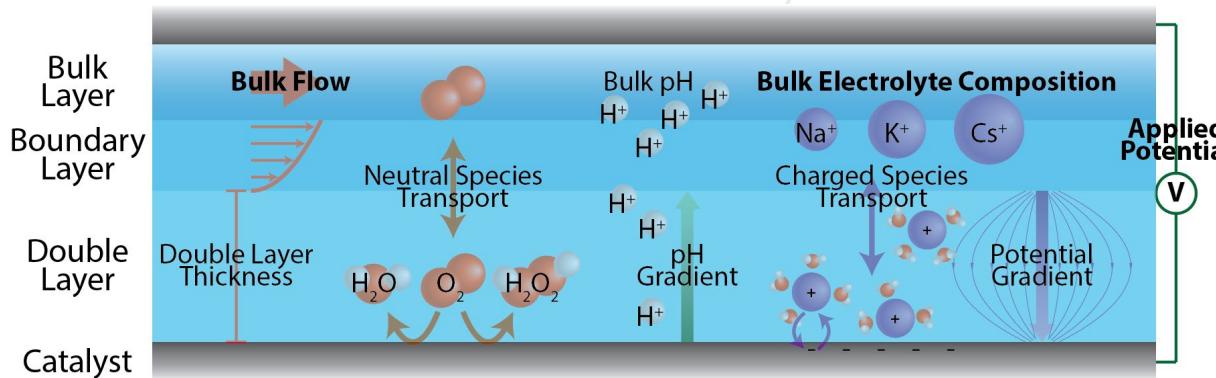


# Mathematical investigation of Bulk and Local Effects in Electrocatalysis



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Eng. Sci. & Applied Math  
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SUSTAINABILITY AND ENERGY



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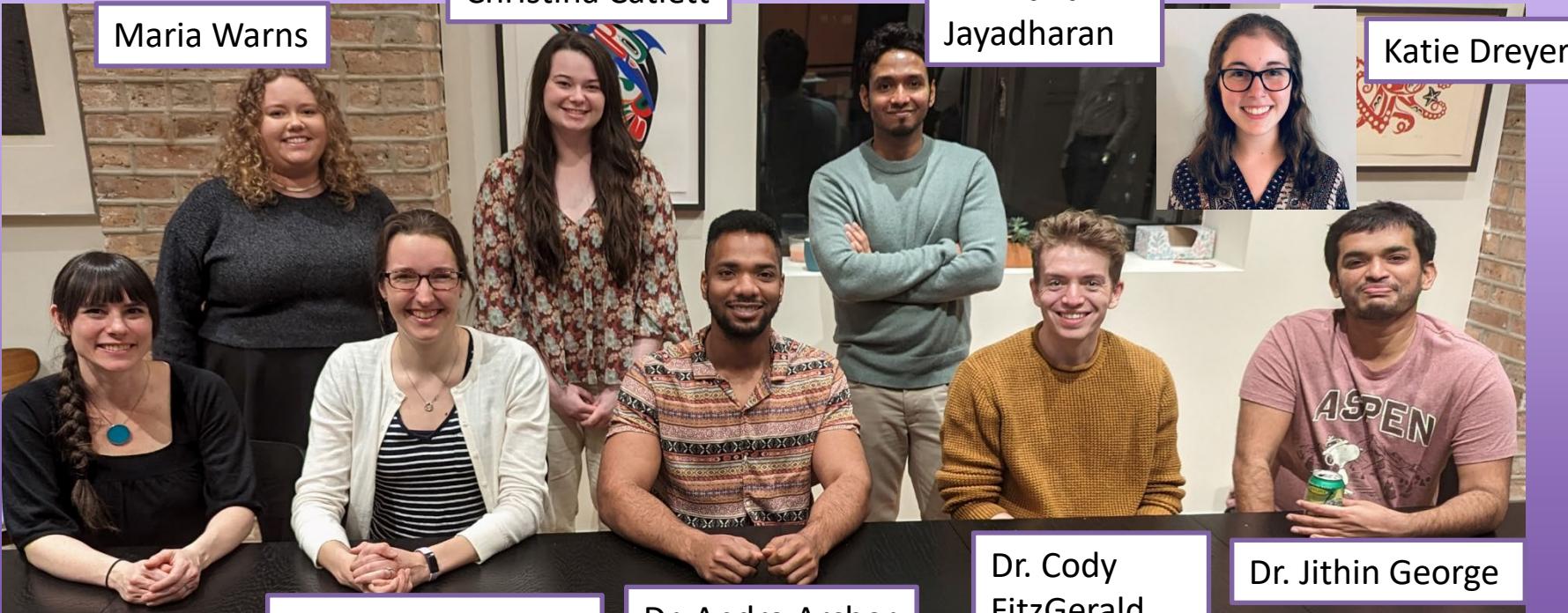
**DEVCOM**  
CHEMICAL BIOLOGICAL  
CENTER

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PAULA M. TRIENENS INSTITUTE FOR  
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**WALDER**  
FOUNDATION



Yifan Zhang



Grace Hooper



Yash Deodhar



Dr. Alasdair  
Hastewell

NU Center for  
Sustainability and  
Engineering  
Resilience

# Reaction Environment Tuning

Prof. Linsey Seitz

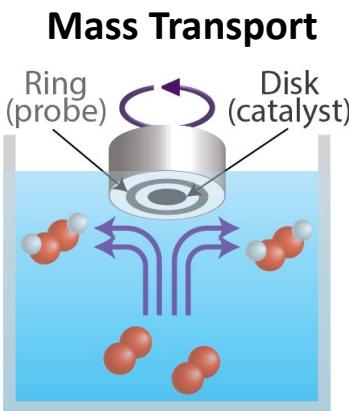
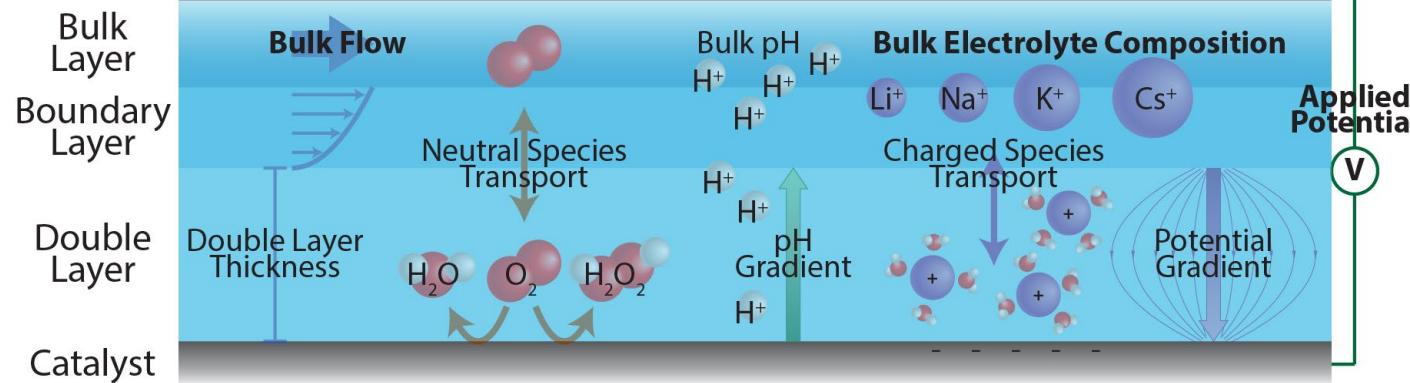
Dr. Jithin George



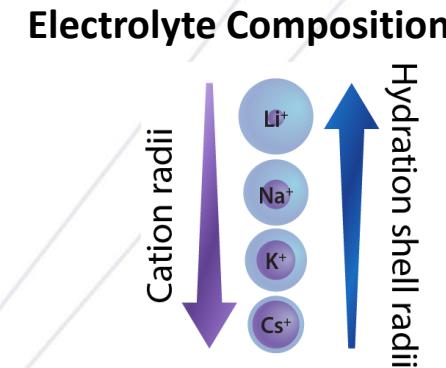
Dr. Brianna Ruggiero



Yash Deodhar

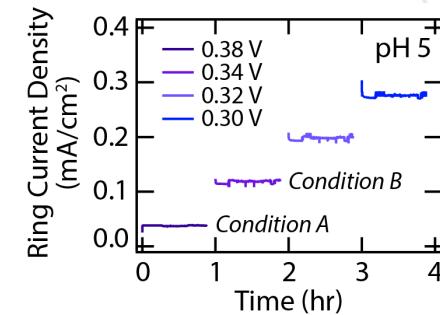


How do **macroscale** system properties...



...affect **microscale** environment at the catalyst surface...

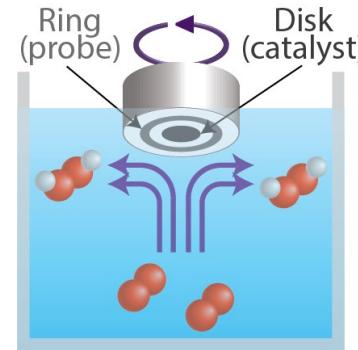
**Operating Potential / Current**



...which impact **reactor performance** and **product selectivity**?



