

# UCB008 - APPLIED CHEMISTRY



## Molecular Spectroscopy Series Lecture - V

### UV-Visible Spectroscopy – Absorption and Intensity Shifts

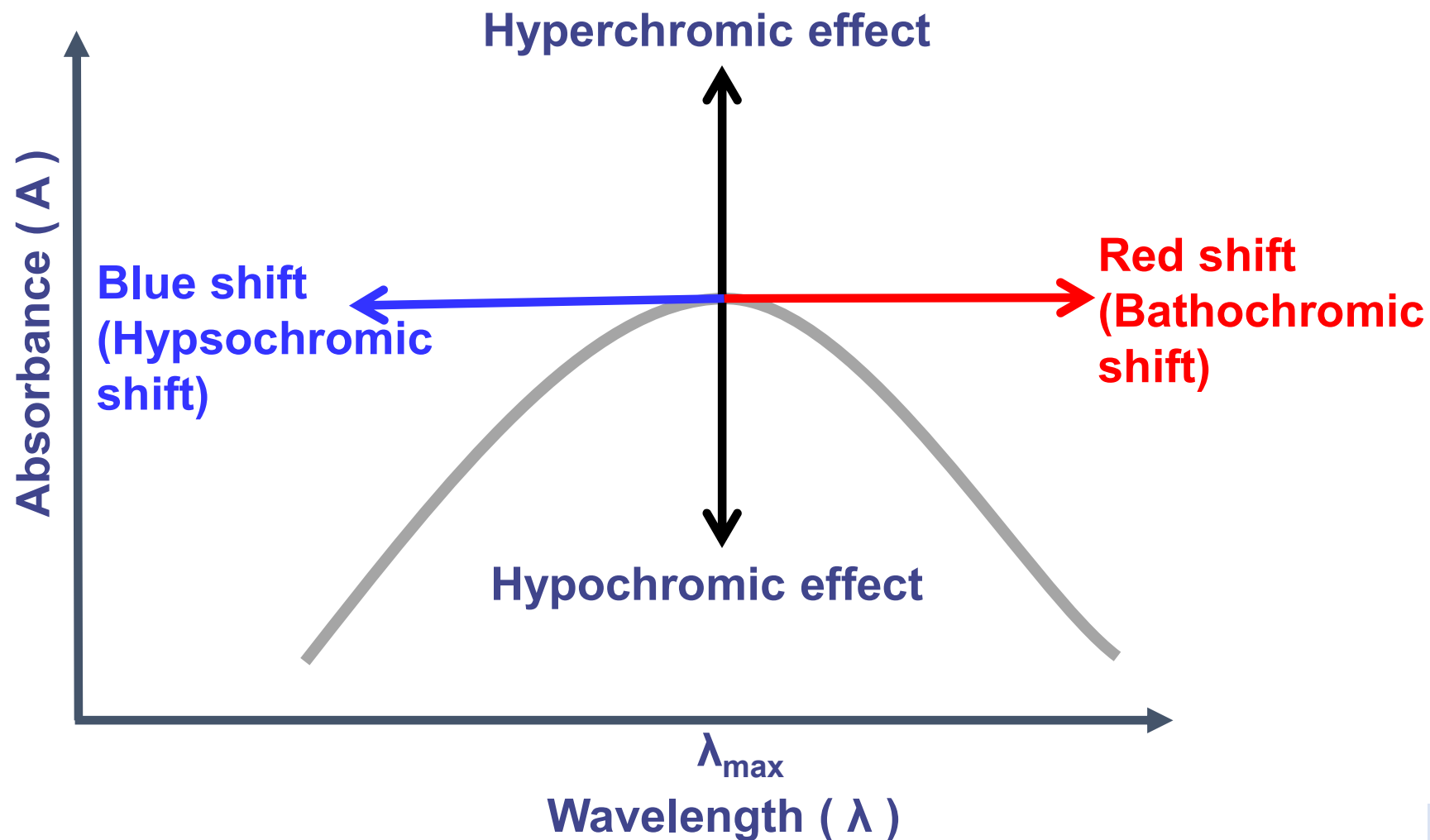
by  
**Prof. Ranjana Prakash**  
School of Chemistry and Biochemistry  
Thapar Institute of Engineering and Technology  
Patiala -147004, India

