



Institute of Energy Process Engineering and
Chemical Engineering



Towards a CFD Model for Industrial Scale Gasification Processes

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Content

- 1 Introduction and Problem Statement
 - The HP POX Pilot Plant
- 2 Results Sydney Bluff-body Flame (HM1, NRBB)
 - Experimental data
 - Results NRBB1
 - Results NRBB2, Mixing of CH₄
 - Results Flame HM1
- 3 Conclusion and further steps

- Partial oxidation/ gasification at high pressure up to 100 bar.
- Investigate of reactive flows with the following characteristics: **turbulent mixing, combustion and downstream reactions.**
- Large-scale reactors of several meters require steady-state solution techniques.

Goal

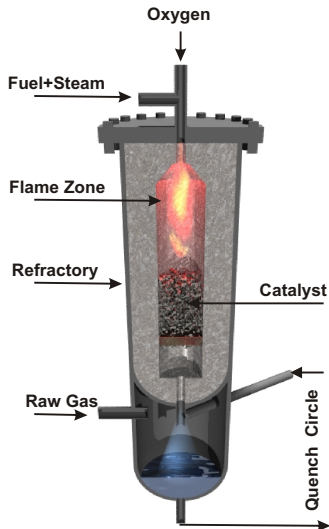
Stable steady-state solver capable of using detailed chemical reaction mechanisms.



HP-POX (High pressure partial oxidation) Pilot Plant, 5 MW thermal power, pressure up to 100 bar

HP-POX

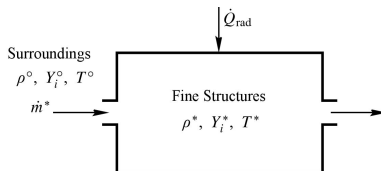
- Use of higher pressure levels to reduce size of plants and investment costs.
- Available modes: non-catalytic (POX), catalytic (ATR), liquid feedstock (MPG)
- Non-premixed educts \Rightarrow flame zone and reforming zone.



- Unsatisfying results of commercial CFD-codes due to combustion models.
- Development of a steady-state OpenFoam solver for chemistry started: **EDCSimpleFoam**

Steps towards that solver

- 1 Steady-state solver for inert mixing using the SIMPLE algorithm for reactingFoam.
- 2 Integration of detailed chemistry mechanisms with the EDC model (Eddy dissipation concept) of Magnussen [MAGNUSSEN, 1989, Gran and Magnussen, 1996]
 - CVODE for stiff ODE
 - CANTERA for chemical source terms
- 3 Basis: EDMSimpleFoam of T. Lucchini



EDC reactor model from [Kleiveland, 2005]

- Consider a cell and assume that the reactions only take place where they are mixed at a molecular level: **fine structure**. The rest is inert: **surrounding fluid**.
- The zones exchange mixture with a rate $\frac{1}{\tau}$ (see eq. (2))

Averaged chemical source terms RR_i for YEqn.h stem from a routine called `EDCSolve(τ , γ)` in class `chemistryModel` :

$$RR_i = \frac{\gamma^2 \rho}{\tau} (y_i^* - y_i^0) \text{ in } \left[\frac{kg}{m^3 s} \right] \quad (1)$$
$$\gamma = C_\gamma \left(\frac{\nu \epsilon}{k^2} \right)^{0.25} \text{ and } \tau = C_\tau \left(\frac{\nu}{\epsilon} \right)^{0.5}$$

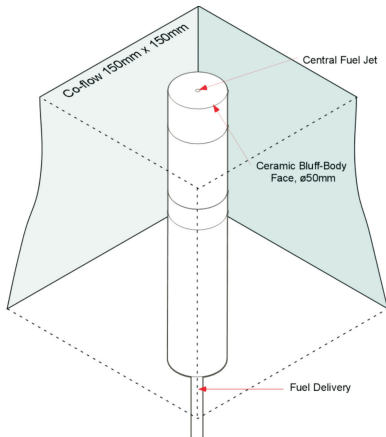
with ρ density, ν viscosity, y_i mass fractions of species i , k , ϵ turbulent kinetic energy and dissipation rate. Where '*' refers to fine structures and '0' to surrounding fluid.