# pH effects on reactor performance



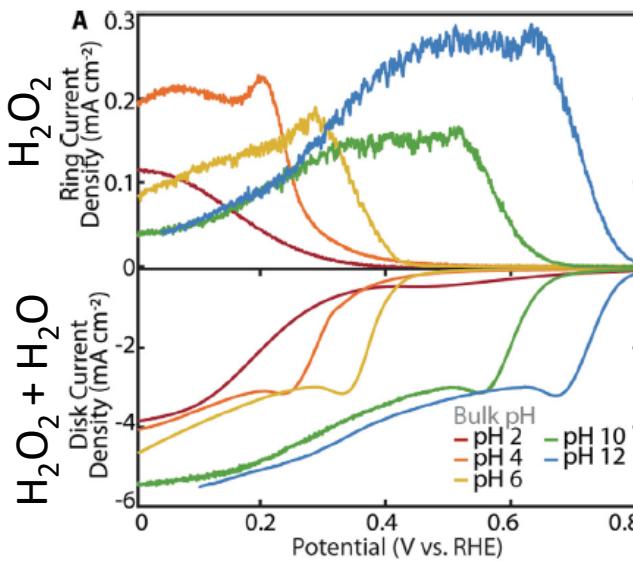
## Experimental Set up:

- Oxygen reduction reaction (ORR) for peroxide production
- Mesoporous carbon black CMK-3 catalyst
- Rotating Ring-Disk electrode
- Iridium oxide local pH sensing

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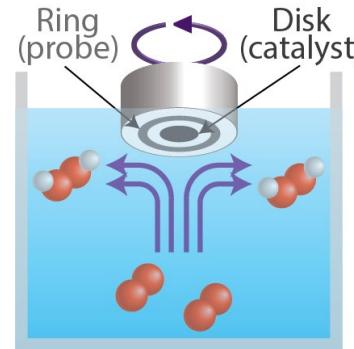


## Linear Sweep Voltammetry





# pH effects on reactor performance



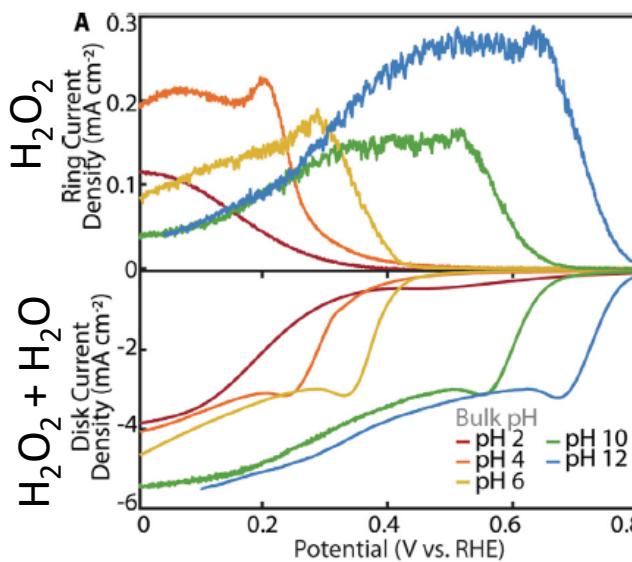
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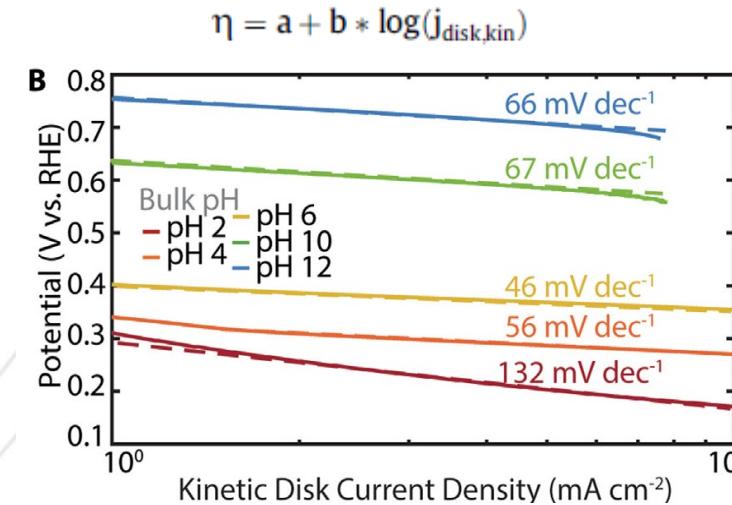
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## Linear Sweep Voltammetry

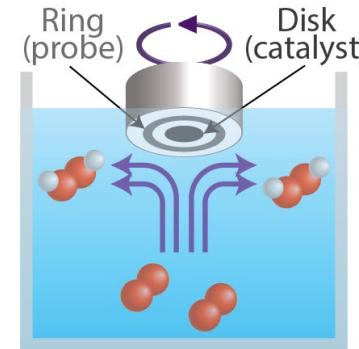


## Tafel slope approximation





# pH effects on reactor performance



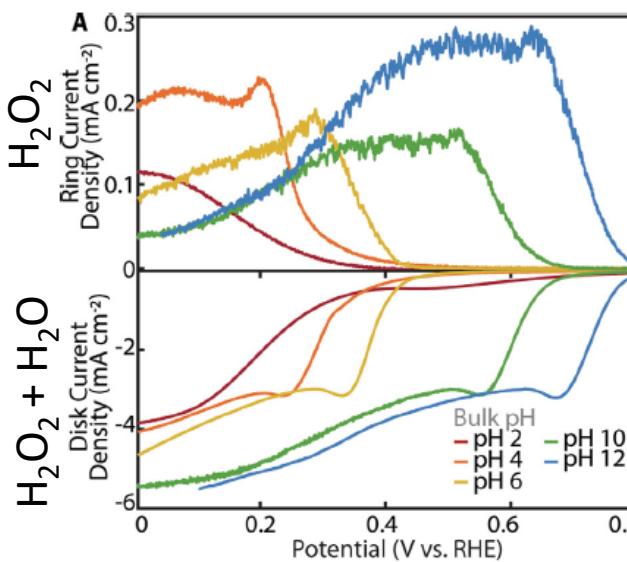
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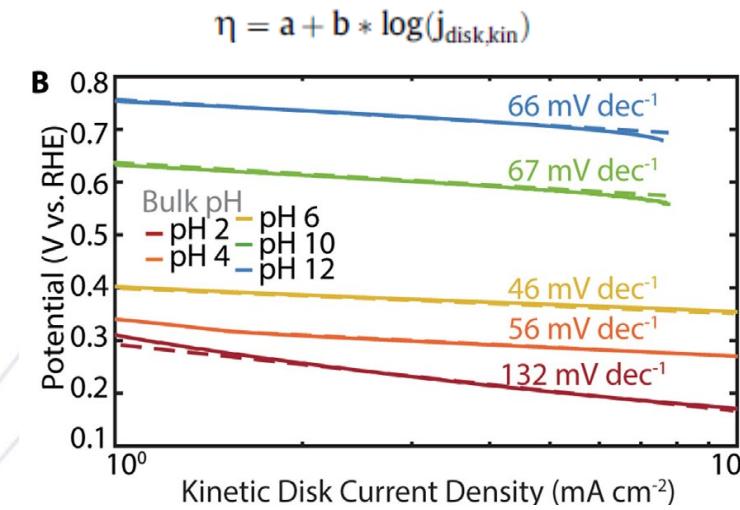
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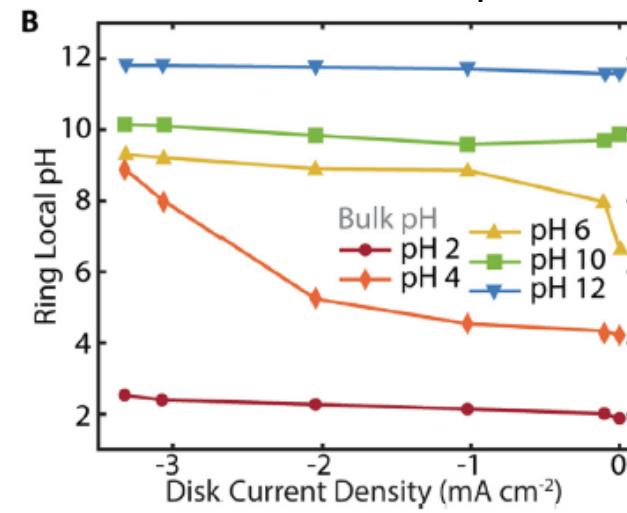
## Linear Sweep Voltammetry



