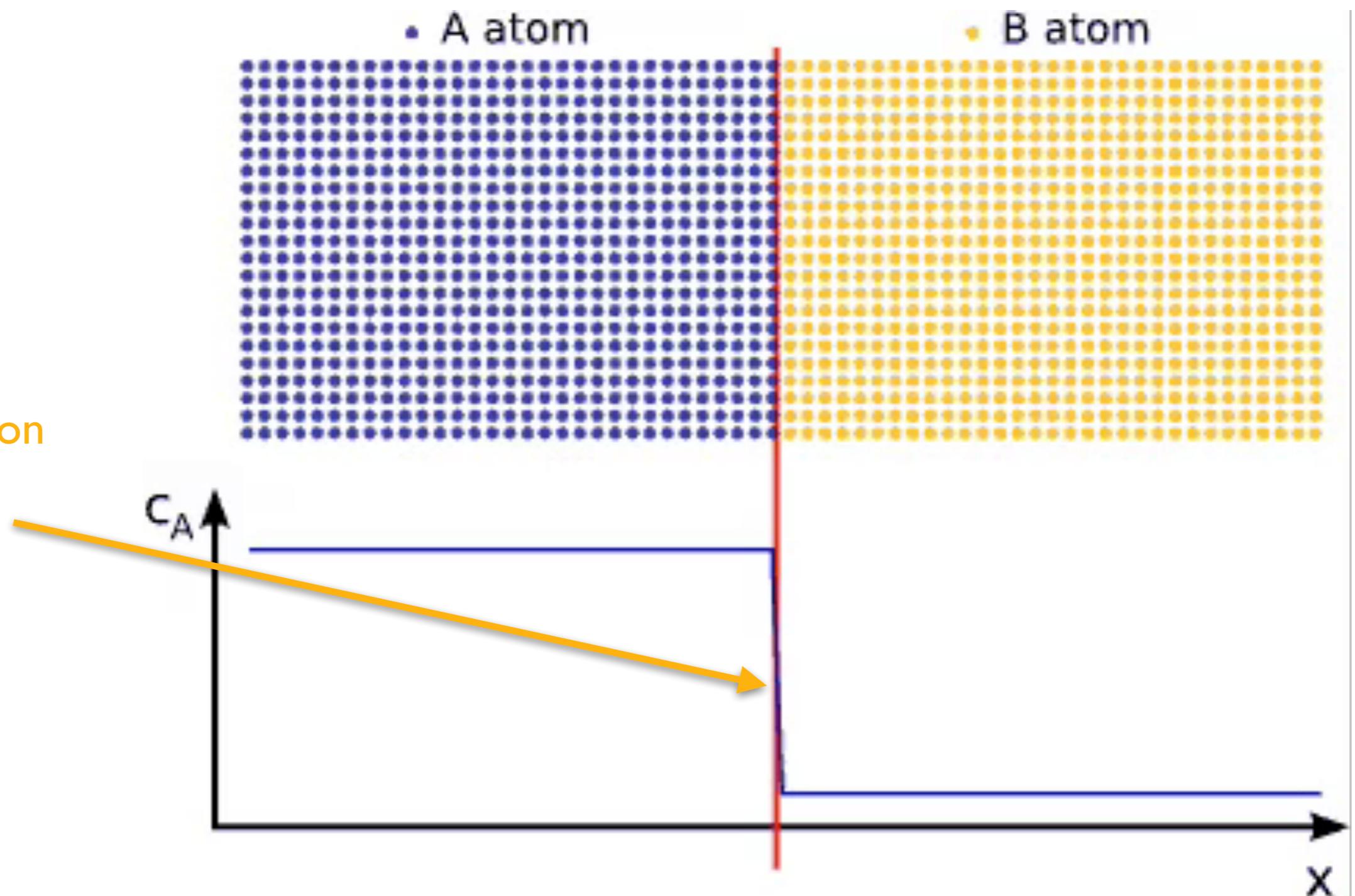
The diffusion process





The diffusion process

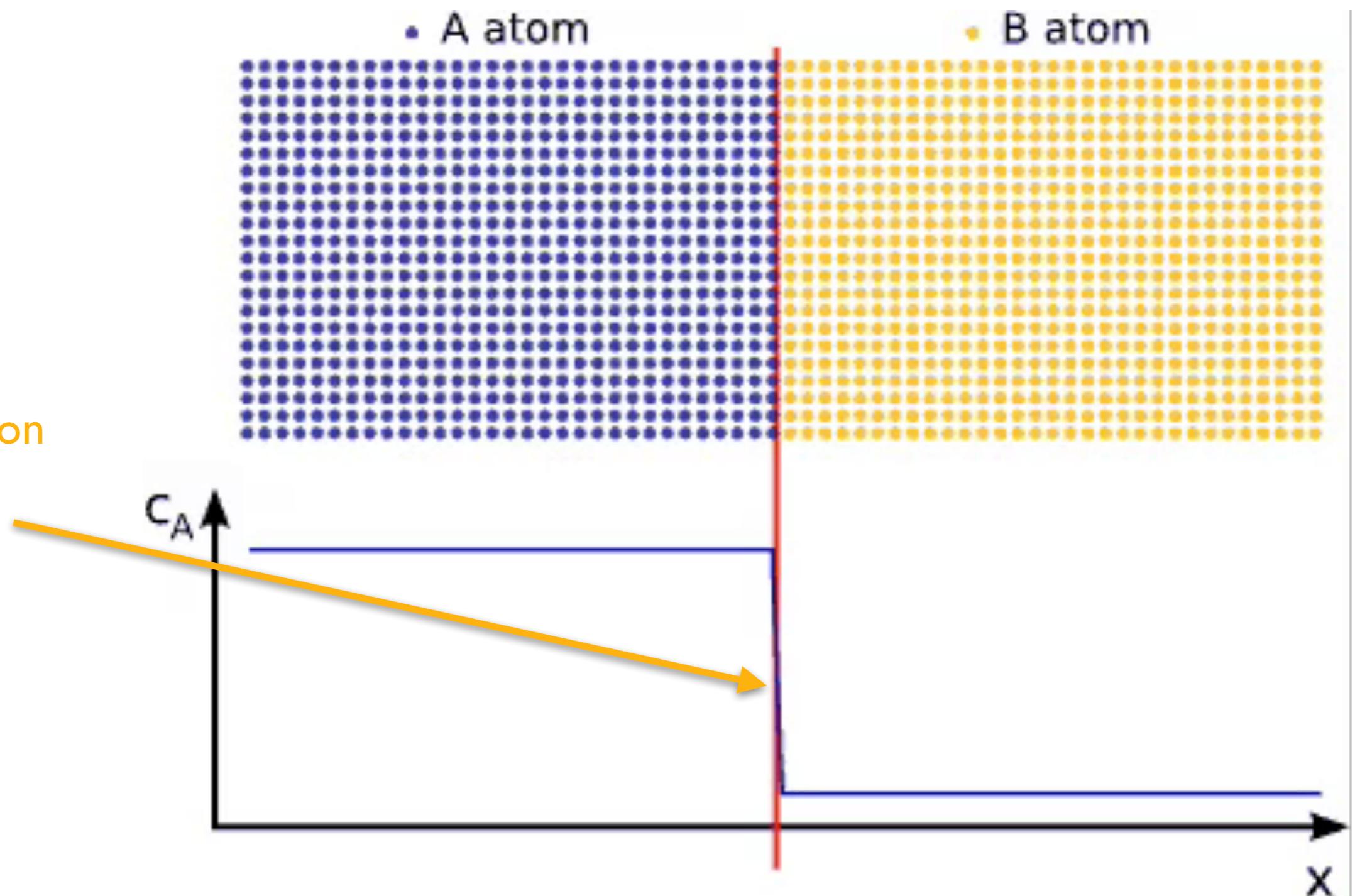
Concentration
gradient

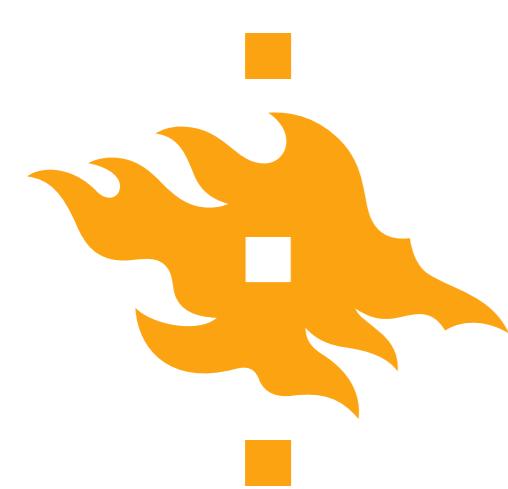




The diffusion process

Concentration
gradient





General concepts of diffusion

- **Diffusion** is a process resulting in mass transport or mixing as a result of the random motion of diffusing particles
- Net motion of mass or transfer of energy is from regions of high concentration to regions of low concentration
- Diffusion reduces concentration gradients



A more quantitative definition

- **Diffusion** occurs when a **conservative property** moves through space at a **rate proportional to a gradient**
- **Conservative property:** A quantity that must be conserved in the system (e.g., mass, energy, momentum)
- **Rate proportional to a gradient:** Movement occurs in direct relationship to the change in concentration
- Consider a one hot piece of metal that is put in contact with a cold piece of metal. Along the interface the change in temperature will be most rapid when the temperature difference is largest



A mathematical definition

- We can now translate the concept of diffusion into mathematical terms.
 - We've just seen “Diffusion occurs when a (1) **conservative property** moves through space at a (2) **rate proportional to a gradient**”
 - If we start with part 2, we can say in comfortable terms that **[transportation rate]** is proportional to **[change in concentration over some distance]**



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- In slightly more quantitative terms, we could say **[flux]** is proportional to **[concentration gradient]**



A mathematical definition

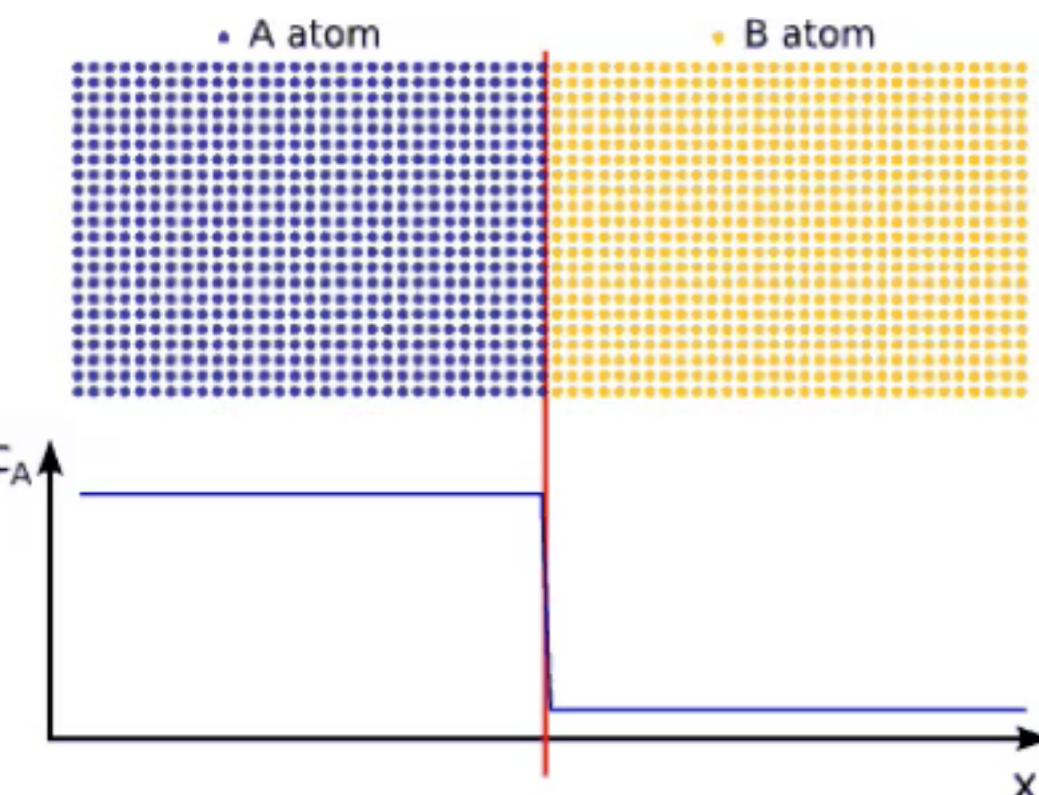
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- Finally, in symbols we can say

$$q \propto \frac{\Delta C}{\Delta x}$$

where q is the mass flux, \propto is the “proportional to” symbol, Δ indicates a change in the symbol that follows, C is the concentration and x is distance



