

## Class overview today - April 25, 2016

- Part I Basic concepts of thermochronology
  - Basic concepts of thermochronology
  - Closure temperature concept
  - Estimating closure temperatures
- Part II Low-temperature thermochronology
  - Definition of low-temperature thermochronology
  - Three common types of low-temperature thermochronometers



### Introduction to Quantitative Geology Lecture 12 Low-temperature thermochronology

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#### Goals of this lecture

Define low-temperature thermochronology

- Introduce three common types of low-temperature thermochronometers
  - Helium dating (The (U-Th)/He method)
  - Fission-track dating (The FT method)
  - Argon dating (The <sup>40</sup>Ar/<sup>39</sup>Ar method)

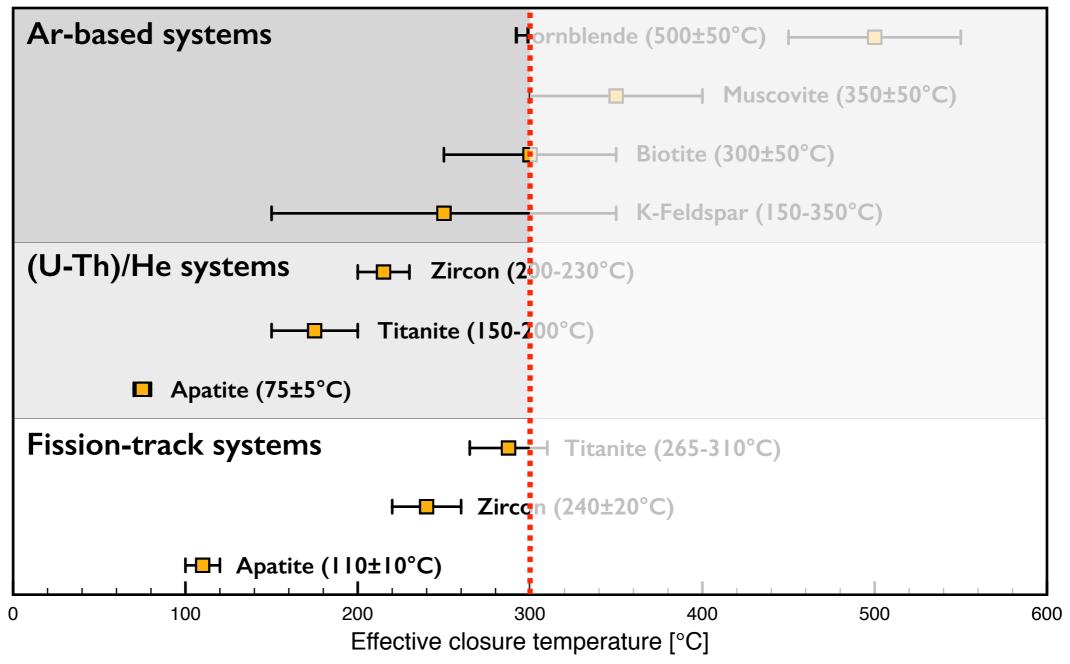


## What is low-temperature thermochronology?

 Low-T thermochronology uses thermochronometers with effective closure temperatures below ~300°C



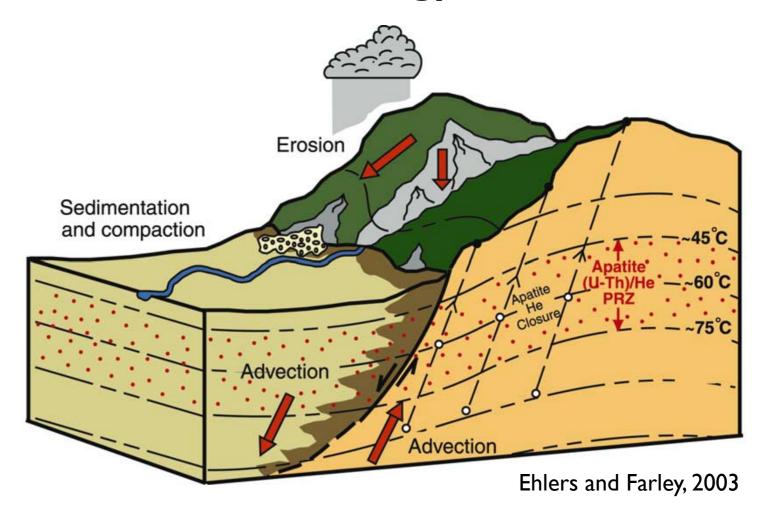
# What is low-temperature thermochronology?