## Tafel slope approximation

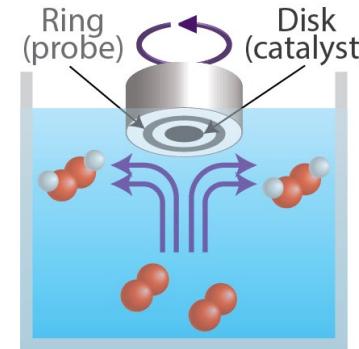


## Local vs bulk pH





# pH effects on reactor performance



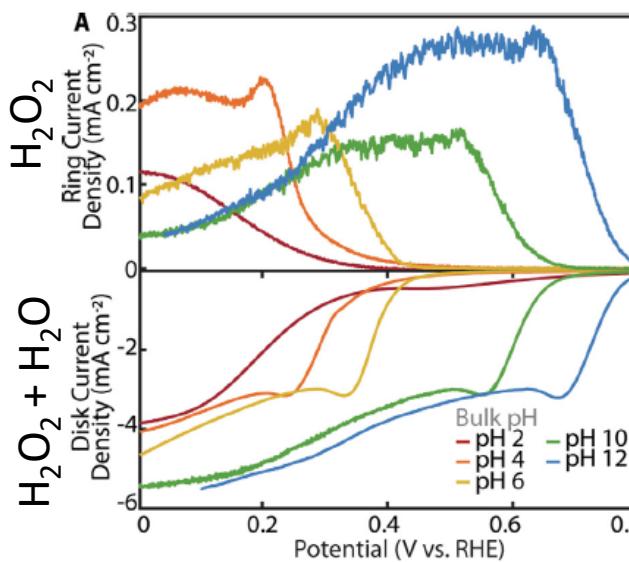
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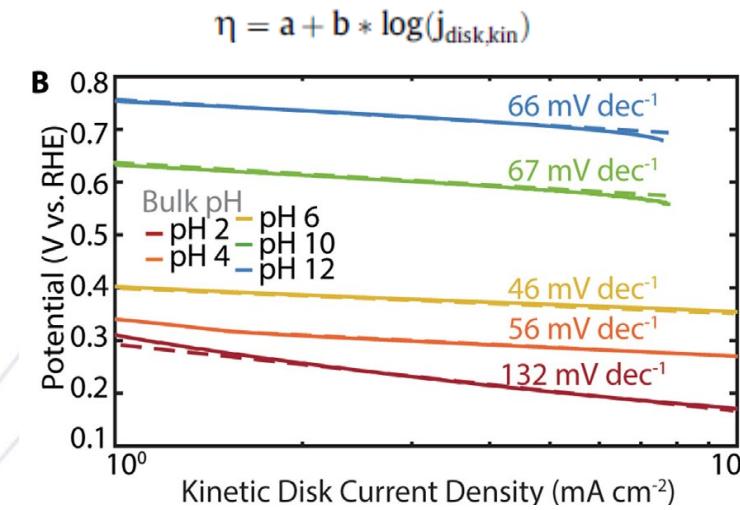
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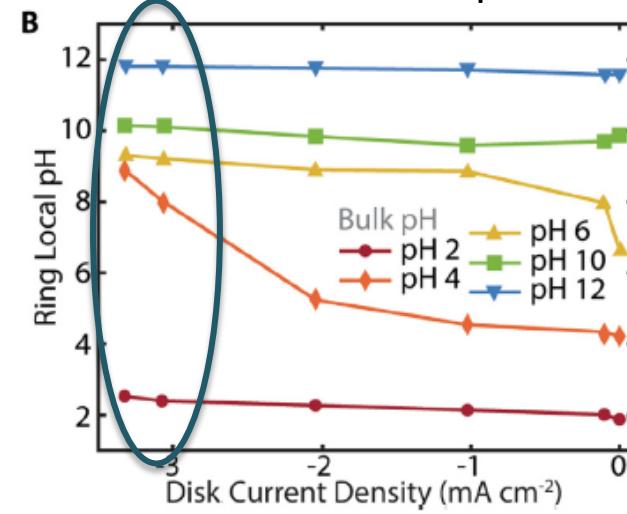
## Linear Sweep Voltammetry



## Tafel slope approximation

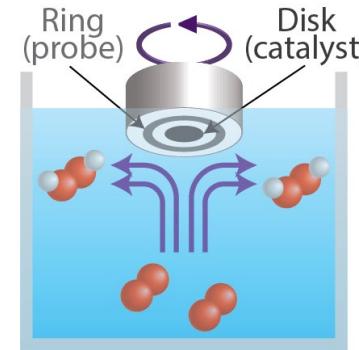


## Local vs bulk pH





# pH effects on reactor performance



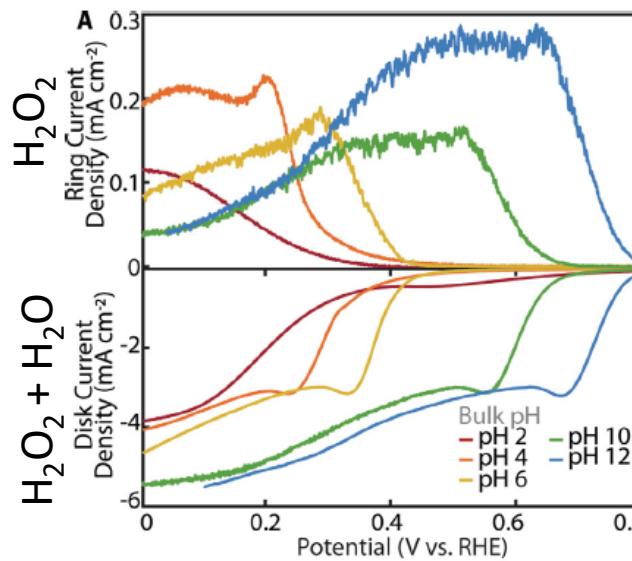
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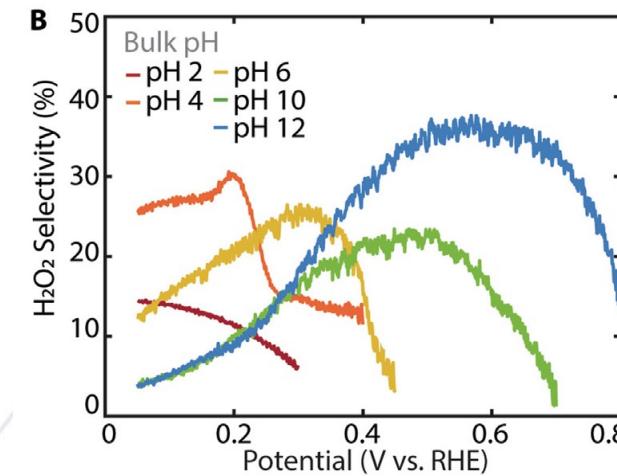
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Linear Sweep Voltammetry

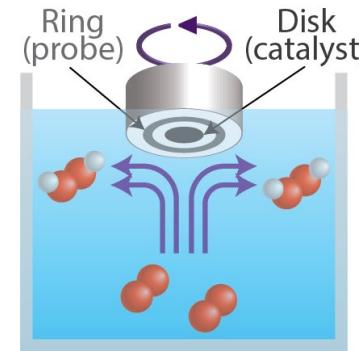


Selectivity





# pH effects on reactor performance



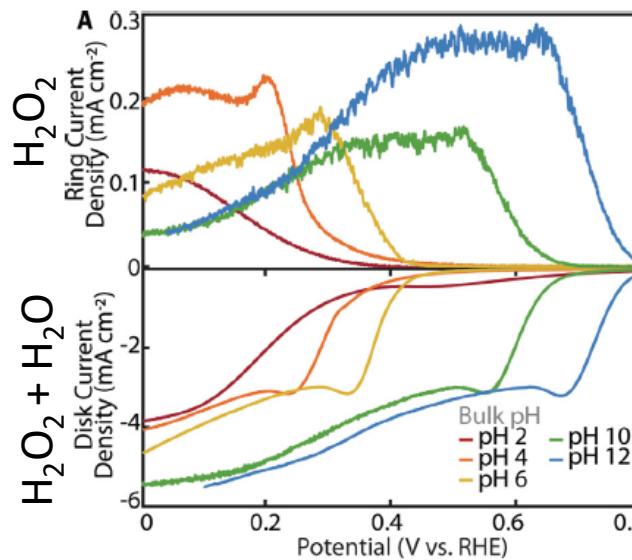
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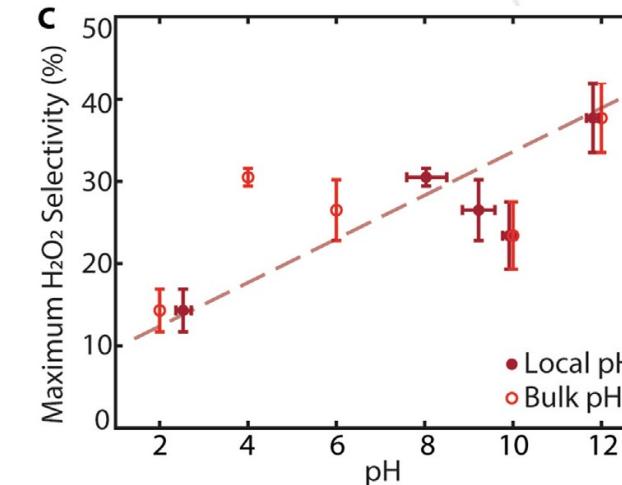
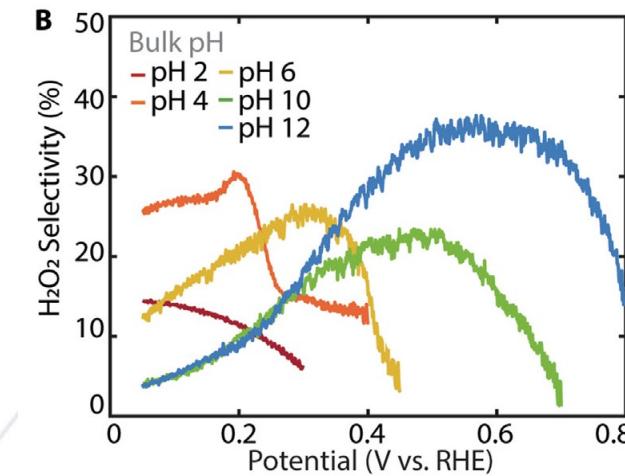
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Linear Sweep Voltammetry