# Learning Outcomes

At the end of this session participants should be able to:

- Distinguish between various shifts and effects associated with UV-visible spectroscopy

## Shifts and Effects

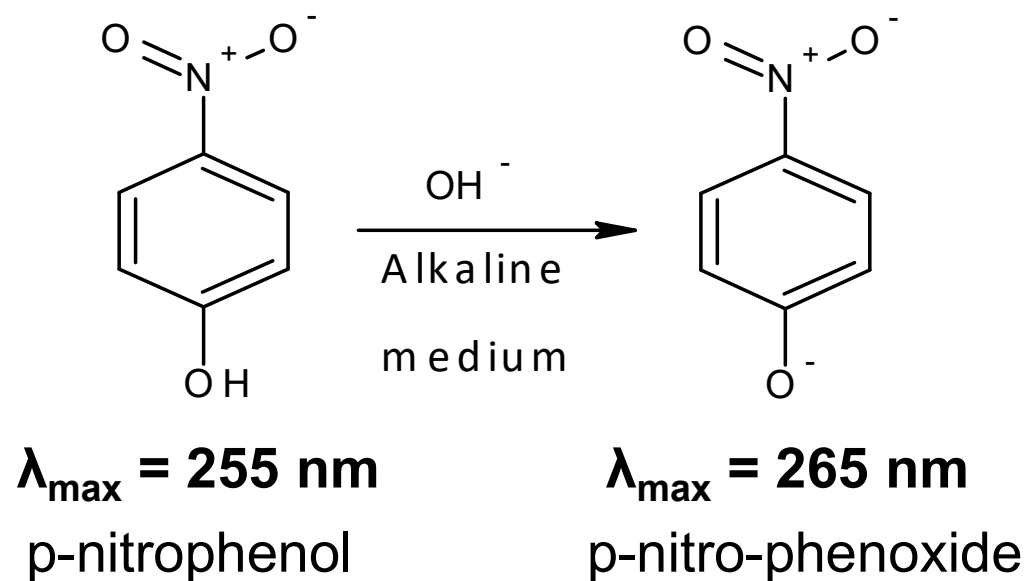


## Bathochromic Shift (Red Shift)

- When absorption maximum ( $\epsilon_{\text{max}}$ ) is shifted towards longer wavelength, it is known as **bathochromic shift** or **red shift**.
- The effect is due to presence of an auxochrome or by the change of solvent.
- For example, an auxochrome group like  $-\text{OH}$ ,  $-\text{OCH}_3$  causes absorption of compound at longer wavelength.

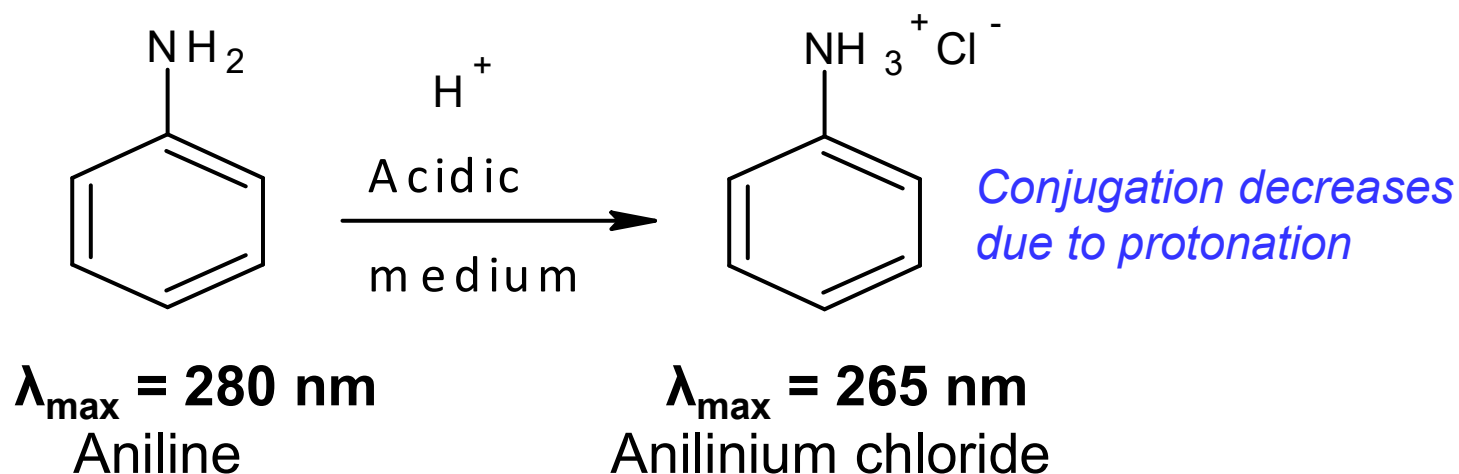
## Bathochromic Shift (Red Shift)

