



Plastic Deformation

Thapar Institute of Engineering & Technology
(Deemed to be University)

Bhadson Road, Patiala, Punjab, Pin-147004

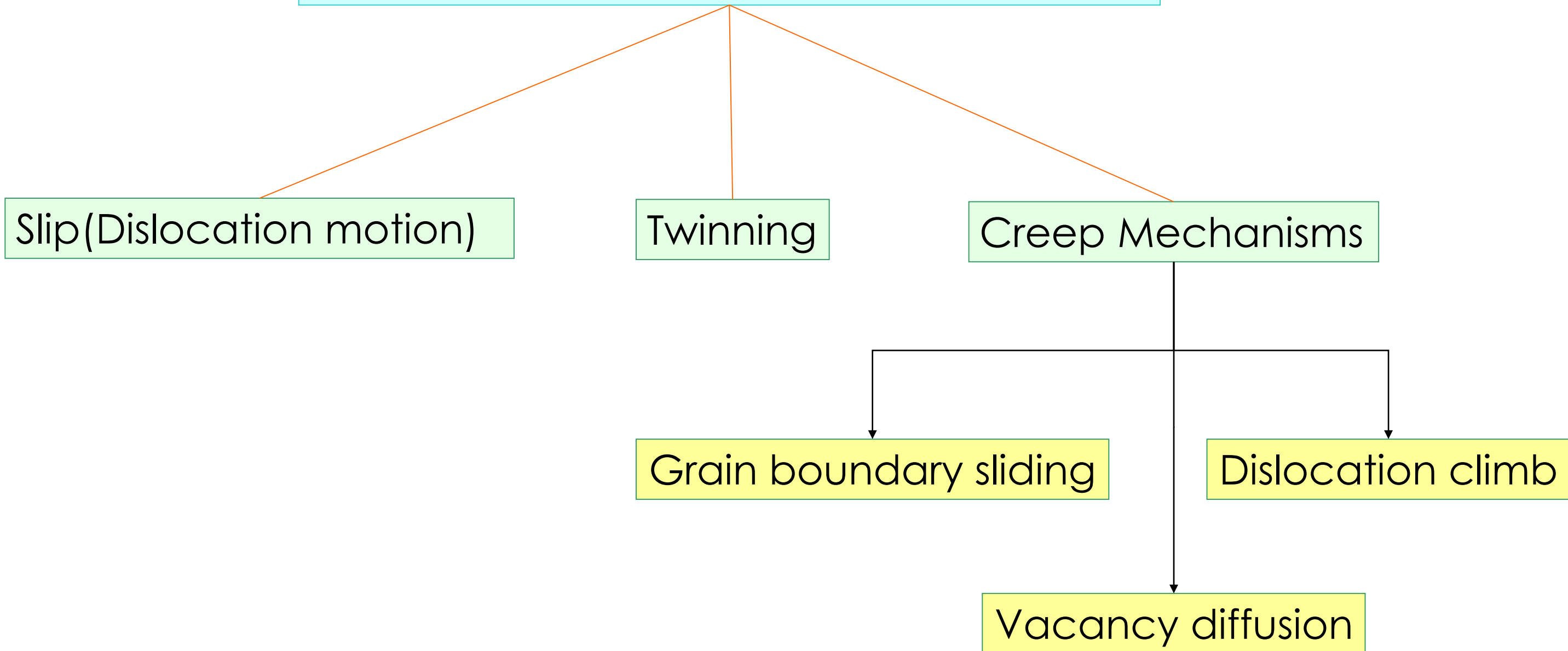
Contact No. : +91-175-2393201

Email : info@thapar.edu



THAPAR INSTITUTE
OF ENGINEERING & TECHNOLOGY
(Deemed to be University)

Plastic Deformation in Crystalline Materials

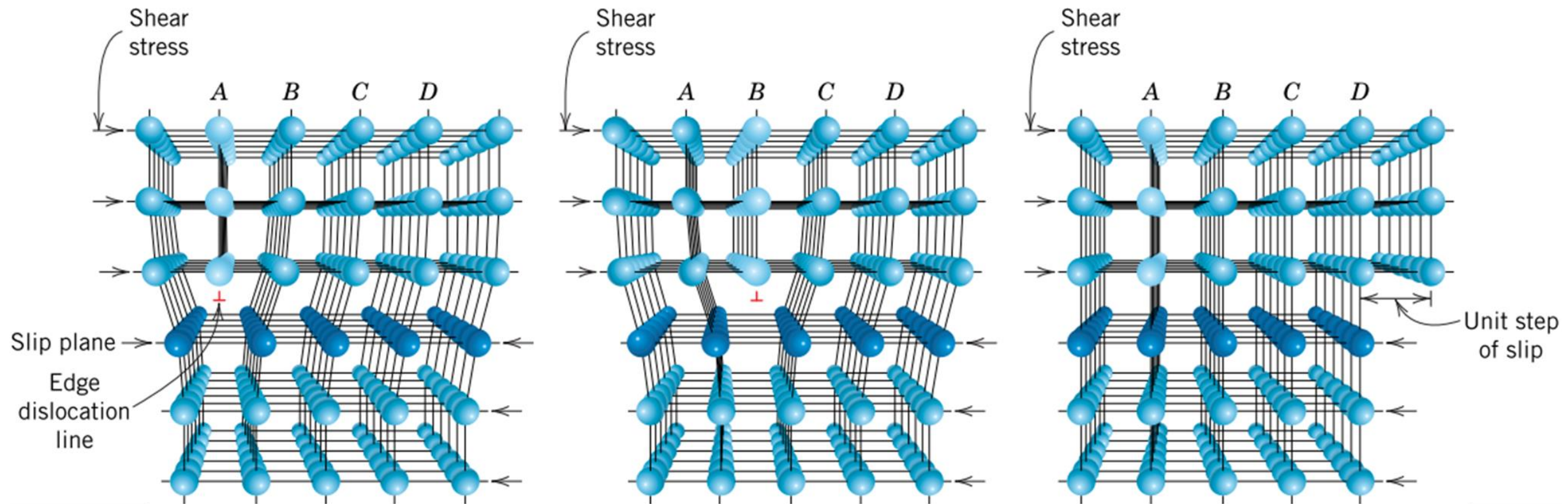


How the plastic deformation takes place at the microscopic level?

What happens to crystal structure after plastic deformation?

Plastic deformation by slip

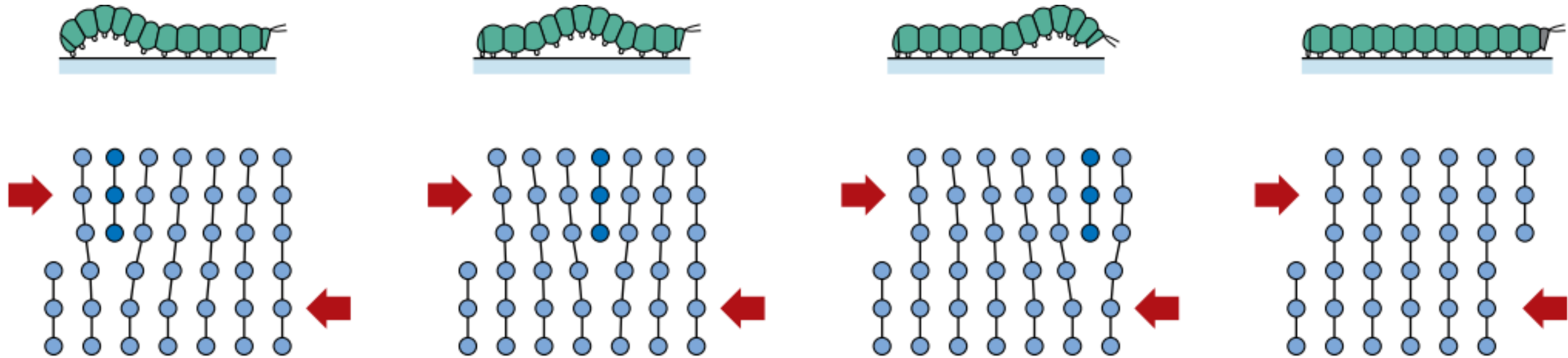
1. Slip is a shear deformation that moves **atoms by many interatomic distances** relative to their initial positions.
2. Steps are created at the surface of the crystal during slip, but **orientation of crystal remains same**.
3. Edge dislocation slides over adjacent plane/half planes of atoms.