Selectivity

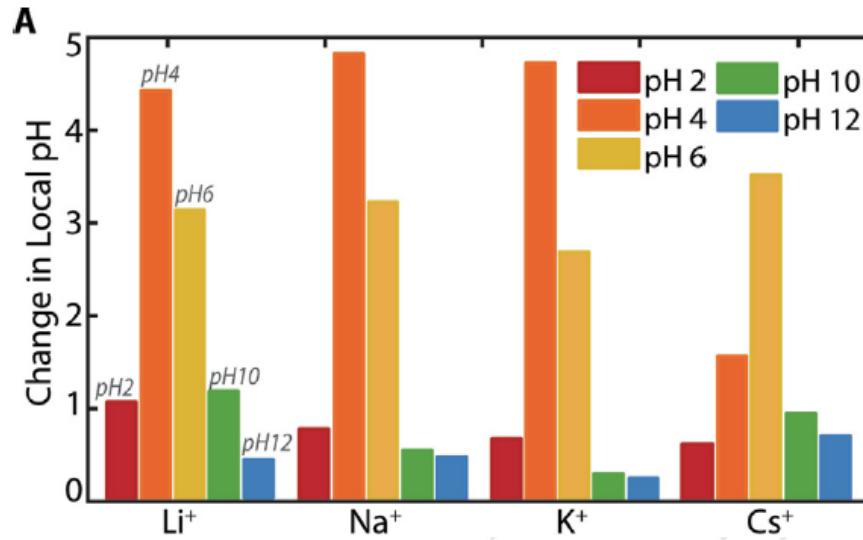
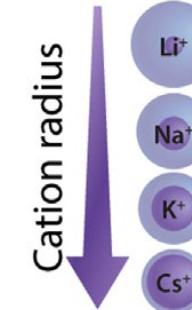
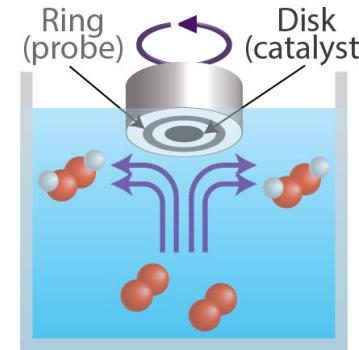




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# Cation effects on reactor performance

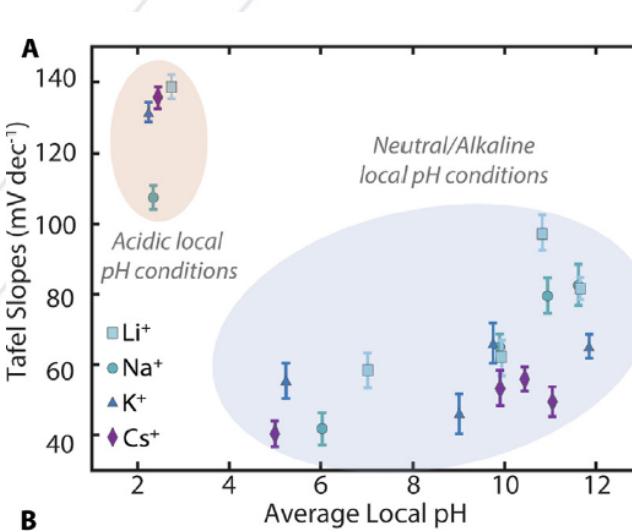
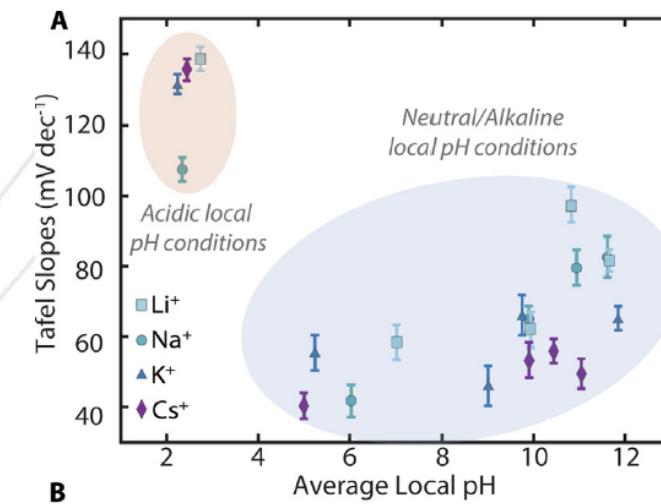
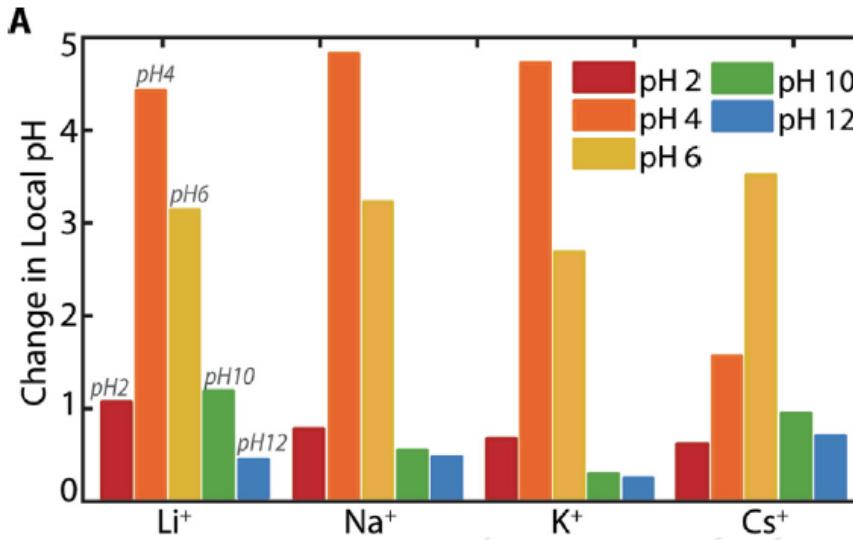
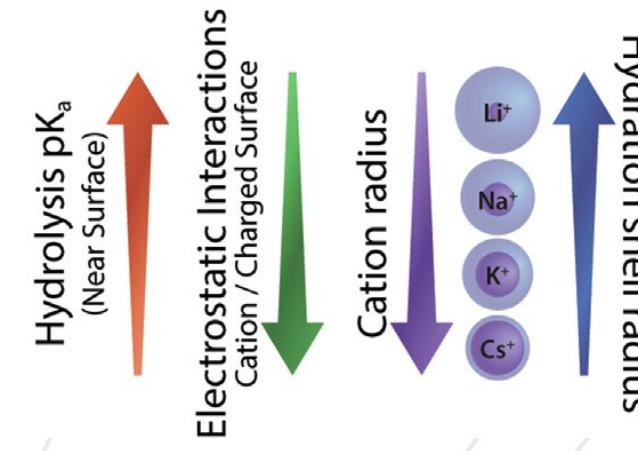
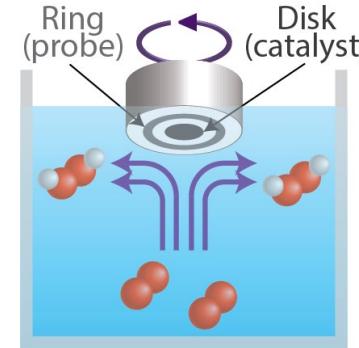




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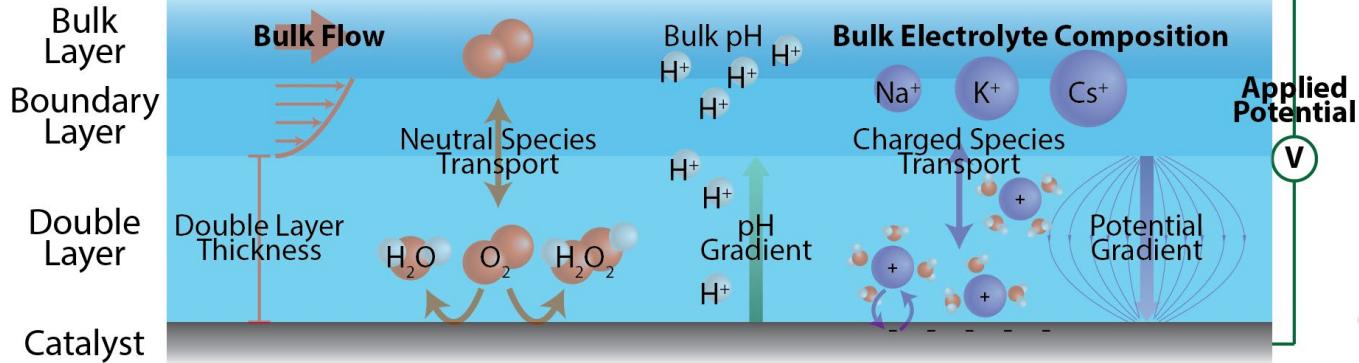