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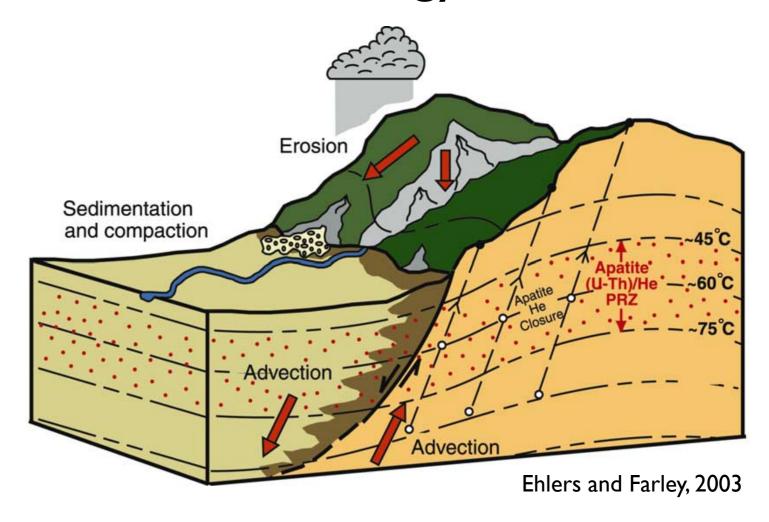
#### Why is thermochronology useful?



- Thermochronometer ages provide a constraint on the time-temperature history of a rock sample
  - In many cases, the age is the time since the sample cooled below the system-specific effective closure temperature



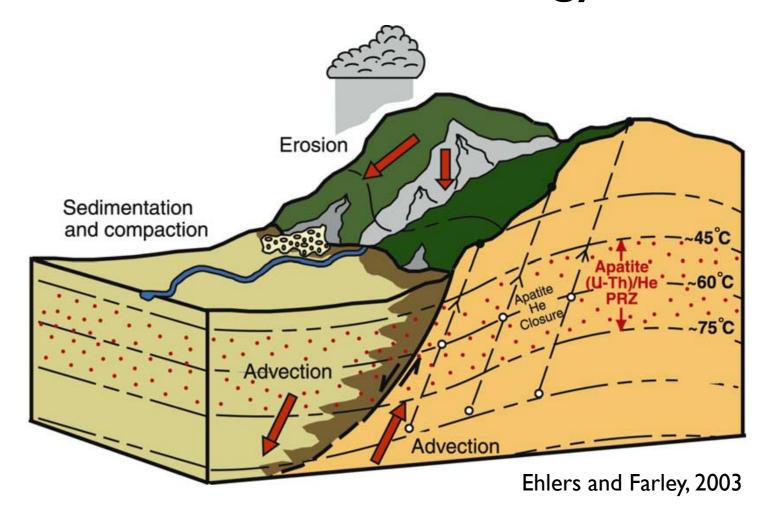
#### Why is thermochronology useful?



• Because the temperatures to which thermochronometers are sensitive generally occur at depths of I to >15 km and ages are typically I to 100's of Ma, they record long-term cooling through the upper part of the crust and can be used to calculate long-term average rates of tectonics and erosion



#### Why is *low-T* thermochronology useful?

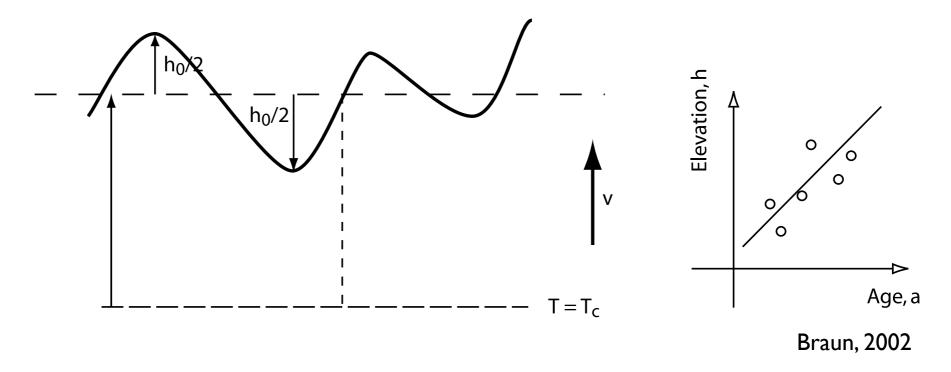


• Low-temperature thermochronometers are unique because of their increased sensitivity to topography, erosional and tectonic processes



