# Solution of the Kohn-Sham equations

We now have an ab-initio, approximate DFT which can be used in practice to compute properties of molecules and condensed systems.

The total energy in terms of the charge density and the single particle orbitals

$$E_{ks} = -\frac{1}{2} \sum_{i=1}^{N} \int_{\mathcal{X}_{i}} \left( \bar{r} \right) \nabla^{2} \psi_{i}(\bar{r}) d\bar{r} + \frac{1}{2} \int_{\mathcal{X}_{i}} d\bar{r} d\bar{r} \frac{f(\bar{r}) f(\bar{r}')}{|\bar{r} - \bar{r}'|} + \int_{\mathcal{X}_{i}} d\bar{r} \int_{\mathcal{X}_{i}} \left( \bar{r} \right) \nabla^{2} \psi_{i}(\bar{r}) d\bar{r} + \frac{1}{2} \int_{\mathcal{X}_{i}} d\bar{r} d\bar{r} \frac{f(\bar{r}) f(\bar{r}')}{|\bar{r} - \bar{r}'|} + \int_{\mathcal{X}_{i}} d\bar{r} \int_{\mathcal{X}_{i}} \left( \bar{r} \right) \nabla^{2} \psi_{i}(\bar{r}) d\bar{r} + \frac{1}{2} \int_{\mathcal{X}_{i}} d\bar{r} d\bar{r} \frac{f(\bar{r}) f(\bar{r}')}{|\bar{r} - \bar{r}'|} + \int_{\mathcal{X}_{i}} d\bar{r} \int_{\mathcal{X}_{i}} \left( \bar{r} \right) \nabla^{2} \psi_{i}(\bar{r}) d\bar{r} + \frac{1}{2} \int_{\mathcal{X}_{i}} d\bar{r} d\bar{r} \frac{f(\bar{r}) f(\bar{r}')}{|\bar{r} - \bar{r}'|} + \int_{\mathcal{X}_{i}} d\bar{r} \int_{\mathcal{X}_{i$$

The Kohn-Sham equations: set of N non linear differential equations for the single particles orbitals; the potential V<sub>KS</sub> depends 'self-consistently' upon the charge density

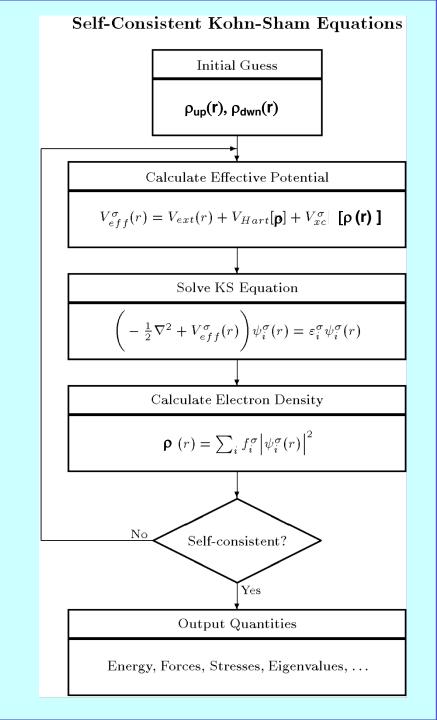
$$\left(-\frac{1}{2}\nabla^{2} + V_{ks}[\rho(\bar{r})]\right)\psi_{i}(\bar{r}) = \mathcal{E}_{i}\psi_{i}(\bar{r}) \quad \text{with } V_{KS} = \frac{SE_{ks}}{S\rho(\bar{r})}$$

Approximations (e.g. LDA, GGA, etc.) for the exchange correlation energy density

 $\varepsilon_{xc}$  and  $E_{xc}[\rho]$  are in general unknown and one needs to resort to approximations