Solve ODEs for an adiabatic ideal reactor (fine scales)

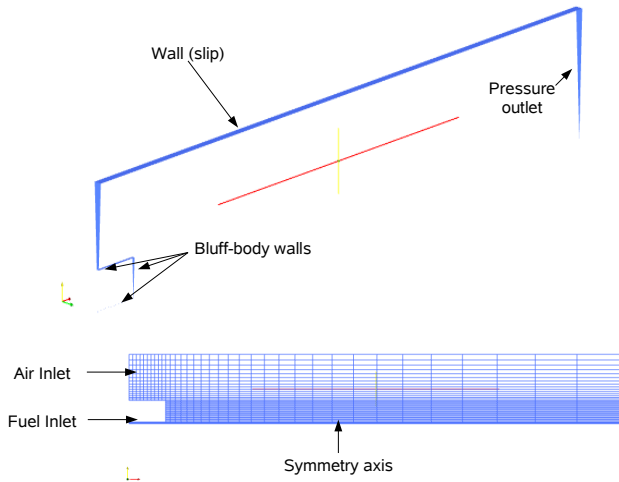
$$\frac{dy_i^*}{dt} = \frac{y_i^0 - y_i^*}{\tau} + \omega_i \quad (2)$$

where $\bar{y}_i = \gamma^3 y_i^* + (1 - \gamma^3) y_i^0$ and ω_i reaction rate

- Until steady-state of the system is reached (condition $\max(\Delta y, \Delta T/1000) < 10^{-5}$).
- This gives us y^* , y^0 for Eqn. (1).
- Integration with CVODE from SUNDIALS package.
- Reaction rates from CANTERA obtained with ATRMech [Zeißler, 2006] (28 Species, 112 Reactions, reduced version of GRIMECH 3.0 [et al,]).



Data from Sandia bluff-body flame HM1 and nonreacting jet NRBB conducted at Sydney university [Dally et al., 1998].



Boundary conditions

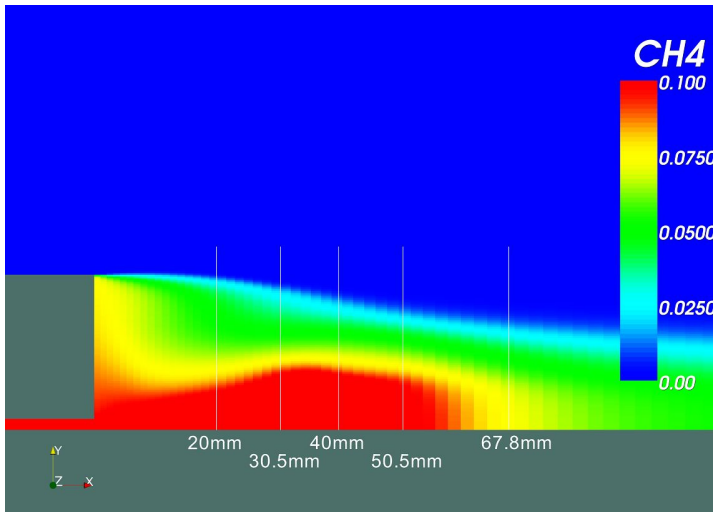
- **Diameters:** $D_{jet} = 3.6\text{mm}$; $D_{channel} = 150\text{mm}$ (300mm), $D_{bluffbody} = 50\text{mm}$
- **Turbulence:** 8.5 % (2.5%) turbulence intensity , 0.135 mm (5.625 mm) mixing length for jet (co-flow)
- **Mesh:** from blockMesh with 1095 and 23640 cells