A mathematical definition



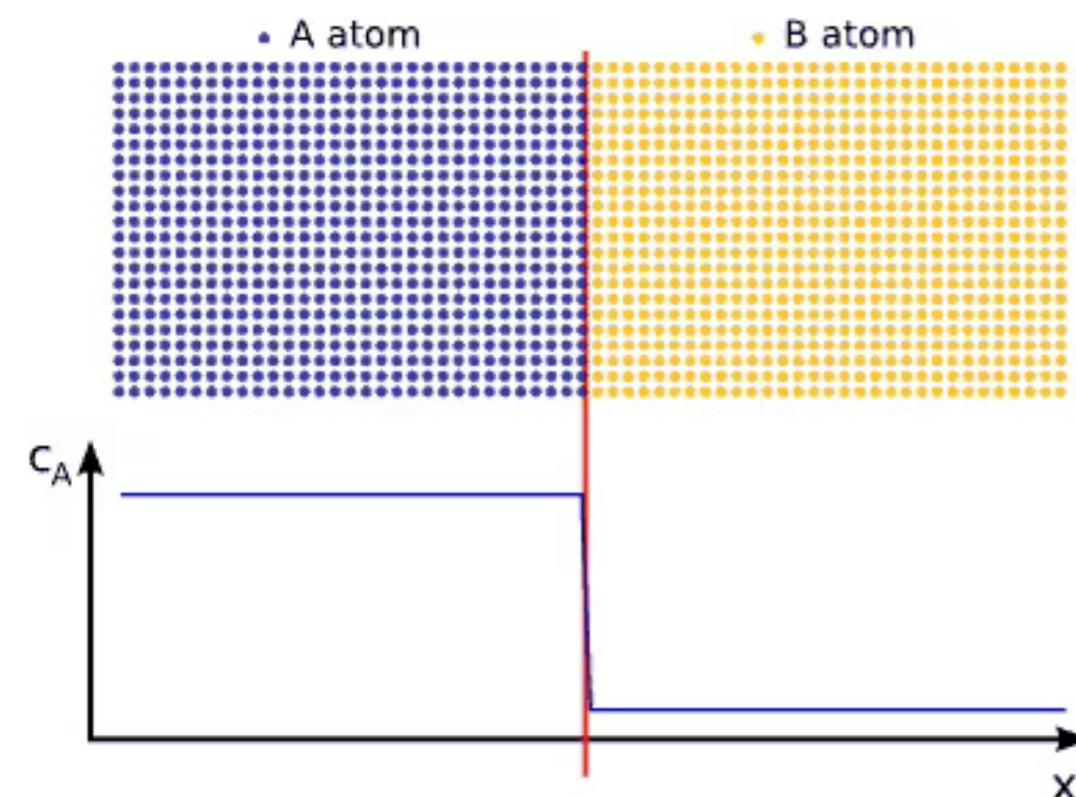
- If transport is directly proportional to the gradient, we can replace the proportional to symbol with a constant
- We can also replace the finite changes Δ with infinitesimal changes ∂
- Keeping the same colour scheme, we see

$$q \propto \frac{\Delta C}{\Delta x} \longrightarrow q = -D \frac{\partial C}{\partial x}$$

where D is a constant called the **diffusion coefficient** or **diffusivity**



A mathematical definition



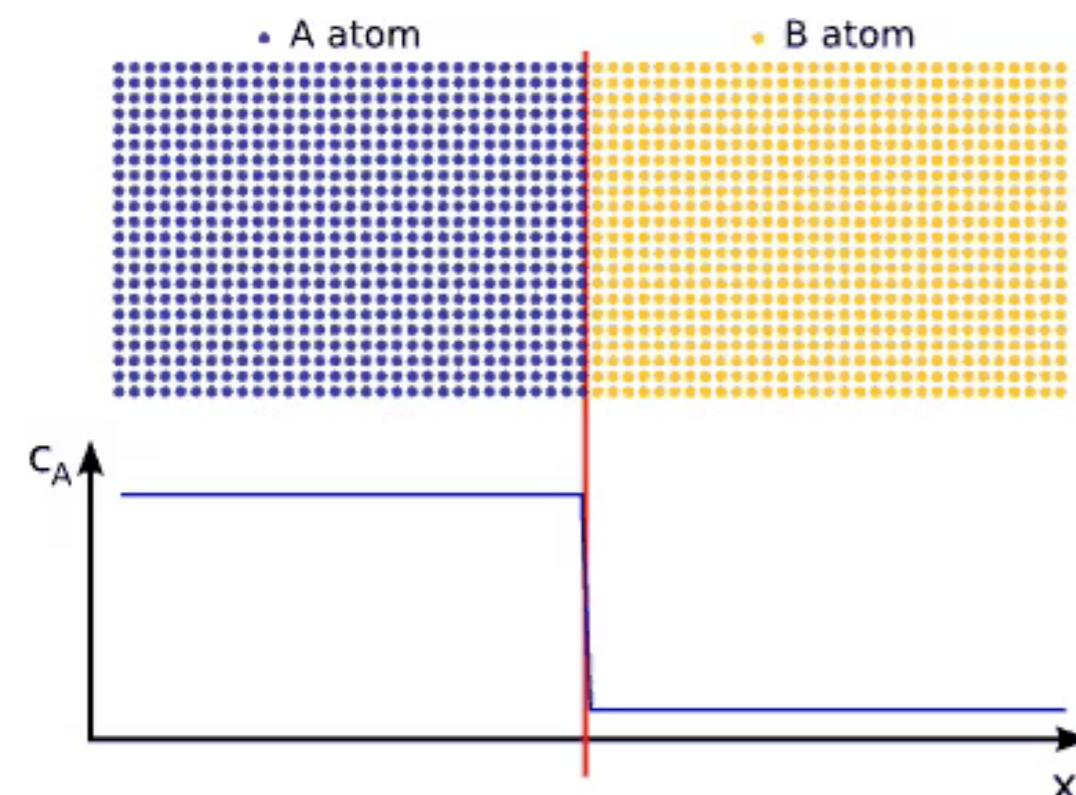
- Consider the example to the left of the concentration of some atoms A and B
- Here, we can formulate the diffusion of atoms of A across the red line with time as

$$q = -D \frac{\partial C_A}{\partial x}$$

where C_A is the **concentration** of atoms of A



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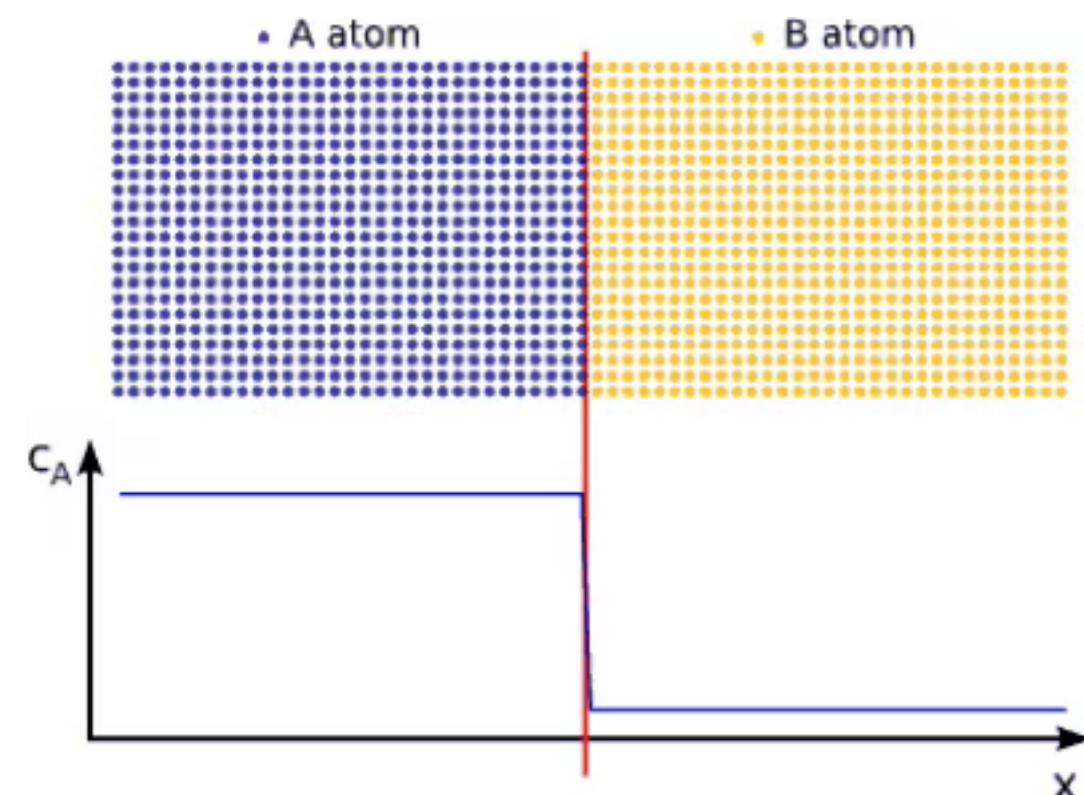
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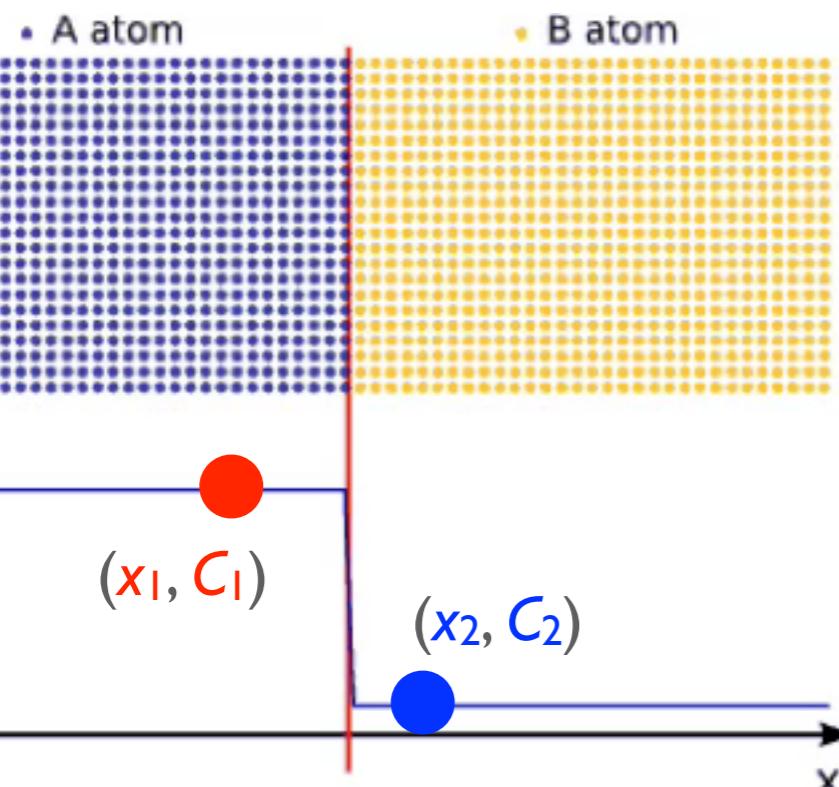
- OK, but why is there a minus sign?

