

PHYS 2C

Discussion Section – 2/05

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Before we Begin:

- Try and **sit next to a student you don't know**
- Introduce yourselves and find out where the other student is from

Today: Lightning Review and 2 Problems

Discussion Problem 1

Calorimetry

By how much will the temperature of a cup (180 g) of coffee at 95 °C be reduced when a 45 g silver spoon (specific heat 0.24 J/g °C) at 25 °C is placed in the coffee and the two are allowed to reach the same temperature?

Assume that the coffee has the same density and specific heat as water (4.184 J/g °C).

Discussion Problem 1 - Solution

Calorimetry

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1 °C

Discussion Problem 2

Linear Temperature Scales

On a linear X temperature scale, water freezes at $-124\text{ }^{\circ}\text{X}$ and boils at $335\text{ }^{\circ}\text{X}$. On a Linear Y temperature scale, water freezes at $-67.00\text{ }^{\circ}\text{Y}$ and boils at $-20.00\text{ }^{\circ}\text{Y}$.

A temperature of $51.00\text{ }^{\circ}\text{Y}$ corresponds to what temperature on the X scale?

Discussion Problem 2 - Solution

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1028.4 $^{\circ}\text{X}$

Discussion Problem 3

Ideal Gas

Oxygen gas having a volume of 1180cm^3 at 42.2°C and $1.02 \times 10^5\text{Pa}$.

Find:

- a) The number of moles of Oxygen present
- b) The final temperature of the sample

Discussion Problem 3 - Solution

Ideal Gas

Oxygen gas having a volume of 1180cm^3 at 42.2°C and $1.02 \times 10^5\text{Pa}$.

Find:

a) The number of moles of Oxygen present

0.046 mol

b) The final temperature of the sample

524.25 K