#### High temperature = no topography sensitivity

#### (a) High T<sub>C</sub> thermochronometers

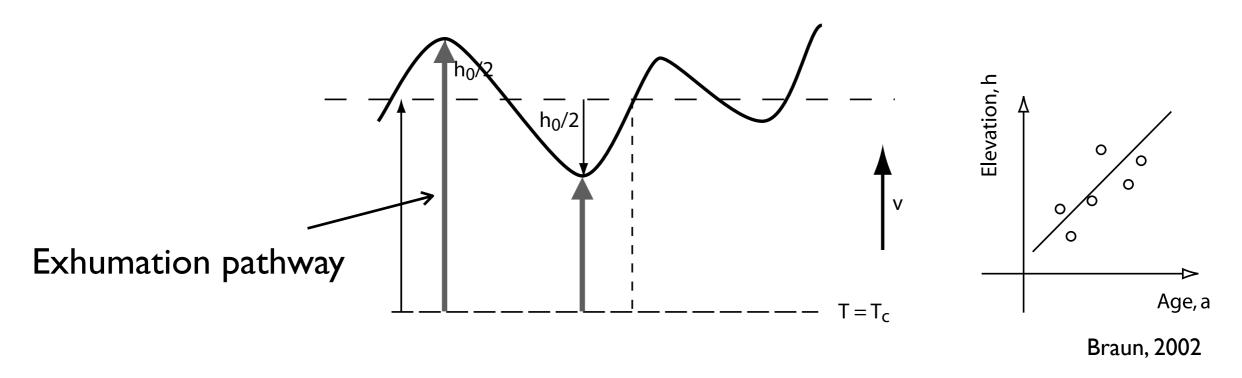


- For thermochronometers with a high effective closure temperature, the closure temperature isotherm will not be influenced by surface topography
  - Note that age will increase with elevation as a result of the topography



#### High temperature = no topography sensitivity

#### (a) High T<sub>C</sub> thermochronometers

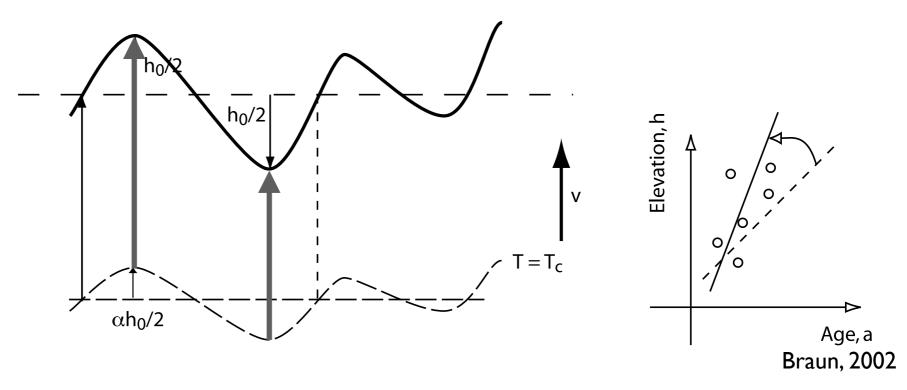


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#### Low-temperature = sensitive to topography

#### (b) Low T<sub>C</sub> thermochronometry

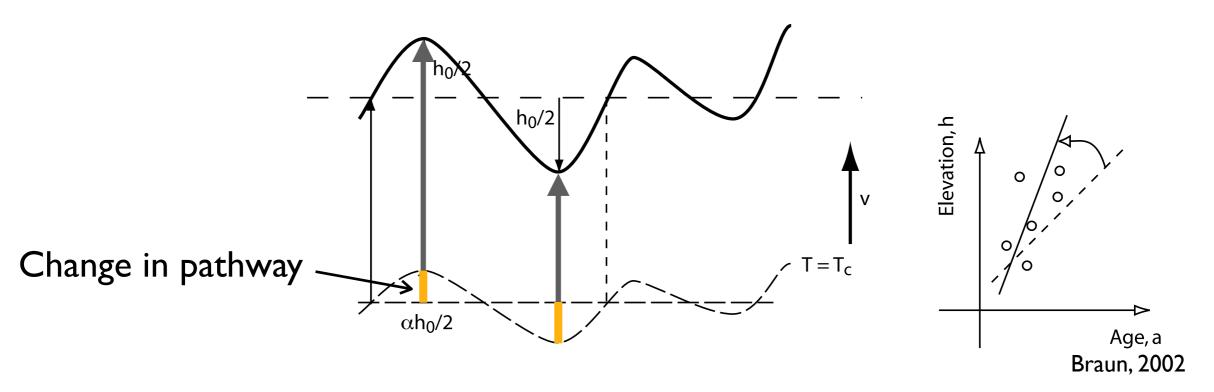


- The effective closure temperature isotherm for lowtemperature thermochronometers will generally be "bent" by the surface topography, changing the age-elevation trend
  - The lower the value of T<sub>c</sub>, the more its geometry will resemble the surface topography