$$q = -D \frac{\partial C_A}{\partial x}$$





A mathematical definition



- OK, but why is there a minus sign?

$$q = -D \frac{\partial C_A}{\partial x}$$

- We can consider a simple case for finite changes at two points: (x_1, C_1) and (x_2, C_2)
- At those points, we could say

$$q = -D \frac{\Delta C}{\Delta x}$$

$$q = -D \frac{C_2 - C_1}{x_2 - x_1}$$

- As you can see, ΔC will be negative while Δx is positive, resulting in a negative gradient



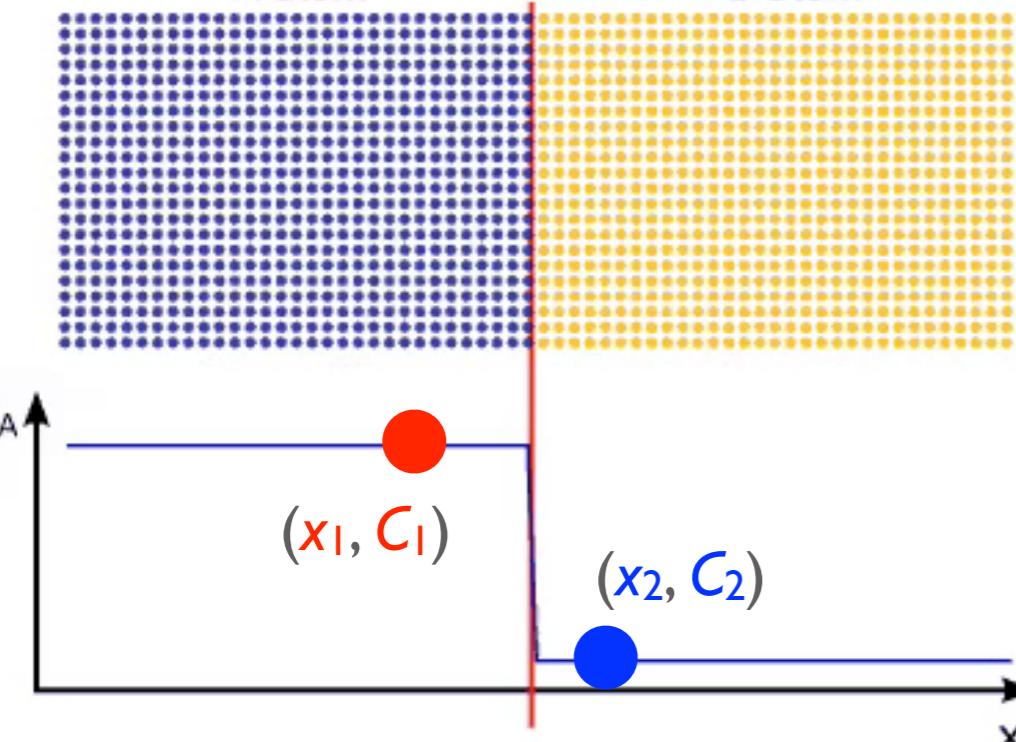
A mathematical definition

Positive flux of A



• A atom

• B atom



- OK, but why is there a minus sign?

$$q = -D \frac{\partial C_A}{\partial x}$$

- Multiplying the negative gradient by $-D$ yields a positive flux q along the x axis, which is what we expect

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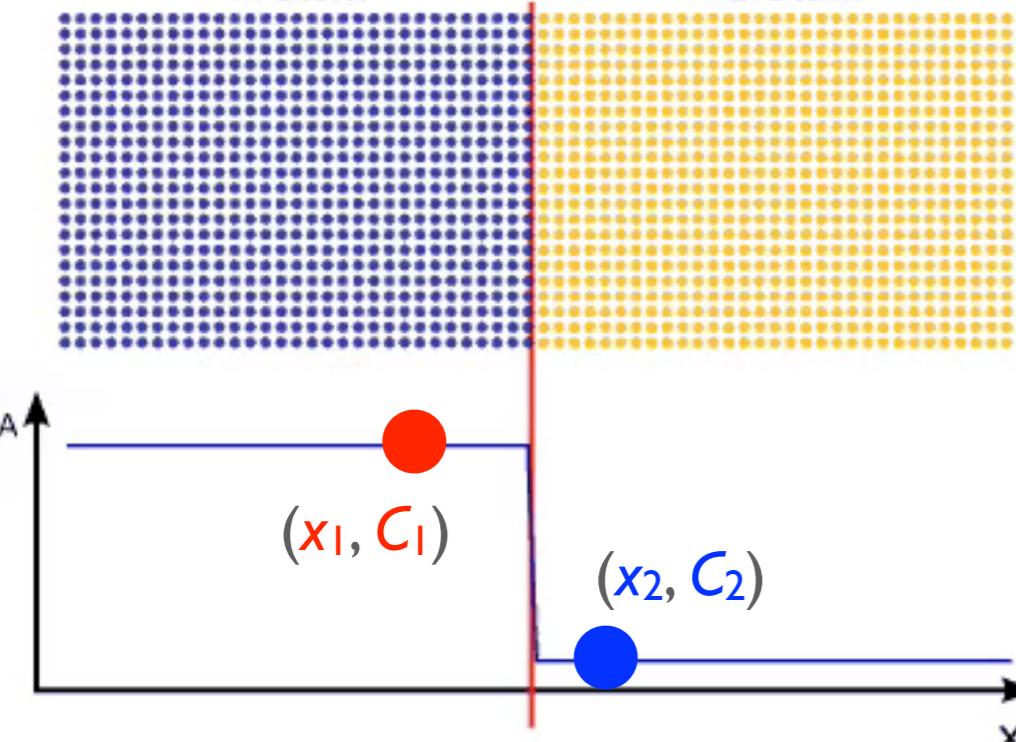
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$$\frac{\Delta C}{\Delta t} = - \frac{\Delta q}{\Delta x}$$

where **t** is time



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$$\frac{\Delta C}{\Delta t} = - \frac{\Delta q}{\Delta x} \leftarrow \text{Conservation of mass/energy}$$

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A mathematical definition

$$\frac{\Delta C}{\Delta t} = - \frac{\Delta q}{\Delta x}$$

- So, how is this a conservation of mass/energy equation?



A mathematical definition

$$\frac{\Delta C}{\Delta t} = - \frac{\Delta q}{\Delta x}$$

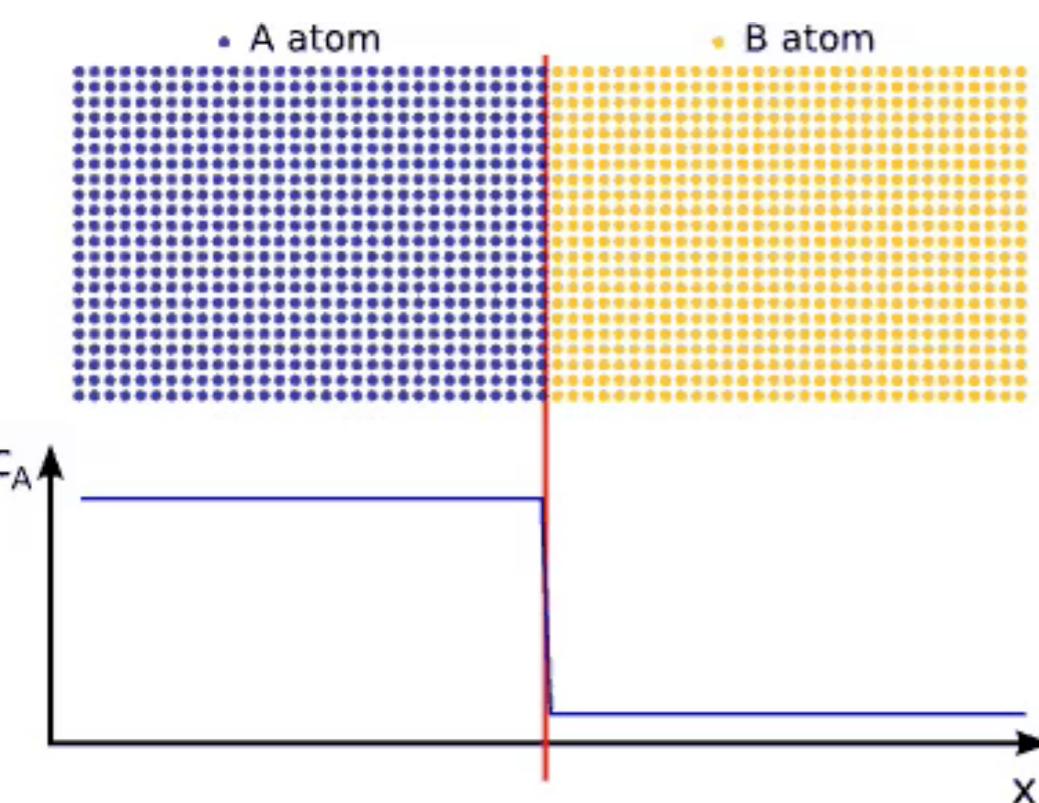
- So, how is this a conservation of mass/energy equation?

$$\frac{\Delta C}{\Delta t} = - \frac{q_2 - q_1}{x_2 - x_1}$$

- Consider the fluxes q_1 and q_2 at two points, x_1 and x_2
- What happens when the flux of mass q_2 at x_2 is larger than the flux q_1 at x_1 ?