- If dislocations can't move, plastic deformation doesn't occur!

Adapted from Fig. 8.1,
Callister & Rethwisch 3e.

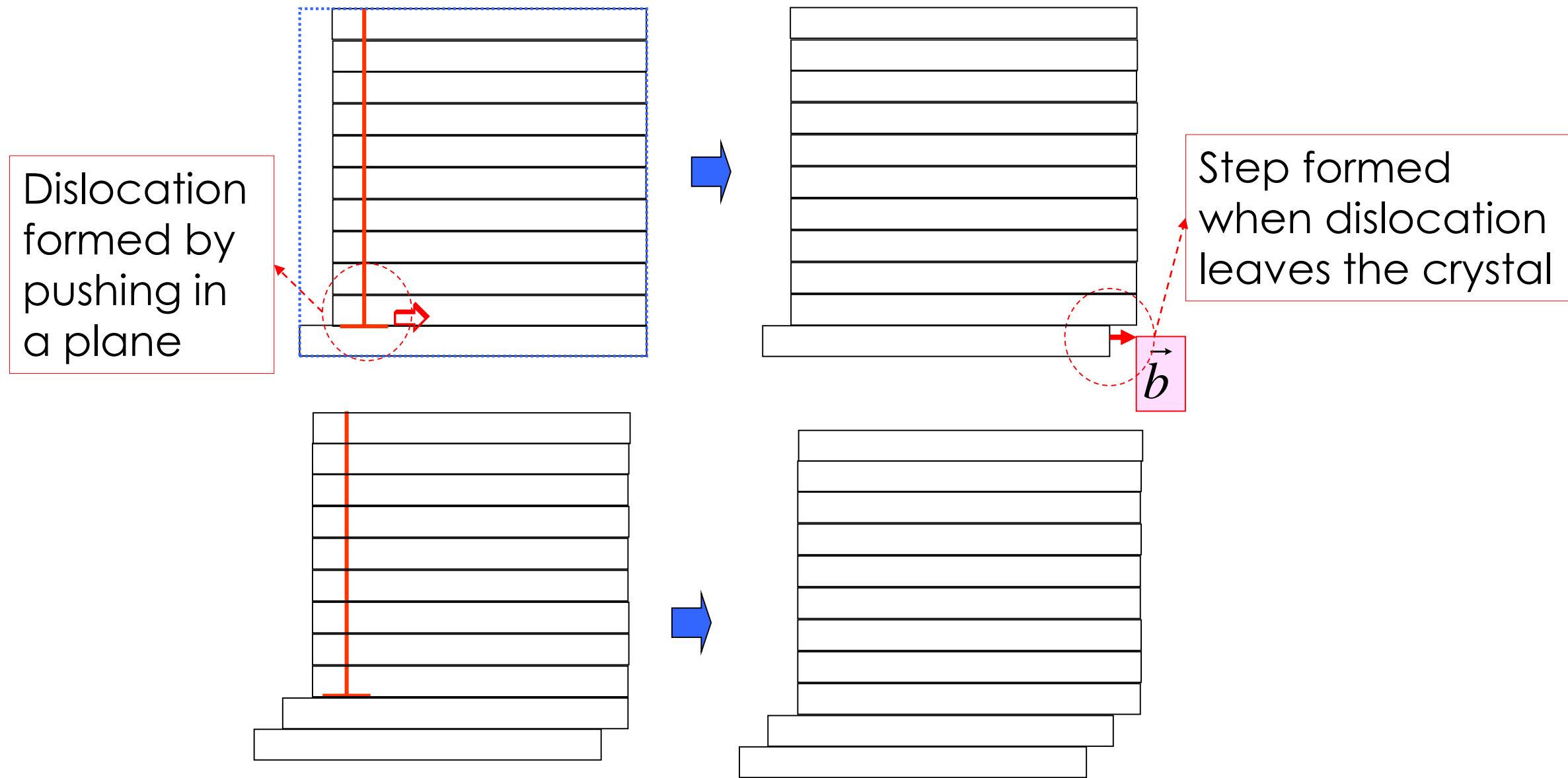
Analogy between caterpillar and dislocation motion.



Plastic deformation → Slip

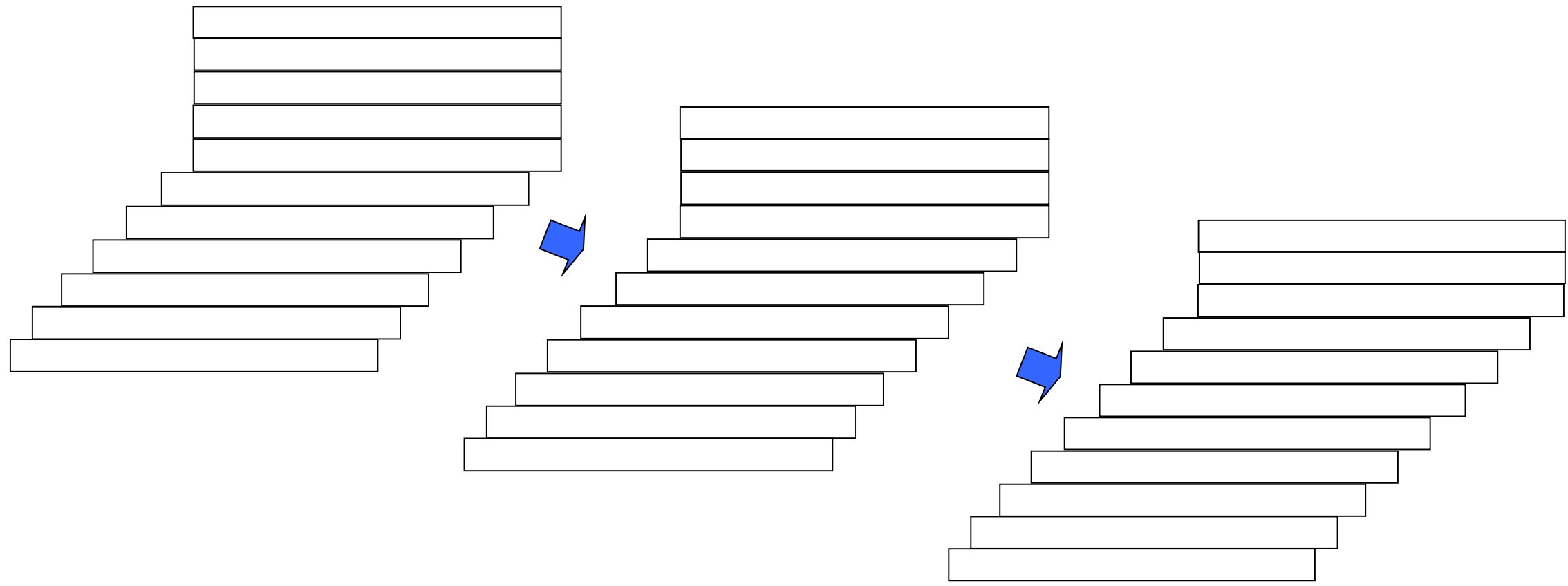
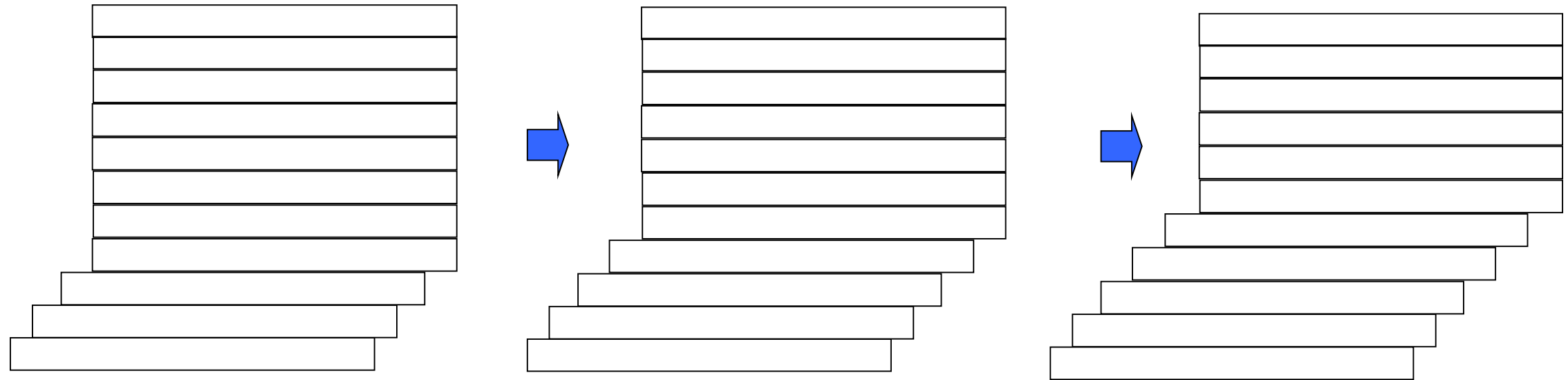
Slip → dislocations