- p-nitrophenol, in alkaline medium, shows **red shift**, because negatively charged oxygen delocalizes more effectively than the unshared pair of electron.



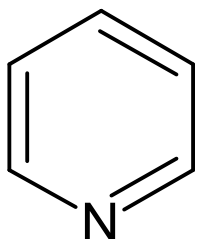
## Hypsochromic Shift (Blue Shift)

- When absorption maximum ( $\epsilon_{\max}$ ) is shifted towards shorter wavelength, it is known as **Hypsochromic shift** or **blue shift**.
- The effect is due to removal of conjugation or by the change of solvent.
- Aniline shows **blue shift** in acidic medium because it loses conjugation.



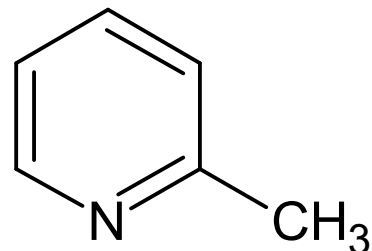
# Hyperchromic effect

- When intensity of absorption maximum ( $\epsilon_{\max}$ ) of a compound increases, it is known as **hyperchromic effect**.



$\lambda_{\max} = 257 \text{ nm}$   
 $\epsilon = 2750$   
 Pyridine

$$A = \epsilon c x$$

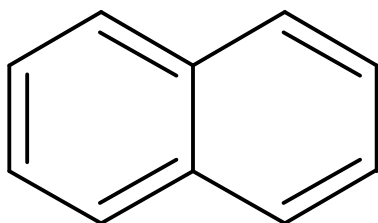


$\lambda_{\max} = 260 \text{ nm}$   
 $\epsilon = 3560$   
 2-methyl pyridine

- Inclusion of an auxochrome also increases intensity of absorption.

## Hypochromic effect

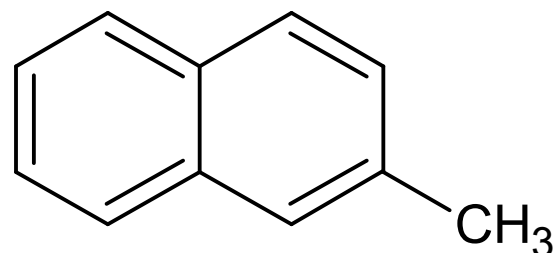
- When intensity of absorption maximum ( $\epsilon_{\max}$ ) of a compound decreases, it is known as **hypochromic effect**.



$$\epsilon = 19000$$

$$\lambda_{\max} = 250 \text{ nm}$$

Naphthalene



$$\epsilon = 10250$$

$$\lambda_{\max} = 237 \text{ nm}$$

2-methyl naphthalene

- $\epsilon_{\max}$  decreases due to the distortion caused by the methyl group.



In the next session.....

- Understand the effect of conjugation on  $\lambda_{\text{max}}$
- Colour in organic compounds