#### Low-temperature = sensitive to topography

(b) Low T<sub>C</sub> thermochronometry



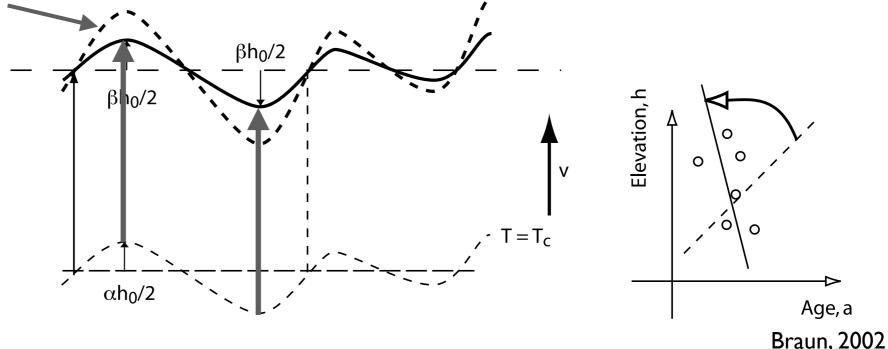
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#### Sensitivity to changing topography

(c) Low Tc thermochronometry + Relief change

Past topography



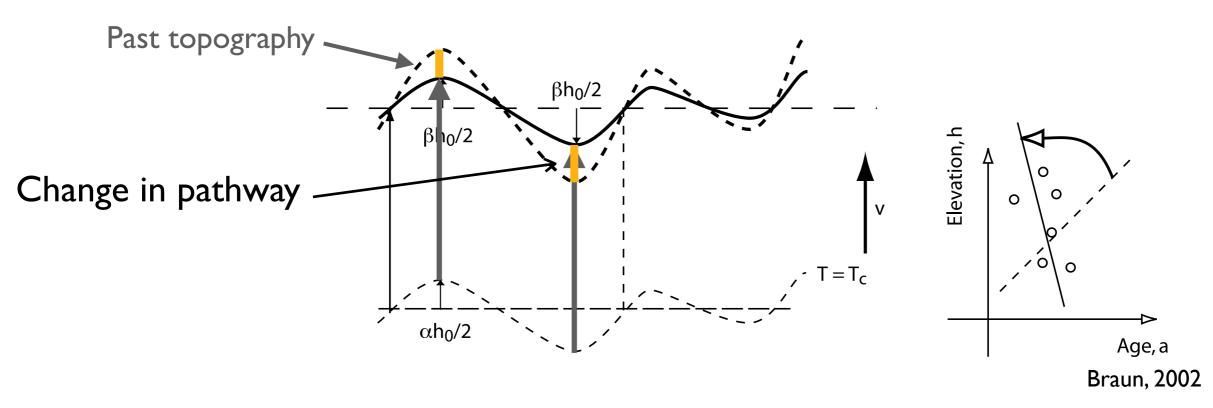
- Because  $T_c$  is sensitive to topography for low-temperature thermochronometers, it is possible to record changes in topography in the past (!)
  - Here, topographic relief decreases and the age-elevation trend gets inverted (older at low elevation)



#### Sensitivity to changing topography

(c) Low Tc thermochronometry + Relief change

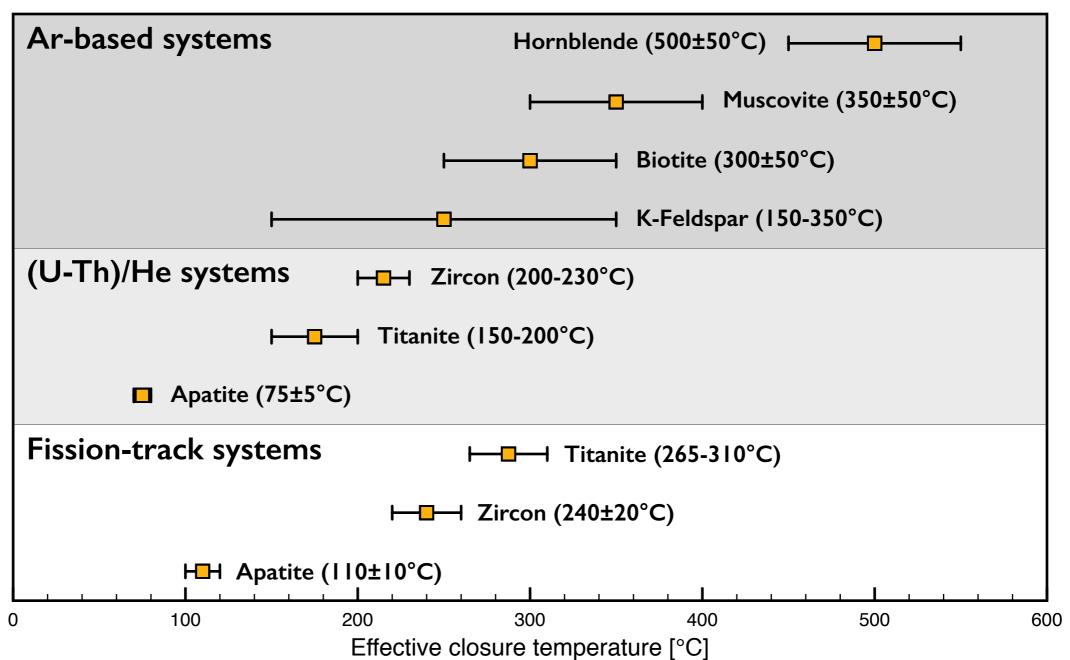
Intro to Quantitative Geology



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#### Common thermochronometers





