

Summary of Useful Z Equations

$$Z = \sum_{A \in N} e^{-E_i/kT} = \sum g_i e^{-E_i/kT}$$

- ① get E_i & g_i from Q.M.
- ② get $Z(T, V)$
- ③ get everything else from Z !!

$$U = -\frac{1}{Z} \frac{\partial Z}{\partial \beta}$$

or

$$U = -\frac{\partial \ln Z}{\partial \beta}$$

or

$$U = \frac{kT^2}{Z} \left(\frac{\partial Z}{\partial T} \right)_V$$

$$N_i = \left(\frac{N_{\text{tot}}}{Z} \right) g_i e^{-E_i/kT}$$

This is population of i^{th} level E_i

$$F = -NkT \ln Z$$

Helmholtz Free Energy ("A" in UHAG)

$$P = - \left(\frac{\partial F}{\partial V} \right)_{T, N} = NkT \left(\frac{\partial \ln Z}{\partial V} \right)_{T, N}$$

$$S = - \left(\frac{\partial F}{\partial T} \right)_{V, N}$$

$$C_v = \left(\frac{\partial U}{\partial T} \right)_V$$

Ideal Diatomic Gas:

$$Z = Z_{\text{transl}} \cdot Z_{\text{rot}} \cdot Z_{\text{vib}}$$

$$Z_{\text{vib}} = \left(\frac{e^{-h\nu/2kT}}{1 - e^{-h\nu/kT}} \right)$$

$kT \gg E$ "classical" limit

$$Z_{\text{rot}} \approx \left(\frac{kT}{h^2/2I} \right)$$

eg. $Z_{\text{transl}} = \frac{V}{h^3} (2\pi mkT)^{3/2}$

For "N" identical, indistinguishable particles

$$Z_{\text{transl}} = \frac{1}{N!} \left(\frac{V}{v_Q} \right)^N$$