

NOTE: All symbols used have their usual meaning. Each question carries 6 marks. Some useful mathematical expressions are given at the end of the questions.

- 1) Consider a hydrogen atom in its ground state.
- Find the Clebsch-Gordan coefficients associated with the coupling of spins of the electron and the proton.
 - Find the transformation matrix which is formed by the C-G coefficients. Check whether this matrix is unitary or not.
- [4+2 = 6 Marks]

- 2) Consider a system of total angular momentum $j=1$. The operators \hat{J}_x , \hat{J}_y and \hat{J}_z (in the usual eigen basis of \hat{J}^2 and \hat{J}_z) are given by

$$\hat{J}_x = \frac{\hbar}{\sqrt{2}} \begin{pmatrix} 0 & 1 & 0 \\ 1 & 0 & 1 \\ 0 & 1 & 0 \end{pmatrix}, \quad \hat{J}_y = \frac{\hbar}{\sqrt{2}} \begin{pmatrix} 0 & -i & 0 \\ i & 0 & -i \\ 0 & i & 0 \end{pmatrix}, \quad \hat{J}_z = \hbar \begin{pmatrix} 1 & 0 & 0 \\ 0 & 0 & 0 \\ 0 & 0 & -1 \end{pmatrix}.$$

- What are the possible values when measuring \hat{J}_x ?
 - Calculate $\langle \hat{J}_z \rangle$ if the system is in the state $\chi = -\hbar$.
 - If the system was initially in the state $|\psi\rangle = \frac{1}{\sqrt{14}} \begin{pmatrix} -\sqrt{3} \\ 2\sqrt{2} \\ \sqrt{3} \end{pmatrix}$, what values will one obtain when measuring \hat{J}_x and with what probabilities?
- [2+1+3 = 6 Marks]

- 3) The wave function of an electron in a hydrogen atom is given by

$$|\psi_{21m_l}\rangle(r, \theta, \phi) = R_{21}(r) \left[\frac{1}{\sqrt{3}} Y_{10}(\theta, \phi) \left| \frac{1}{2}, +\frac{1}{2} \right\rangle + \sqrt{\frac{2}{3}} Y_{11}(\theta, \phi) \left| \frac{1}{2}, -\frac{1}{2} \right\rangle \right].$$

Where $\left| \frac{1}{2}, \pm \frac{1}{2} \right\rangle$ are the spin state vectors.

- Calculate $\hat{J}_z |\psi_{21m_l}\rangle(r, \theta, \phi)$, i.e., the z -component of the electron's total angular momentum.
 - If you measure \hat{J}^2 , what values will you obtain? What are the corresponding probabilities?
 - If you measure the z -component of the electron's orbital angular momentum, what values will you obtain? What are the corresponding probabilities?
 - Calculate $\langle \psi_{21m_l} | \hat{L}_- | \psi_{21m_l} \rangle$ and $\langle \psi_{21m_l} | \hat{S}_z | \psi_{21m_l} \rangle$.
- [1+2+1+2 = 6 Marks]

$$\begin{aligned} &+i\hbar J_x \\ &-i\hbar J_y \\ &= 2\hbar J_z \end{aligned}$$

$$\begin{aligned} &\sqrt{2} J_x + J_y \\ &= 2 J_z \end{aligned}$$

$$\begin{aligned} &+i\hbar J_x \\ &-i\hbar J_y \end{aligned}$$

- 4) Consider the electron's spin operator: $\hat{S} = \frac{\hbar}{2} \hat{\sigma}$, where the components of $\hat{\sigma}$ are the Pauli spin matrices.
- a) Find the eigenvalues and eigenstates of \hat{S} in the direction of a unit vector \vec{n} ; assume \vec{n} lies in the xz plane.
- b) Find the probability of measuring $\hat{S}_z = +\frac{\hbar}{2}$.