Production of alpha particles

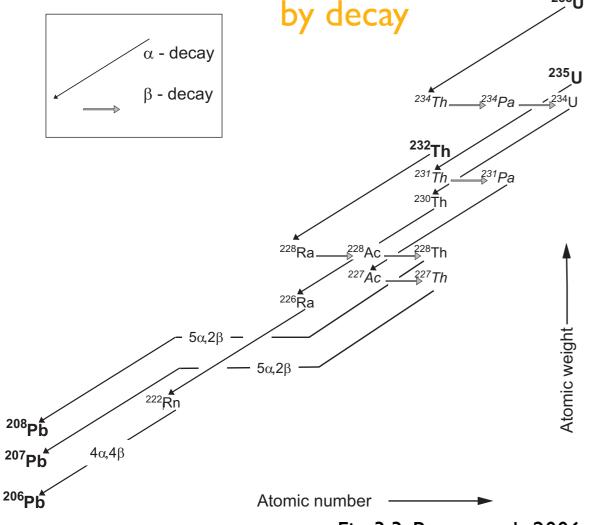


Fig. 3.3, Braun et al., 2006

- (U-Th)/He thermochronology is based on the production and accumulation of <sup>4</sup>He from parent isotopes <sup>238</sup>U, <sup>235</sup>U,
   <sup>232</sup>Th and <sup>147</sup>Sm
- ${}^{4}$ He ( $\alpha$  particles) produced during decay chains
  - $^{238}U$   $8~\alpha$  decays
  - $^{235}U$  7  $\alpha$  decays
  - $^{232}$ Th 6  $\alpha$  decays
  - $^{147}$ Sm  $^{1}\alpha$  decay



Production of alpha particles

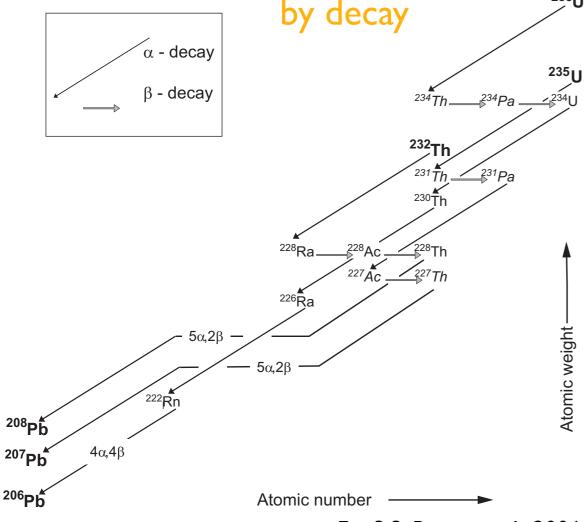


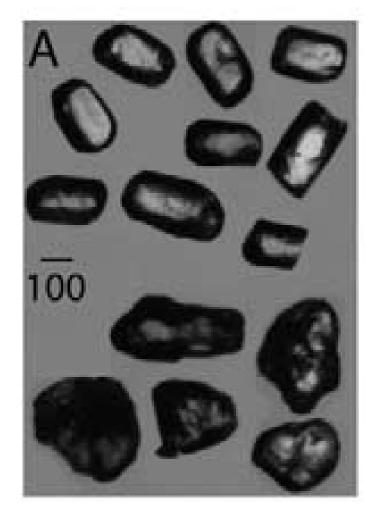
Fig. 3.3, Braun et al., 2006

• Ignoring the contribution of <sup>147</sup>Sm, we can say that the production of <sup>4</sup>He is

<sup>4</sup>He = 
$$8 \times {}^{238}\text{U}(e^{\lambda_{238}t} - 1)$$
  
+  $7 \times \frac{{}^{238}\text{U}}{137.88}(e^{\lambda_{235}t} - 1)$   
+  $6 \times {}^{232}\text{Th}(e^{\lambda_{232}t} - 1)$ 

where  ${}^4\text{He}$ ,  ${}^{238}\text{U}$  and  ${}^{232}\text{Th}$  are the present-day abundances of those isotopes, t is the He age and the  $\lambda$  values are the decay constants





