

[illegible]

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He states that atoms may not exist in the free state but may exist in the combined state in the form of molecules.



Atoms combine to form molecules

O_2 is an element while CO_2 & H_2O are compounds.

Diagram illustrating the classification of substances:

- Substances are divided into two categories:
 - Element** (Simpler substance)
 - Compound** (Complex substance)

Antoine L. Lavoisier laid the foundation of chemical sciences by establishing on invariant laws of chemical combinations.

1. Law of chemical Combination

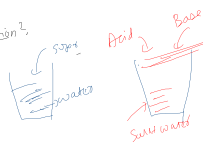
Elements — substances whose molecules are made up of only one type of atoms.

Compounds : Substances whose molecules are made up of more than one type of atoms are called compounds.

- Properties of compound are different from the properties of elements they are made up of.

- The combination takes place via chemical reaction following certain laws called laws of Chemical Combination

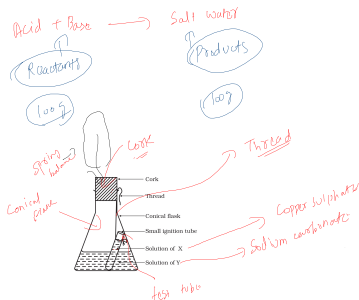
Chemical reaction?



When two or more substances combine to form an entirely different product

Antoine Lavoisier gave 1st Law of Chemical combination

LAW OF CONSERVATION OF MASS

 $\rightarrow X_{L^m}$

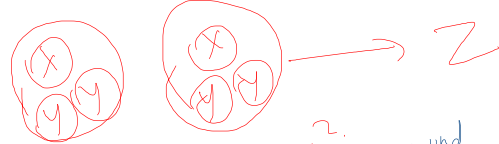
(2) $\times 999$

Copper Sulphate + Sodium Carbonate

In any chemical reaction, the total mass of reactants is equal to the total mass of the products.

Law of conservation of mass states that mass can neither be created nor destroyed in a chemical reaction.

Copper carbonate + Sodium sulphate

 z^{-2}

- 2.
- a) compound
 - b) Molecule
 - c) Element
 - d) Both a & b

✓ mass can created in chemical reaction.

11. If $(x^3 + mx^2 + bx + 6)$ has $(x - 2)$ as a factor and leaves a remainder 3 when divided by $(x - 3)$, find the values of a and b .

→ $P(n) = n^3 + 4n^2 + 6n + 4$
 $(n-2)$ is a factor

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$$20 \text{ g} + 100 \text{ g} + 20 \text{ g} + 100 \text{ g}$$

$$2a + b + 7 = 0$$

$2\pi + 7\pi = 9\pi$