

1. a)

$$H_N\{s_i\} = -h \sum_{i=1}^N s_i$$

Thus:

$$\begin{aligned} Z_N &= \sum_{s_1=\pm 1} \sum_{s_2=\pm 1} \cdots \sum_{s_N=\pm 1} \exp \left\{ -\beta h \sum_{i=1}^N s_i \right\} \\ &= \sum_{s_1=\pm 1} \sum_{s_2=\pm 1} \cdots \sum_{s_{N-1}=\pm 1} \exp \left\{ -\beta h \sum_{i=1}^{N-1} s_i \right\} \left(\sum_{s_N=\pm 1} \exp \{ -\beta h s_N \} \right) \\ &= \sum_{s_1=\pm 1} \sum_{s_2=\pm 1} \cdots \sum_{s_{N-1}=\pm 1} \exp \left\{ -\beta h \sum_{i=1}^{N-1} s_i \right\} (e^{\beta h} + e^{-\beta h}) \\ &= \sum_{s_1=\pm 1} \sum_{s_2=\pm 1} \cdots \sum_{s_{N-1}=\pm 1} \exp \left\{ -\beta h \sum_{i=1}^{N-1} s_i \right\} 2 \cosh(\beta h) \\ &= 2 \cosh(\beta h) \sum_{s_1=\pm 1} \sum_{s_2=\pm 1} \cdots \sum_{s_{N-1}=\pm 1} \exp \left\{ -\beta h \sum_{i=1}^{N-1} s_i \right\} \\ &= 2 \cosh(\beta h) Z_{N-1} \end{aligned}$$

And $Z_1 = e^{\beta h} + e^{-\beta h} = 2 \cosh(\beta h)$. It follows that:

$$Z_N = [2 \cosh(\beta h)]^N$$

Now for the average magnetization $\langle s_i \rangle$:

$$\begin{aligned} \langle s_i \rangle &= \frac{1}{Z_N} \sum_{s_1=\pm 1} s_1 \sum_{s_2=\pm 1} s_2 \cdots \sum_{s_N=\pm 1} s_N \exp \left\{ -\beta h \sum_{i=1}^N s_i \right\} \\ &= \frac{1}{Z_N} \sum_{s_1=\pm 1} s_1 \sum_{s_2=\pm 1} s_2 \cdots \sum_{s_{N-1}=\pm 1} \exp \left\{ -\beta h \sum_{i=1}^{N-1} s_i \right\} \sum_{s_N=\pm 1} s_N \exp \{ -\beta h s_N \} \\ &= \frac{1}{Z_N} \sum_{s_1=\pm 1} s_1 \sum_{s_2=\pm 1} s_2 \cdots \sum_{s_{N-1}=\pm 1} \exp \left\{ -\beta h \sum_{i=1}^{N-1} s_i \right\} [-2 \sinh(\beta h)] \\ &\quad \vdots \\ &= \frac{[-2 \sinh(\beta h)]^N}{Z_N} \\ &= [-\tanh(\beta h)]^N \end{aligned}$$

2. The canonical partition function for an ideal gas is:

$$\begin{aligned}
 Z_N &= \frac{1}{h^3 N!} \int d^{3N} p d^{3N} q e^{-\beta H} \\
 &= \frac{1}{\lambda^{3N} N!} \int d^{3N} q \exp \left\{ -\beta \sum_{i < j} u_{ij} \right\} \\
 &= \frac{1}{\lambda^{3N} N!} \int d^{3N} q e^{-\beta \times 0} \\
 &= \frac{1}{\lambda^{3N} N!} \int d^{3N} q \\
 &= \frac{V^N}{\lambda^{3N} N!}
 \end{aligned}$$

Since there is no interaction between ideal gas molecules. We can use this to evaluate the grand partition function for an ensemble of ideal gas samples:

$$\begin{aligned}
 \mathfrak{Z} &= \sum_{N=0}^{\infty} z^N \frac{V^N}{\lambda^{3N} N!} \\
 &= \sum_{N=0}^{\infty} \frac{(zV/\lambda^3)^N}{N!} \\
 &= e^{zV/\lambda^3}
 \end{aligned}$$

We also have that:

$$P = \frac{k_B T}{V} \ln \mathfrak{Z} = z \frac{k_B T}{\lambda^3} \Rightarrow \frac{P}{k_B T} = \frac{z}{\lambda^3}$$

and:

$$\langle N \rangle = z \frac{\partial}{\partial z} \left[\frac{zV}{\lambda^3} \right] = \frac{zV}{\lambda^3} \Rightarrow \frac{\langle N \rangle}{V} = \frac{z}{\lambda^3}$$

From which we deduce:

$$\frac{P}{k_B T} = \frac{\langle N \rangle}{V} \Rightarrow PV = \langle N \rangle k_B T$$

