

Recitation Class 1

Zexi Li

lzx12138@sjtu.edu.cn

2021.05.18

Outline

Chapter 1: Crystalline structure of solids

Chapter 2: Quantum Mechanics

Chapter 3: Introduction to the Quantum Theory of Solids

Table of Contents

Chapter 1: Crystalline structure of solids

Chapter 2: Quantum Mechanics

Chapter 3: Introduction to the Quantum Theory of Solids

Semiconductor Materials

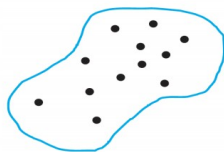
Conductors	Semiconductors	Insulators
$< 10^{-3} \Omega \cdot \text{cm}$	$10^{-3} - 10^9 \Omega \cdot \text{cm}$	$> 10^9 \Omega \cdot \text{cm}$
Metals (Au, Al, Cu, Hg, ...)	Si, Ge, GaAs, InP, ...	SiO ₂ , HfO ₂ , ...
Solids, liquids (Hg)	Solids	Solids, liquids gases

Table: Semiconductor

Semiconductors are the materials that have resistivities between $10^{-3} - 10^9 \Omega \cdot \text{cm}$ depending on light illumination, temperature, electric field, magnetic field and impurities.

Type of Solids

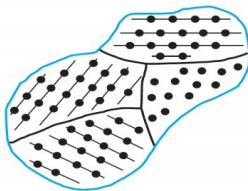
■ Amorphous



(a)

All atoms or ions are periodically ranged in a short range (a few atoms)

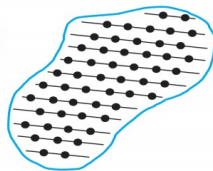
■ Polycrystals



(b)

Multiple crystalline grains randomly packed

■ Single crystals

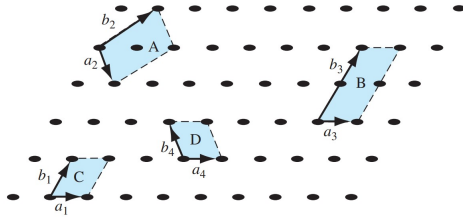


(c)

All atoms or ions are periodically ranged in a long range (μm scale)

All semiconductors covered in this course are assumed to be single crystalline.

Primitive and unit cell



Unit Cell: small volume of the crystal that can be used to reproduce the entire crystal.

A unit cell is not a unique entity

Primitive Cell: the smallest unit cell that can be repeated to form the lattice.

Lattice types

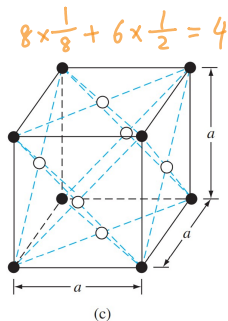
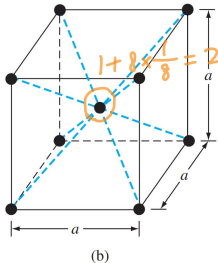
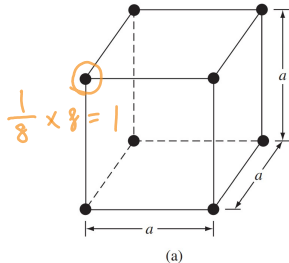


Figure: (a) simple cubic(sc), (b) body-centered cubic(bcc), (c) face-centered cubic(fcc)

#number of atoms per unit cell

$$\text{Volume Density} = \frac{\# \text{ atoms per unit cell}}{\text{volume of unit cell}}$$

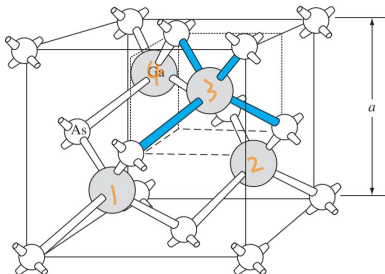
$$\text{Surface Density} = \frac{\# \text{ atoms per lattice plane}}{\text{area of lattice plane}}$$

(Example after reviewing Miller index)

The diamond structure

The diamond structure all atoms are of the same species

The zincblende structure two different types of atoms. e.g, GaAs.



$$\begin{aligned} &4 + 6 \times \frac{1}{2} \\ &+ 8 \times \frac{1}{8} \\ &= 8 \end{aligned}$$

The diamond structure: To help remember

Method 1:

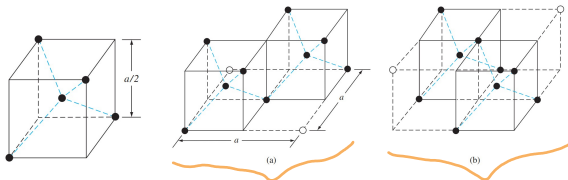
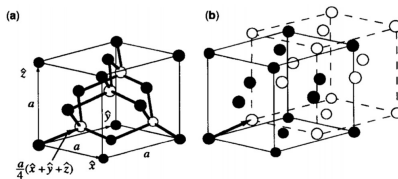


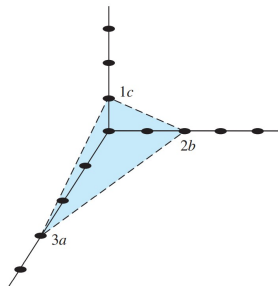
Figure: Tetrahedral structure & (a) bottom half, (b) top half

Method 2:



Equivalent to two face-centered cubics sliding $1/4$ diagonal length along a diagonal.