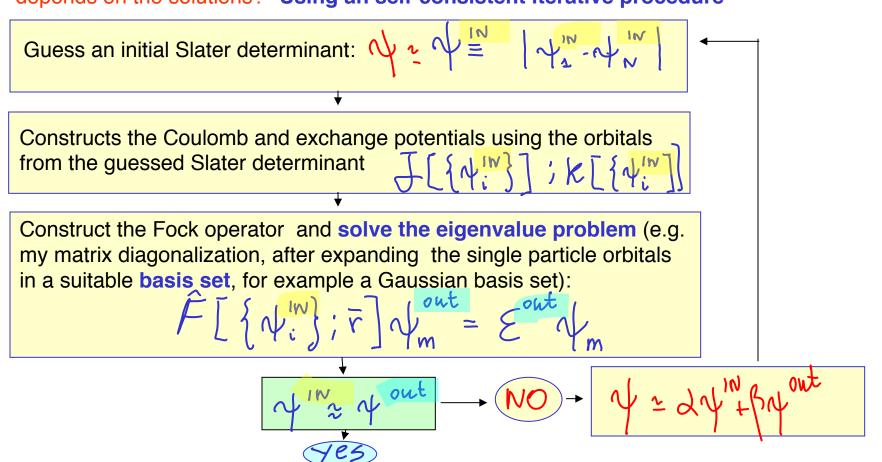
## **Solving KS equations**

- Choose approximation for exchange and correlation potential
- •Choose atomic configuration, i.e. positions of the nuclei **R**<sub>I</sub>
- •Choose:
  - how to treat atomic cores (allelectron, frozen core, pseudopotentials)
  - basis set to expand single particle orbitals
- Start self-consistent cycle by "guessing" a set of single particle orbitals (i.e. the expansion coefficients in terms of a given basis set) and thus an input charge density  $\rho$ .
- •Solve KS eigenvalue problem in an iterative manner.



## Compare with solution of Hartree-Fock equations

How do we solve the Hartree-Fock equations, i.e. an eigenvalue problem where the operator depends on the solutions? **Using an self-consistent iterative procedure** 



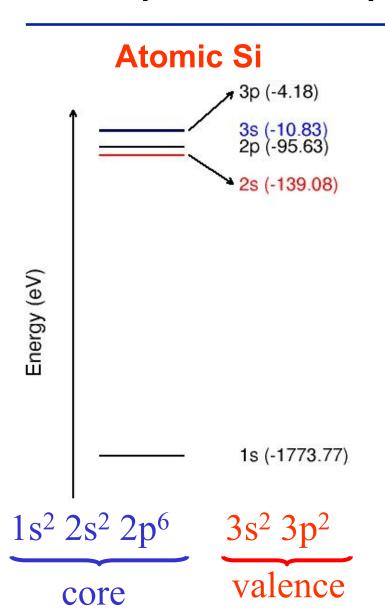
Self-consistency attained; Compute total energies and orbital energies

#### Representation of atomic cores

- All-electron (AE) methods: all electrons in the system are treated explicitly; some of them may be considered "frozen" in the configuration they have in the atom

- Pseudopotential (PP) method: only valence electrons are treated explicitly
  - Core states strongly bound to nuclei atomic-like
  - Valence states (treated explicitly) they undergo changes when going from an atom to a molecule or a condensed system; the way they change determine the bonding, electronic and optical properties, magnetism, .....

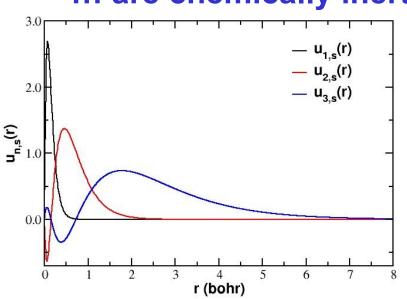
## Pseudopotentials and pseudo-wavefunctions (I)