Ehlers and Farley, 2003

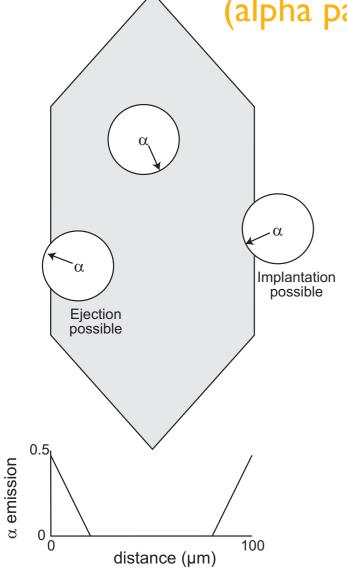
Nice, datable apatites

Not-so-nice apatites

Ages are calculated by measuring the <sup>4</sup>He concentration by heating and degassing the mineral sample, then separately measuring the U and Th concentrations, for example by using an inductively coupled plasma mass spectrometer (ICP-MS)



Potential ejection of <sup>4</sup>He (alpha particles)



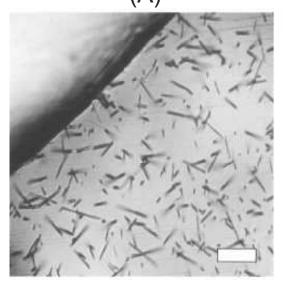
- Selected mineral grains for dating should be high-quality, euhedral minerals free of mineral inclusions with a prismatic crystal form
  - Why does the crystal form matter?
     Alpha particles travel ~20 µm when created and may be ejected from or injected to the sample crystal
  - We can correct for this!

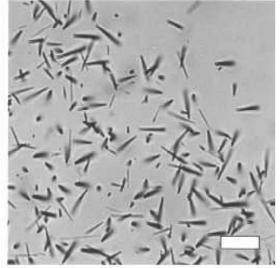
Fig. 3.4, Braun et al., 2006



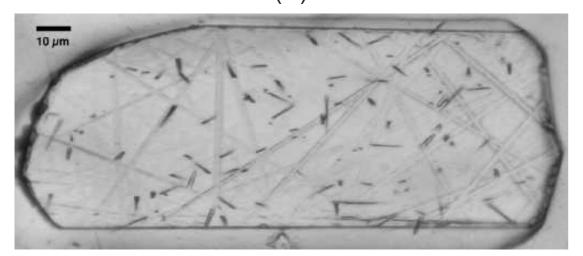
#### Fission-track dating - FT method

# Etched fission tracks in apatite (A) (B)





(C)

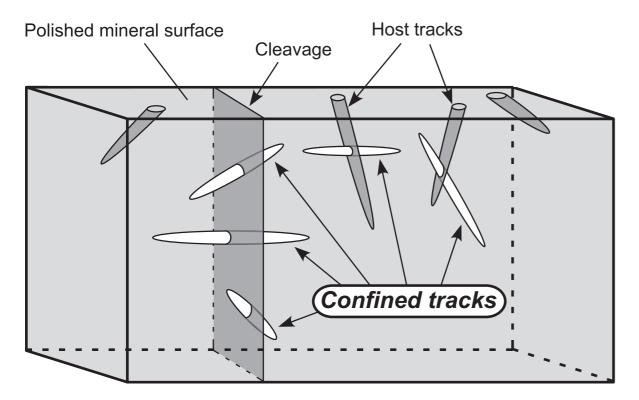


Tagami and O'Sullivan, 2005

- Fission-track dating is based on measuring the <u>accumulation of damage</u> <u>trails in a host crystal</u> as the result of spontaneous fission of <sup>238</sup>U
  - Fission splits the <sup>238</sup>U atom into two fragments that repel and damage the crystal lattice over the distance they travel
  - In apatite, fresh fission tracks are ~16
     μm long and ~11 μm long in zircon
- Similar to diffusive loss of <sup>4</sup>He, these damage trails will be repaired, or anneal, at temperatures above T<sub>c</sub>



#### Fission-track dating - FT method



Tagami and O'Sullivan, 2005

- To be visible under a microscope, tracks must be chemically etched and enlarged
- At this point, tracks can be manually (or automatically) counted to determine the track density
- The FT age can be calculated as

$$t = \frac{1}{\lambda_{\rm D}} \ln \left( \frac{\lambda_{\rm D}}{\lambda_{\rm f}} \frac{N_{\rm s}}{238 \,{\rm U}} + 1 \right)$$

where  $\lambda_D$  is the <sup>238</sup>U decay constant,  $\lambda_f$  is the fission decay constant,  $N_s$  is the number of spontaneous fission tracks in the sample and <sup>238</sup>U is the number of <sup>238</sup>U atoms