Crystalline Plane and Miller Index



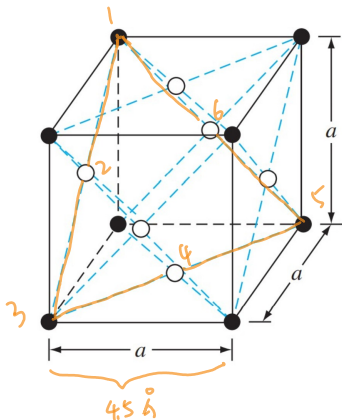
$$(3, 2, 1) \xrightarrow{\text{Reciprocal}} \left(\frac{1}{3}, \frac{1}{2}, 1\right) \xrightarrow{\text{multiply lcd}} (2, 3, 6)$$

Any parallel plane is entirely equivalent to any other.

The $[hkl]$ direction is perpendicular to the (hkl) plane.

Example: Surface density

The lattice constant of a single crystal is 4.50 \AA . Calculate the surface density of atoms ($\#$ per cm^2) on the plane (111) for face-centered cubic lattice.



$$3 \times \frac{1}{6} + 3 \times \frac{1}{2} = 2$$

$$\frac{\sqrt{3}}{4} (\sqrt{2} a)^2$$

$$\text{Surface density} = \frac{2}{\frac{\sqrt{3}}{4} (\sqrt{2} a)^2}$$

Doping

n-type semiconductors: Charge carriers are negative, i.e. electrons doped by donor-type of dopants.

p-type semiconductors: Charge carriers are positive, i.e. holes doped by acceptor-type of dopants.

Table of Contents

Chapter 1: Crystalline structure of solids

Chapter 2: Quantum Mechanics

Chapter 3: Introduction to the Quantum Theory of Solids

Basic concepts

Wave function:

$$\Psi(x)$$

Probability density function:

$$|\Psi(x)|^2 = \Psi(x) \cdot \Psi^*(x)$$

Schrodinger Equation:

$$-\frac{\hbar^2}{2m} \cdot \frac{\partial^2 \Psi(x, t)}{\partial x^2} + V(x) \Psi(x, t) = j\hbar \frac{\partial \Psi(x, t)}{\partial t}$$

$$\boxed{\frac{\partial^2 \psi(x)}{\partial x^2} + \frac{2m}{\hbar^2} (E - V(x)) \psi(x) = 0}$$

E : total energy of the particle.

Infinite quantum well

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + V(x)\Psi = E\Psi, \quad \begin{cases} V(x) = +\infty, & x \leq 0 \text{ or } x \geq a \\ V(x) = 0, & 0 < x < a \end{cases}$$

General solution:

$$\Psi(x) = Ae^{-ikx} + Be^{ikx}$$

Boundary condition:

$$\Psi(x)|_{x=0,a} = 0$$

$$\int_0^a \Psi(x)\Psi^*(x) dx = 1$$

conclusion:

$$k = \frac{n\pi}{a}, n = 0, \pm 1, \pm 2, \dots$$

$$E = \frac{k^2 \hbar^2}{2m} = \frac{n^2 \pi^2 \hbar^2}{2ma^2}$$

Finite quantum well

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + V(x)\Psi = E\Psi, \quad \begin{cases} V(x) = \overset{+\infty}{V_0}, & x \leq 0 \text{ or } x \geq a \\ V(x) = 0, & 0 < x < a \end{cases}$$

General solution:

$$\Psi(x) = \begin{cases} Ae^{-ik_1 x} + Be^{ik_1 x}, & k_1 = \sqrt{\frac{2m(E-V_0)}{\hbar^2}}, \quad x \leq 0 \text{ or } x \geq a \\ Ce^{-ik_2 x} + De^{ik_2 x}, & k_2 = \sqrt{\frac{2mE}{\hbar^2}}, \quad 0 < x < a \end{cases}$$

Boundary condition:

$$\Psi(x)|_{x=0} \text{ continuous}$$

$$\Psi(x)|_{x=a} \text{ continuous}$$

$$\int_{-\infty}^{\infty} \Psi(x)\Psi^*(x) dx = 1$$

$$E < V_0$$

$$k_1 \quad \underline{A \cdot e^{-kx} + B e^{kx}}$$

$$E > V_0$$

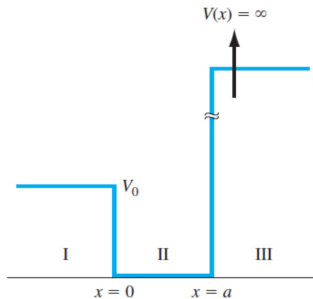
$$k_1$$

Note: depending on the relationship between E and V_0 , $\Psi(x)$ is different.