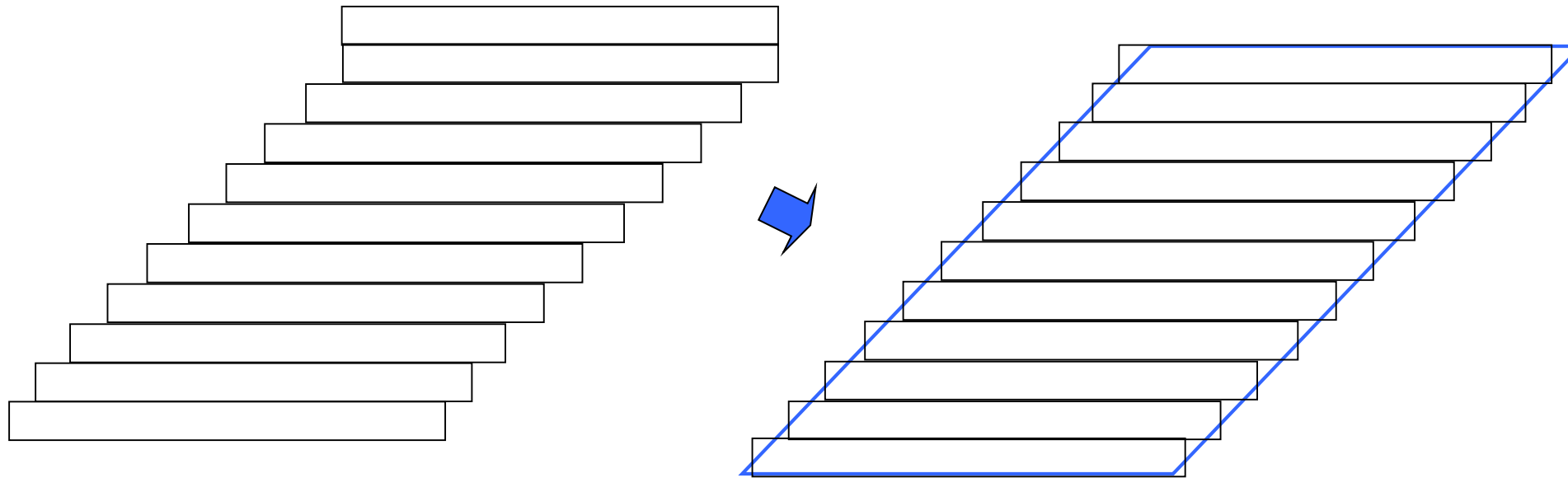
Plastic deformation requires movement of dislocations on the slip plane.



Now visualize dislocations being punched in on successive planes → moving and finally leaving the crystal

Slip movement

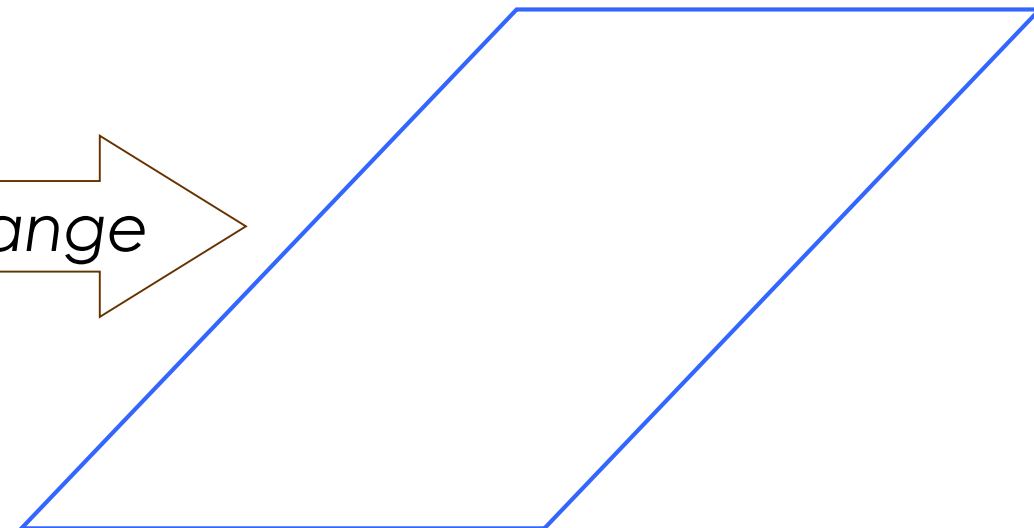




This sequence of events finally leads to deformed shape which can be approximated to a rhombus