# Argon dating - 40Ar/39Ar method

- Argon dating is based on the decay of 40K to radiogenic 40Ar
  - Potassium is one of the most abundant elements in the crust, making argon dating one of the more common thermochronology methods
- 40Ar/39Ar dating is used on white micas, biotite, K-feldspar and amphiboles



## Argon dating - 40Ar/39Ar method

- 40Ar/39Ar ages are found by <u>irradiating a sample (and standard)</u> with fast neutrons, producing <sup>39</sup>Ar from <sup>39</sup>K in the sample
- The <sup>40</sup>Ar/<sup>39</sup>Ar ratio is then measured as samples are either degassed entirely or step heated (next slide)
- The <sup>40</sup>Ar/<sup>39</sup>Ar age can be calculated as

$$t = \frac{1}{\lambda} \ln \left( 1 + J \frac{^{40}\text{Ar}}{^{39}\text{Ar}} \right)$$

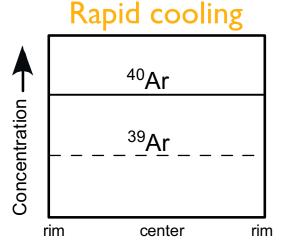
where  $\lambda$  is the decay constant of  $^{40}$ K,  $^{40}$ Ar/ $^{39}$ Ar is the measured sample  $^{40}$ Ar/ $^{39}$ Ar ratio and J is the irradiation factor

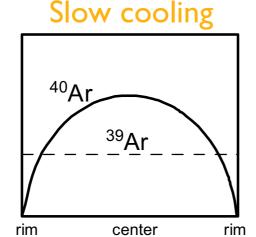
$$J = \frac{e^{\lambda t} - 1}{^{40}\text{Ar}/^{39}\text{Ar}}$$

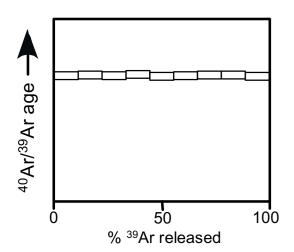
where t is a known age for a standard and  $^{40}$ Ar/ $^{39}$ Ar is its measured  $^{40}$ Ar/ $^{39}$ Ar ratio

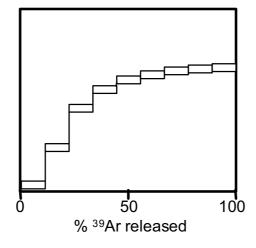


#### Argon dating - Step heating









Harrison and Zeitler, 2005

- Step heating of <sup>40</sup>Ar/<sup>39</sup>Ar samples involves stepwise heating of samples to gradually release Ar as the sample temperature increases
- With this, it is possible to see the <sup>40</sup>Ar distribution in the sample, which is a function of the sample cooling history



#### Argon dating - Step heating

#### <sup>40</sup>Ar/<sup>39</sup>Ar age spectra

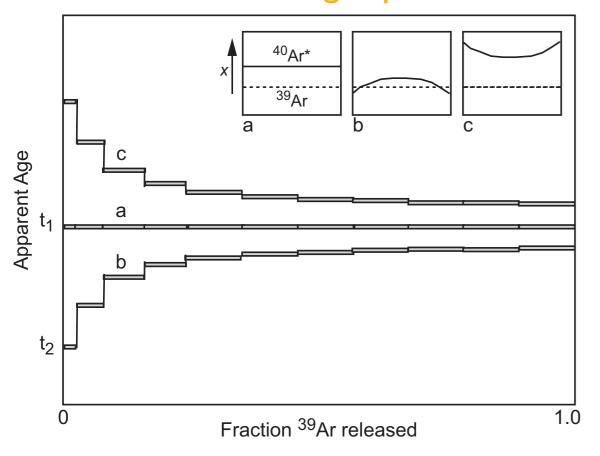
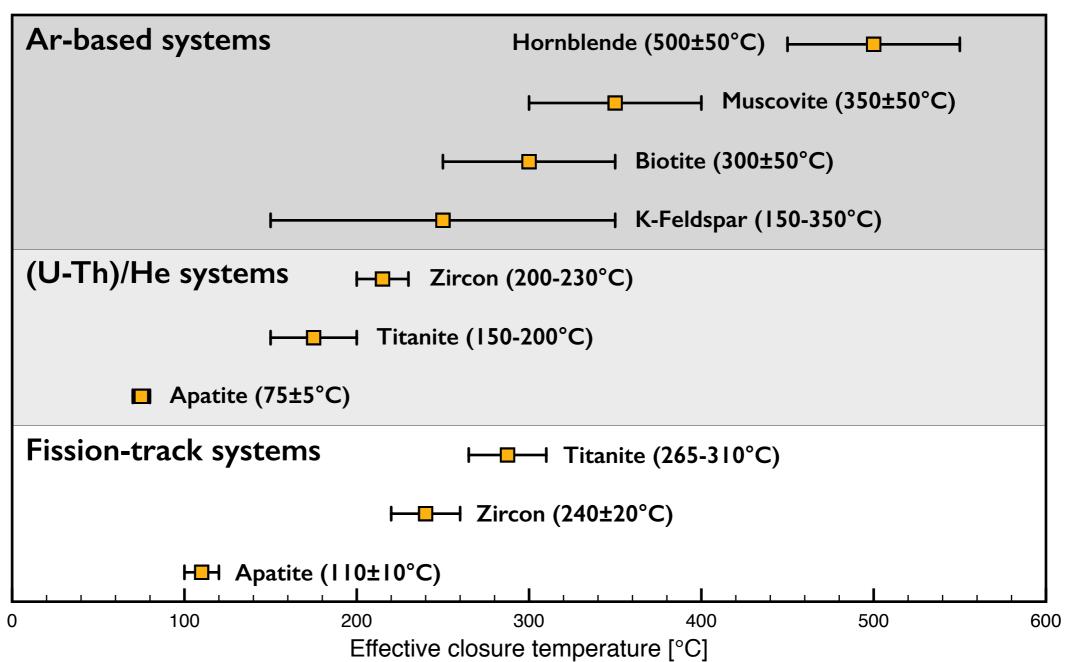


