

Lecture summary

- Fluorescence, angular distribution and polarization (revisit)
- Atomic parity non-conservation
- Wigner-Eckart Theorem
- Optical pumping
- Wigner's 6-j symbol

Homework

1. In the hydrogen atom, the $2^2P_{3/2}$ ($F = 1, M_F = +1$) state can decay to three possible states in the ground-level $1^2S_{1/2}$: $F = 1, M_F = +1$; $F = 1, M_F = 0$; $F = 0, M_F = 0$.
 - a. Compute the branching ratios of all three decay channels using the 3-j, but not 6-j symbols.
 - b. Compute the branching ratios of all three decay channels using both the 3-j and 6-j symbols.

$$\begin{aligned}
 \langle n' F' M' J' I | r_q | n F M J I \rangle \\
 = (-1)^{J'+I-M_{F'}} \cdot \sqrt{(2F+1)(2F'+1)} \cdot \begin{Bmatrix} J' & F' & I \\ F & J & 1 \end{Bmatrix} \\
 \cdot \begin{pmatrix} F & 1 & F' \\ M_F & q & -M_{F'} \end{pmatrix} \cdot \langle n' J' || r || n J \rangle
 \end{aligned}$$

- c. Calculate the spatial distribution of the fluorescence emissions from this excited state.

Reading Assignments:

Putting a spin on light and atoms: how to build a better magnetometer

<https://phys.org/news/2010-09-atoms-magnetometer.html>