

93102R



# Scholarship 2022 Chemistry

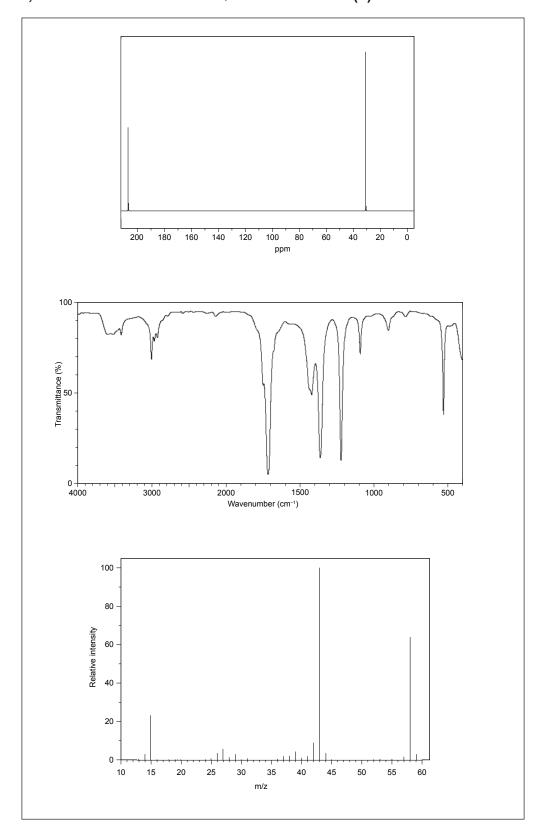
# RESOURCE BOOKLET

Refer to this booklet to answer the questions for Scholarship Chemistry.

Check that this booklet has pages 2–4 in the correct order and that none of these pages is blank.

YOU MAY KEEP THIS BOOKLET AT THE END OF THE EXAMINATION.

## <sup>13</sup>C NMR, MS, AND IR SPECTRA FOR QUESTION FOUR (a)



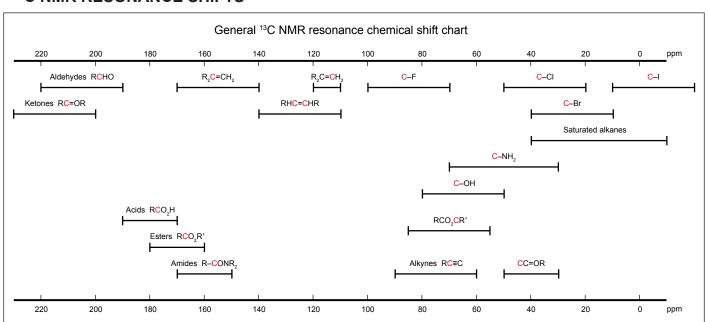
### SPECTROSCOPY DATA SHEET

### **INFRARED SPECTROSCOPY**

Functional group	Vibration	Wavenumber/ cm <sup>-1</sup>
Alkane	C-H stretch	2950-2800 (s)
Alkene	C=C-H stretch	3100-3010 (s)
Aikerie	C=C stretch	1690-1630 (m)
	C-F stretch	1400-1000 (s)
Alkyl	C-Cl stretch	785-540 (m-w)
halide	C-Br stretch	650-510 (s-m)
	C-I stretch	600-485 (s-m)
Alcohol	O-H stretch	3600-3300 (s)
AICOHOI	C-O stretch	1260-1000 (s)
	N–H stretch (1 per bond)	3500-3300 (s-w)
Amine	N-H bend	1640-1500 (s)
	C-N stretch	1200-1025 (s)

Functional group	Vibration	Wavenumber/ cm <sup>-1</sup>					
Aldehyde	C=O stretch	1725 (s)					
Ketone	C=O stretch	1715 (s)					
	O-H stretch	3400 (s)					
Carboxylic acid	C=O stretch	1730-1700 (s)					
aoia	C-O stretch	1320-1210 (s)					
Acid	C=O stretch	1810-1775 (s)					
chloride	C-CI stretch	730-550 (s-m)					
Ester	C=O stretch	1750-1735 (s)					
ESIEI	C-O stretch	1260-1160 (s)					
Amide	N–H stretch	3500-3200 (s)					
	C=O stretch	1680-1630 (s)					

### <sup>13</sup>C NMR RESONANCE SHIFTS



# PERIODIC TABLE OF THE ELEMENTS

		4)	_		4)	7		_	0		•	~		4)			_	~			
18	2	H	4.0	10	Ne	20.2	18	Ar	40.0	36	Kr	83.8	54	X	13]	98	R	222	118	Og	
			17	6	<b></b>	19.0	17	C	35.5	35	Br	79.9	53	_	127	85	At	210	117	L	
			91	8	0	16.0	16	S	32.1	34	Se	79.0	52	Te	128	84	Po	210	116	Lv	
			15	7	Z	14.0	15	Ь	31.0	33	As	74.9	51	$\mathbf{S}\mathbf{p}$	122	83	Bi	209	115	Mc	
			14	9	C	12.0	14	Si	28.1	32	Ge	72.6	50	Sn	119	82	Pb	207	114	I	
			13	5	В	10.8	13	Al	27.0	31	Ga	69.7	49	In	115	81	Ι	204	113	$\mathbf{N}\mathbf{h}$	
									12	30	$\mathbf{Z}\mathbf{n}$	65.4	48	Cd	112	80	$_{ m Hg}$	201	112	$\mathbf{C}\mathbf{n}$	277
									II	29	Cn	9.89	47	Ag	108	62	Au	197	1111	Rg	272
			mass						10	28	Z	58.7	46	Pd	106	82	Pt	195	110	Ds	271
			Relative atomic mass						6	27	$C_0$	58.9	45	Rh	103	77	Ir	192	109	Mt	268
			Relativ						8	26	Fe	55.9	44	Ru	101	92	Os	190	108	Hs	265
	1	Н	1.0						_	25	Mn	54.9	43	Tc	6.86	75	Re	186	107	Bh	264
·	umber								9	24	Cr	52.0	42	Mo	95.9	74	M	184	106	S	263
	Atomic number								5	23	>	50.9	41	Sp	92.9	73	Ta	181	105	Db	262
	7								4	22	Ξ	47.9	40	$\mathbf{Zr}$	91.2	72	Ht	179	104	Rf	261
									3	21	Sc	45.0	6	X	88.9	.1	Lu	175	103	$\Gamma$ r	262
			7	4	Be	9.0	12		24.3	20	Ca	40.1	38	Sr	87.6	56	Ba	137	88	Ra	226
			I	3	Ľ	6.9	11	Na	23.0	19			37	Rb	85.5	55	Cs			Fr	223

	57	58	59	09	61	62	63	64	65	99	29	89	69	70
Lanthanide	La	Ce	Pr	Nd	Pm	Sm	Eu	Сd	$\mathbf{T}\mathbf{b}$	Dy	Ho	Er	Tm	Yb
Series	139	140	141	144	147	150	152	157	159	163	165	167	169	173
	68	06	91	92	93	94	95	96	76	86	66	100	101	102
Actinide	Ac	$\mathbf{L}\mathbf{h}$	Pa	n	dN	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No
Series	227	232	231	238	237	239	241	244	249	251	252	257	258	259