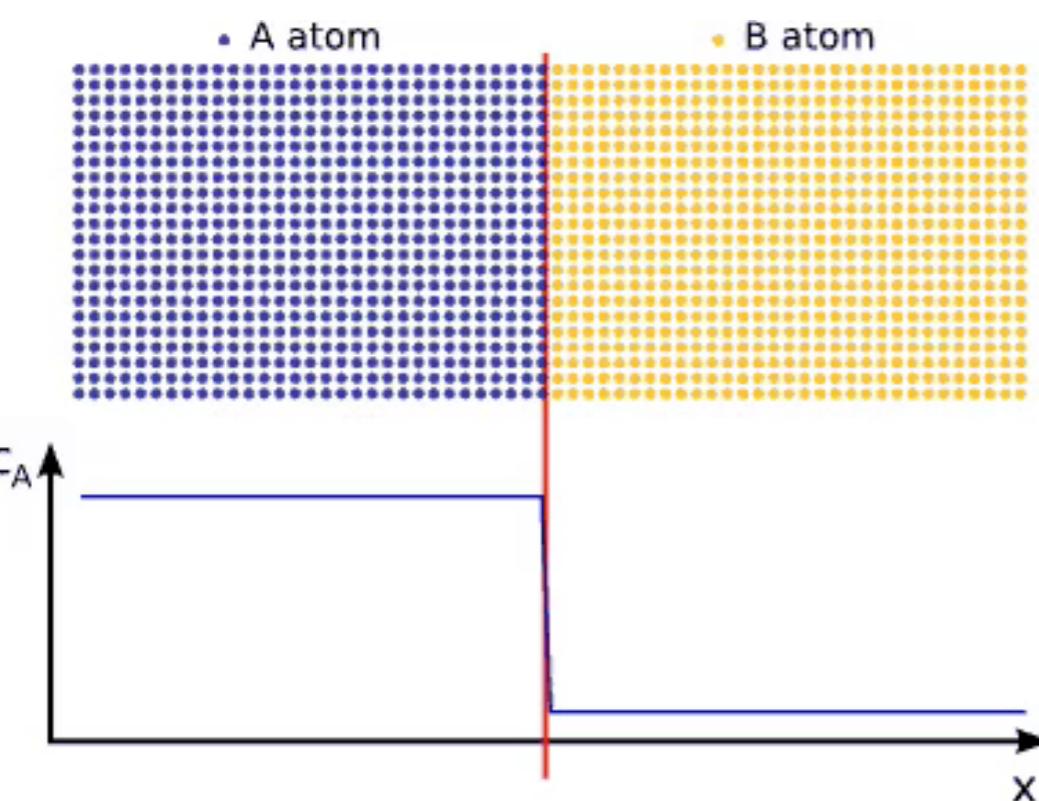
A mathematical definition



- If we again replace the finite changes Δ with infinitesimal changes ∂ , we can describe our example on the left
$$\frac{\partial C_A}{\partial t} = -\frac{\partial q}{\partial x}$$
- Essentially, all this says is that the concentration of A will change based on the flux across a reference face at position x minus the flux across a reference face at position $x + dx$



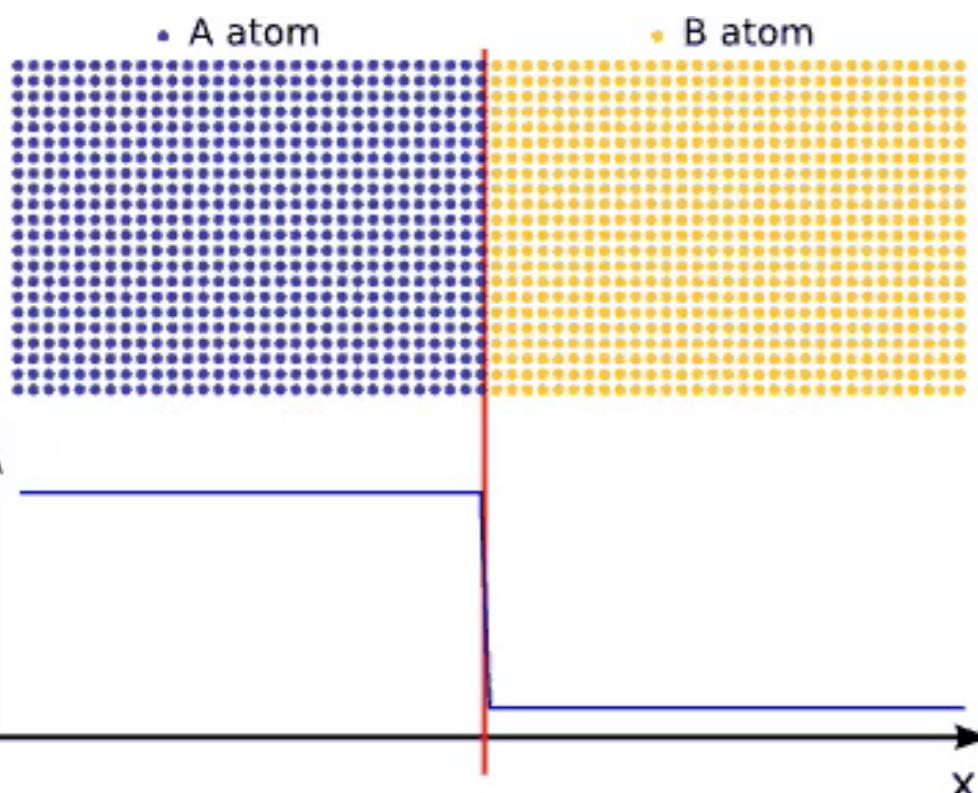
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A mathematical definition



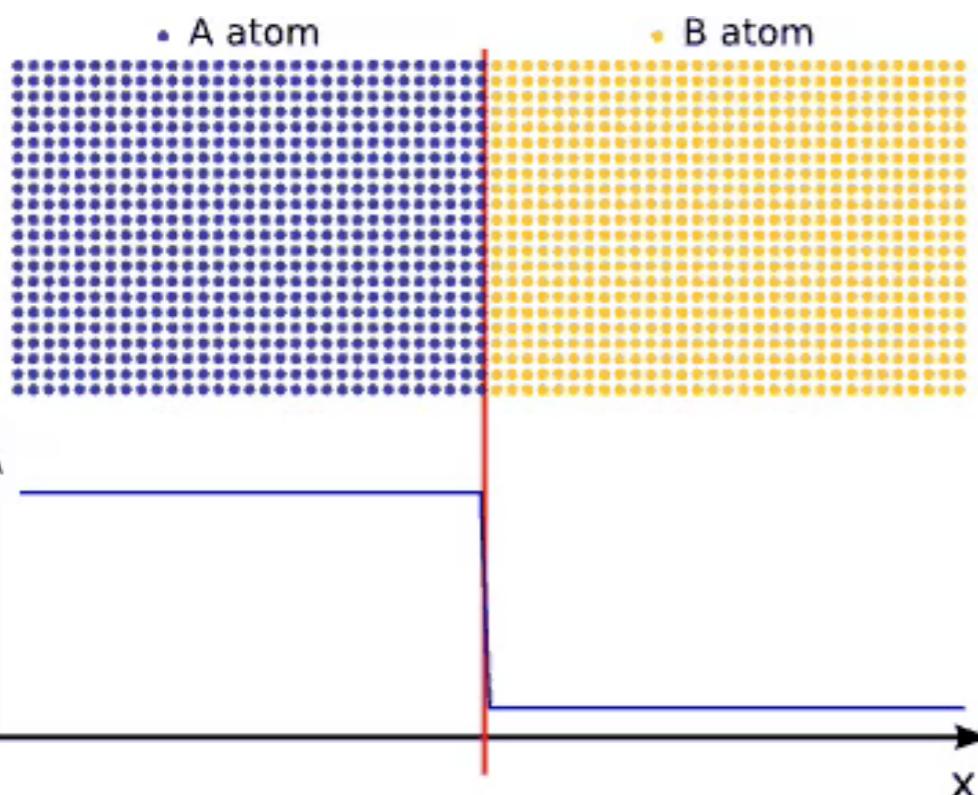
- In part 2 of today's lectures (Lecture 6), we'll see how to combine the two equations we've just seen into the diffusion equation and how to solve that equation

$$q = -D \frac{\partial C_A}{\partial x} + \frac{\partial C_A}{\partial t} = -\frac{\partial q}{\partial x} = \text{diffusion}$$

Note: This is not the actual mathematical relationship for diffusion



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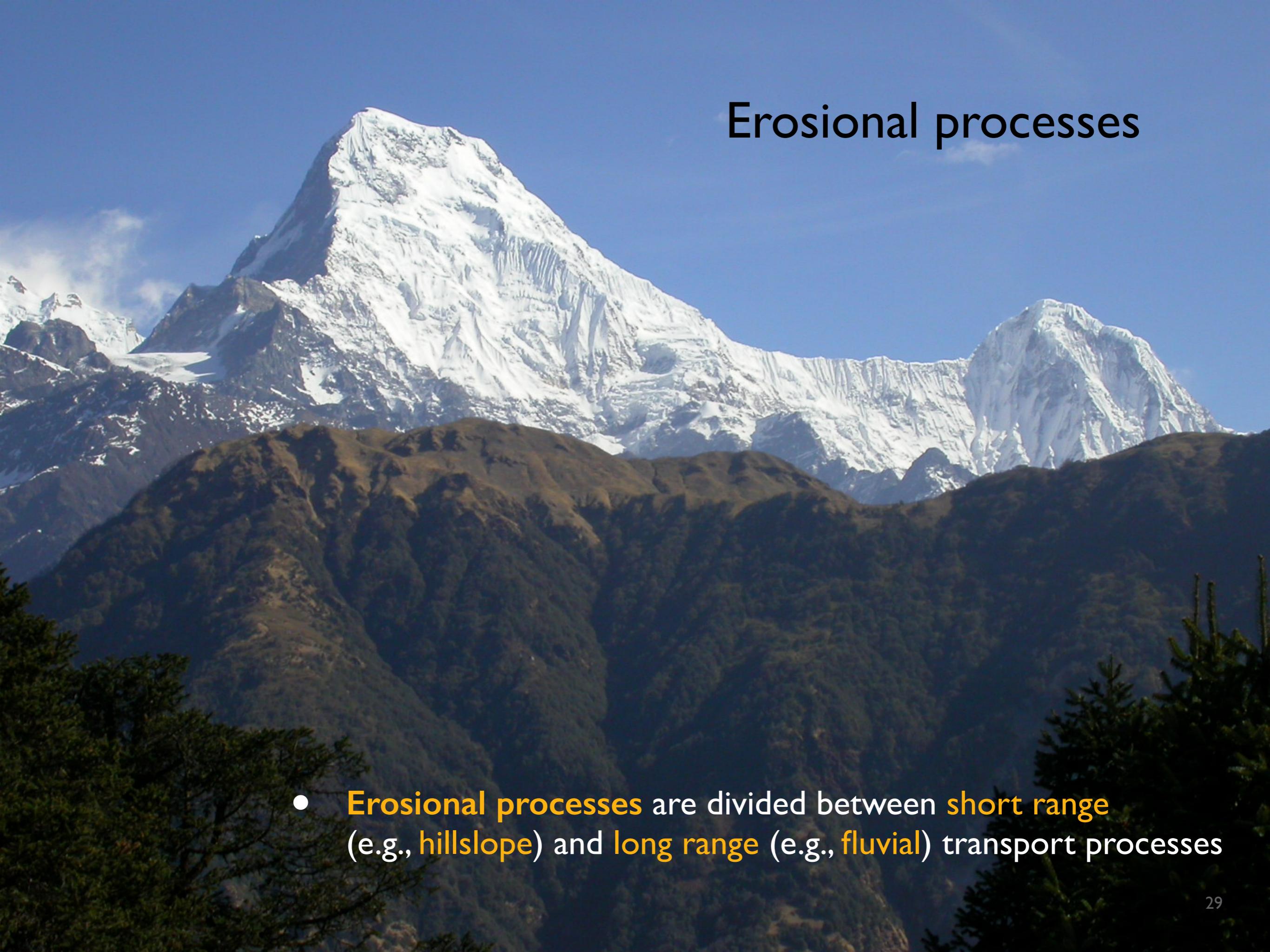
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General concepts of diffusion

- So our definitions of diffusion to this point are OK for true diffusion processes, but there are also numerous geological processes that are not themselves diffusion processes, but result in diffusion-like behaviour
- **Hillslope diffusion** is a name given to the overall behaviour of various surface processes that transfer mass on hillslopes in a diffusion-like manner

A scenic view of a mountain range under a clear blue sky. In the foreground, dark green hills are visible. Behind them, majestic mountains rise, their peaks and ridges covered in white snow. The sky is a vibrant blue with a few wispy clouds.

