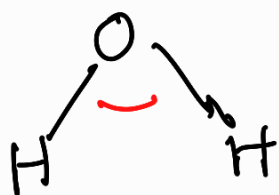


CHM616 - Assignment #1

Due Date: 30/1/2022

Q1. For H_2O & H_2O_2 Molecules, Construct both xyz & z-matrix form geometries and perform RHF/6-31G calculation. Report nuclear repulsion energy & RHF energy for each calculation. Verify that xyz & z-matrix calc for each molecule gives same results. The geometry for each molecule is given below.

H_2O :



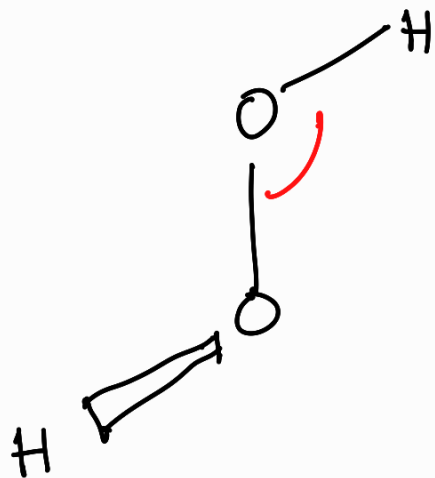
$$R(\text{OH}) = 0.95 \text{ \AA}$$

$$A(\text{HOH}) = 104.5^\circ$$

xyz: Molecule must be in xy plane,

with oxygen at the origin.

H₂O₂:



$$R(O-O) = 1.47 \text{ \AA}$$

$$R(O-H) = 0.95 \text{ \AA}$$

$$A(HOO) = 95^\circ$$

$$D(H-O-O-H) = 90^\circ$$

XYZ: One oxygen at origin, the other oxygen along +ve z-axis;
One (O-O-H) must be in xz-plane & the other O-O-H must be in zy-plane.

(contd)

Q2. Setup RHF calculations on Ne atom for the following basis sets. Note that basis set is to be set using \$BASIS input group.

- ① MINI
- ② MIDI
- ③ STO-3G (STO with NGAUSS=3)
- ④ 3-21G (N21 with NGAUSS=3)
- ⑤ 4-31G (N31 - - - = 4)
- ⑥ 6-31G (N31 - - - = 6)
- ⑦ 6-311G (N311 - - - = 6)
- ⑧ CC-pVDZ (GBASIS = CCD)
- ⑨ CC-pVTZ (- , - = CCT)
- ⑩ CC-pVQZ (- , - = CCQ)

For each basis set, collect following info from output file.

- ① No of Basis functions (size)
- ② RHF energy

Prepare the following table sorted with increasing basis set size

<u>Basis</u>	<u># of Basis Funs</u>	<u>RHF Energy</u>
.	.	.
.	.	.
.	.	.

Make a graph of # of Basis Funs
of x-axis & RHF Energy on y-axis.

Q3. Construct Symbolic Z-matrix

for the following molecules &

perform RHF/6-31G calculation.