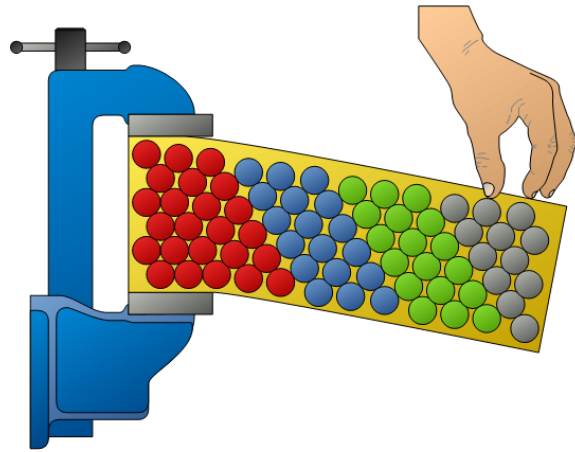


Net shape change

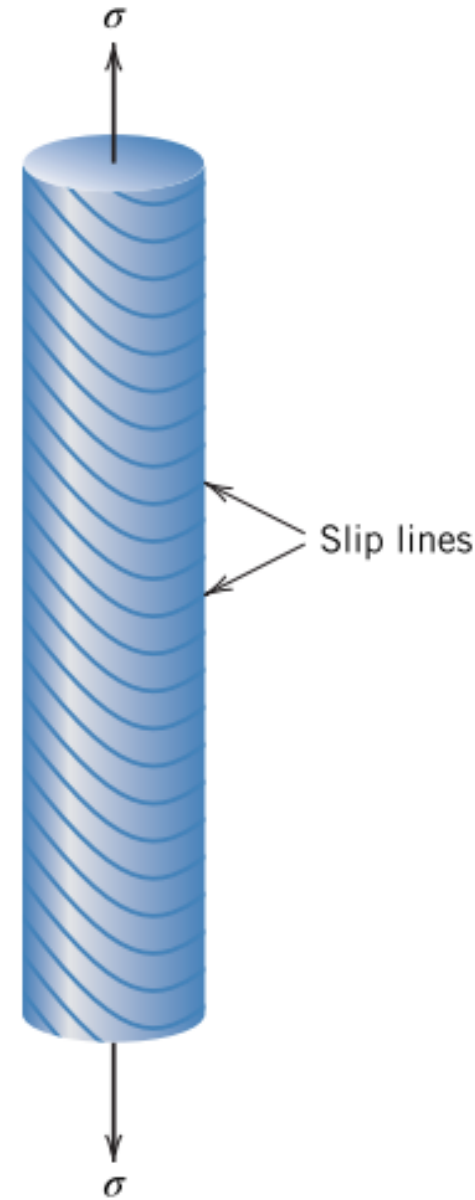
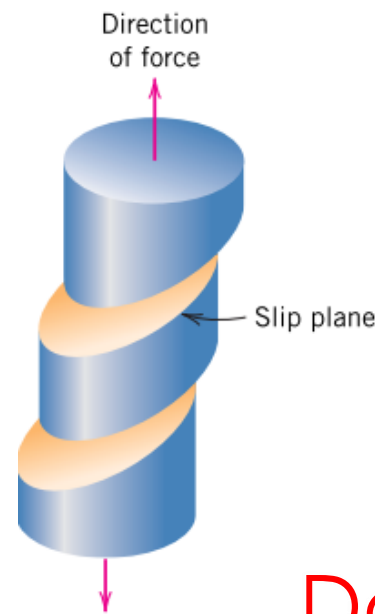
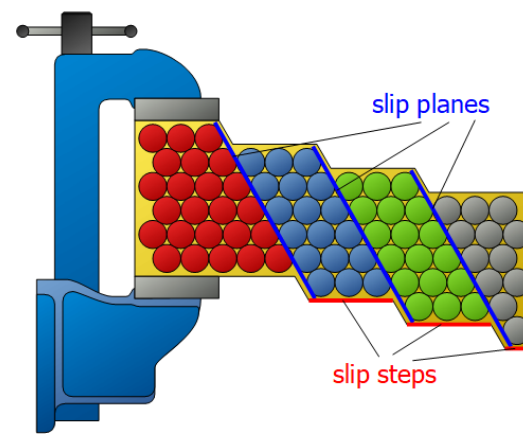


Slip is movement of planes

Before slip



After slip



Visual observation of slip planes



Does this movement takes place in any plane and direction?

Crystallography determines the Burgers vector:

Fundamental lattice translational vector lying on the slip plane.

We need to know:

Slip plane (most densely packed plane) → Dislocations move on this plane

Slip direction (Burger vector) → Dislocations move along this direction

Crystal system	Direction of Burger vector (this is the direction of slip)
Monoatomic FCC	$\frac{1}{2}\langle 110 \rangle$
Monoatomic BCC	$\frac{1}{2}\langle 111 \rangle$
Monoatomic SC	$\langle 100 \rangle$
NaCl type structure	$\frac{1}{2}\langle 110 \rangle$
CsCl type structure	$\langle 100 \rangle$
DC type structure	$\frac{1}{2}\langle 110 \rangle$

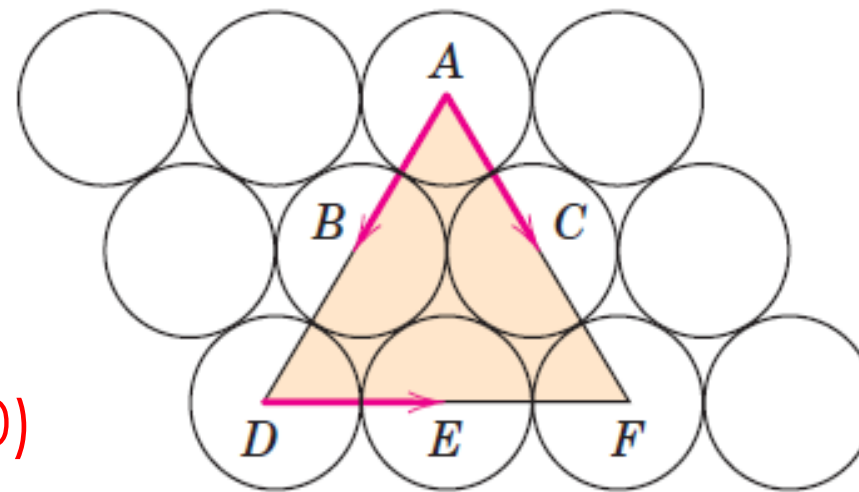
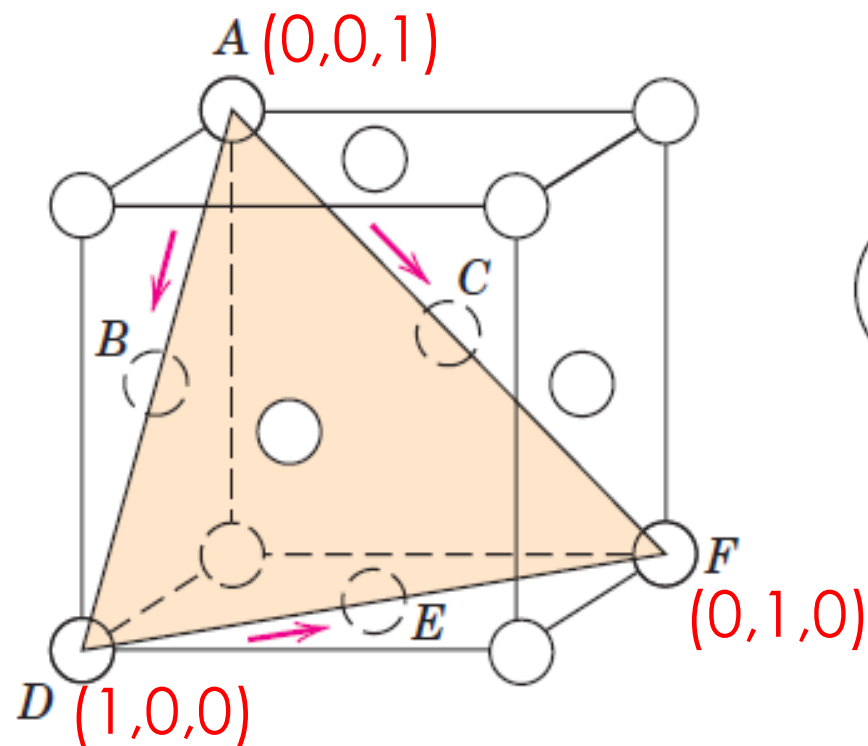
For a FCC system

Slip plane: $\{111\}$ planes (most densely packed planes)

Slip direction: $\langle 110 \rangle$

Slip system
 $\{111\} \langle 110 \rangle$

Consider (111) plane; On this plane exist $[10\bar{1}]$, $[\bar{1}10]$ and $[0\bar{1}1]$ directions.



Three possible slip directions on same slip plane

$\{111\}$ planes: There are 4 unique members.

On each plane three possible $\langle 110 \rangle$ type directions.

