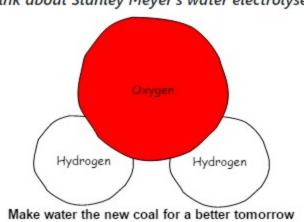
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## **HOLOMORPHE**

Think about Stanley Meyer's water electrolyser!



## Comparison of electrolysers

Name	Theoretical power consumption per kilogram of produced hydrogen (kWh/nm^3)	Theoretical electrical power (MW)	Theoretical installation surface (m^2)	Used materials	Theoretical operating temperature (°C)	Theoretical life time (hours)
Alkaline electrolyser	4,5	4	100	Anode (Ni, Co, Fe) / Cathode (Ni, C) / Medium (Stainless steel)	90	80 000
PEM electrolyser	4,5	1	25	Anode (Platine) / Cathode (Iridium) / Medium (Titane)	80	20 000
SOEC electrolyser	3,5	2,7	300	Anode (Ni+ZrO2) / Cathode (Nickel, Cobalt, Nickel de Cermet)	1000	100 000
Stanley Meyer's electrolyser (Holomorphe – Hydrogen container)	0,000626488	0,00004	13,846156	Anode et cathode (stainless steel)	25	876 000

Theoretical flow (m^3/h)	Theoretical ratio of flow per surface ((m^3/h)/m^2)	Theoretical ratio of power per surface (W/m^2)	Theoretical ratio of flow per power ((m^3/h)/W)	Type of electrolyte	Type of water	Theoretical yield (%)
800	8	40 000	0,0002	Sodium hydroxyde (NaOH) – Liquid	Alkaline water	77
200	8	40 000	0,0002	Solid polymere	Pure water	80
750	2,5	9 000	0,00028	Electrolyte waterproof for gas	Water steam	85
63,8479895	4,6112429688	2,8888884395	1,5961997375	No	All types of water (fresh, salted, rain, waste)	478 860

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