

SAFETY DATA SHEET HYDRATED LIME

OSHA Hazard Communication Standard 29 CFR 1910.1200. Prepared to GHS Rev03.

SECTION 1. IDENTIFICATION

Product Name Hydrated Lime **Chemical Name** Calcium Hydroxide

Synonyms Chemical Hydrate, Commercial Hydrate, Industrial Hydrate, Hydrate Lime,

Hydrate Tailings

Uses Soil Stabilization, Fixation, Neutralization, Desulphurization, Agriculture, Water

Treatment

Distributor Mintek Resources

3725 Pentagon Blvd.

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Beavercreek, OH 45431 Phone: 937-431-0218

Emergency Contact VelocityEHS: (800) 255-3924 (MIS8507735)

SECTION 2. HAZARDS IDENTIFICATION

Classification of the substance or mixture



GHS08 Health Hazard



Signal Word Danger

Hazard-Determining Calcium Oxide, Calcium Carbonate, Calcium Hydroxide

Components of Labeling

H303 May be harmful if swallowed. **Hazard Statements**

> H335 May cause respiratory irritation.

H373 May Cause damage to organs (lungs) through repeated or prolonged exposure.

Precautionary P101 If medical advice is needed, have product container or label at hand.

P102 Keep out of reach of children. **Statements** P260

> P264 Wash hands thoroughly after handling. P270 Do not eat, drink when using this product.

P280 Wear protective gloves/protective clothing/eye protection/face protection.

P284 Wear respiratory protection.

Do not breathe dust.

P303 If on skin (or hair): Take off immediately all contaminated clothing. Response

Rinse skin with water/shower. Wash contaminated clothing before reuse.

P304 If inhaled: Remove person to fresh air and keep comfortable for breathing.

P305 If in eyes: Rinse cautiously with water for several minutes



SECTION 3. COMPOSITION

Component	Formula	% Wt.	CAS No.	PEL
Calcium Hydroxide	Ca(OH) ₂	>90	1305-62-0	5 mg/m³
Silica-Crystalline Quartz	SiO ₂	<1%	14808-60-7	10 mg/m₃ respirable

SECTION 4. FIRST AID MEASURES

Inhalation Acute: May cause respiratory irritation.

Chronic: Respiratory tract irritation, coughing, burning sensation.

Eyes Acute: Causes serious eye damage.

Chronic: Pain, watering, redness.

Skin Acute: Causes skin irritation.

Chronic: Pain or irritation, redness, blistering may occur.

Ingestion Acute: No known effects.

Chronic: Burning sensation, abdominal cramps and pain, vomiting.

Treatments

Inhalation Move victim to fresh air. Seek medical attention if necessary. If breathing has

stopped, give artificial respiration.

Eyes Immediately flush eyes with large amounts of water for at least 15 minutes.

Pull back the eyelid to make sure all the lime dust has been washed out. Seek

medical attention immediately. Do not rub eyes.

Skin Flush exposed area with large amounts of water. Seek medical attention

immediately.

Ingestion Give large quantities of water or fruit juice. Do not induce vomiting. Seek

medical attention immediately. Never give anything by mouth if victim is rapidly

losing consciousness or is unconscious or convulsing.

SECTION 5. FIRE FIGHTING MEASURES

Flash Point Non-flammable

Autoignition Temperature Non-flammable

Inflammability Limits None, Non-combustible solid, but will support combustion by liberation

of oxygen.

Explosion RiskNone by itself, but heat produced by reaction with strong acids can generate

steam and pressure.

Hazardous Combustion

Products

Decomposes to produce calcium oxide (CaO), which can react with water to

produce steam and pressure.

Extinguishing Media Use dry chemical fire extinguisher. Do not use water or halogenated

compounds, except that large amounts of water may be used to deluge small

quantities of lime kiln dust. Use appropriate extinguishing media for

surrounding fire conditions.

Fire Fighting Instructions Keep personnel away from and upwind of fire. Wear full fire-fighting turn-out

gear (full Bunker gear), and respiratory protection (self-contained breathing

apparatus).



SECTION 6. ACCIDENTAL RELEASE MEASURES

Individual and Collective Avoid creating conditions which release dust – use mechanical vacuums to

Precautions remove dust from workspaces.

Avoid inhalation of Dust Wear respiratory protection – minimum NIOSH N-95 Dust Mask.

Cleaning Methods (Leaks &

Spills)

Use personal protective equipment (eyes, skin, and inhalation, see Section 8). Use dry methods (vacuuming, sweeping) to collect spilled materials. Avoid generating dust. For large spills, evacuate area downwind of clean-up area operations to minimize dust exposure. For small spills, store spilled materials in dry, sealed plastic or metal containers. Dust residue on surfaces may be

washed with water.

Precautions for the Protection

of the Environment

May not be released into surface waters without controls (increases pH).