# Cation effects on reactor performance

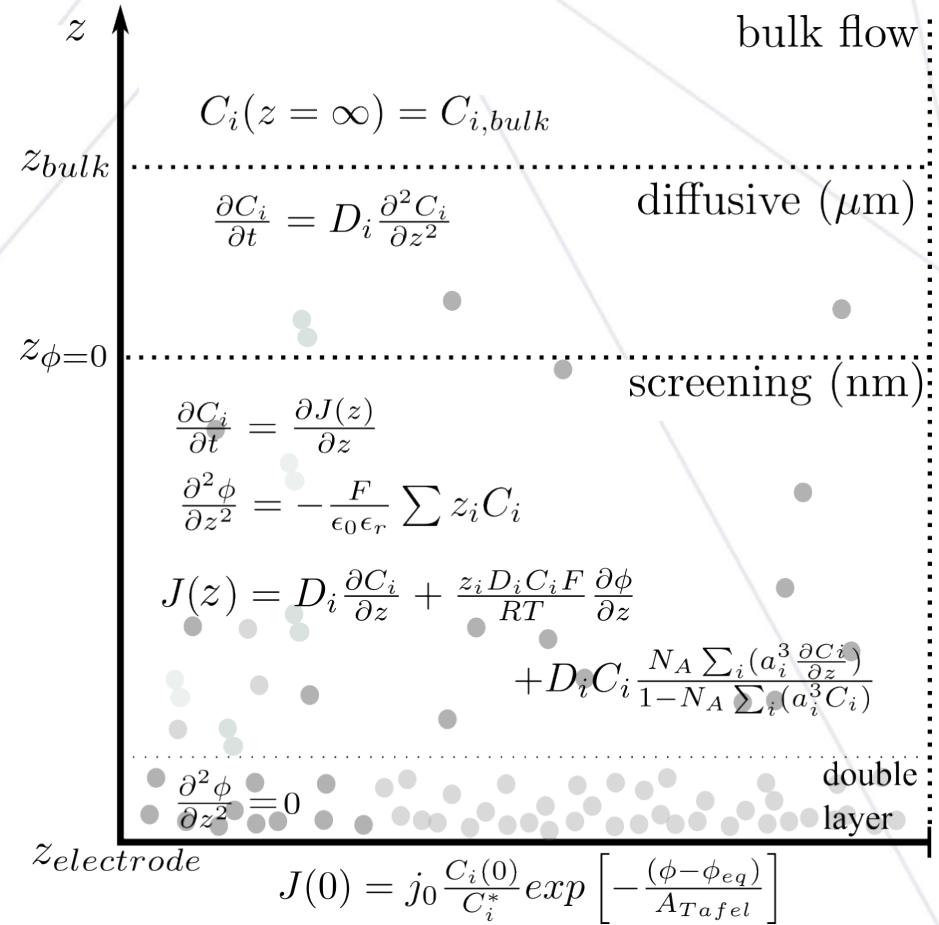




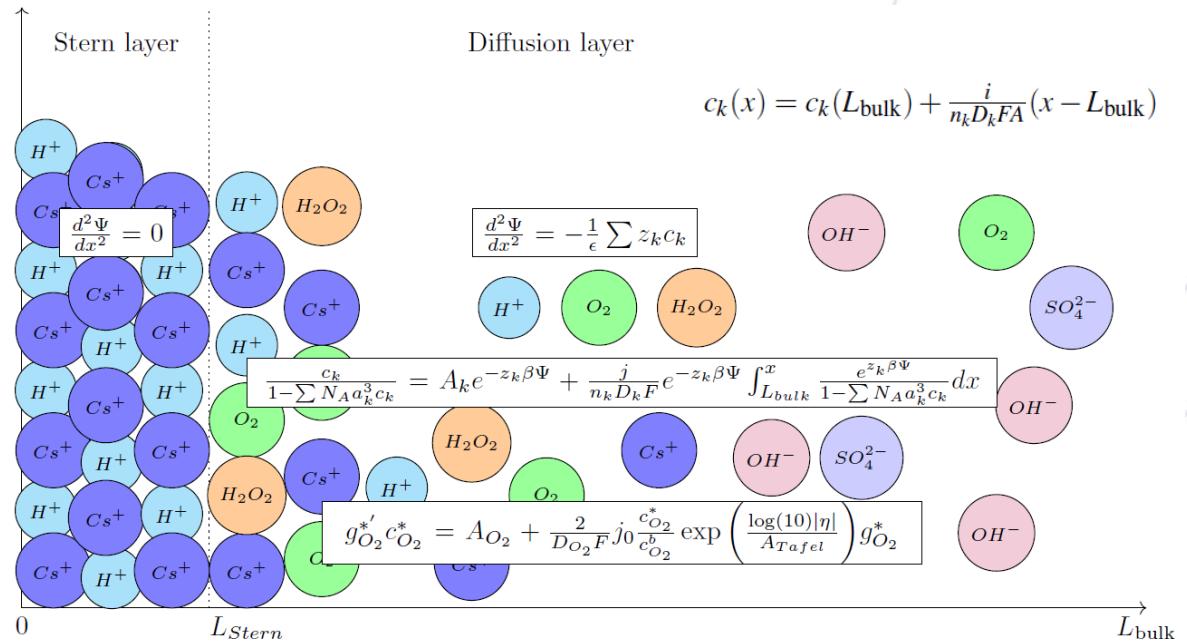
# Electrocatalytic reactor modeling



- Solve at steady state  $\frac{\partial C}{\partial t} \approx 0$
- Modified Poisson-Boltzmann equation
- Butler-Volmer boundary

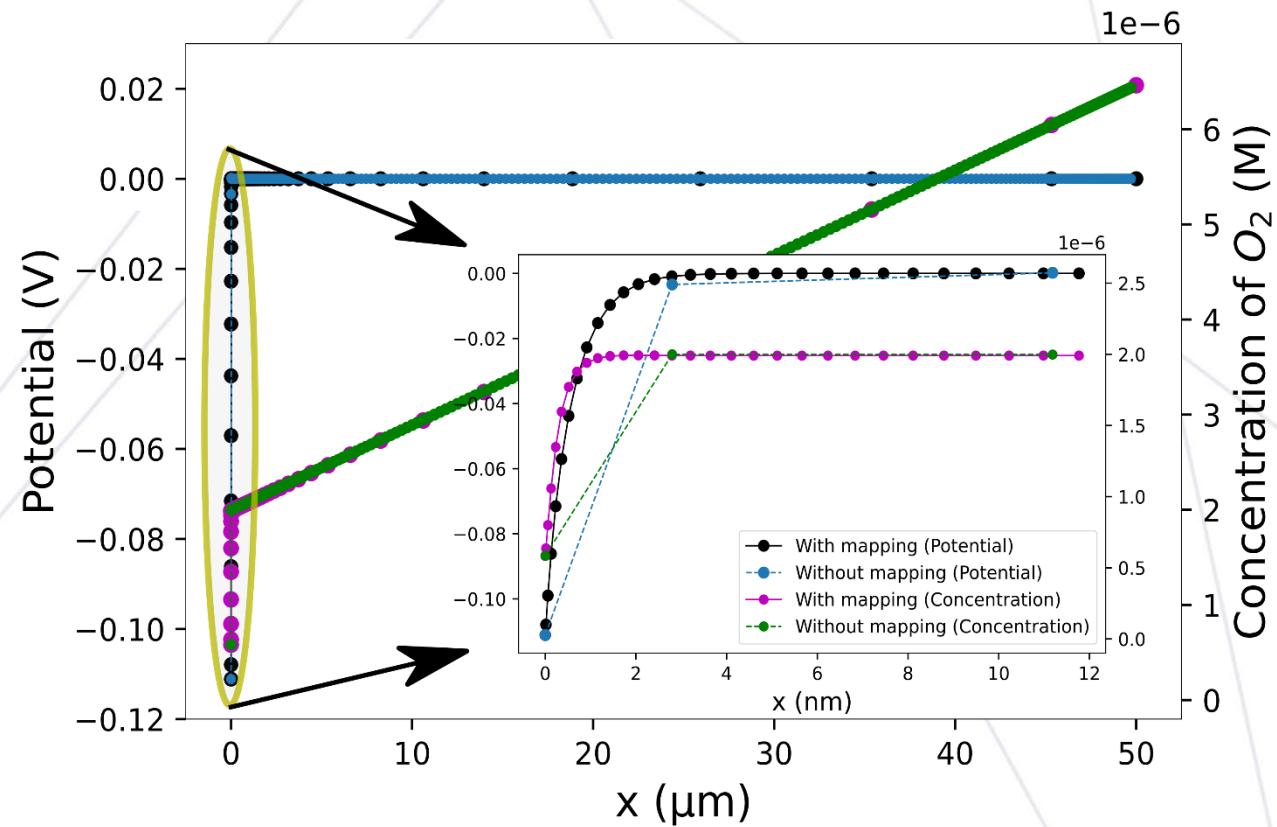
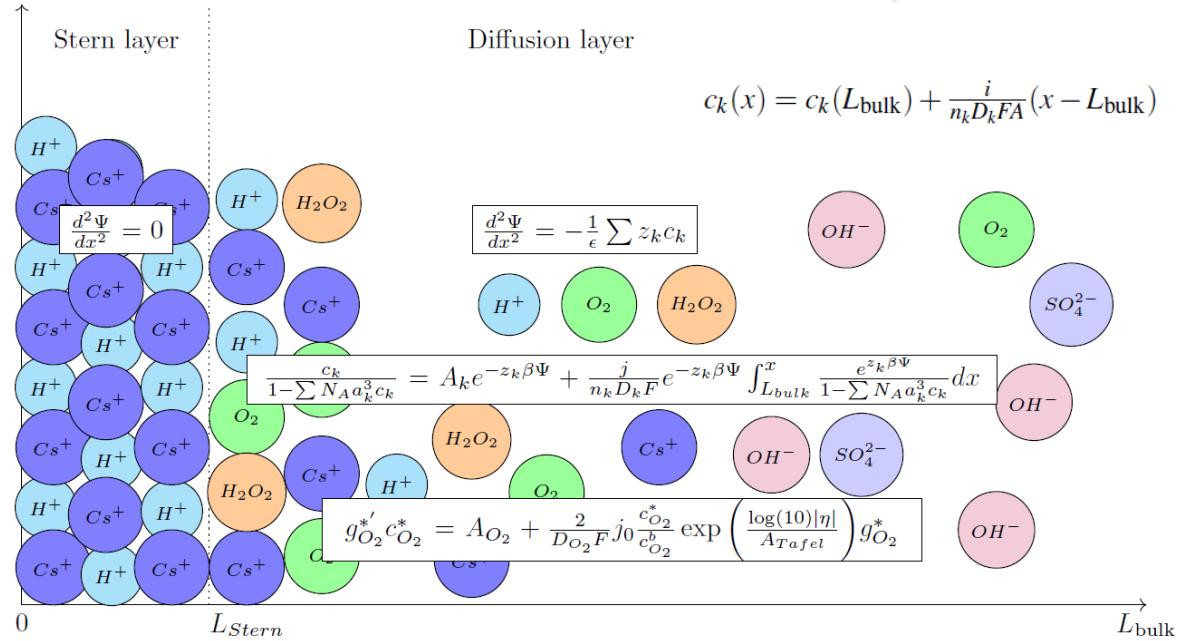