#### Core electrons...

highly localized very low energy compared to valence states

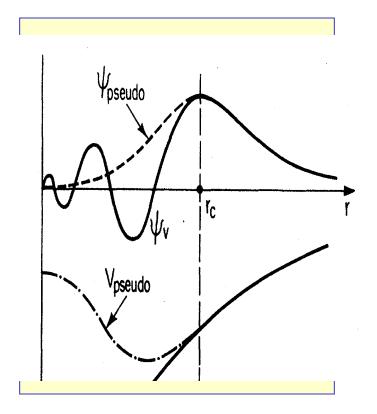
### ... are chemically inert



Valence wave functions must be orthogonal to the core wave functions

### Pseudopotentials and pseudo-wavefunctions (II)

- Idea: construct 'pseudo-atom' which has the valence states of the real atom as its lowest electronic states
- Preserves scattering properties and total energy differences for the real atom
- Removal of orbital nodes makes plane-wave expansion feasible
- Possible limitations: Can the pseudo-atom correctly describe the bonding in different environments?
  - → transferability of pseudo-potentials in different chemical environments needs to be checked.



### Transferability of pseudopotentials: atoms

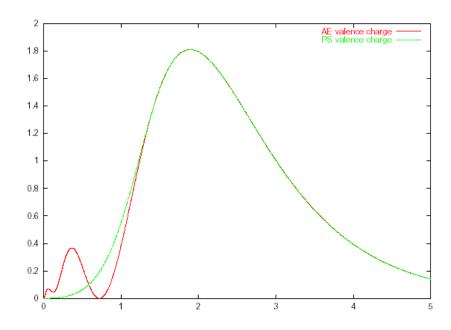
- Compute the energy of two different configurations with AE and PP methods for the chosen atom
- Compute the difference in energy using both AE and PP methods
- For the pseudopotential to be transferable: △E AE = △E PP

Example of configurations for the SI atom:

```
\begin{array}{l} \rightarrow 3s^2 \ 3p^2 \quad \text{(reference)} \\ \rightarrow 3s^2 \ 3p^1 \ 3d^1 \\ \rightarrow 3s^1 \ 3p^3 \\ \rightarrow 3s^1 \ 3p^2 \ 3d^1 \\ \rightarrow 3s^0 \ 3p^3 \ 3d^1 \end{array}
```

## Pseudopotentials and pseudo-wavefunctions (III)

#### Comparison between all electron and pseudo-charge density



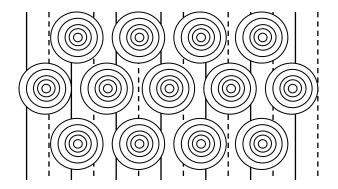
## Four choices for single particle orbital representation in modern DFT calculations

#### Plane waves

- The simplicity of Fourier Expansions
- The speed of Fast Fourier Transforms
- Requires smooth pseudopotentials

#### Localized orbitals

- The intuitive appeal of atomic-like states
- Simplest interpretation in tight-binding form
- Gaussian basis widely used for finite systems (within all quantum in chemistry

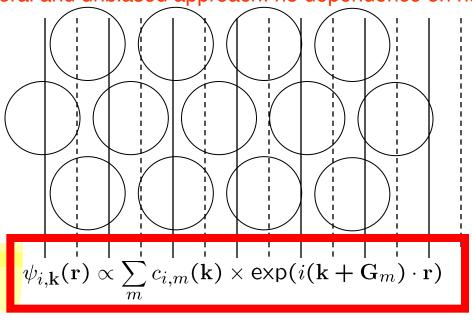


#### Augmented methods

- "Best of both (plane waves and localized orbital) worlds" also most demanding
- Requires matching technique to match functions inside and outside spheres.
- Real Space representation of wavefunctions, charge densities and potentials (not as popular as the other three methods)