Waste Disposal Dispose according to federal, provincial/state, and local environmental

regulations.

SECTION 7. HANDLING & STORAGE

Handling In open air or in ventilated places, avoid skin and eye contact, avoid creating

airborne dust.

Storage Store in dry places sheltered from humidity. Keep away from acids. Keep out

of reach of children.

SECTION 8. EXPOSURE CONTROLS/PERSONAL PROTECTION

Occupational Exposure Limits

	OSHA PEL (mg/m³)	ACGIH TLV (mg/m³)	Ont. Reg. 833 TWAEV (mg/m³)
Calcium Hydroxide	5	2	2
Silica - Crystalline Quartz	2.5 (total dust), 0.8 (respirable)	0.5 (respirable)	0.1

Engineering Controls

Use ventilation and dust collection to control exposure to below applicable

limits.

Individual Protection Measures (Personal Protective Equipment):

Respiratory Protection Wear NIOSH N-95 Dust Mask.

Eye Protection Eye protection (chemical goggles, safety glasses and/or face shield) should be

worn where there is a risk of lime exposure. Contact lenses should not be work

when working with lime products.

Hand Protection Use clean dry gloves.

Skin Protection Cover body with suitable clothes (long sleeves shirts and trousers). Use over

the angle waterproof caustic resistant footwear.

SECTION 9. PHYSICAL & CHEMICAL PROPERTIES

Appearance Solid, brown/white/tan/gray granular

Odor Odorless

Odor Threshold Not Applicable

pH 12.4 graduated solution at 25°C



Melting Point 1076°F (580°C) **Boiling Point** 5162°F (2850°C) Flash Point Not Applicable **Evaporation Rate** Not Applicable **Flammability** Not Applicable Upper/Lower Flammability Not Applicable Vapor Pressure (+to) Non-Volatile Vapor Density (air=ml) Non-Volatile 700-900 kg/ m³ **Relative Density** Solubility in Water 0.165g/100g **Partition Coefficient** Not Applicable **Auto-ignition Temperature** Not Applicable **Decomposition Temperature** Not Applicable Viscosity Not Applicable

SECTION 10. STABILITY & REACTIVITY

Stability The product is stable.

Decomposition Temperature None

Reactivity No specific test data related to reactivity available for this product.

Conditions to Avoid Vicinity of incompatible materials.

Incompatibility • Acids

Reactive Fluoridated

Brominated or Phosphorous CompoundsAluminum (may form hydrogen gas)

Reactive Powdered Metals
 Organic Acid Anhydrides
 Nitro-organic Compounds
 Interhalogenated Compounds

Hazardous Decomposition Products None

SECTION 11. TOXICOLOGICAL INFORMATION

Toxicity This product is not listed by MSA, OSHA, or IARC as a carcinogen, but this

product may contain crystalline silica, which has been classified by IARC as (Group 1) carcinogenic to humans when inhaled in the form of quartz or cristobalite. No reported Carcinogenicity, Reproductive Effects, Teratogenicity

or Mutagenicity.

Exposure Limits Refer to Section 8

Irritancy Can cause severe irritation of eyes, skin, respiratory tract, and gastrointestinal

tract.

Chronic Exposure: Inhalation of silica can cause a chronic lung disorder, silicosis.



SECTION 12. ECOLOGICAL INFORMATION

Alkaline substance that increases pH to 12.4 in a saturated water solution at 25°C.

Calcium hydroxide gradually reacts with CO2 in air to form calcium carbonate (CaCO3).

Calcium carbonate is ecologically neutral.

Uncontrolled spillage in surface waters should be avoided since the increase pH could be detrimental to fish.

Harmful to aquatic life in high concentration.

SECTION 13. DISPOSAL CONSIDERATIONS

Dispose according to federal, provincial/state, and local environmental regulations.

SECTION 14. TRANSPORT INFORMATION

 ${\bf Classification} \qquad \qquad {\bf TDG:} \qquad {\bf Not\; listed\; for\; ground\; transportation}.$

HMR: Not listed for ground transportation.

TDG: Transportation of Dangerous Goods Regulation (Canada)

HMR: Hazardous Materials Regulation (USA)

SECTION 15. Regulatory Information

Symbol WHMIS RATING

D2A, E

NFPA RATING

HEALTH-3 SPECIFIC HAZARD - ALK FLASH POINTS-0 REACTIVITY-1

HMIS RATING

HEALTH-2 SPECIFIC HAZARD - ALK FLASH POINTS-0 REACTIVITY-1

SECTION 16. OTHER INFORMATION

 Original Prepared
 05/13/2013

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 9/15/2023

Revision # 2

Calciment can be removed from vehicles using rags dampened with dilute vinegar. After applying dilute vinegar, vehicles (especially chrome surfaces) must be washed with water.

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