

## Problem D1

Consider a particle in a harmonic oscillator with energy levels  $nhf$ , where  $n = 0, 1, 2, \dots$ , at temperature  $T$ .

### Question a:

Show that the partition function  $Z$  of the particle is

$$Z \simeq \left( \frac{kT}{hf} \right)^\kappa$$

for  $kT \gg hf$  and give a value for  $\kappa$ .

Consider now a particle, which can sit in two harmonic oscillators. Oscillator  $A$  has energy levels  $nhf$ , and oscillator  $B$  has energy levels  $\epsilon_0 + nhf$ , where  $\epsilon_0 > 0$  and  $n = 0, 1, 2, \dots$ . The particle can hop freely between the two oscillators, and the available states for the particle are therefore the harmonic oscillator levels in  $A$  and  $B$ .

### Question b:

Give an expression for the partition function for the particle, when the temperature is  $T$ . What is the probability  $P_A$  that the particle sits in oscillator  $A$ ?

### Question c:

Give an expression for the average energy  $E(T)$  of the particle. Show that  $E(T) \approx a + bT$  when  $kT$  is large compared to both  $hf$  and  $\epsilon_0$ , and give values for  $a$  and  $b$ .

## Problem D2

Consider a single particle, which can occupy any of the 6 energy levels shown in figure 1. Three of the energy levels are at the energy  $\epsilon_1$ , two are at the energy  $\epsilon_2$ , and one is at the energy  $\epsilon_3$ . Here,  $\epsilon_3 > \epsilon_2 > \epsilon_1$ . The system is in thermal equilibrium with a reservoir at temperature  $T$ .

### Question a:

Give an expression for the partition function  $Z$  of the system.

### Question b:

Compute the mean energy  $\bar{E}$  of the system.

### Question c:

Compute the limits  $T \rightarrow 0$  and  $T \rightarrow \infty$  of  $\bar{E}$  and explain both limits physically.

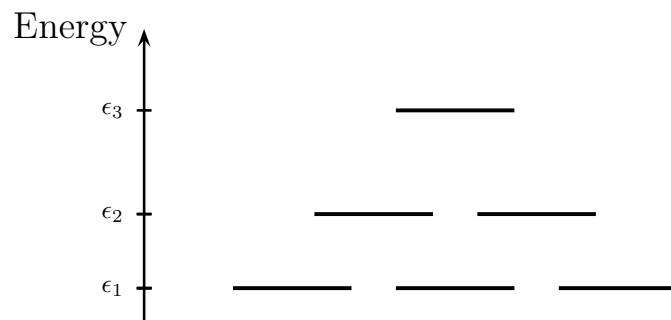


Figure 1: The 6 energy levels of the considered system.

## Hand-in D

An alternative definition of the entropy is

$$S = -k \sum_r p_r \ln(p_r), \quad (1)$$

where  $p_r$  is the probability of the system to be in microstate  $r$  and the sum is over all microstates accessible to the system. We take Eq. (1) to hold for any probability distribution  $p_r$  with  $\sum_r p_r = 1$  and  $p_r \geq 0$ .

### Question a:

For an isolated system in equilibrium,  $p_r = p_0 \equiv 1/\Omega$  for all accessible states  $r$ . Show that in this case, Eq. (1) reduces to our basic definition given by Schroeder Eq. (2.45), i.e.  $S = S_0 \equiv k \ln(\Omega)$ .

### Question b:

Take a system with another probability distribution  $p_r$  and entropy  $S$ . Show that

$$S_0 - S = k \sum_r p_0 \frac{p_r}{p_0} \ln \left( \frac{p_r}{p_0} \right). \quad (2)$$

### Question c:

Show that we can write

$$S_0 - S = k p_0 \sum_r f(p_r/p_0) \quad (3)$$

with  $f(x) = x \ln(x) - x + 1$ .

### Question d:

Show that  $f(x) > 0$  for all  $x \geq 0$ , except for  $x = 1$  where  $f(x) = 0$ . *Hint:* Look at the sign of  $f'(x)$ .

### Question e:

Prove that  $S \leq S_0$  for all probability distributions  $p_r$  over microstates, and that the equality sign only holds for  $p_r = p_0$ .

You have now shown that the entropy is maximum when the probability is constant for all accessible states for an isolated system, i.e. when it is in equilibrium.