# Mathematical/Computational analysis



- Solve outer diffusive regime explicitly
- Integrate inward from bulk switch from PDE-> ODE

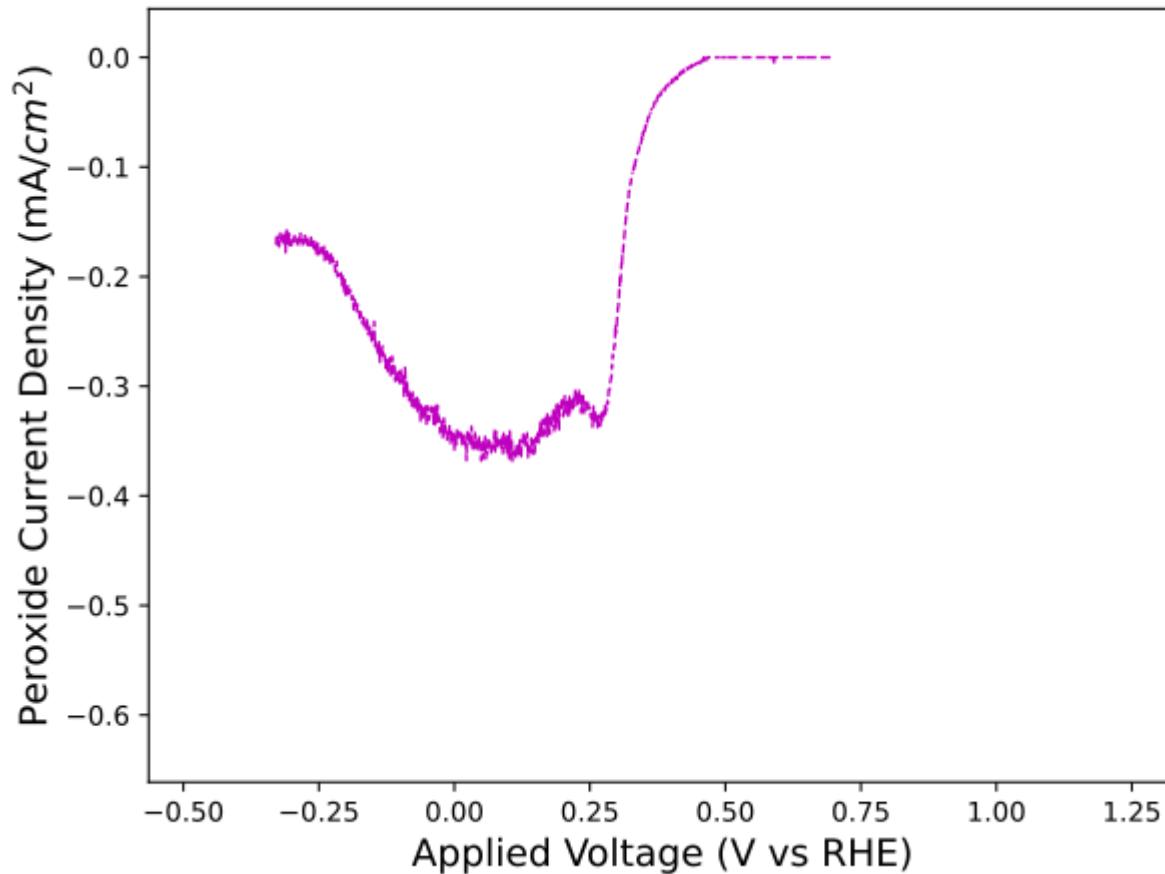
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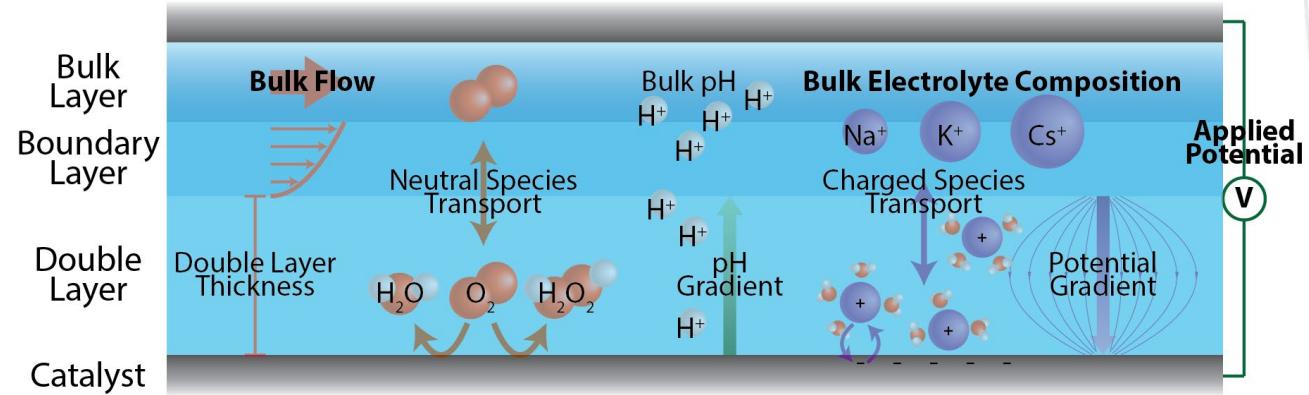
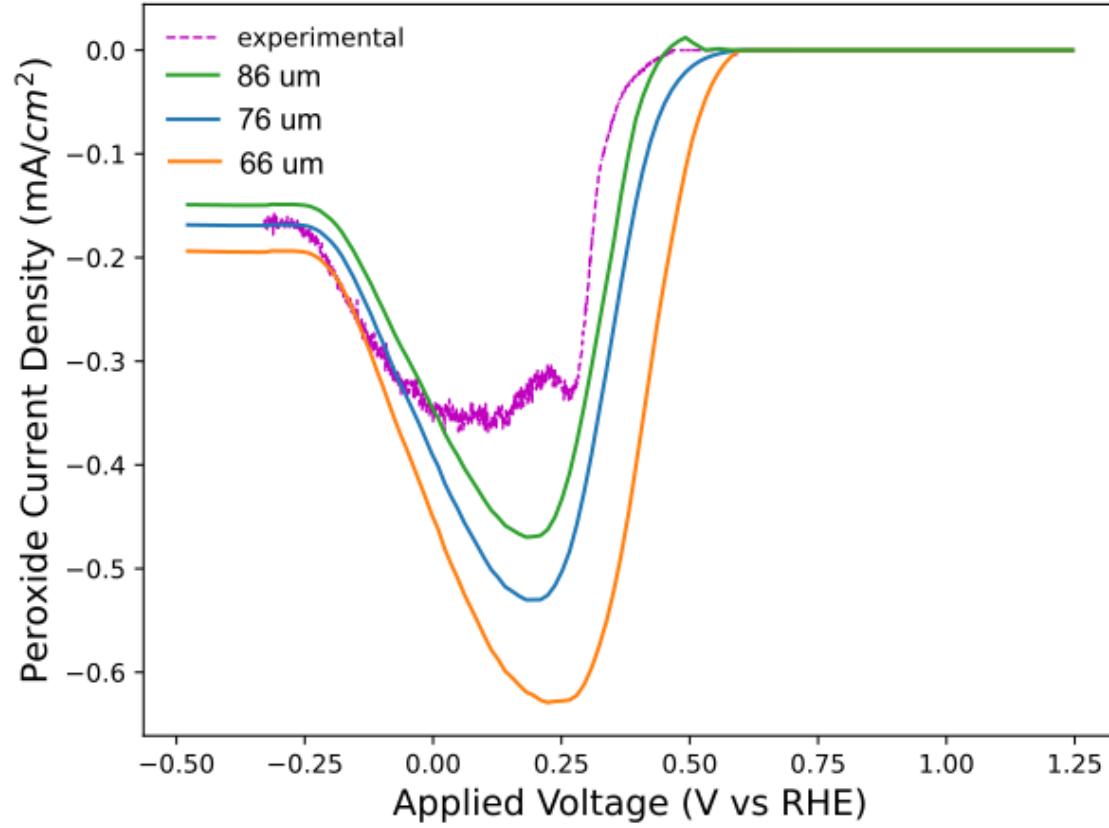
- Nonlinear mapping of space to resolve  $\sim$ nm
- Spectral methods for spatial resolution