Use some sensible values for c-c,

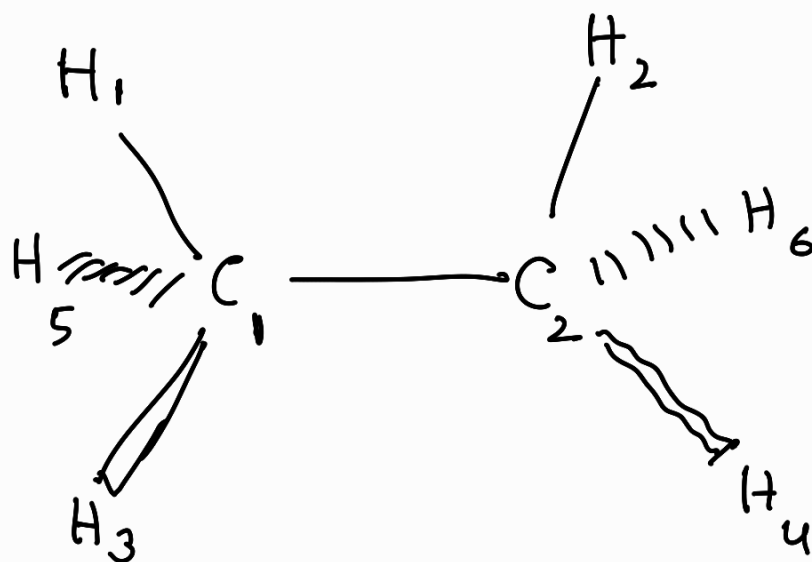
c=c, c-h, c=o, c-o bonds,

and angles and dihedrals. Rough

geometry is indicated in these

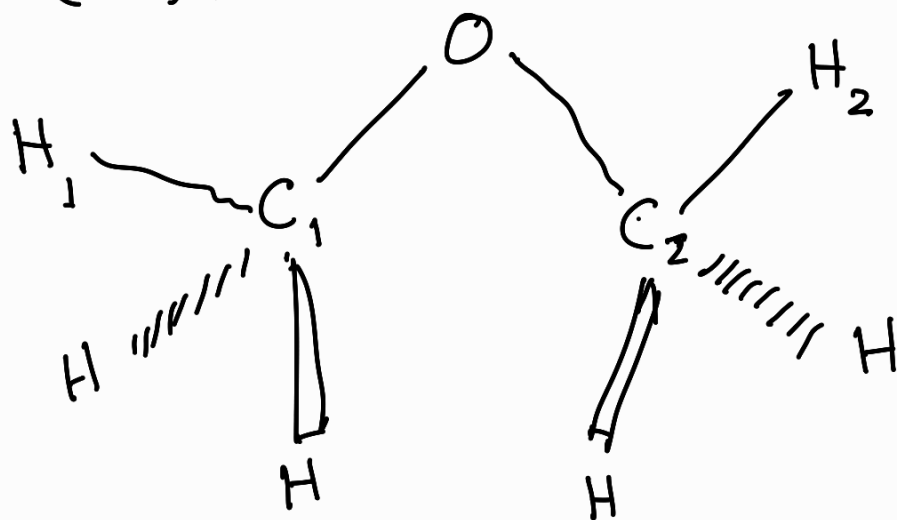
figures.

(a) C_2H_6

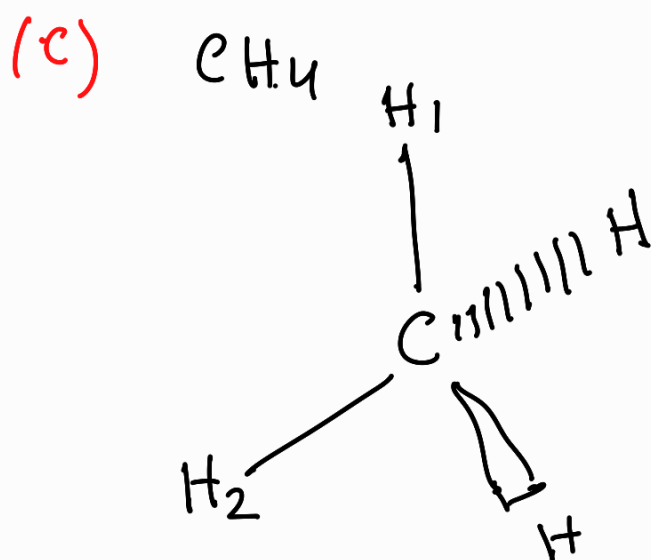


Here , H_1, C_1, C_2, H_2 are in plane
 H_3 & H_4 are in front & H_5 & H_6 at
 back.

(b) $(CH_3)_2O$



Here , H_1, H_2, C_1, C_2 & O are in
 a plane.

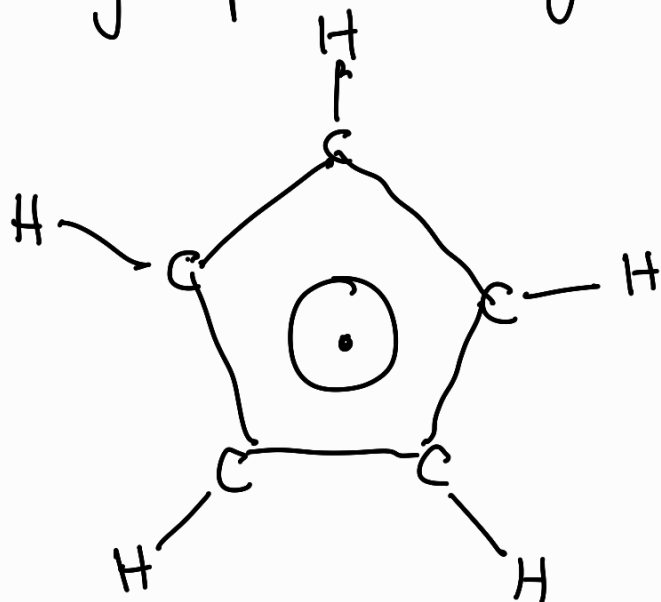


(d) C_2H_2

(Linear)



(e) Cyclopentadienyl anion ($C_5H_5^-$)



(D_{5h} Symmetry)

Use of dummy atoms is encouraged.