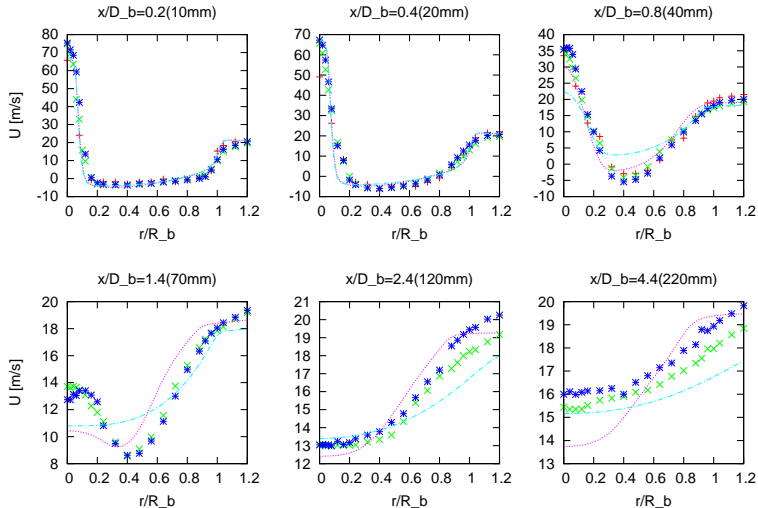
Three different inlet conditions:

- NRBB 1 (velocity): $U_{jet} = 61\text{m/s}$; $U_{coflow} = 20\text{ m/s}$
- NRBB 2 (inert mixing): $U_{jet} = 50\text{m/s}$; $U_{coflow} = 20\text{ m/s}$
- HM1 (reacting): $U_{jet} = 118\text{m/s}$ $U_{coflow} = 40\text{m/s}$

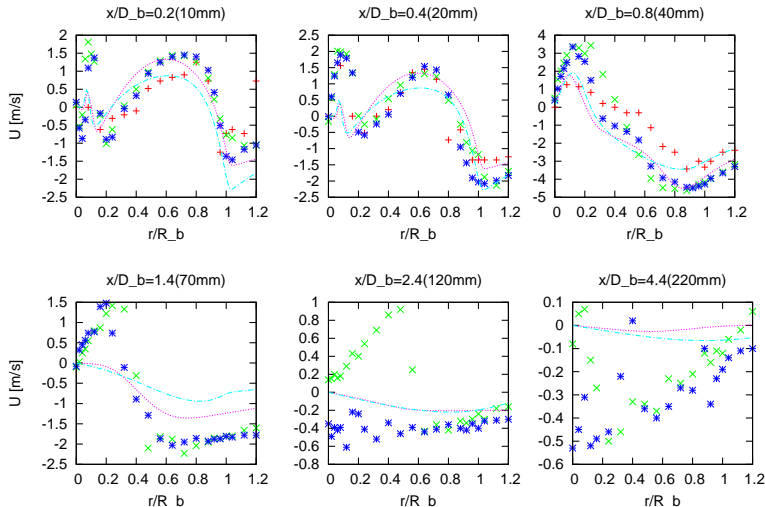
Other models:

- $k - \epsilon$ /RKE turbulence model
- Non-reacting cases without energy equation