# Current density of $\text{H}_2\text{O}_2$ production



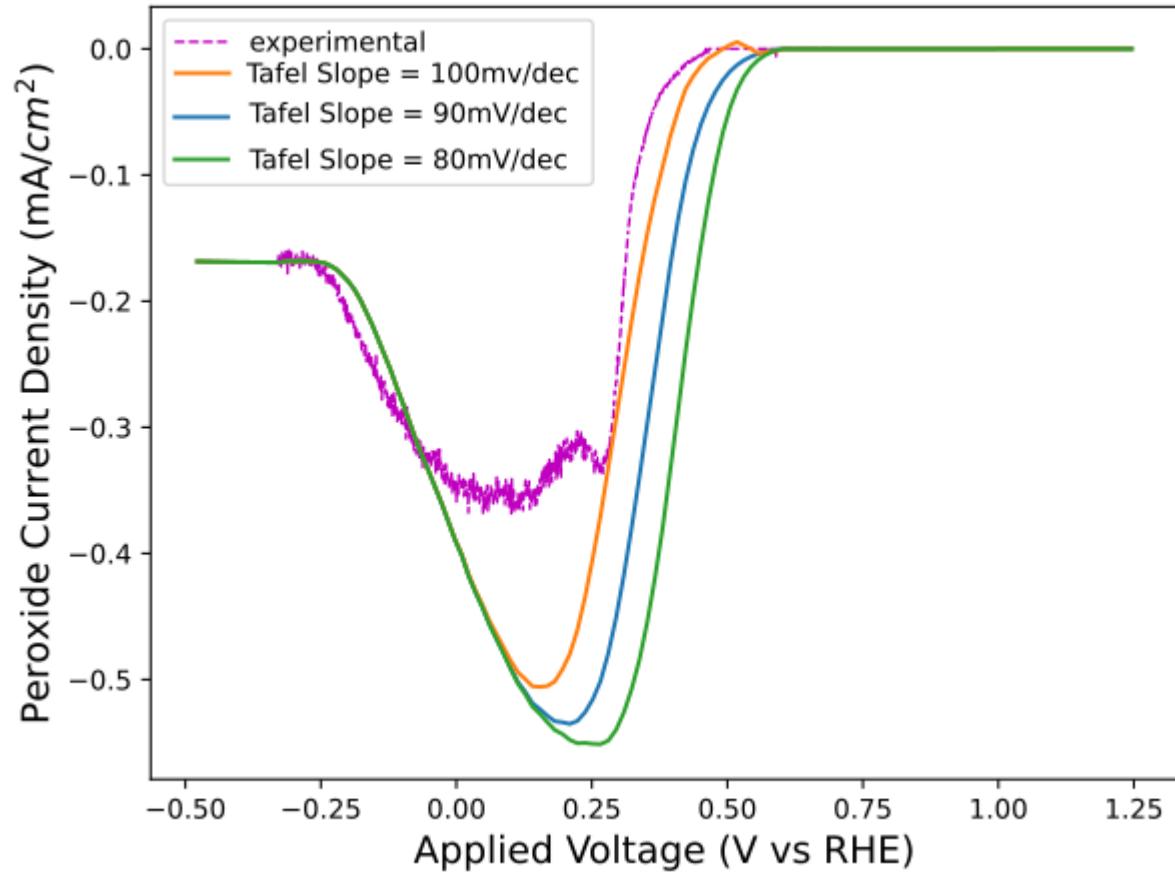
- pH 4
- Cs+ ions
- Goals:
  - Match behavior in experiment
  - Disambiguate steric interactions, transport, and reaction kinetic effects

# Varying length of diffusive region



- Sets transport (diffusion) limited current density
- decrease in current at higher voltage

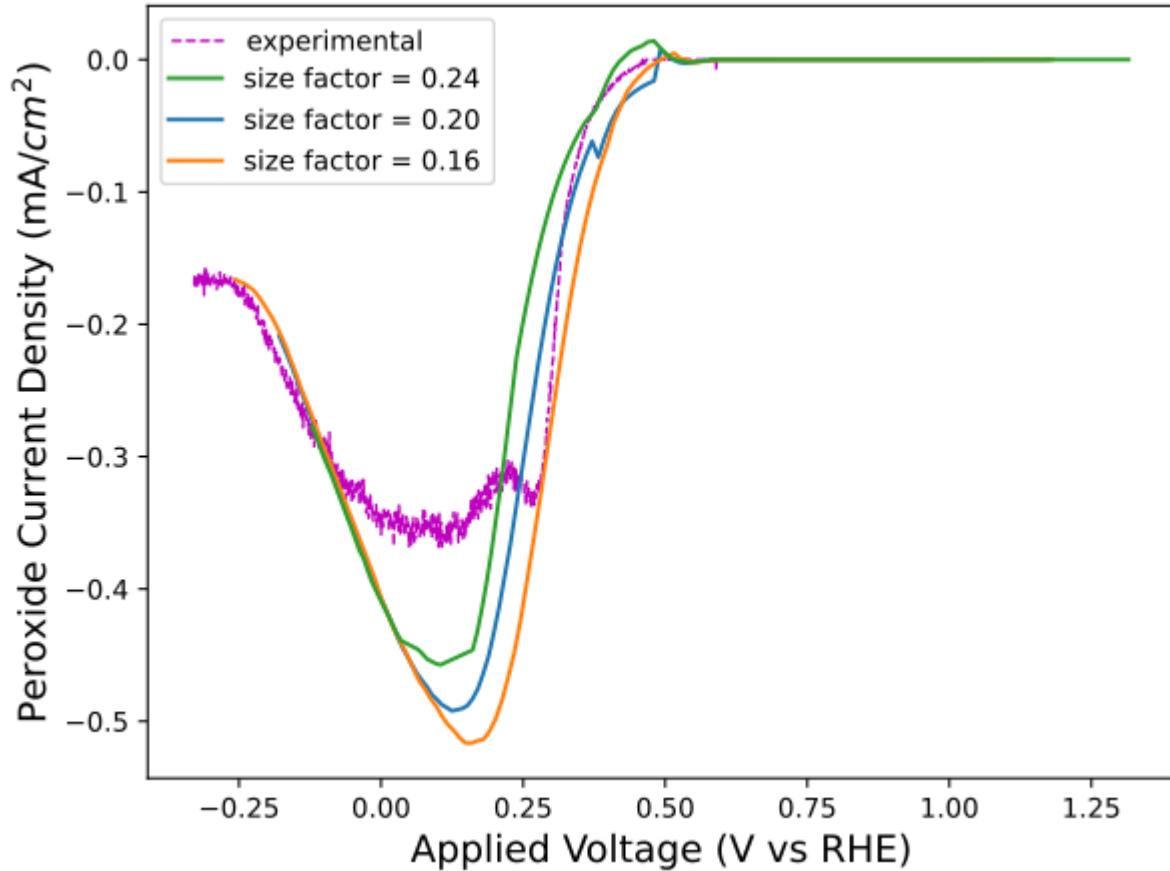
# Varying kinetic reaction rate as Tafel slope



$$J(0) = j_0 \frac{C_i(0)}{C_i^*} \exp \left[ -\frac{(\phi - \phi_{eq})}{A_{Tafel}} \right]$$

- Effects both slope after onset potential
- and decrease in current at higher voltage

