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In [84]: ### from thermostate import State, Q_, units
                    from math import pi
                    print('Jason Secula')
                    print('Homework 3.3')
                    print('
                    substance = 'air'
                    R Cnst = Q (8.3145498*(10**3),'(kJ)/(K*kmole)')
                    R_Bar = Q_(28970, '(kg)/(kmole)') #Had to Convert from Joules, ThermoState
                    R = (R Cnst/R Bar) \#R = (8.31(kJ)/(kmol-K))/(28.97kg/kmol)
                    \#^{^R} = 0.2870055160510874 \text{ kilogram kelvin}
                   T 1 = Q (27.0, 'degC')
                    p_1 = Q_(101.0, 'kPa')
                    V 1 = Q (566.0, 'cm**3')
                    r = Q (8.0, 'dimensionless')
                    T_4 = Q_{(1726.0, 'degC')}
                    #^ ^Given-^-^-^-^-^-^-^-^-
                    #State 1
                    st 1 = State(substance, T=T 1, p=p 1)
                    #State 2
                    V 2 = (V 1/r)
                    v = ((V 1/V 2)*st 1.v)
                    st_2 = State(substance, s=st_1.s, v=v_2)
                    #State 3
                    st_3 = State(substance, T=T_1, v=v_2)
                    #State 4
                    st 4 = State(substance, T=T 4, s=st 3.s)
                    denominator = (p_1*V_1).to('kJ') #Dimensionality issue, had to convert file
                    m = (denominator/(R*T 1)) #Mass
                    print ('The mass was found to be = {}'.format(m))
                    Wnet = (((st_3.u -st_4.u) - (st_2.u - st_1.u)))
                    print('1) The Net Work Output = {}'.format(Wnet)) #W_net in kJ/kg
                    eta = (1-((st 4.u-st 1.u)/(st 3.u-st 4.u)))
                    print(('2) The Thermal Efficiency is = {}').format(eta))
                    MEP = (Wnet/(V 2-V 1)) #.to('bar') #V d = V btc-V tdc
                   print('3) The MEP is = {}'.format(MEP))
                    #ColdAirStandardAnalysis
                    print('4) The cold air is as followed')
                    cv = ((st_3.u-st_2.u)/(st_3.T-st_2.T)) \#DeltaU=cv*DeltaT >> Cv = DeltaU/DeltaU=cv*DeltaT >> Cv = DeltaU/DeltaU=cv*DeltaU=cv*DeltaT >> Cv = DeltaU/DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*DeltaU=cv*
                    #^ ^ ^CV = 715.8716134760114 joule / kelvin / kilogram^ ^ ^ ^ ^
                    cp = (cv+R)
                    #^ ^ ^CP = 1002.8771295270988 joule / kelvin / kilogram^_^_^_^_
                    \#k = (cp/cv)
                    k = (1+(cv/R))
                    \#k = 3.4942782401038777 dimensionless
                    #One
                    v_1ca = ((R*T_1)/p_1) #ca = Cold Air
                    #Two
                   T 2ca = (T 1*(r**(k-1)))
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p 2ca = (p 1*(r**k))
v 2ca = (v 1ca*r)
#Three
v 3ca = v 2ca
p 3ca = (p 2ca*((v 2ca/v 3ca)**k)).to('kg/(m*s**2)') #Page 512 in Text Boo
T 3ca = (T 2ca*(p 3ca/p 2ca))
#Four
p 4ca = p 3ca.to('kg/(m*s**2)')
v_4ca = (r*v_3ca) #Original: (p_4ca/(R*T_4)): Demensionality Error...
\#^{^} r = (v_4/v_3) for an otto system.
W 12 = (m*(cv*(T 1-T 2ca))).to('kJ')
W_34 = (m*(p_3ca*(v_4ca-v_4ca))).to('kJ')
Q in = (m*(cv*(T 3ca-T 2ca))).to('kJ') #Originally used this for Eta, but
Wnet ca = (W 12 - W 34)
print(' 1.b) The Net Work Output = {}'.format(Wnet ca)) #W net in kJ/kg
eta ca = (1-(1/(r**(k-1))))
print(' 2.b) The Thermal Efficiency is = {}'.format(eta ca))
MEP ca = (((Wnet ca/(v 1ca-v 2ca)))*m)
print(' 3.b) The MEP is = {}'.format(MEP ca))
```

Jason Secula Homework 3.3

The mass was found to be = 0.0006636043338180686 kilogram

- 1) The Net Work Output = -1340853.8127436675 joule / kilogram
- 2) The Thermal Efficiency is = 2.0001209117904235 dimensionless
- 3) The MEP is = 2707.428193323912 joule / centimeter ** 3 / kilogram
- 4) The cold air is as followed
- 1.b) The Net Work Output = -25.363297626730944 kilojoule
- 2.b) The Thermal Efficiency is = 0.9944096075523767 dimensionless
- 3.b) The MEP is = 0.0028190947579642568 kilogram ** 2 * kilopascal