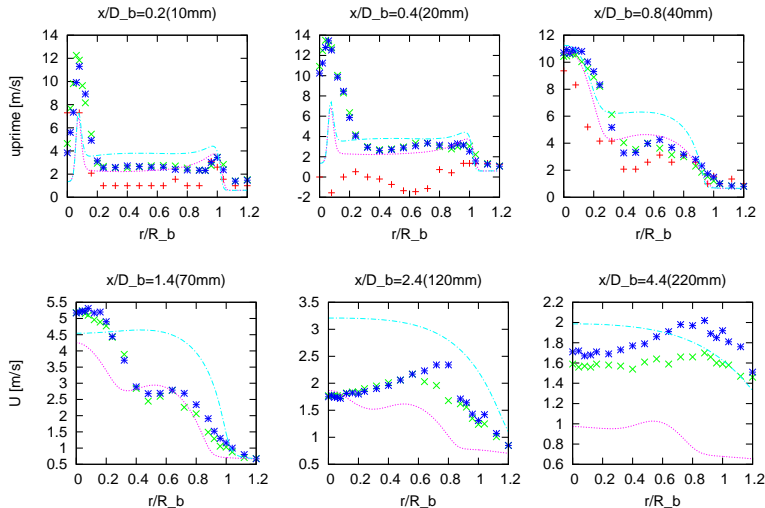




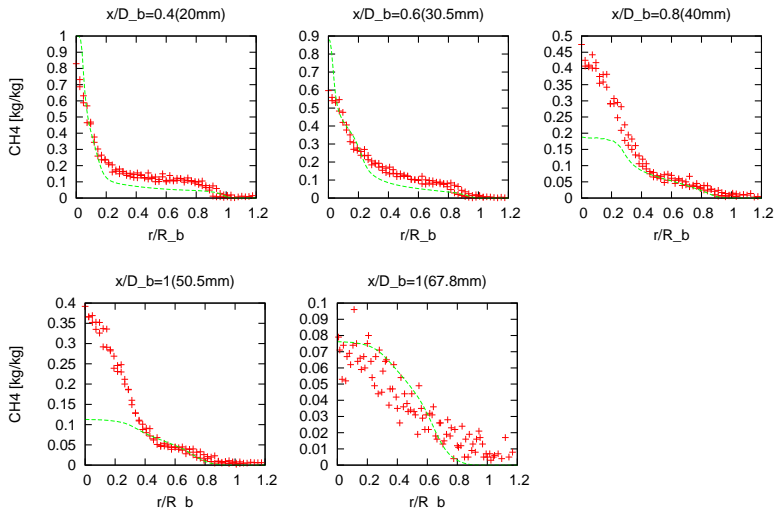
Axial velocity U_x , + HM1(exp.), lines EDCSimpleFoam, blue: geometry large; pink: geometry slim (as shown on page 11)



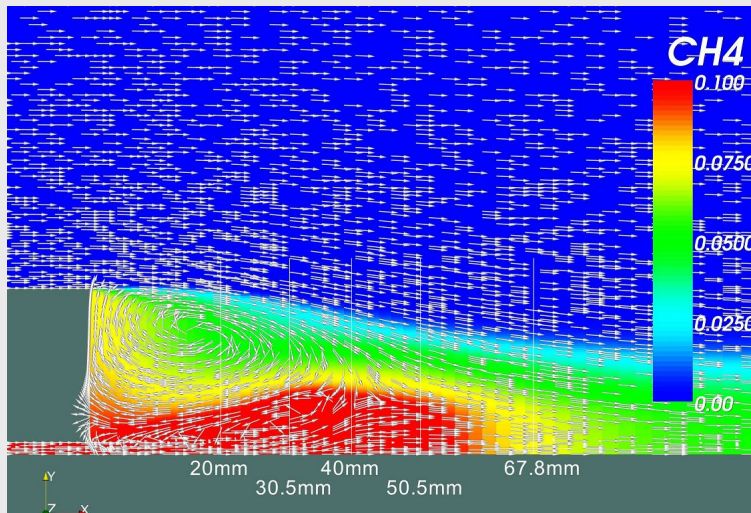
Radial velocity U_y , + HM1(exp.), lines EDCSimpleFoam, blue: geometry large; pink: geometry slim



Fluctuation of velocity U' , + HM1(exp.), lines EDCSimpleFoam,
blue: geometry large; pink: geometry slim

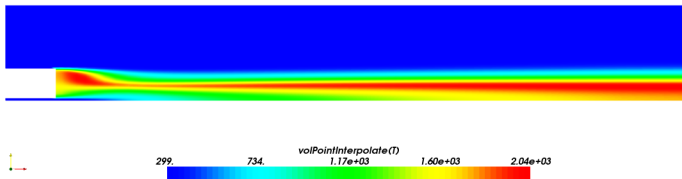


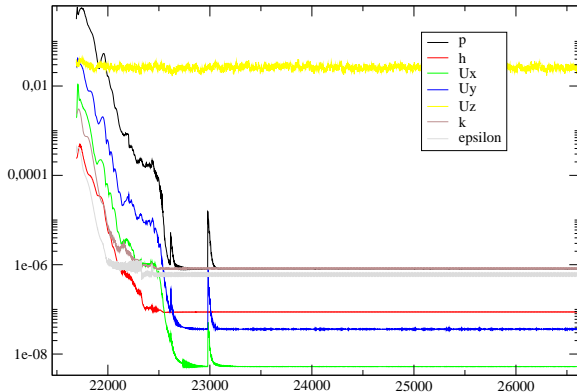
Mass fraction CH4, + NRBB(exp.), lines EDCSimpleFoam