[4+2 = 6 Marks]

- 5) Consider a hydrogen atom whose wave function is given at time $t = 0$ by

$$\Psi(\vec{r}, 0) = \frac{A}{\sqrt{\pi}} \left(\frac{1}{a_0} \right)^{3/2} e^{-r/a_0} + \frac{1}{\sqrt{2\pi}} \left(\frac{z - \sqrt{2}x}{r} \right) R_{21}(r),$$

Where A is a real constant, a_0 is the Bohr radius, and $R_{21}(r)$ is the radial wave function:

$$R_{21}(r) = \frac{1}{\sqrt{6}} \left(\frac{1}{a_0} \right)^{3/2} \left(\frac{r}{2a_0} \right) e^{-r/2a_0}.$$

- a) Write down $\Psi(\vec{r}, 0)$ in terms of $\sum_{n,l,m} \psi_{nlm}(\vec{r})$ where $\psi_{nlm}(\vec{r}) = R_{nl}(r) Y_{lm}(\theta, \phi)$ is the hydrogen atom wave function

- b) Find A so that $\Psi(\vec{r}, 0)$ is normalized, i.e., $\int \Psi_{n'l'm'}^*(\vec{r}) \Psi_{nlm}(\vec{r}) d^3r = \delta_{n'n} \delta_{l'l} \delta_{m'm}$.

- c) Write down the wave function $\Psi(\vec{r}, t)$ at some later time t .

- d) Is $\Psi(\vec{r}, 0)$ an eigenfunction of \hat{L}^2 and \hat{L}_z ? If yes, what are the eigenvalues?

- e) If a measurement of the energy is made, what value(s) could be found and with what probability?

- f) What is the probability that a measurement of \hat{L}_z yields $1\hbar$? [1+1+1+1+1+1 = 6 Marks]

- 6) Consider a system of two spinless particles of reduced mass μ that is subject to a finite, central potential well

$$V(r) = \begin{cases} -V_0 & \text{for } 0 \leq r \leq a \\ 0 & \text{for } r > a \end{cases}$$

Where V_0 is positive. The purpose of this problem is to show how to find the minimum value of V_0 so that the potential well has one $l = 0$ bound state.

- a) Find the solution of the radial Schrödinger equation in both regions, $0 \leq r \leq a$ and $r > a$, in the case where the particle has zero angular momentum and its energy is located in the range $-V_0 < E < 0$.
- b) Show that the continuity condition of the radial function at $r = a$ can be reduced to a transcendental equation in E .
- c) Find the minimum value of V_0 so that the system has (i) one bound state, and (ii) two bound states. [Hint: Use the continuity condition in (b) OR the method of graphical solution of the transcendental equation derived in (b)].

[2+2+2 = 6 Marks]

Table 1: Spherical Harmonics and their expressions in Cartesian coordinates.

$Y_{lm}(\theta, \varphi)$	$Y_{lm}(x, y, z)$
$Y_{00}(\theta, \varphi) = \frac{1}{\sqrt{4\pi}}$	$Y_{00}(x, y, z) = \frac{1}{\sqrt{4\pi}}$
$Y_{10}(\theta, \varphi) = \sqrt{\frac{3}{4\pi}} \cos \theta$	$Y_{10}(x, y, z) = \sqrt{\frac{3}{4\pi}} \frac{z}{r}$
$Y_{1,\pm 1}(\theta, \varphi) = \mp \sqrt{\frac{3}{8\pi}} e^{\pm i\varphi} \sin \theta$	$Y_{1,\pm 1}(x, y, z) = \mp \sqrt{\frac{3}{8\pi}} \frac{x \pm iy}{r}$
$Y_{20}(\theta, \varphi) = \sqrt{\frac{5}{16\pi}} (3 \cos^2 \theta - 1)$	$Y_{20}(x, y, z) = \sqrt{\frac{5}{16\pi}} \frac{3z^2 - r^2}{r^2}$
$Y_{2,\pm 1}(\theta, \varphi) = \mp \sqrt{\frac{15}{8\pi}} e^{\pm i\varphi} \sin \theta \cos \theta$	$Y_{2,\pm 1}(x, y, z) = \mp \sqrt{\frac{15}{8\pi}} \frac{(x \pm iy)z}{r^2}$
$Y_{2,\pm 2}(\theta, \varphi) = \sqrt{\frac{15}{32\pi}} e^{\pm 2i\varphi} \sin^2 \theta$	$Y_{2,\pm 2}(x, y, z) = \mp \sqrt{\frac{15}{32\pi}} \frac{x^2 - y^2 \pm 2ixy}{r^2}$