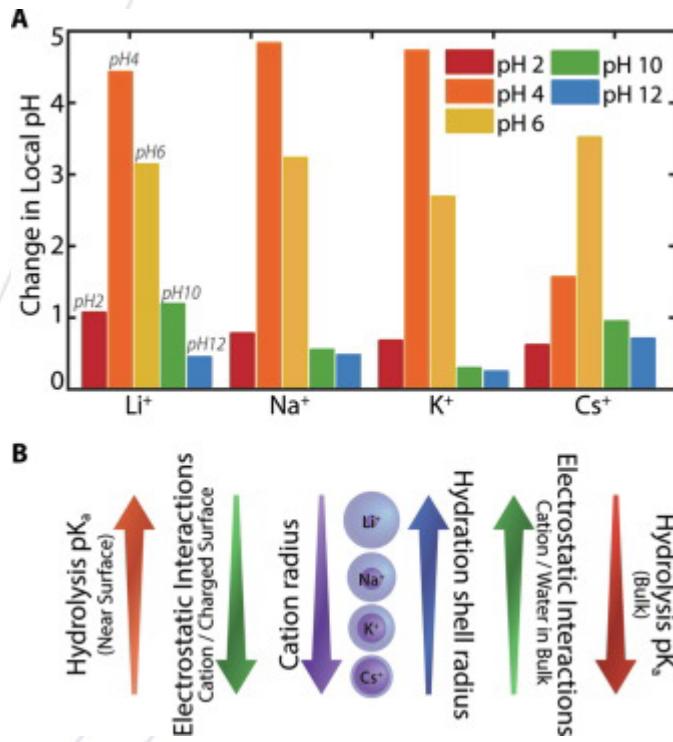
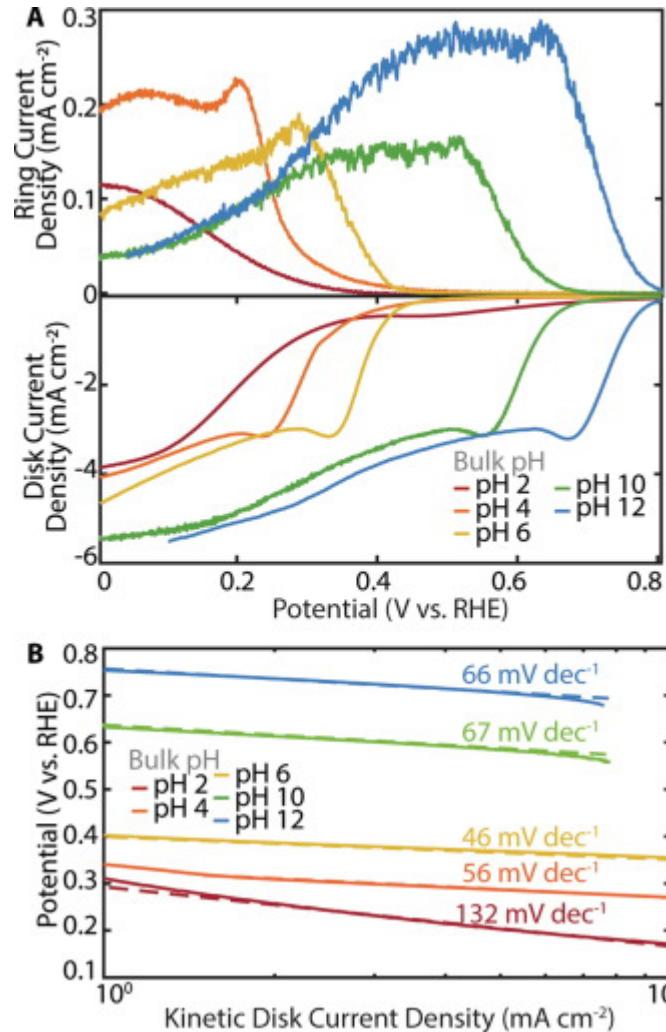
# Varying effective radius of ions



$$J(x) = D_i \frac{\partial C_i}{\partial z} + \frac{z_i D_i C_i F}{RT} \frac{\partial \phi}{\partial z} + D_i C_i \frac{N_A \sum_i (a_i^3 \frac{\partial C_i}{\partial z})}{1 - N_A \sum_i (a_i^3 C_i)}$$

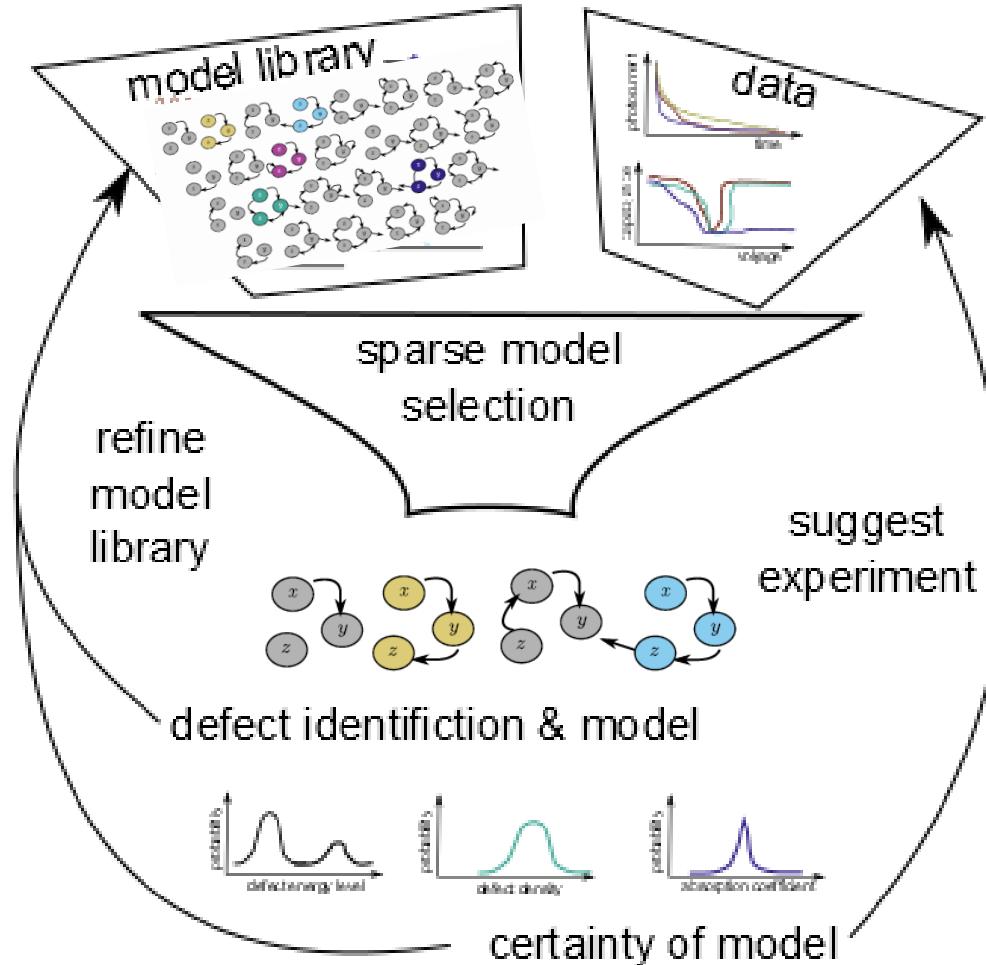
- Larger ions block current due to steric exclusion
- Onset further left.

# Next steps over more conditions:



- Fit data to curves to determine most likely parameter set(s)
- Study across different ions and pH

# Mathematical methods for ChemBio



- Spatial modeling of electrochemical catalysis
- **Data-driven discovery of dynamical systems**



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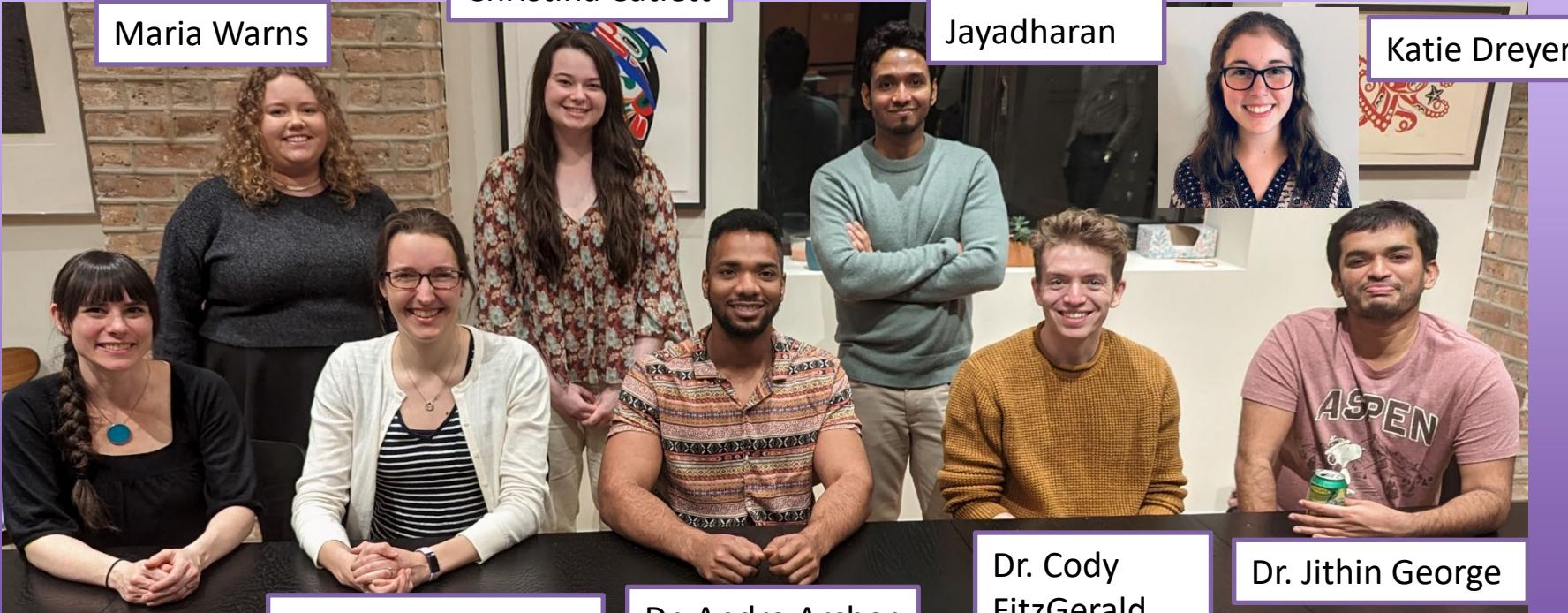


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