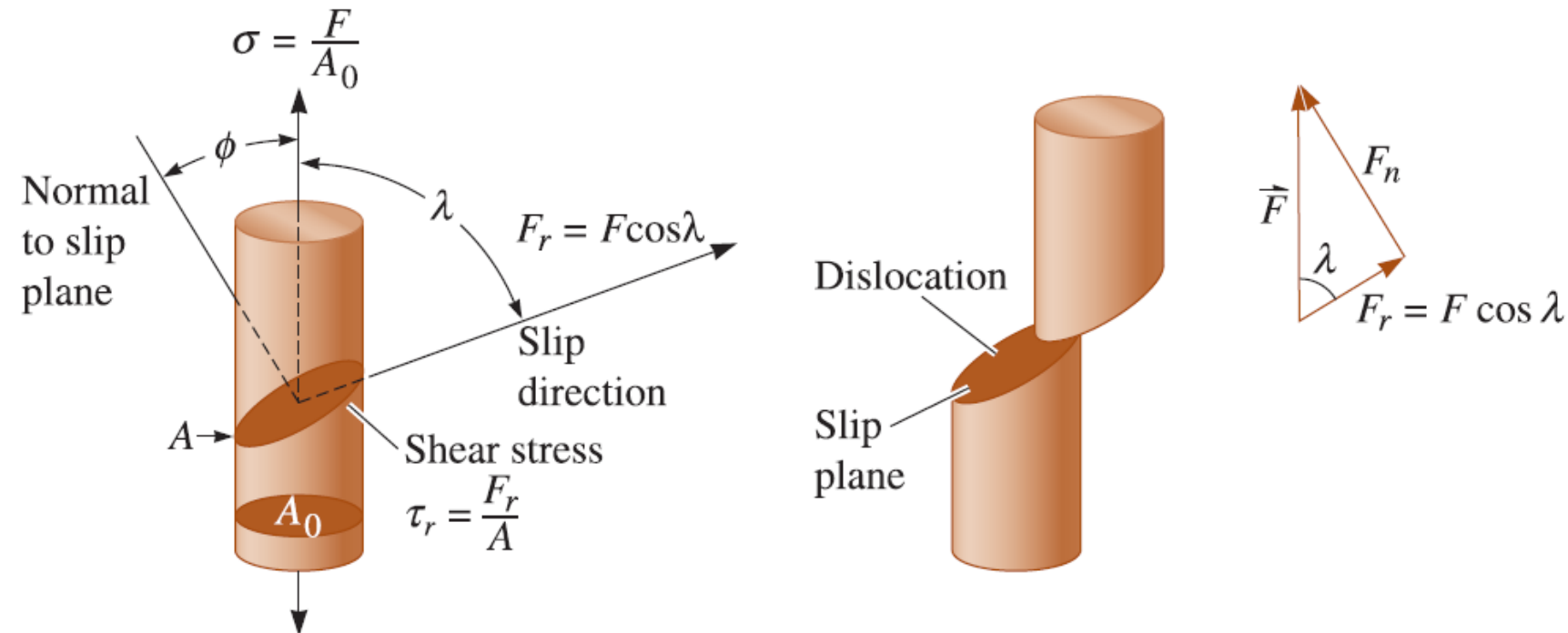
Total **12 possible combinations**.

A combination of a slip direction lying on a slip plane is called a slip system.

This is described in terms of a family of directions and a family of planes.

<i>Metals</i>	<i>Slip Plane</i>	<i>Slip Direction</i>	<i>Number of Slip Systems</i>
<i>Face-Centered Cubic</i>			
Cu, Al, Ni, Ag, Au	{111}	$\langle 110 \rangle$	12
<i>Body-Centered Cubic</i>			
α -Fe, W, Mo	{110}	$\langle 111 \rangle$	12
α -Fe, W	{211}	$\langle 111 \rangle$	12
α -Fe, K	{321}	$\langle 111 \rangle$	24
<i>Hexagonal Close-Packed</i>			
Cd, Zn, Mg, Ti, Be	{0001}	$\langle 11\bar{2}0 \rangle$	3
Ti, Mg, Zr	{10 $\bar{1}0$ }	$\langle 11\bar{2}0 \rangle$	3
Ti, Mg	{10 $\bar{1}1$ }	$\langle 11\bar{2}0 \rangle$	6

HCP metals have fewer active slip systems and hence, are normally quite brittle.



Full amount of force is not used to move the dislocation along slip direction on the slip plane.

Although tensile stress is σ ,

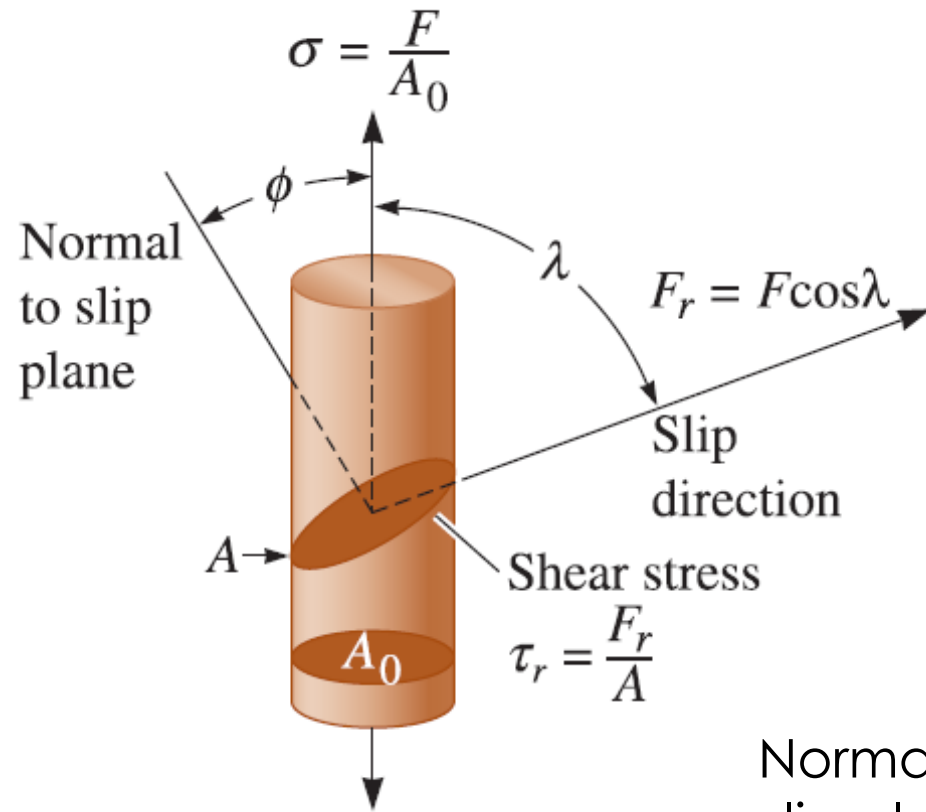
The shear stress responsible to move the dislocation is τ_r (Resolved shear stress).

$$\tau_r = \frac{F_r}{A} = \frac{F \cos \lambda}{\frac{A_0}{\cos \phi}} = \frac{F}{A_0} \cos \lambda \cos \phi$$