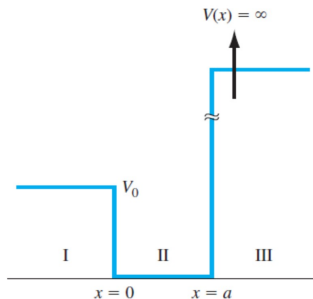
Example

Consider the one-dimensional potential function shown in the figure below. Assume the total energy of an electron is $E < V_0$.

- Write the wave solutions that apply in each region.
- Write the set of equations that result from applying the boundary conditions.
- Show explicitly why, or why not, the energy levels of the electron are quantized.



Example



$$\Psi(x) = \begin{cases} Ae^{-ik_1x} + Be^{ik_1x}, & x \leq 0 \\ Ce^{-ik_2x} + De^{ik_2x}, & 0 < x < a \\ 0, & x \geq a \end{cases}$$

$$\vec{E} = \dots$$

$$\begin{cases} \Psi(x)|_{x=0} \text{ continuous} \\ \Psi(x)|_{x=a} \text{ continuous} \\ \int_{-\infty}^{\infty} \Psi(x)\Psi^*(x) dx = 1 \end{cases}$$

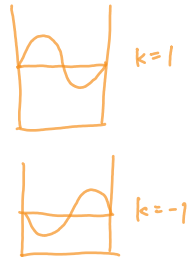
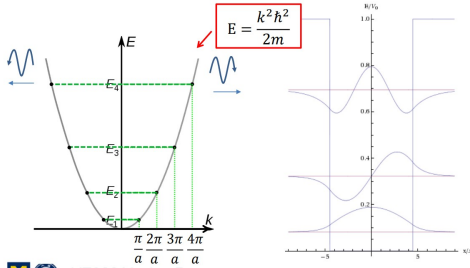
Table of Contents

Chapter 1: Crystalline structure of solids

Chapter 2: Quantum Mechanics

Chapter 3: Introduction to the Quantum Theory of Solids

Energy bands



For same energy level, the k can have two values, Because the wave can move to positive and negative directions.

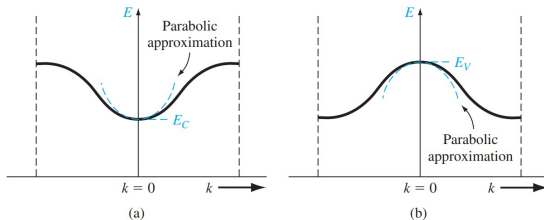
Effective mass

$$E = \frac{p^2}{2m} = \frac{k^2 \hbar^2}{2m}$$

Taking derivative:

$$\frac{dE}{dk} = \frac{\hbar^2 k}{m}$$

$$\frac{1}{\hbar^2} \frac{d^2 E}{dk^2} = \frac{1}{m}$$



$$E = C \cdot k^2$$

Figure: (a) The conduction band in reduced k space, and the parabolic approximation. (b) The valence band in reduced k space, and the parabolic approximation

$$E - E_c = C_1 k^2$$

Effective mass

$$\boxed{\frac{1}{\hbar^2} \frac{d^2 E}{dk^2} = \frac{2C_1}{\hbar^2} = \frac{1}{m^*}}$$

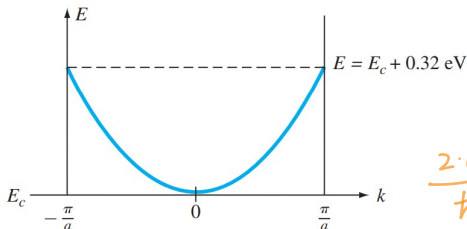
$$E = E(k) = E_c + \frac{\hbar^2}{2m_n^*} (k - k_1)^2$$

$$E = E(k) = E_v - \frac{\hbar^2}{2m_p^*} (k - k_2)^2$$

Example

A simplified E versus k curve for an electron in the conduction band is given. The value of a is 10 \AA . Determine the relative effective mass m^*/m_0 .

1.175



Answer:

$$E - E_c = C \cdot k^2 = 0.32 \text{ eV}$$

$$k = \frac{\pi}{a}$$

C

$$\frac{2 \cdot C}{\hbar^2} = \frac{1}{m^*}$$

m^*

$$m_0 = 9.1 \times 10^{-31}$$

End