

Compressible flow

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- So far we have worked predominantly with incompressible flows.
- That is convenient because:
 - ρ is constant and can be freely moved in and out of differential operators.
 - and the mass conservation eqn simplifies to $\nabla \cdot \boldsymbol{v} = 0$
- We have even applied it to air.
 - - everyday experience tells us it is pretty compressible.
 - - balloons
 - - sound waves (pressure/density waves)

Sound propagation

- I am sure you know that sound is pressure waves.
 - Changes in density
 - Changes in pressure
 - Small velocities
- We have previously talked about
 - Mechanical balances
 - Mass balance
 - Relationship between pressure and density? (Compressibility/Hooke/... Ch1)
- Do we have what we need to let us understand sound waves?

Lets revisit

- We have
 - $\rho \frac{D\mathbf{v}}{Dt} = \mathbf{f} + \nabla \cdot \boldsymbol{\sigma}$ *momentum conservation*
 - $\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \mathbf{v}) = 0$ *mass conservation*
- Assume ideal (viscosity free) fluid
 - $\boldsymbol{\sigma} = -p\mathbf{I}$ and $\nabla \cdot \boldsymbol{\sigma} = -\nabla p$

- So we have:

- $\frac{\partial \mathbf{v}}{\partial t} + (\mathbf{v} \cdot \nabla) \mathbf{v} = \mathbf{g} - \frac{1}{\rho} \nabla p$

- Euler equations for ideal compressible flow

- $\rho \frac{D\mathbf{v}}{Dt} = \mathbf{f} + \nabla \cdot \boldsymbol{\sigma}$

- $\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \mathbf{v}) = 0$

- $\nabla \cdot \boldsymbol{\sigma} = -\nabla p$

Now consider a small perturbation (SOUND)

- Start from hydrostatic equilibrium.

- $p = p_0 \quad | \quad \rho = \rho_0 \quad | \quad v = 0$

$$1. \quad \frac{\partial v}{\partial t} + (v \cdot \nabla)v = \cancel{g} - \frac{1}{\rho} \nabla p$$

- Consider small perturbation

- $p = p_0 + \Delta p$
 - $\rho = \rho_0 + \Delta \rho$
 - $v = \Delta v$

$$2. \quad \frac{\partial \rho}{\partial t} + \nabla \cdot (\rho v) = 0$$

- $\frac{\partial \rho}{\partial t} + (v \cdot \nabla)\rho + \rho(\nabla \cdot v) = 0$

- Step 1: Insert in (1)

- $\frac{\partial v}{\partial t} + (\cancel{v \cdot \nabla})v = -\frac{1}{\rho_0} \nabla \Delta p$

$$3. \quad (v \cdot \nabla)\Delta \rho$$

Eqn14.12

Step 2: insert in the mass balance eqn...

- $\frac{\partial \rho}{\partial t} + \nabla \cdot (\rho \mathbf{v}) = 0$

- $\nabla \cdot (\rho \mathbf{v}) =$

- $(\mathbf{v} \cdot \nabla) \rho + \rho (\nabla \cdot \mathbf{v})$
- $= (\mathbf{v} \cdot \nabla) \Delta \rho + (\rho_0 + \Delta \rho) (\nabla \cdot \mathbf{v})$
- $= \cancel{(\mathbf{v} \cdot \nabla) \Delta \rho} + \rho_0 (\nabla \cdot \mathbf{v}) + \cancel{\Delta \rho (\nabla \cdot \mathbf{v})}$
- $= \text{small} * \text{small} + \text{big} * \text{small} + \text{small} * \text{small}$

- small perturb

- $p = p_0 + \Delta p$
- $\rho = \rho_0 + \Delta \rho$
- $\mathbf{v} = \Delta \mathbf{v}$

- HENCE: $\frac{\partial \Delta \rho}{\partial t} = -\rho_0 (\nabla \cdot \mathbf{v})$

- Apply $\frac{\partial}{\partial t}$ to (B)

- $\frac{\partial^2 \Delta \rho}{\partial t^2} = -\rho_0 \left(\nabla \cdot \frac{\partial}{\partial t} \mathbf{v} \right)$

$$A. \quad \frac{\partial \mathbf{v}}{\partial t} = -\frac{1}{\rho_0} \nabla \Delta p$$

$$B. \quad \frac{\partial \Delta \rho}{\partial t} = -\rho_0 (\nabla \cdot \mathbf{v})$$

- Insert A on RHS

- $\frac{\partial^2 \Delta \rho}{\partial t^2} = \nabla^2 \Delta p$

- Really close to a wave eqn. now
 - Needs relationship between p and rho

- barotropic equation of state

- $p = p(\rho)$

- Linearize:

- $\Delta p \approx \left. \frac{dp}{d\rho} \right|_0 \Delta \rho$

- $\Delta p \approx \frac{K_0}{\rho_0} \Delta \rho$

- *(this is how we defined equilib. bulk modulus)*

- Inserting gives:

- $\frac{\partial^2 \Delta \rho}{\partial t^2} = \frac{K_0}{\rho_0} \nabla^2 \Delta \rho$

- Speed of sound is $c_0 = \sqrt{\frac{K_0}{\rho_0}}$

- $\frac{\partial^2 \Delta \rho}{\partial t^2} = \nabla^2 \Delta p$

Fluid	T	c_0
Glycerol	25	1920
Sea water	20	1521
Fresh water	20	1482
Lube oil	25	1461
Mercury	25	1449
Ethanol	25	1145
Hydrogen	27	1310
Helium	0	973
Water vapor	100	478
Neon	30	461
Humid air	20	345
Dry air	20	343
Oxygen	30	332
Argon	0	308
Nitrogen	27	363
Unit	$^{\circ}\text{C}$	m s^{-1}

Empirical sound speeds in various liquids (above) and gases (below). The temperature of the measurement is also listed.