Erosional processes

- **Erosional processes** are divided between **short range** (e.g., **hillslope**) and **long range** (e.g., **fluvial**) transport processes



Hillslope processes

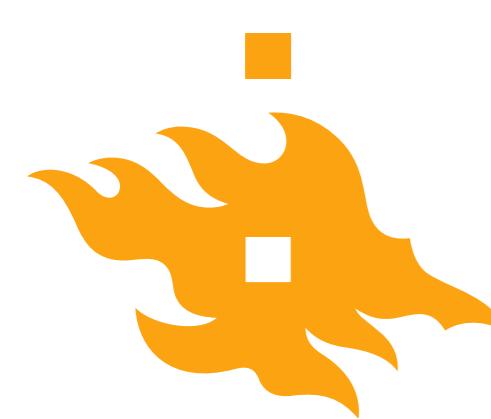
- **Hillslope processes** comprise the different types of mass movements that occur on hillslopes
- **Slides** refer to cohesive blocks of material moving on a well-defined surface of sliding
- **Flows** move entirely by differential shearing within the transported mass with no clear plane at the base of the flow
- **Heave** results from disrupting forces acting perpendicular to the ground surface by expansion of the material



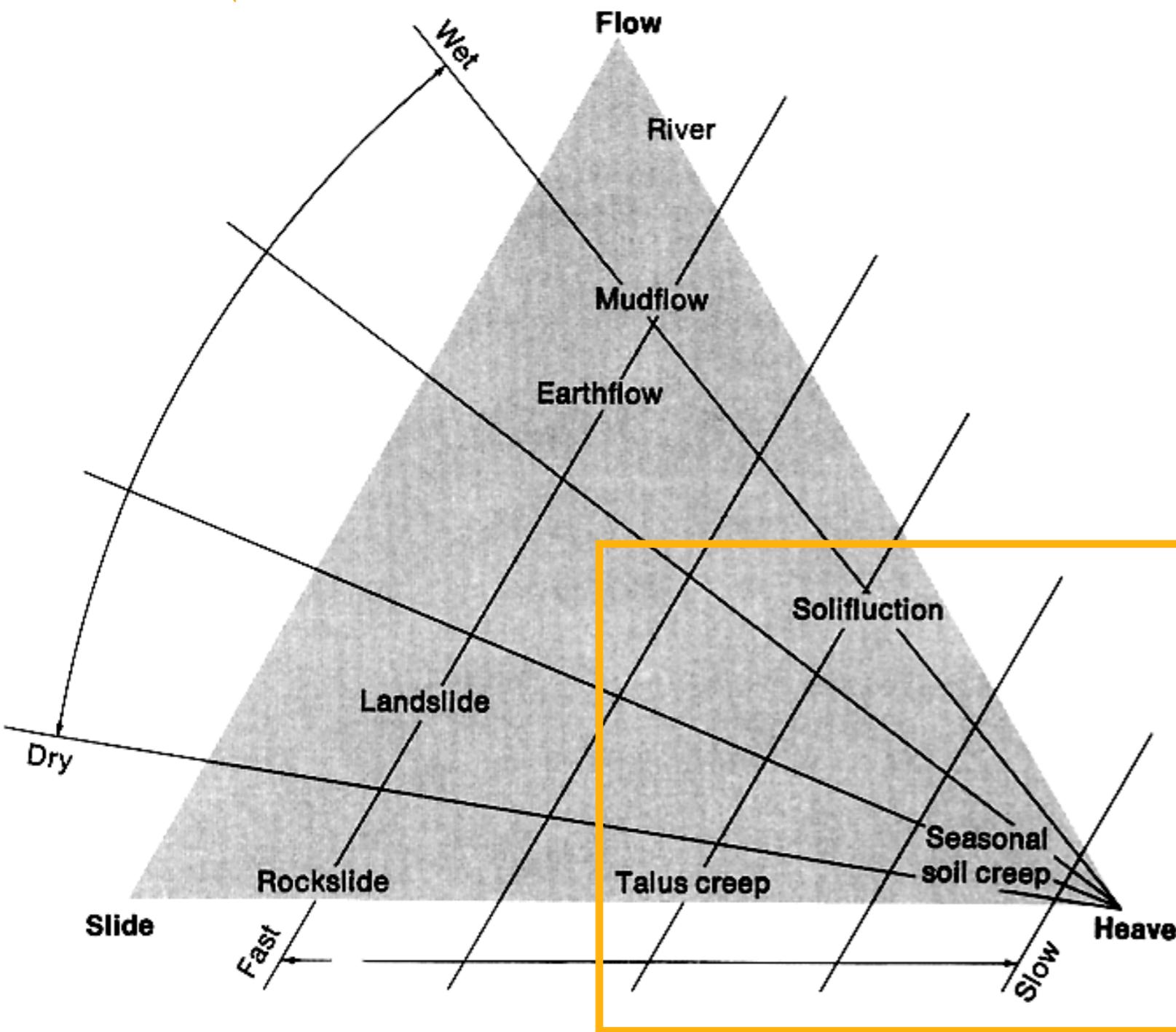
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Our focus



Mass movement processes



- Creep is almost too slow to monitor



Heave and creep

- **Creep:** The extremely slow movement of material in response to gravity
- **Heave:** The vertical movement of unconsolidated particles in response to expansion and contraction, resulting in a net downslope movement on even the slightest slopes
- **Seasonal creep or soil creep** is periodically aided by heaving



Heave and creep

Nearly vertical
Romney shale
displaced by
seasonal creep

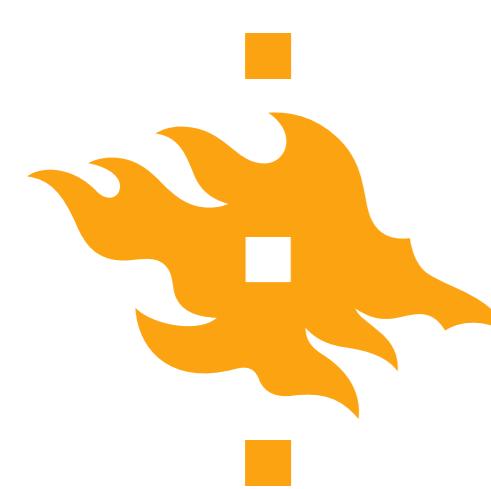




Heave and creep



Fig. 4.29, Ritter et al., 2002



How does heaving work?

- Near-surface material moves perpendicular to the surface during **expansion (E)**
- Expansion can result from swelling or freezing
- In theory, particles settle vertically downward during **contraction (C)**
- In reality, particle settling is not vertical, but follows a path closer to **D**

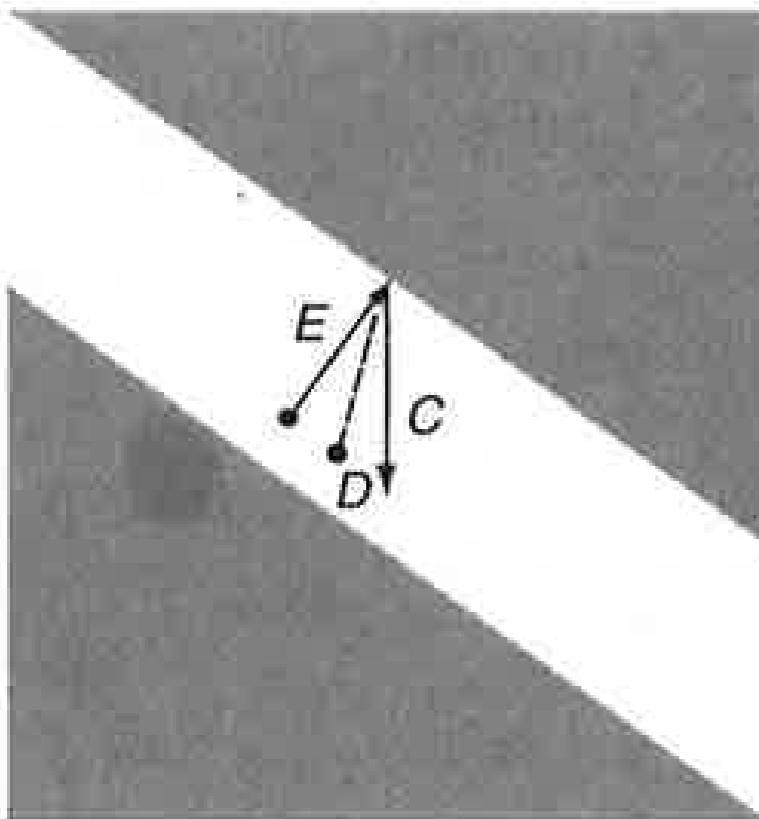


Fig. 4.30, Ritter et al., 2002



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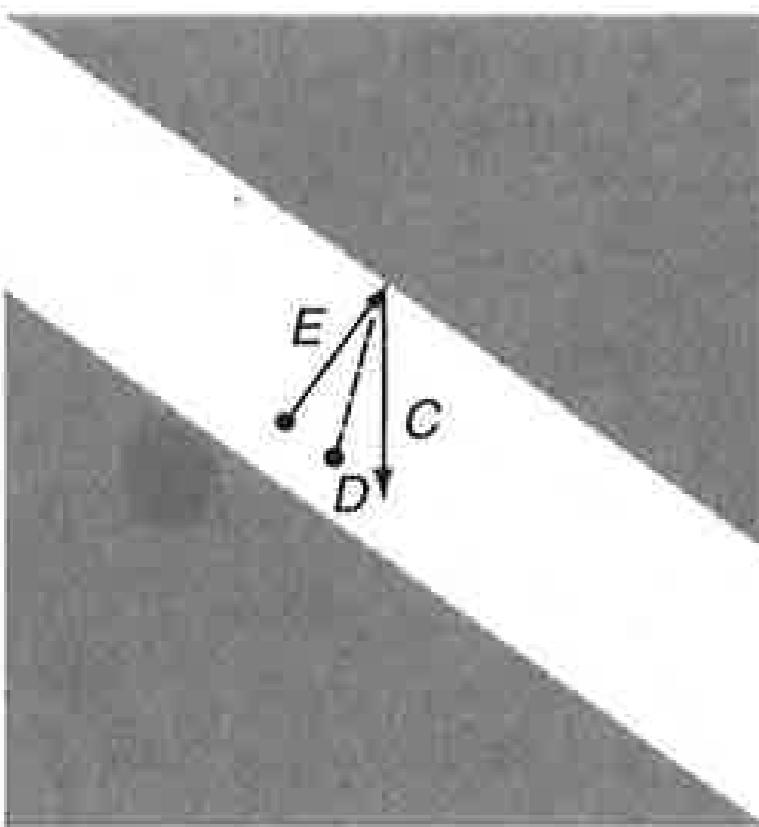
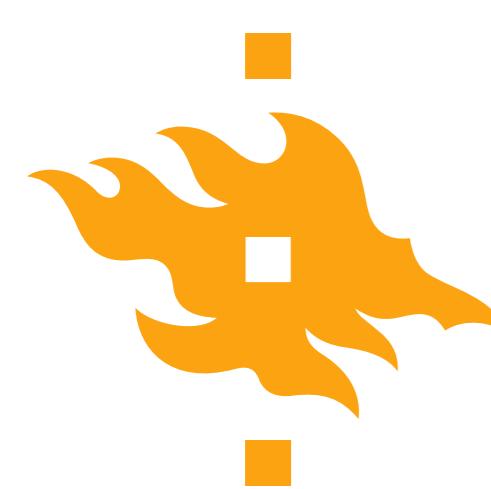


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- Based on this concept, what do you think will influence the rates of creep?



How does heaving work?