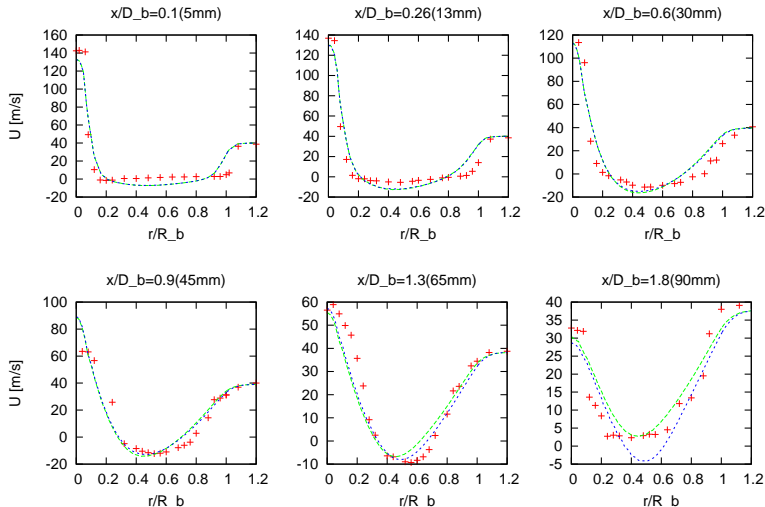
40mm, 50.5mm: CH₄-under-predicted due to vortex

- Calculation of HM1 flame
- Up to now only solution for coarse mesh
- Difficult at start-up, under-relaxation important

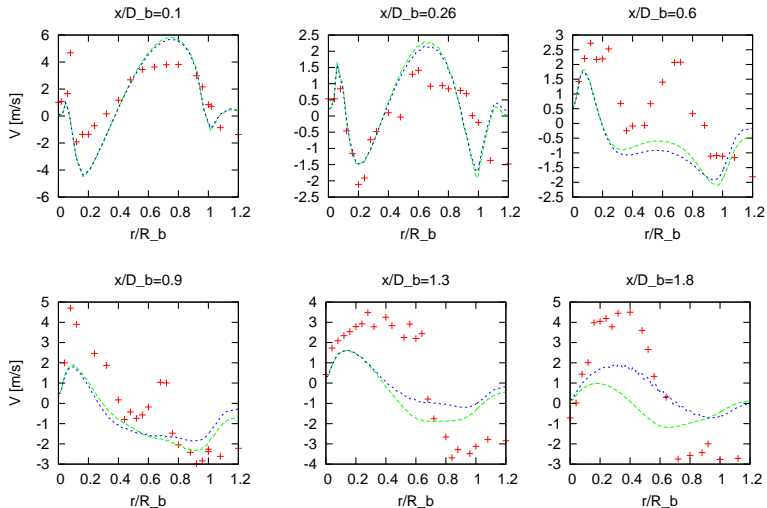




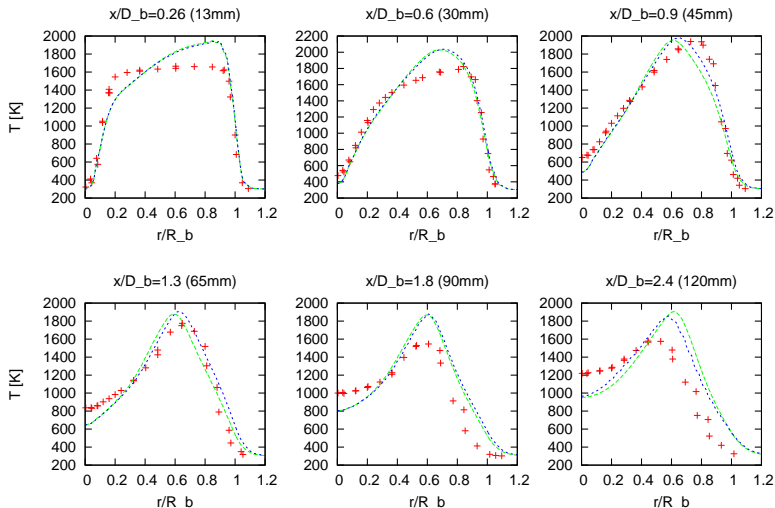
Residuals for calculation on coarse grid