Resolved shear stress

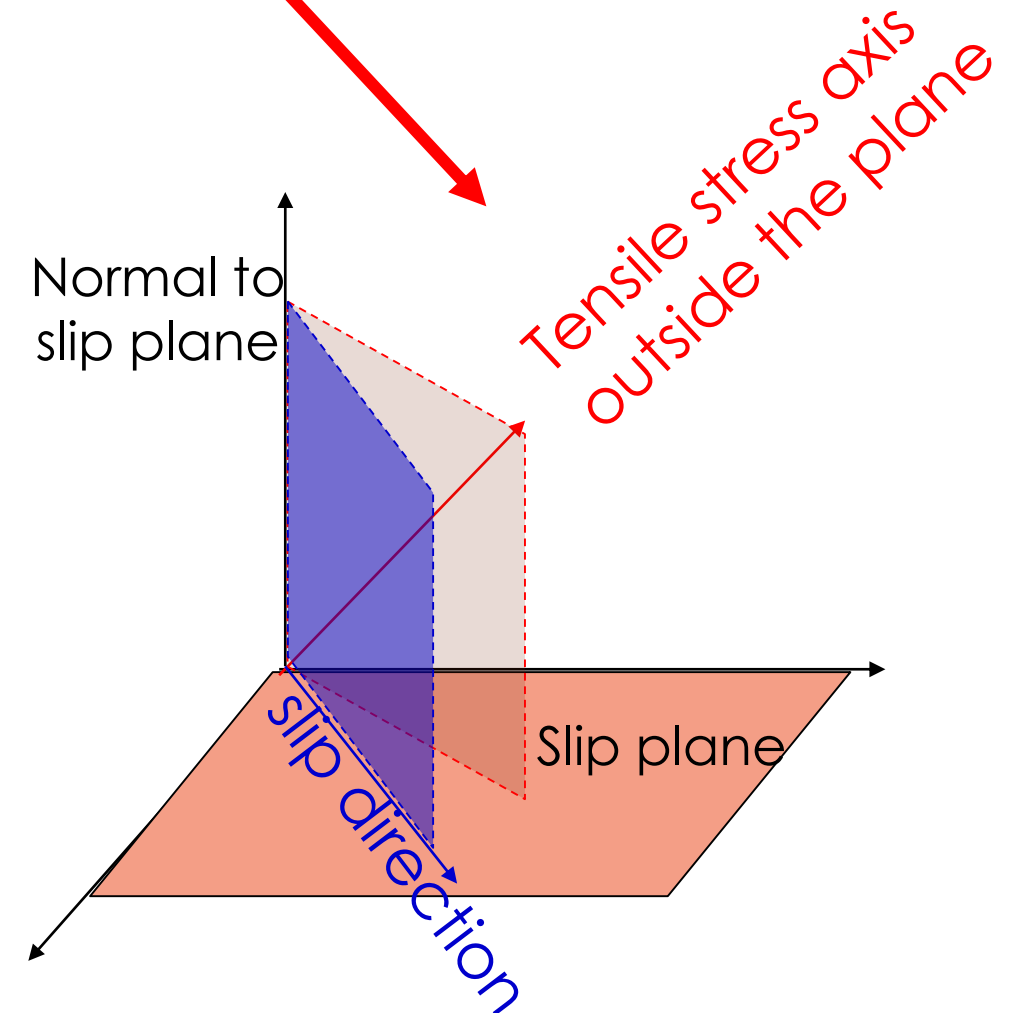
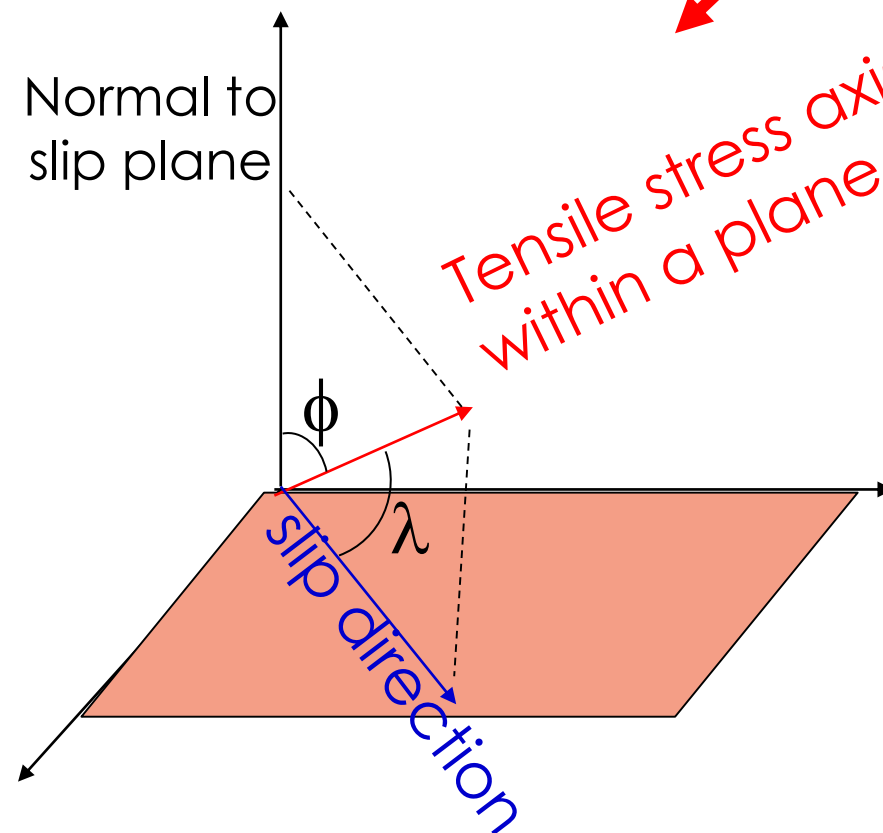
$$\tau_r = \sigma \cos \lambda \cos \phi$$

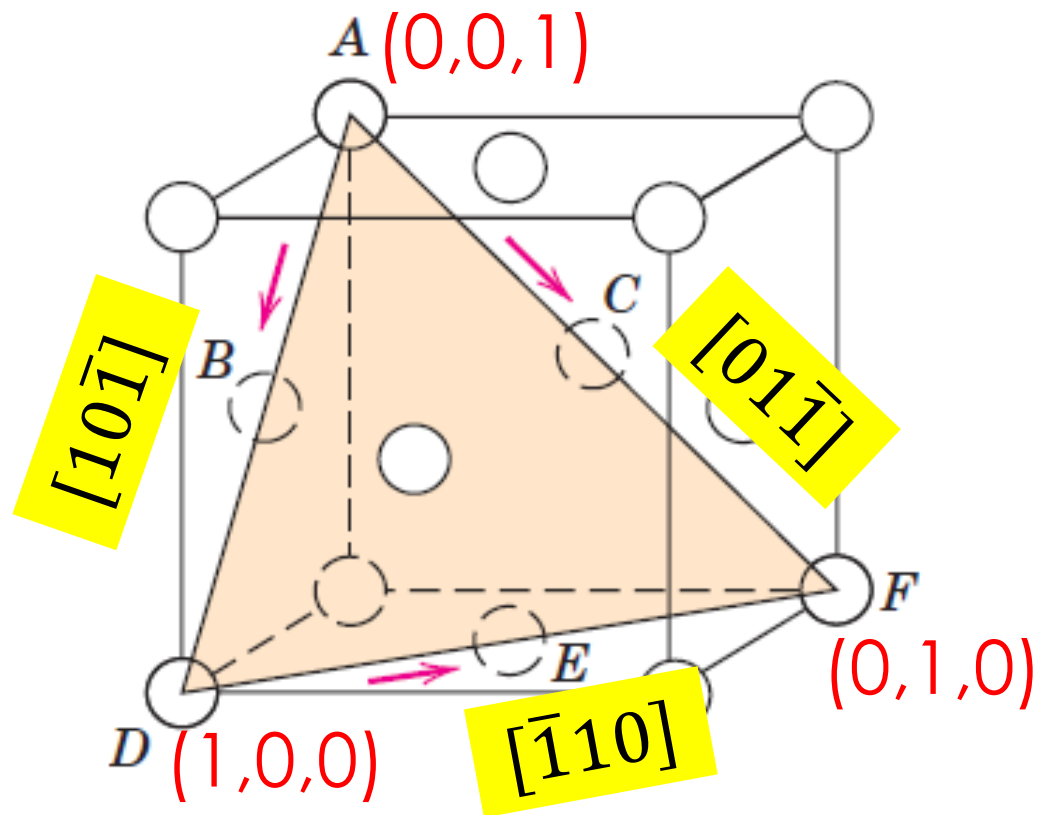
Minimum amount of shear stress capable of moving a dislocation is critical resolved shear stress τ_{CRSS}



Is ALWAYS $(\phi + \lambda) = 90^\circ$?

No, not necessarily.





If we apply a tensile stress, the amount of shear stress along these possible slip directions is not same.

- Depends on the angles λ and ϕ
($\tau_r = \sigma \cos \lambda \cos \phi$)