Fig. 3.1, Braun et al., 2006

- As we have seen on the previous slide,
  - (a) flat age spectra indicate rapid cooling of a rock sample (at time  $t_1$ , here)
  - (b) spectra with lower concentrations initially either indicate partial reheating of the sample at time t<sub>2</sub> or slow cooling from t<sub>1</sub> to t<sub>2</sub>
  - (c) an unexpected behavior with higher Ar concentrations initially (i.e., near the rim of the grain)!
  - This "excess" Ar may have been taken up from surrounding minerals



#### Common thermochronometers





 Why is low-temperature thermochronology a particularly interesting tool for those interested in geomorphology or active tectonics?

 How is are (U-Th)/He or <sup>40</sup>Ar/<sup>39</sup>Ar methods different from fission-track dating?



 Why is low-temperature thermochronology a particularly interesting tool for those interested in geomorphology or active tectonics?

 How is are (U-Th)/He or <sup>40</sup>Ar/<sup>39</sup>Ar methods different from fission-track dating?



#### Lab and final project primer

- The final two laboratory exercises will be based on thermochronology
  - The exercises will be <u>divided into two parts</u>, with the second exercise building on what you will have done the previous week
  - As usual, you will modify a Python code to produce some plots and provide short answers to some related questions
  - The questions you will answer for the write-ups for these two labs will be relatively simple, only to let me know that you were able to do the requested tasks, because...



#### Lab and final project primer

- ...you will expand on the work you do in the final two labs in a <u>formal written report</u>
- The report will be no longer than 8 typed pages (single spaced) including figures and references
- The idea is to describe some background on the data you will work with, the concept for its interpretation and your results/ conclusions
- The structure for the report will be described in detail on the final laboratory exercise handout



#### References

- Braun, J. (2002), Quantifying the effect of recent relief changes on age-elevation relationships, Earth and Planetary Science Letters, 200(3-4), 331–343.
- Braun, J., der Beek, van, P., & Batt, G. E. (2006). Quantitative Thermochronology. Cambridge University Press.
- Coutand, I., Whipp, D. M., Grujic, D., Bernet, M., Fellin, M. G., Bookhagen, B., et al. (2014). Geometry and kinematics of the Main Himalayan Thrust and Neogene crustal exhumation in the Bhutanese Himalaya derived from inversion of multithermochronologic data. *Journal of Geophysical Research: Solid Earth.* doi: 10.1002/2013JB010891
- Ehlers, T.A., & Farley, K.A. (2003). Apatite (U-Th)/He thermochronometry; methods and applications to problems in tectonic and surface processes. *Earth and Planetary Science Letters*, 206(1-2), 1–14.
- Harrison, T. M., and P. K. Zeitler (2005), Fundamentals of Noble Gas Thermochronometry, in *Low-Temperature Thermochronology: Techniques, Interpretations and Applications*, vol. 58, edited by P.W. Reiners and T.A. Ehlers, pp. 123–149, Mineralogical Society of America.
- Tagami, T., & O'Sullivan, P. B. (2005). Fundamentals of Fission-Track Thermochronology. In P.W. Reiners & T.A. Ehlers (Eds.), Low-Temperature Thermochronology: Techniques, Interpretations and Applications (Vol. 58, pp. 19–47). Mineralogical Society of America.