#### Plane wave basis sets

The most general and unbiased approach: no dependence on nuclei positions



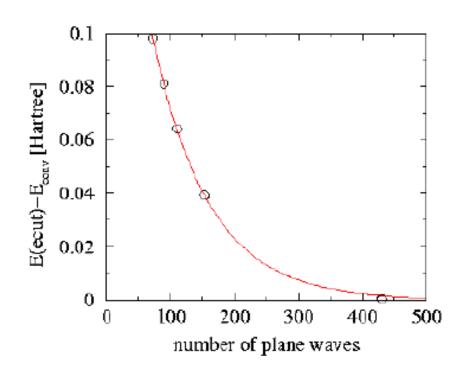
$$\sum_{m'} H_{m,m'}(\mathbf{k}) c_{i,m'}(\mathbf{k}) = \varepsilon_i(\mathbf{k}) c_{i,m}(\mathbf{k})$$

$$H_{m,m'}(\mathbf{k}) = \frac{\hbar^2}{2m_e} |\mathbf{k} + \mathbf{G}_m|^2 \delta_{m,m'} + V_{eff}(\mathbf{G}_m - \mathbf{G}_{m'}).$$

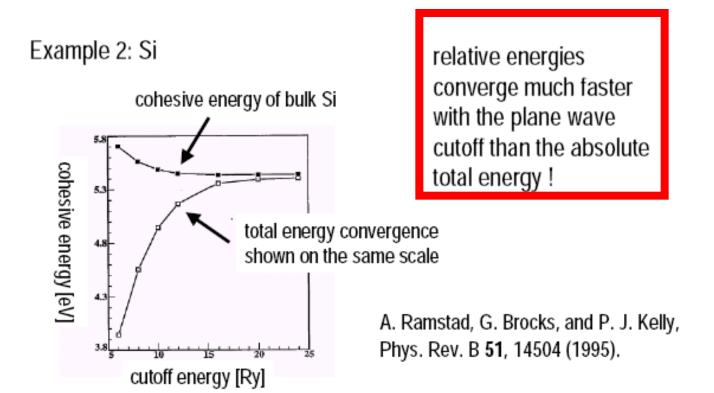
 The problem is the atoms! High Fourier components. Solution: use pseudopotentials.

## Convergence as a function of # of PWs (or PW energy cutoff)

Example 1: absolute convergence for GaAs, LDA, Hamann pseudopotential, data points at E<sub>cut</sub> = 7, 8, 10, 20 Ry

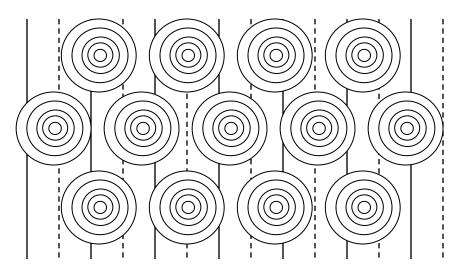


## Convergence as a function of # of PWs (or PW energy cutoff)



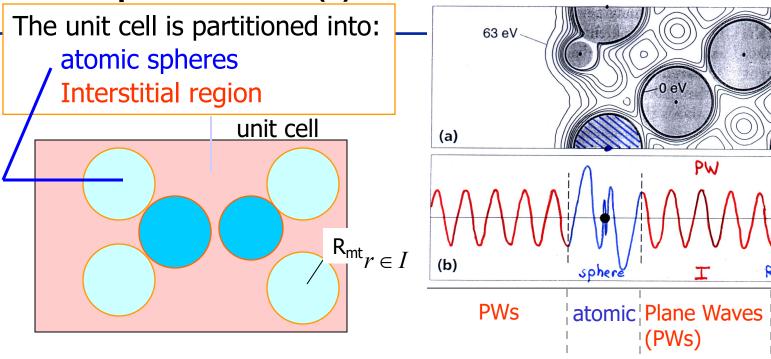
## Augmented plane waves (I)

(L)APW method



- Augmentation: represent the wave function inside given spheres centered on atoms in spherical harmonics, and as plane wave outside.
  - "Best of both worlds"
  - Requires matching inside and outside functions
  - Most general form can approach arbitrarily precision as it does not require the use of pseudopotentials.