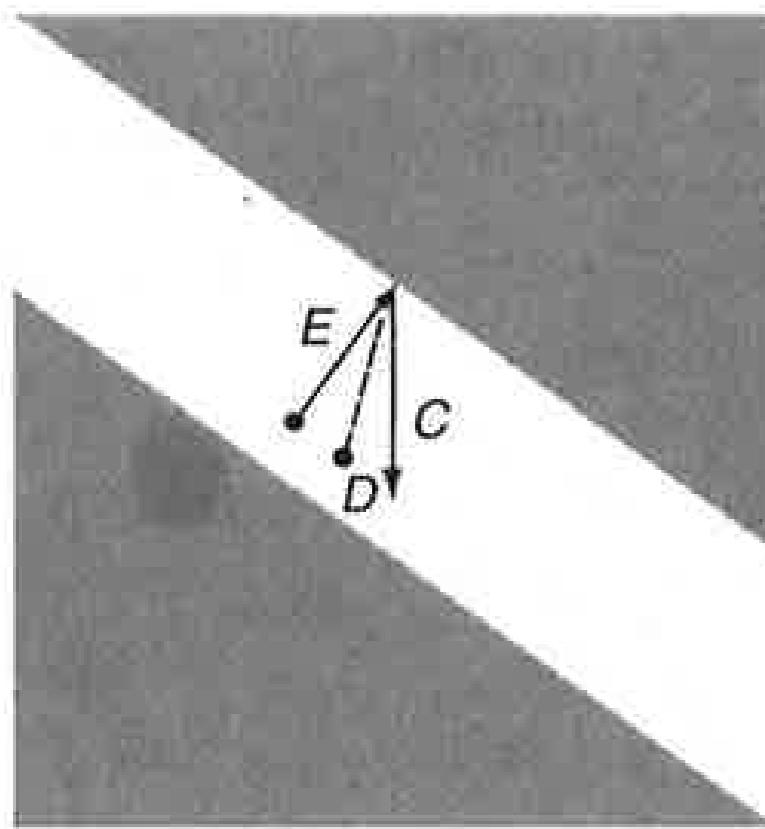


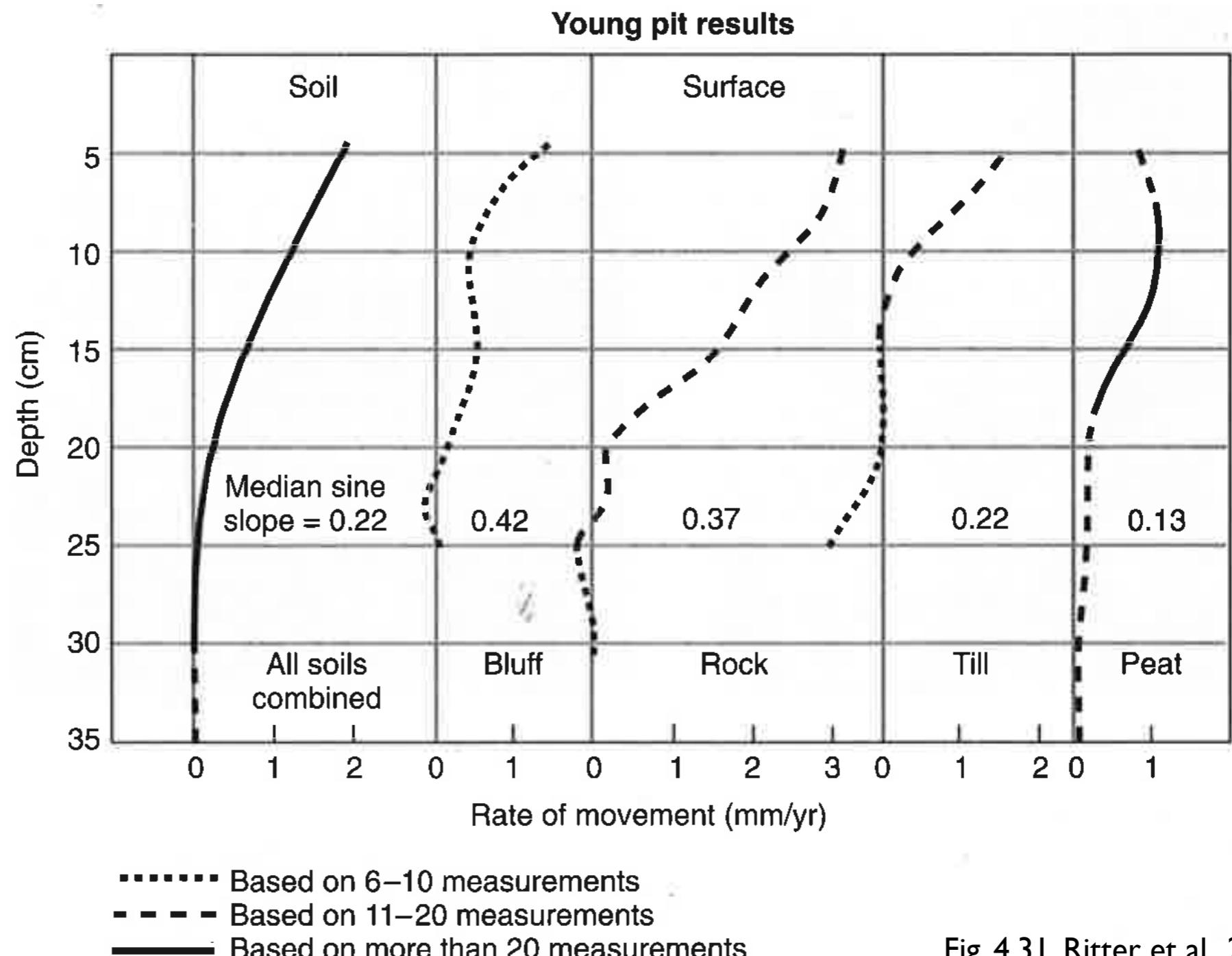
Fig. 4.30, Ritter et al., 2002

- Near-surface material moves perpendicular to the surface during **expansion (E)**
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- Based on this concept, what do you think will influence the rates of creep?
Slope angle, soil/regolith moisture, particle size/composition



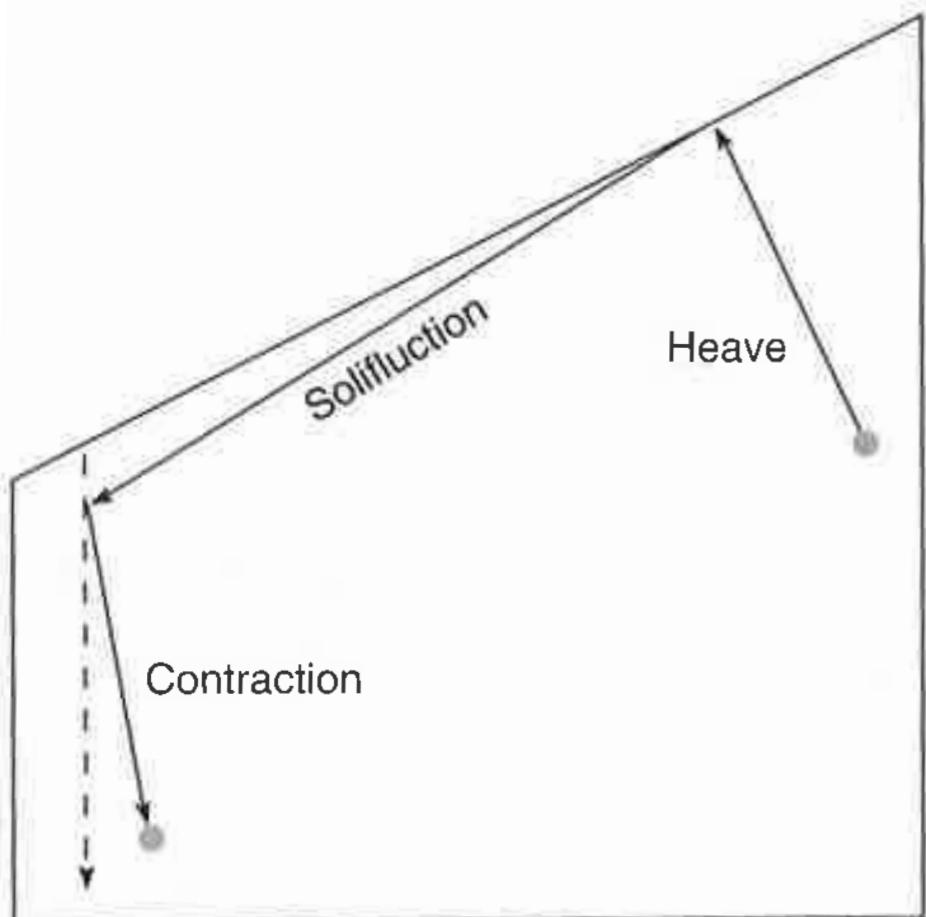
Mass movement by heaving

Creep rates
vary with
depth and soil/
regolith
composition





Frost creep and solifluction

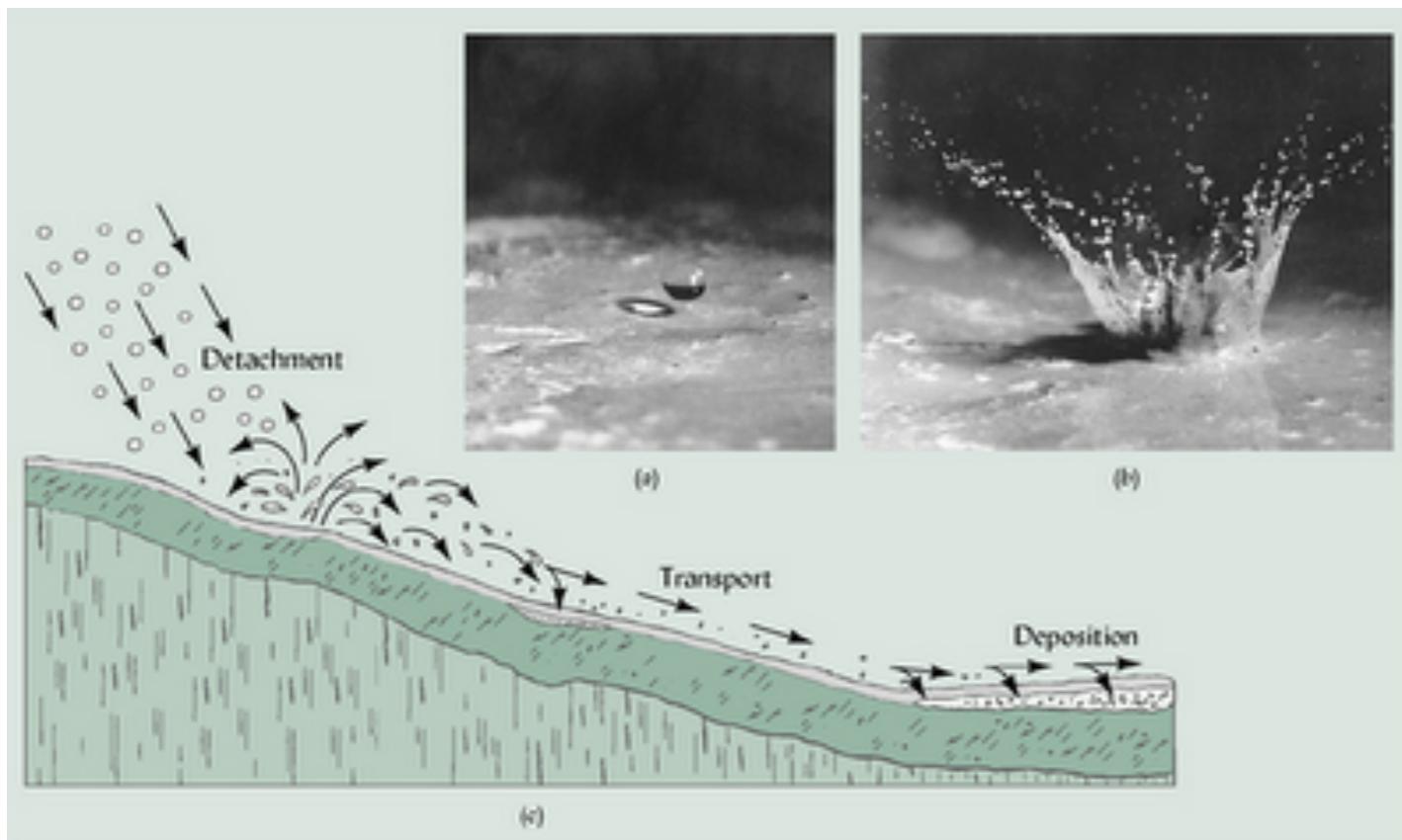


- **Solifluction** occurs in saturated soils, often in periglacial regions
 - In periglacial settings, **frost heave** leads to expansion of the near-surface material
 - During warm periods, saturated material at the surface flows downslope above the impermeable permafrost beneath