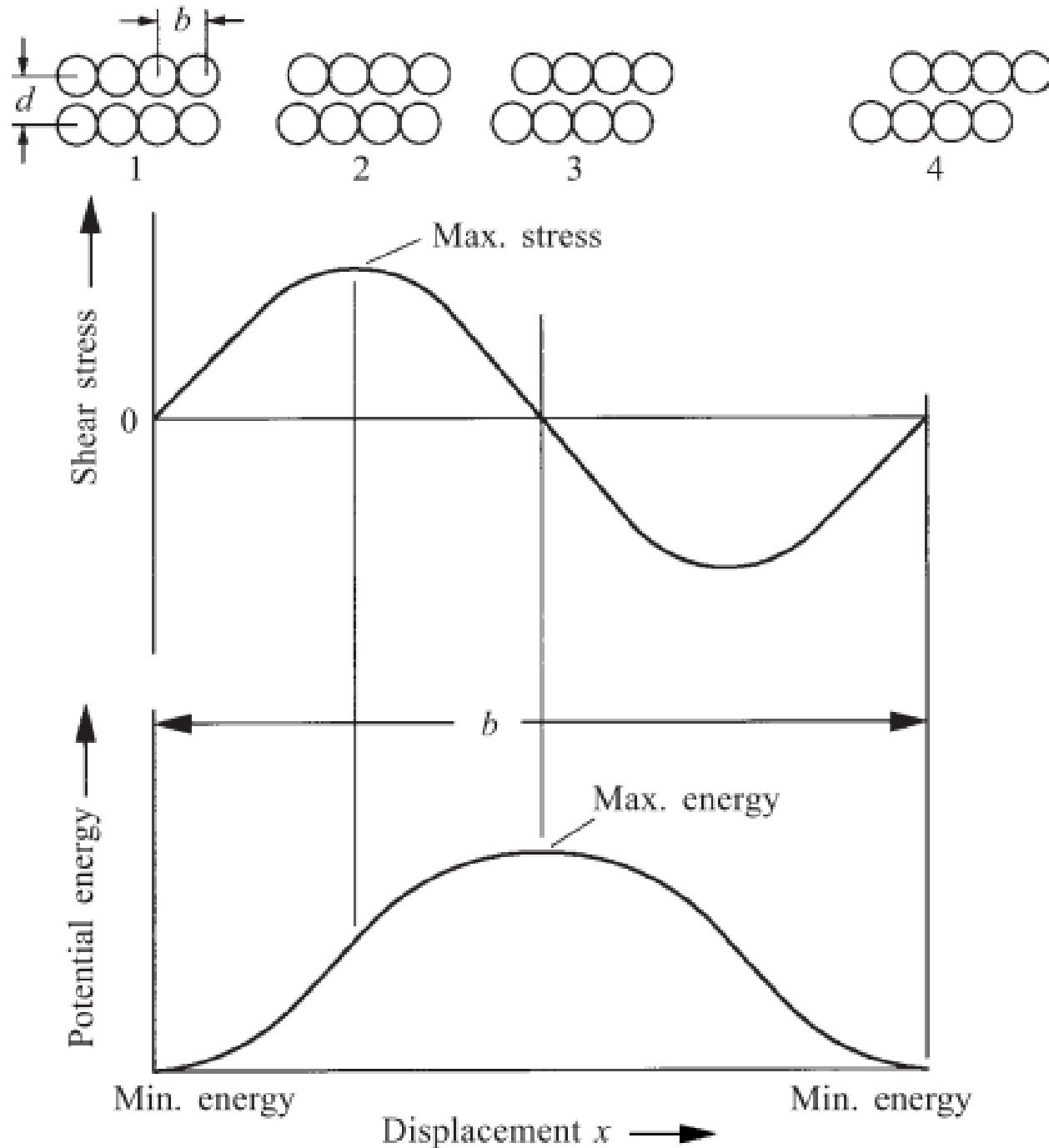
Dislocations will move along the direction which has the maximum of the resolved shear stress among all possible directions.

$$\tau_{CRSS} = \sigma_y (\cos \lambda \cos \phi)_{max}$$

A single crystal of aluminum is oriented for a tensile test such that its slip plane normal makes an angle of 28.18° with the tensile axis. Three possible slip directions make angles of 62.48° , 72.08° , and 81.18° with the same tensile axis. (a) Which of these three slip directions is most favored? (b) If plastic deformation begins at a tensile stress of 1.95 MPa, determine the critical resolved shear stress for aluminum.

Stress to move a dislocation in **ideal** crystals

16



$$\tau = k \sin\left(\frac{2\pi x}{b}\right)$$

$$\frac{2\pi x}{b} \ll 1$$

$\therefore \sin(\theta) = \theta$ if θ is very small

$$\tau = k\left(\frac{2\pi x}{b}\right)$$

$$\tau = \frac{\mu b}{2\pi d} \sin\left(\frac{2\pi x}{b}\right)$$

Now for maximum τ

$$\tau_{CRSS} = \frac{\mu}{6}$$

$b=d$ for many crystals

$K = \text{constant}$

$b = \text{interatomic distance}$

$d = \text{interplanar spacing}$

Now,

$$\mu = \frac{\tau}{\theta} ; \theta = \frac{x}{d}$$

$$\Rightarrow \tau = \frac{\mu x}{d}$$

$$\Rightarrow k = \frac{\mu b}{2\pi d}$$

Do real crystals follow the above theoretical prediction?

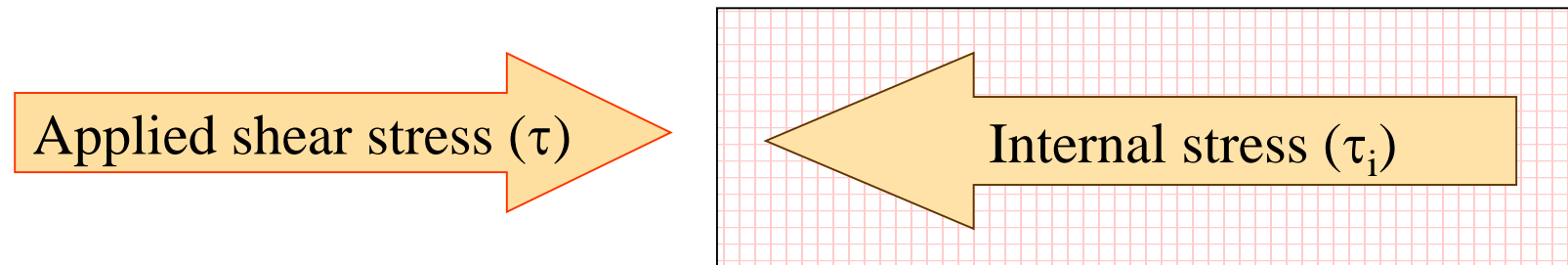
Stress to move a dislocation in real crystals

17

Crystal	Structure	Shear Modulus μ (MN/m ²)	Theoretical τ (MN/m ²)	CRSS, τ (MN/m ²)	μ/τ (This should be 6)
Copper	FCC	44000	7300	0.5	88000
Aluminum	FCC	25000	4100	0.75	33000
Gold	FCC	26000	4300	0.5	52000
Nickel	FCC	70000	11600	5	14000
Silver	FCC	26000	4300	0.5	52000
Iron	BCC	70000	11600	15	4700
Zinc	HCP	33000	5500	0.3	110000

Real crystal do not follow the above result
and deforms at much lower stress.

- Ideal crystal assumes no dislocations, but real crystal have dislocations present in the crystal.
- The dislocations does not move simultaneously.
- They move by small adjustment in the bond lengths in the dislocation region.
- Dislocations already have stress field and helps for movement.



We need to consider a model which uses stress field already present in the crystal

- Peierls & Nabarro consider **presence of dislocations** and calculate the stress to move it called as **Peierls-Nabarro stress (PN stress)**.
- Width of the dislocation is considered as a basis for the ease of motion of a dislocation in the model which is a function of the bonding in the material.