## **Augmented plane waves (II)**



#### Basis set:

PW: 
$$e^{i(\vec{k}+\vec{K}).\vec{r}}$$

Atomic partial waves

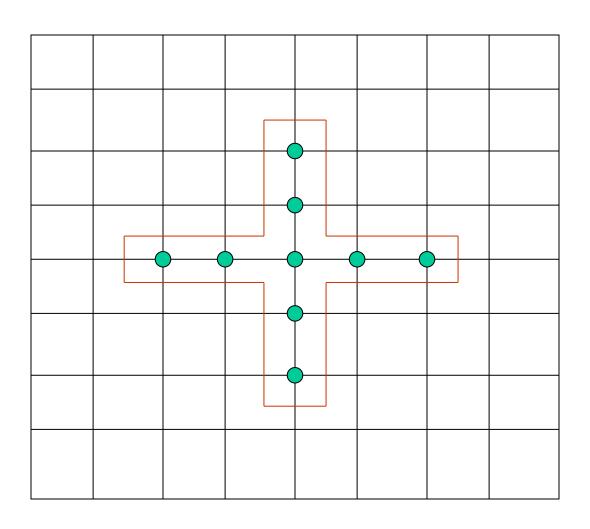
$$\sum_{\ell m} A_{\ell m}^K u_\ell(r',\varepsilon) Y_{\ell m}(\hat{r}')$$

join

 $u_l(r,\varepsilon)$  are the numerical solutions of the radial Schrödinger equation in a given spherical potential for a particular energy  $\varepsilon$   $A_{lm}{}^{K}$  coefficients for matching the PW

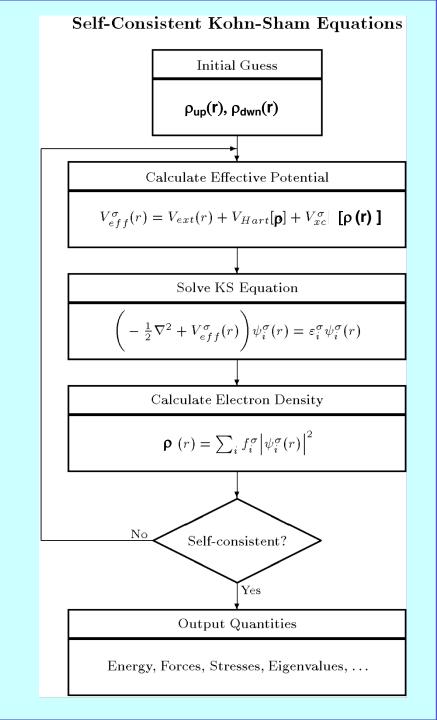
### Finite-difference representation of KS eigenvalue problem:

## Example: 2-d 4<sup>th</sup> order Laplacian



## Solving KS equations

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  - basis set to expand single particle orbitals
- Start self-consistent cycle by "guessing" a set of single particle orbitals (i.e. the expansion coefficients in terms of a given basis set) and thus an input charge density  $\rho$ .
- Solve KS eigenvalue problem in an iterative manner.



## Iterative solutions of KS eigenvalue problem

Linear mixing (or following the direction of "steepest descent")

$$g_{in}^{in} = \lambda g_{i}^{out} + (1-\lambda)g_{i}^{in} = g_{i}^{in} + \lambda(g_{i}^{out} - g_{i}^{in})$$

• How do we choose  $\alpha$  ?

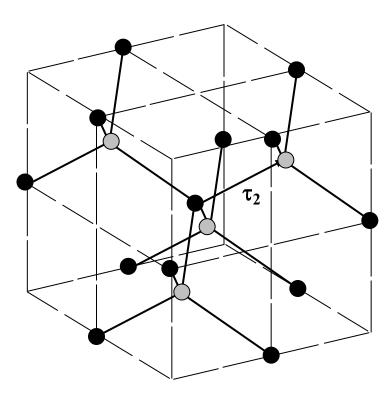
 $\alpha$  is related to the response function of a system and to a measure of the polarizability. Linear mixing with large a works better for rigid, strongly bound systems; however convergence can be difficult to obtain for "soft" systems such as a metal surface.