Fig. 11.14b, Ritter et al., 2002



Rain splash

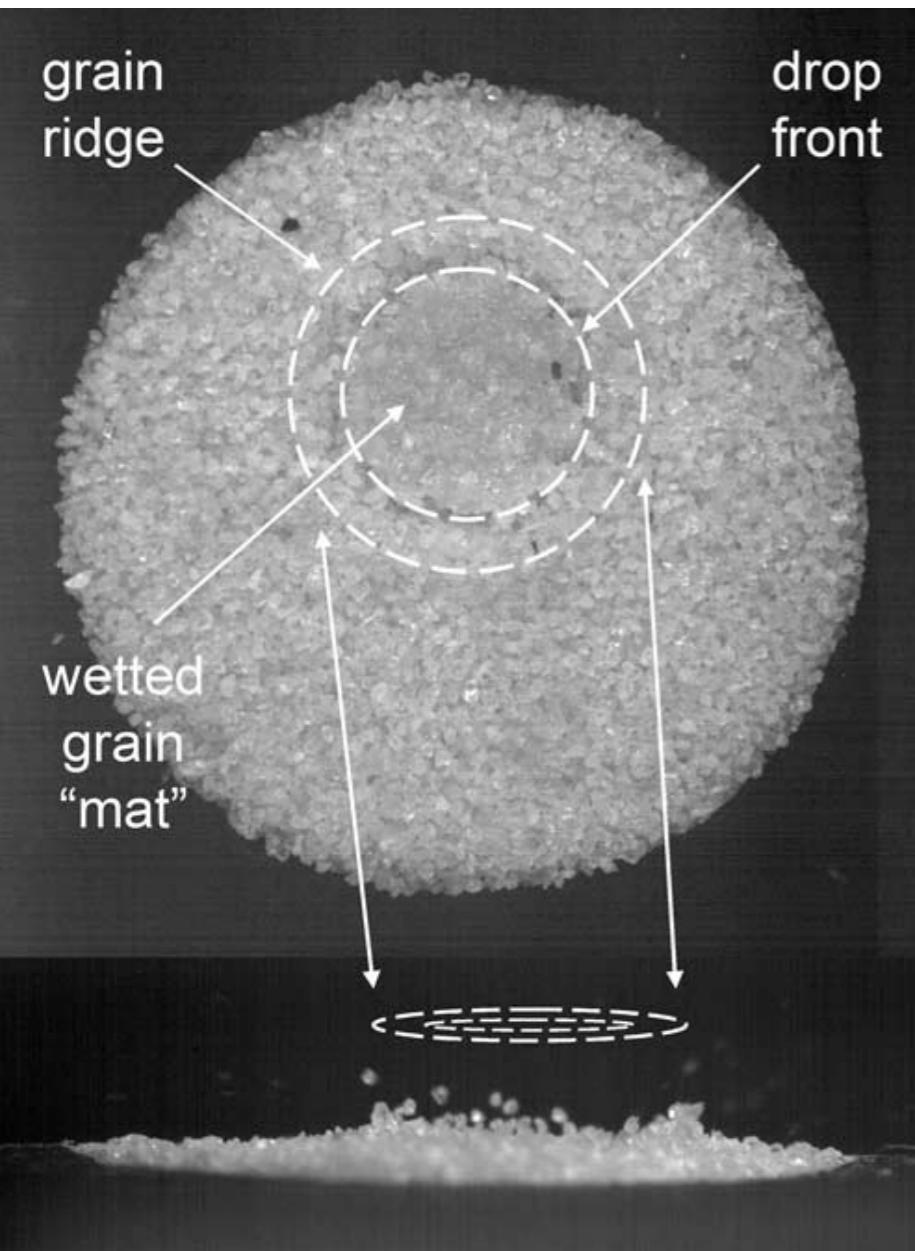


<http://geofaculty.uwyo.edu/neil/>

- Rain splash transport refers to the downslope drift of grains on a sloped surface as a result of displacement by raindrop impacts
- Although this process can produce significant downslope mass transport, it is generally less significant than heave

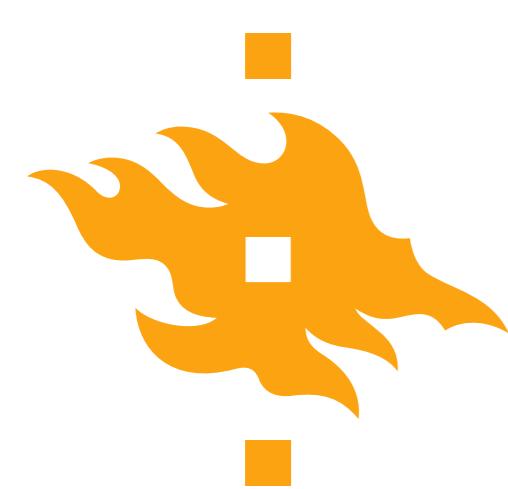


Studying rain splash

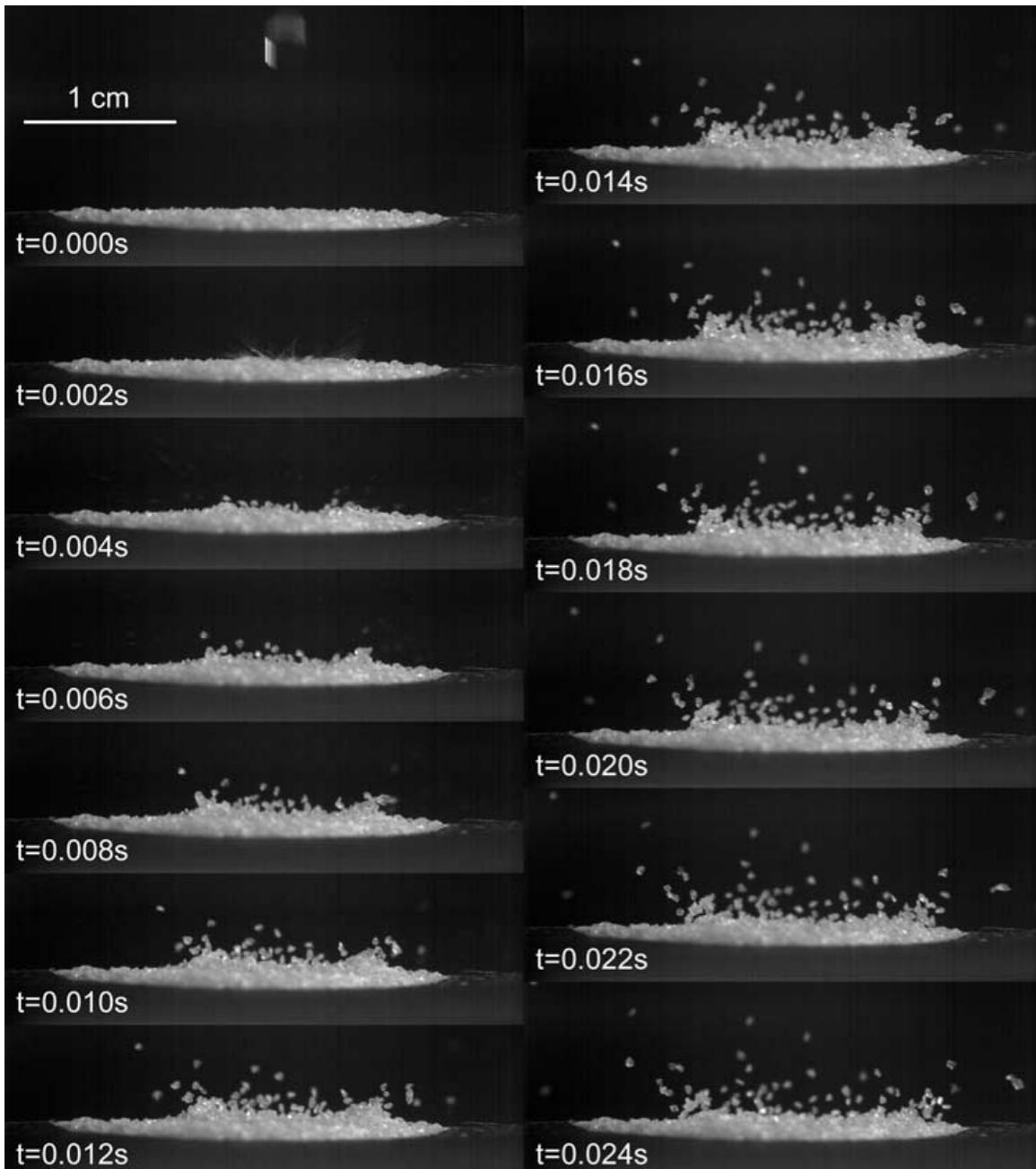


- Experimental setup:
 - “Rain drops” released from a syringe ~5 m above a dry sand target
 - Drops travel down a pipe to avoid interference by wind
 - Various drop sizes (2-4 mm), sand grain sizes (0.18 - 0.84 mm) and hillslope angles
 - High-speed camera used to capture raindrop impact and sand grain motion