Table 2: First few radial wave functions $R_{nl}(r)$ of the hydrogen atom

$R_{10}(r) = 2a_0^{-3/2} e^{-r/a_0}$	$R_{21}(r) = \frac{1}{\sqrt{6}a_0^3} \frac{r}{2a_0} e^{-r/2a_0}$
$R_{20}(r) = \frac{1}{\sqrt{2}a_0^3} \left(1 - \frac{r}{2a_0}\right) e^{-r/2a_0}$	$R_{31}(r) = \frac{8}{9\sqrt{6}a_0^3} \left(1 - \frac{r}{6a_0}\right) \left(\frac{r}{3a_0}\right) e^{-r/3a_0}$
$R_{30}(r) = \frac{2}{3\sqrt{3}a_0^3} \left(1 - \frac{2r}{3a_0} + \frac{2r^2}{27a_0^2}\right) e^{-r/3a_0}$	$R_{32}(r) = \frac{4}{9\sqrt{30}a_0^3} \left(\frac{r}{3a_0}\right)^2 e^{-r/3a_0}$

Table 3: First few Legendre polynomials and associated Legendre functions.

Legendre Polynomials	Associated Legendre Functions
$P_0(\cos \theta) = 1$	$P_1^1(\cos \theta) = \sin \theta$
$P_1(\cos \theta) = \cos \theta$	$P_2^1(\cos \theta) = 3 \cos \theta \sin \theta$
$P_2(\cos \theta) = (3 \cos^2 \theta - 1)/2$	$P_2^2(\cos \theta) = 3 \sin^2 \theta$
$P_3(\cos \theta) = (5 \cos^3 \theta - 3 \cos \theta)/2$	$P_3^1(\cos \theta) = 3 \sin \theta (5 \cos^3 \theta - 1)/2$
$P_4(\cos \theta) = (35 \cos^4 \theta - 30 \cos^2 \theta + 3)/8$	$P_3^2(\cos \theta) = 15 \sin^2 \theta \cos \theta$
$P_5(\cos \theta) = (63 \cos^5 \theta - 70 \cos^3 \theta + 15 \cos \theta)/8$	$P_3^3(\cos \theta) = 15 \sin^3 \theta$

$-Y_{111} = \sqrt{\frac{3}{8\pi}} x$
 $Y_{111} = \sqrt{\frac{3}{8\pi}} y$

Table 4: First few Laguerre polynomials and associated Laguerre polynomials.

Laguerre polynomials $L_k(r)$	Associated Laguerre polynomials $L_k^N(r)$
$L_0 = 1$	$L_1^1 = -1$
$L_1 = 1 - r$	$L_2^1 = -4 + 2r, \quad L_2^2 = 2$
$L_2 = 2 - 4r + r^2$	$L_3^1 = -18 + 18r - 3r^2, \quad L_3^2 = 18 - 6r, \quad L_3^3 = -6$
$L_3 = 6 - 18r + 9r^2 - r^3$	$L_4^1 = -96 + 144r - 48r^2 + 4r^3,$
$L_4 = 24 - 96r + 72r^2 - 16r^3 + r^4$	$L_4^2 = 144 - 96r + 12r^2, \quad L_4^3 = 24r - 96, \quad L_4^4 = 24$
$L_5 = 120 - 600r + 600r^2 - 200r^3$	$L_5^1 = -600 + 1200r - 600r^2 + 100r^3 - 5r^4,$
$+ 25r^4 - r^5$	$L_5^2 = 1200 - 1200r + 300r^2 - 20r^3,$
	$L_5^3 = -1200 + 600r - 60r^2, \quad L_5^4 = 600 - 120r,$
	$L_5^5 = -120$