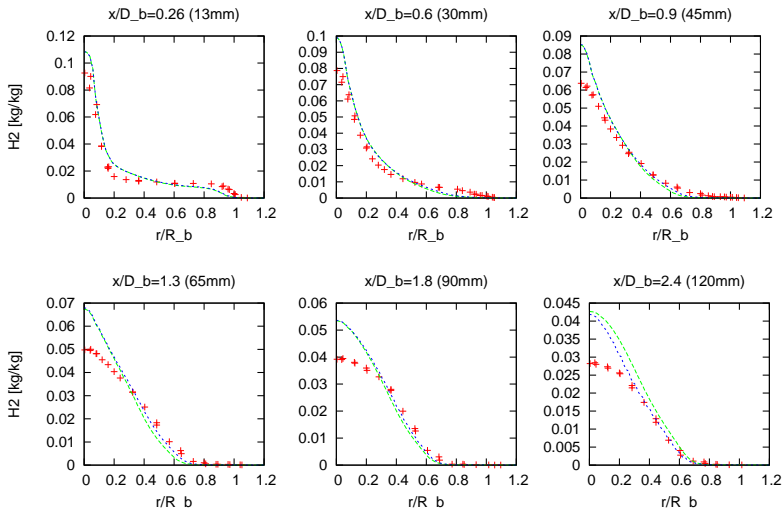
Axial velocity U , + HM1(exp.), lines EDCSimpleFoam



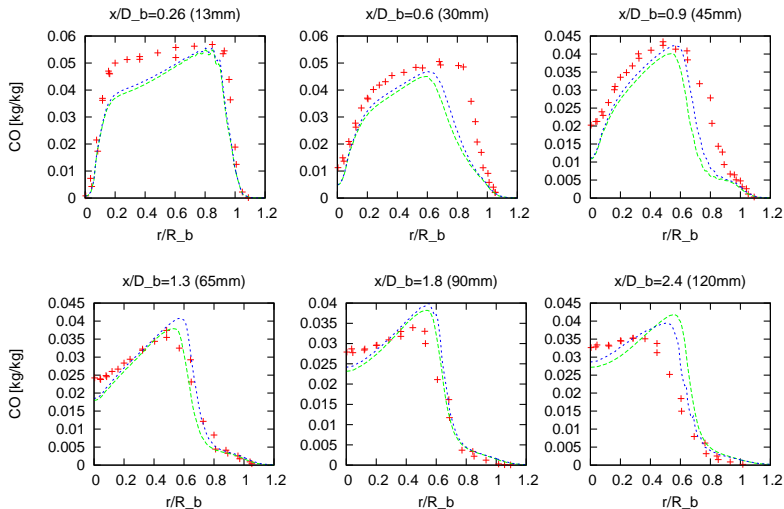
Radial Velocity V , + HM1(exp.), lines EDCSimpleFoam



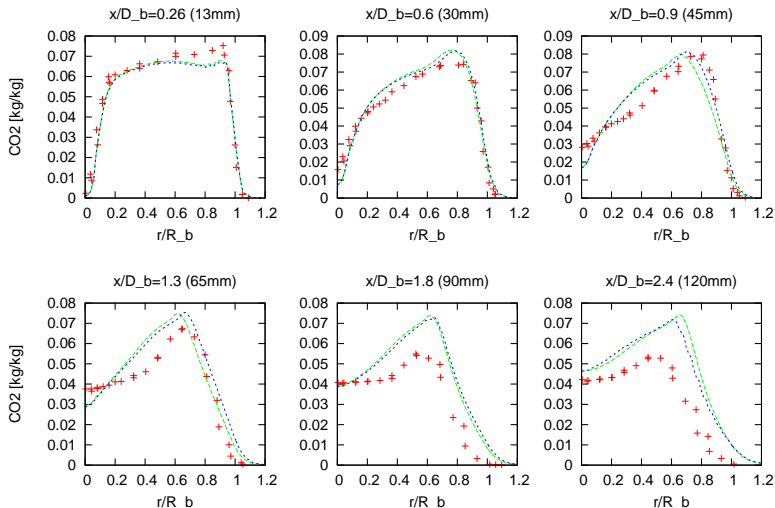
Temperature, + HM1(exp.), lines EDCSimpleFoam