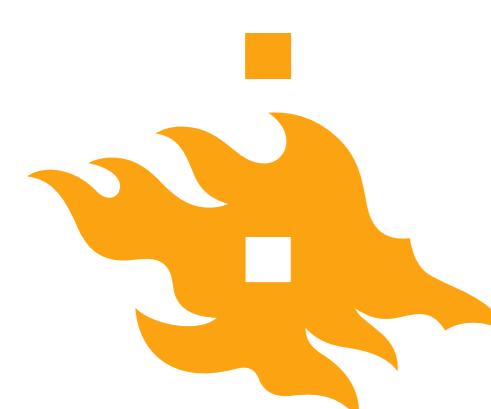
Furbish et al., 2007



Studying rain splash

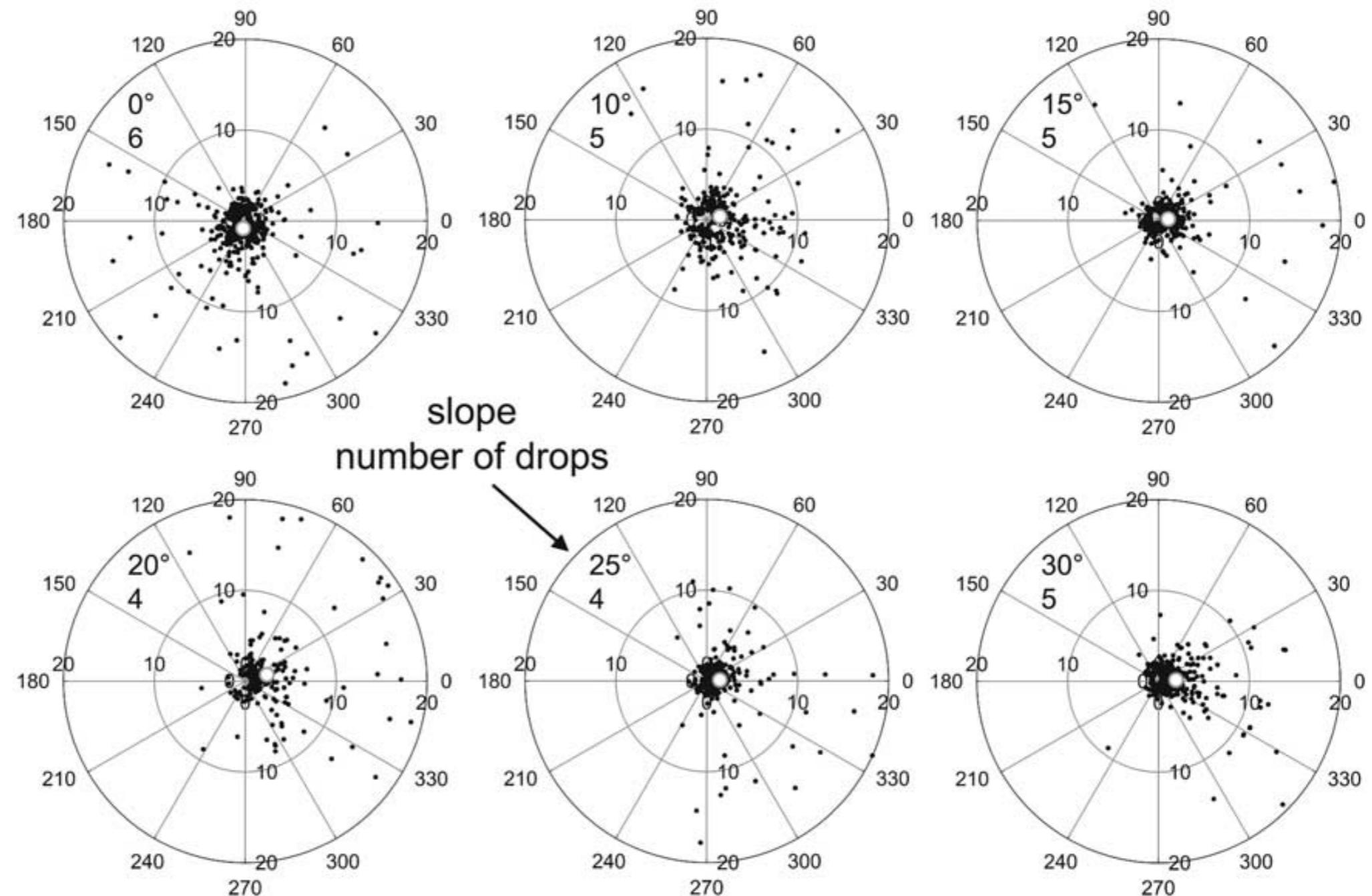


- Dry sand grains are displaced following raindrop impact
- Miniature bolide impacts (?)

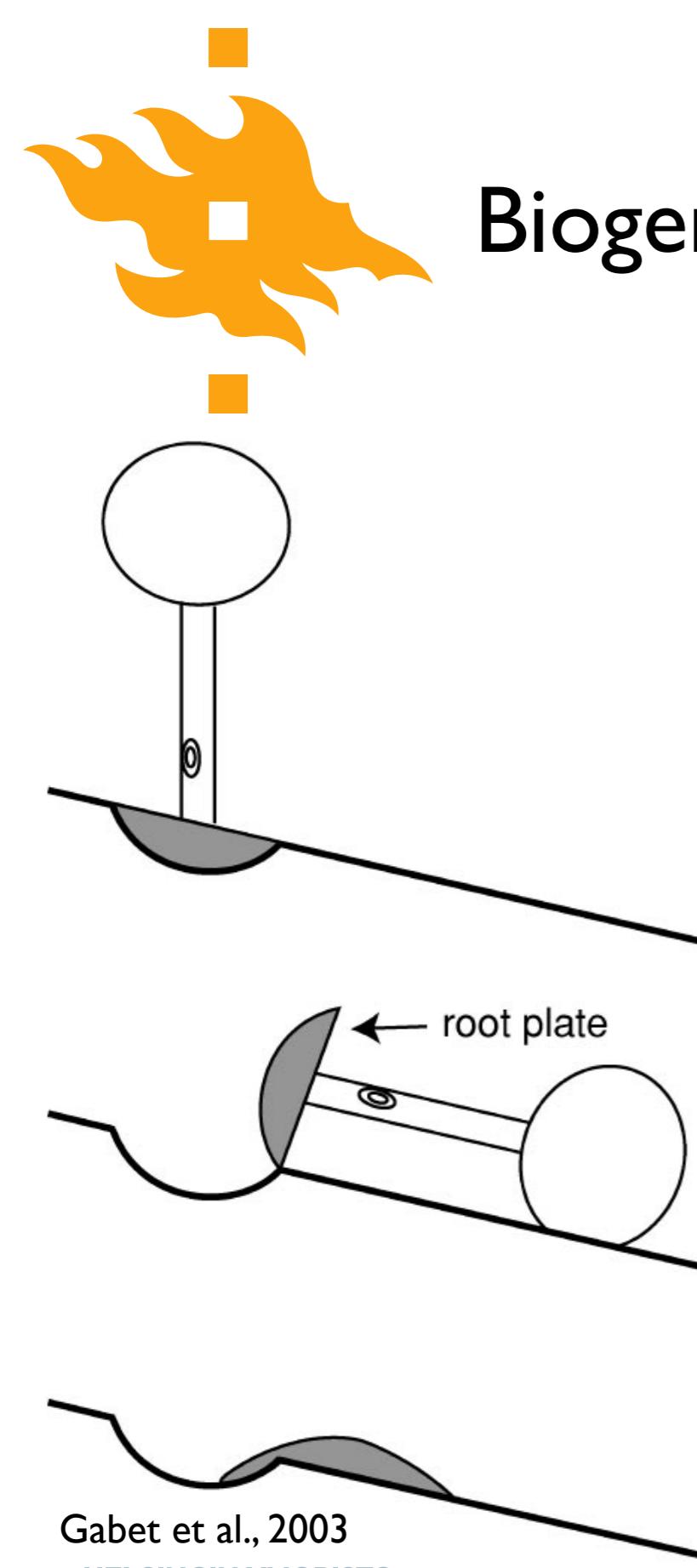


Studying rain splash

More particles drift downslope as slope angle increase



Furbish et al., 2007

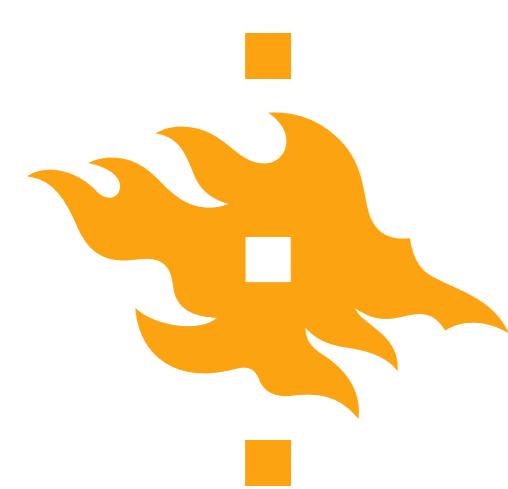


Biogenic transport: Tree throw

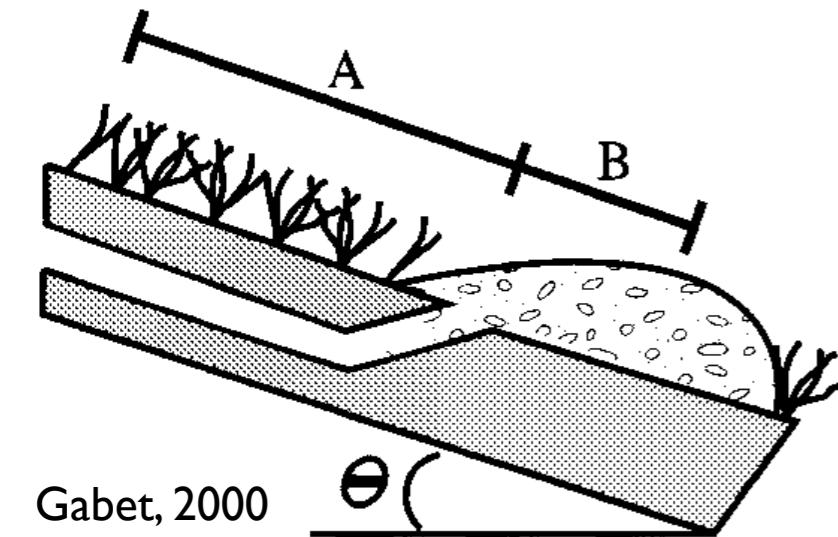
- Falling trees also displace sediment/soil and can produce downslope motion
- When trees fall, its root mass rotates soil and rock upward
- Gradually, this soil/rock falls down beneath the root mass as it decays

Gabet et al., 2003

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UNIVERSITY OF HELSINKI



Biogenic transport: Gopher holes



- Gophers dig underground tunnels parallel to the surface and displace sediment both under and above ground
- On slopes, this sediment is displaced downslope, resulting in mass movement
- Locally, this process can be the dominant mechanism for sediment transport



Common features

- What do all of these sediment transport processes have in common?
- In each case, the rate of transport is strongly dependent on the hillslope angle
 - Steeper slopes result in faster downslope transport
 - In other words, the flux of mass is proportional to the topographic gradient
- This suggests these erosional processes can be modelled as **diffusive**



Recap

- **What are the two components of diffusion processes?**
- **How does soil creep result in diffusion of soil or regolith?**
- **What are the main factors controlling the rate of hillslope diffusion?**



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References

- Furbish, D. J., Hamner, K. K., Schmeeckle, M., Borosund, M. N., & Mudd, S. M. (2007). Rain splash of dry sand revealed by high-speed imaging and sticky paper splash targets. *J. Geophys. Res.*, 112(F1), F01001. doi: 10.1029/2006JF000498
- Gabet, E. J. (2000). Gopher bioturbation: Field evidence for non-linear hillslope diffusion. *Earth Surface Processes and Landforms*, 25(13), 1419–1428.
- Gabet, E. J., Reichman, O. J., & Seabloom, E. W. (2003). THE EFFECTS OF BIOTURBATION ON SOIL PROCESSES AND SEDIMENT TRANSPORT. *Annual Review of Earth and Planetary Sciences*, 31(1), 249–273. doi:10.1146/annurev.earth.31.100901.141314
- Ritter, D. F., Kochel, R. C., & Miller, J. R. (2002). *Process Geomorphology* (4 ed.). McGraw-Hill Higher Education.
- Shuster, D. L., Flowers, R. M., & Farley, K. A. (2006). The influence of natural radiation damage on helium diffusion kinetics in apatite. *Earth and Planetary Science Letters*, 249(3-4), 148–161.