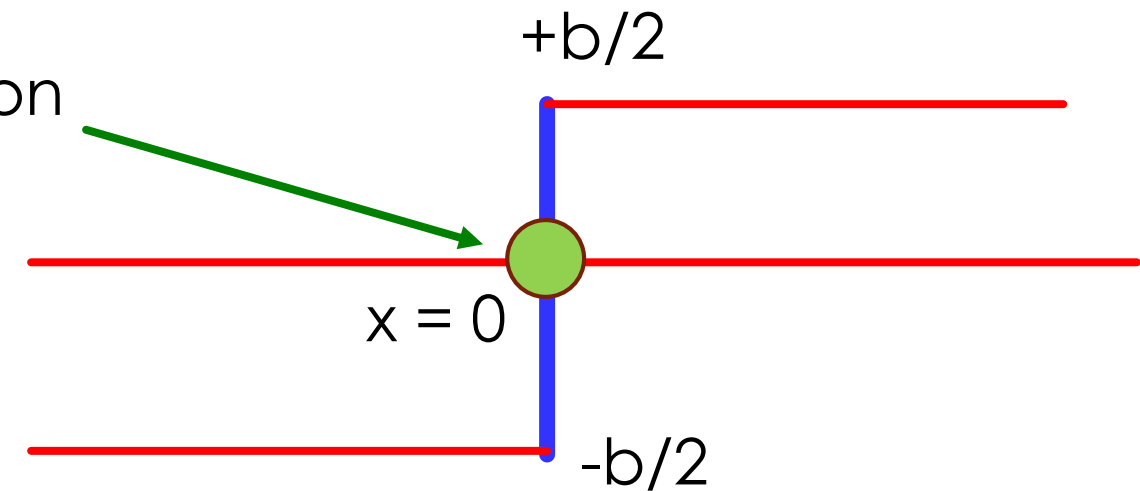
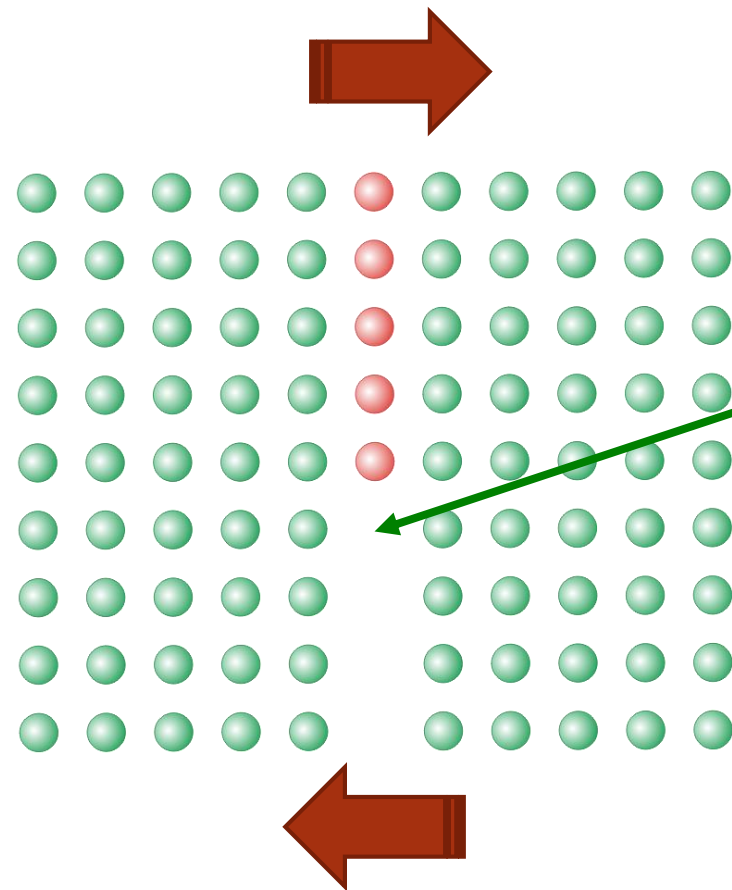
$$\tau_{PN} = \mu e^{-\left(\frac{2\pi w}{b}\right)}$$

- μ → Shear modulus of the crystal
- b → $|\mathbf{b}|$
- w → Width of the dislocation !!!
(dimension over which dislocation is relaxed)

Stress to move a dislocation in real crystals

20

Width of dislocation: Few atomic distance
e.g. 10-20 planes, Diamond



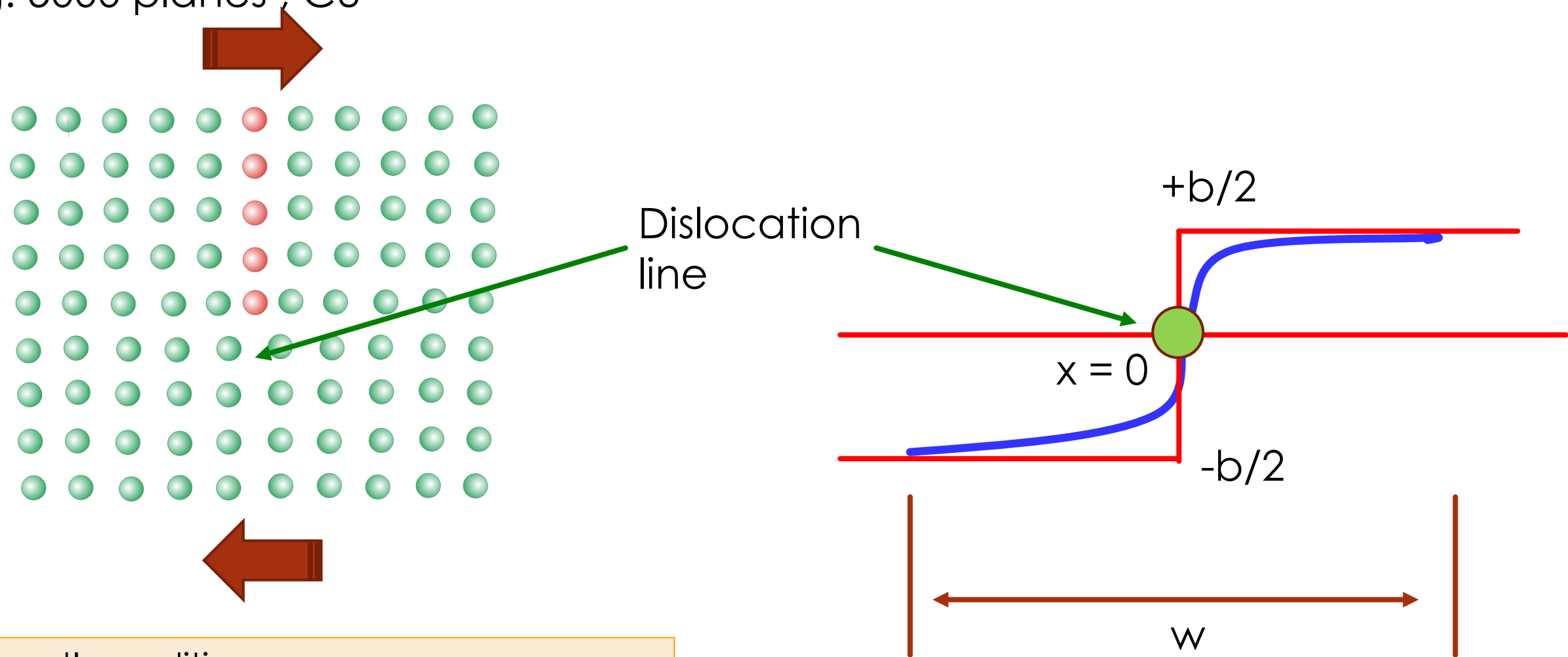
w is almost zero here

- Unrelaxed condition- *stiff*
- **Narrow** width of *displacement fields*
- \Rightarrow atomic adjustments required (*for any one atom*) for dislocation motion are large

Stress to move a dislocation in real crystals

21

Width of dislocation: Many interatomic distance
e.g. 5000 planes, Cu



- 'Relaxed' condition
- Large width of displacement fields
- \Rightarrow atomic adjustments required for dislocation motion are small

Effect of w on τ_{PN}				
w	0	b	$5b$	$10b$
τ_{PN}	μ	$\mu / 400$	$\mu / 10^{14}$	$\mu / 10^{27}$

Hence,

- Narrow dislocations are more difficult to move than wide ones
- Dislocations with larger b are more difficult to move

Nature of chemical bonding in the crystal determines the → extent of relaxation & the width of the dislocation

Covalent crystals

- Strong and directional bonds → small relaxation (low w) → high τ_{PN}
- Usually fail by brittle fracture before τ_{PN} is reached

Metallic crystals

- Weaker and non-directional bonds → large relaxation (high w) → low τ_{PN}
- E.g. Cu can be cold worked to large strains

Ionic crystals

- Moderate and non-directional bonds
- Surface cracks usually lead to brittle fracture
- Large b (NaCl: $b = 3.95\text{\AA}$) → more difficult to move

1. Plastic deformation takes place by the movement of slip planes.
2. Dislocation movement takes place along a specific direction and plane called as slip system.
3. Stress to move dislocations is affected by the presence of dislocations.
4. Narrow (stiff) dislocations are difficult to move.
5. Wide (relaxed) dislocations are easy to move.