3. We have:

$$\begin{aligned}\frac{1}{\lambda^3} \sum_{l=1}^{\infty} b_l z^l &= \frac{P}{k_B T} = \frac{1}{\lambda^3} \sum_{l=1}^{\infty} a_l \times \left(\lambda^3 \frac{\langle N \rangle}{V} \right)^l \\ \Rightarrow \sum_{l=1}^{\infty} b_l z^l &= \sum_{l=1}^{\infty} a_l \times \left(\sum_{k=1}^{\infty} k b_k z^k \right)^l\end{aligned}$$

Now we match coefficients of z . The only way to get a z term on the right handside is if $l = k = 1$:

$$b_1 z = a_1 \times b_1 z \Rightarrow a_1 = 1$$

Also note that:

$$b_1 = \frac{1}{V} \int_V d^3 r = 1 \Rightarrow a_1 = b_1 = 1$$

For the z^2 terms, the only way to get z^2 on the right is if $l = 1$ and $k = 2$ or if $l = 2$ and $k = 1$

$$\begin{aligned}b_2 z^2 &= a_1 \times (2b_2 z^2) + a_2 \times (b_1 z^1)^2 \\ \Rightarrow b_2 &= 2b_2 + a_2 \\ \Rightarrow a_2 &= -b_2\end{aligned}$$

Where I used the fact that $a_1 = b_1 = 1$. For the z^3 terms, the only way to get z^3 on the right is if $l = l$ and $k = 3$ or if $l = 2$ and $k = 1, 2$ (cross multiplied terms) and $l = 3$. This is because 3 is a prime number.

$$\begin{aligned}b_3 z^3 &= a_1 \times (3b_3 z^3) + a_2 \times (b_1 z \times 2b_2 z^2 + 2b_2 z^2 \times b_1 z) + a_3 \times (b_1 z)^3 \\ \Rightarrow b_3 &= 3b_3 + (-b_2)(4b_2) + a_3 \\ \Rightarrow a_3 &= 4b_2^2 - 2b_3\end{aligned}$$

4. We wish to find $a_2 = -b_2$, to this end note that:

$$\begin{aligned} b_2 &= \frac{\lambda^3}{2!V\lambda^{3 \times 2}} \int d^3 q_1 \int d^3 q_2 f_{12} \\ &= \frac{1}{2V\lambda^3} \int d^3 q_1 \int d^3 (q_2 - q_1) [e^{-\beta V(|q_2 - q_1|)} - 1] \\ &= \frac{1}{2V\lambda^3} \int d^3 q_1 \int d^3 x [e^{-\beta V(|x|)} - 1] \end{aligned}$$

Where $x = q_2 - q_1$. Provided that the interaction range of the potential is short the integrals can be treated as approximately independent. This follows from the fact that if q_1 starts differing largely from q_2 , the integrand dependent on x becomes negligible ($V \rightarrow 0$). Thus the integral over x cannot vary much with changing q_1 . Hence:

$$\begin{aligned} b_2 &\approx \frac{1}{2V\lambda^3} \left(\int d^3 q_1 \right) \left(\int d^3 x [e^{-\beta V(|x|)} - 1] \right) \\ &= \frac{1}{2\lambda^3} \int d^3 x [e^{-\beta V(|x|)} - 1] \\ &= \frac{1}{2\lambda^3} \int_0^{2\pi} d\phi \int_0^\pi \sin \theta d\theta \int_0^\infty r^2 dr [e^{-\beta V(r)} - 1] \\ &= \frac{2\pi}{\lambda^3} \int_0^\infty r^2 dr [e^{-\beta V(r)} - 1] \end{aligned}$$

Now we can use this to evaluate a_2 .

$$\begin{aligned} a_2 = -b_2 &= -\frac{2\pi}{\lambda^3} \int_0^\infty (e^{-\beta V(r)} - 1) r^2 dr \\ &= -\frac{2\pi}{\lambda^3} \int_0^\alpha (0 - 1) r^2 dr - \frac{2\pi}{\lambda^3} \int_\alpha^\infty (1 - 1) r^2 dr \\ &= \frac{2\pi}{\lambda^3} \int_0^\alpha r^2 dr \\ &= \frac{2\pi\alpha^3}{3\lambda^3} \end{aligned}$$

5. Doing a similar calculation to the above:

$$\begin{aligned} a_2 &= -\frac{2\pi}{\lambda^3} \int_0^\infty (e^{-\beta V(r)} - 1) r^2 dr \\ &= -\frac{2\pi}{\lambda^3} \left[\int_0^\alpha (0 - 1) r^2 dr + \int_\alpha^{2\alpha} (e^{\beta v_0/kT} - 1) r^2 dr - \int_{2\alpha}^\infty (1 - 1) r^2 dr \right] \\ &= -\frac{2\pi}{\lambda^3} \left[-\frac{\alpha^3}{3} + (e^{\beta v_0} - 1) \left(\frac{8\alpha^3}{3} - \frac{\alpha^3}{3} \right) \right] \\ &= -\frac{2\pi\alpha^3(7e^{\beta v_0/kT} - 8)}{3\lambda^3} \end{aligned}$$

It can be seen from this that a_2 is negative for low T , and is positive for high T when the exponential is large and small respectively. Thus a_2 does change sign with temperature.