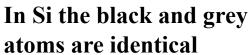
### Eigenvalue problem

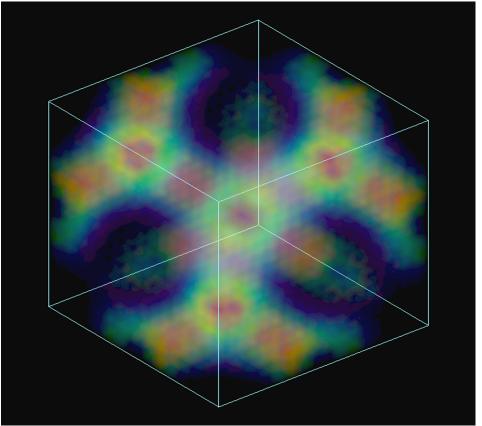
- Direct methods:
  - Full diagonalization of the Hamiltonian matrix in a chosen basis set: all eigenvalues  $\varepsilon_{i,k}$  and eigenvectors  $\Psi_{i,k}$  are computed
  - In general, suitable for methods with atom-centered, localized orbitals only
  - Practical up to a Hamiltonian matrix size of ~10,000 basis functions
- Variational approaches:
  - Diagonalization problem can be presented as a minimization problem for a quadratic form (the total energy)
  - Steepest descent and conjugate gradients methods

## Electron density in solids, e.g. Silicon

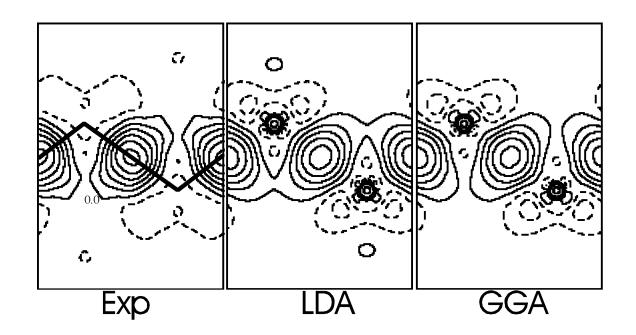
From: "Electronic Structure: Basic Theory and Practical Methods", R. M. Martin, Cambridge University Press, 2004 –







## Electron density in Silicon: comparison with experiment



- Electron density <u>difference</u> from sum of atoms
  - Experimental density from electron scattering
  - Calculations with two different functionals and LAPW basis
    - J. M. Zuo, P. Blaha, and K. Schwarz, J. Phys. Cond. Mat. 9, 7541 (1997).
  - Very similar results with pseudopotentials
    - O. H. Nielsen and R. M. Martin (1995)

## Comparison of structural parameters obtained using different basis sets.

	Method	С		Si		CaF <sub>2</sub>		bcc Fe		
		$\mid a \mid$	B	$\mid a \mid$	B	a	B	$\mid a \mid$	B	m
	$NCPP^a$	3.54	460	5.39	98	5.21	90	$2.75^{c}$	226 <sup>c</sup>	
٧	$PAW^a$	3.54	460	5.38	98	5.34	100			
1	$PAW^b$	3.54	460	5.40	95	5.34	101	2.75	247	2.00
	$USPP^b$	3.54	461	5.40	95	5.34	101	2.72	237	2.08
	$LAPW^a$	3.54	470	5.41	98	5.33	110	$2.72^{d}$	$245^d$	$2.04^d$
	$EXP^a$	3.56	443	5.43	99	5.45	85-90	$2.87^{d}$	$172^d$	$2.12^d$

- a lattice constant; B bulk modulus; m magnetization
- aHolzwarth, et al.; bKresse & Joubert; cCho & Scheffler; dStizrude, et al.