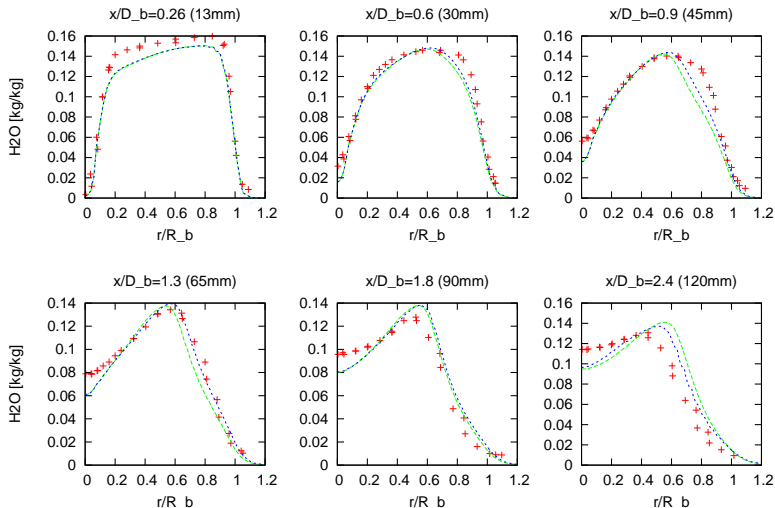
Hydrogen (H₂), + HM1(exp.), lines EDCSimpleFoam



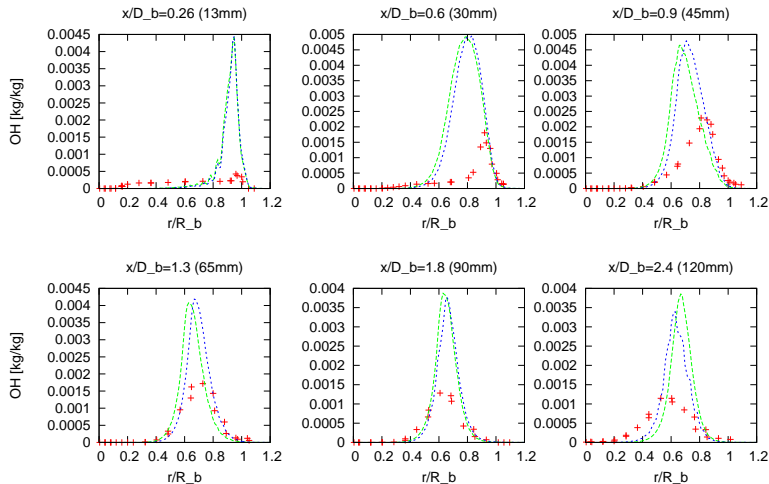
Carbon monoxide (CO), + HM1(exp.), lines EDCSimpleFoam



Carbon dioxide (CO₂), + HM1(exp.), lines EDCSimpleFoam



Hydrogen (H₂O), + HM1(exp.), lines EDCSimpleFoam



Hydrogen (OH), + HM1(exp.), lines EDCSimpleFoam

- Fluctuation and residual of U_z
- Temperature at mixing point where shear-stresses are highest (jet inlet)
- EDCFoam for unsteady calculation doesn't give same results yet

- Further study of the flame and checking of the results
- Stable solution for fine mesh
- Checking of unsteady solver EDCFoam
- Reduction of chemistry (ISAT and/or ILDM)
- Application to the whole HP-POX reactor
- Optimization and readability of code
- New class canteraThermo for chemistry, viscosity, diffusivity,
...
- **Virtuhcon** launch in 2009 will contribute to the development of high temperature conversion processes, too (see flyers).

- Tommaso Lucchini
- Lurgi GmbH
- all OpenFOAM developers
- the community
- Prof. Ertesvag



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