

Assignment-0

Q1.a) Show that  $D_{\max} = \frac{b}{T}$

Let the wavelength be  $\lambda$

Planck energy density is given by  $\rightarrow$

$$u(\lambda) d\lambda = \frac{8\pi hc}{\lambda^5} \frac{1}{e^{[hc/\lambda k_B T]} - 1} d\lambda$$

for planck energy density to be maximum,

$$\frac{du}{d\lambda} = 0$$

$$u = \frac{8\pi hc}{\lambda^5} \frac{1}{e^{[hc/\lambda k_B T]} - 1}$$

$$\frac{du}{d\lambda} = 0$$

$$0 = \frac{8\pi hc}{\lambda^6} (-5) \frac{1}{e^{[hc/\lambda k_B T]} - 1} - \frac{8\pi hc}{\lambda^5} \cdot \frac{\rho}{(e^{[hc/\lambda k_B T]} - 1)^2} \cdot \left(\frac{hc}{k_B T}\right) \left(\frac{-1}{\rho}\right)$$

$$5 = \frac{1}{(e^{[hc/\lambda k_B T]} - 1)} \cdot \frac{hc}{k_B T} \cdot \frac{1}{\lambda} \cdot e^{[hc/\lambda k_B T]}$$

$$\text{Let } \frac{hc}{k_B T} = a$$

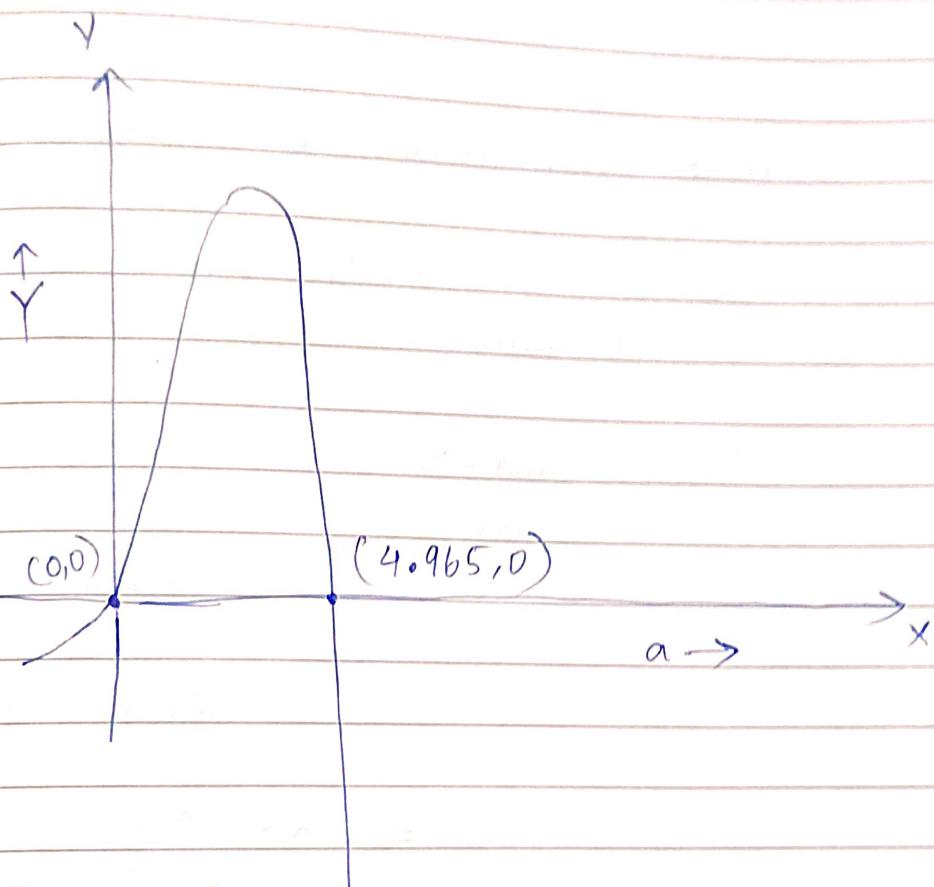
Then, we have  $\rightarrow$

$$5 = \frac{1}{(e^a - 1)} \cdot a e^a$$

$$5(e^a - 1) = ae^a$$

$$5(e^a - 1) - ae^a = 0$$

On plotting the graph of  $Y = 5(e^a - 1) - ae^a$ , we get  $\rightarrow$



Clearly,  $Y = 0$  when  $a = 0, 4.965$   
since we have to find  $x_{\max}$ ,  
 $\therefore a \neq 0 \& a = 4.965$

$$\therefore a = \frac{hc}{\lambda K_B T}$$

$$\frac{hc}{\lambda_{\max} K_B T} = 4.965$$

$$\Rightarrow \lambda_{\max} = \frac{hc}{(4.965) K_B} \cdot \frac{1}{T}$$

We can take  $b = \frac{hc}{4.965 k_B}$ , then  $\rightarrow$

$$\lambda_{\text{max}} = \frac{b}{T}$$

Let  $\lambda_{\text{max}} = \delta_{\text{max}}$  (as per question)  
 $\therefore \delta_{\text{max}} = \frac{b}{T}$  (hence proved)

b) surface temperature of a star = ?

Intensity at  $\lambda = 446 \text{ nm}$

Intensity = ?

Wien's Law  $\rightarrow$

$$\lambda_{\text{max}} = \frac{b}{T} = \frac{hc}{4.965 k_B} \cdot \frac{1}{T}$$

$$= 2.89777 \times 10^{-3} \cdot \frac{1}{T}$$

$$T = \frac{2.89777 \times 10^{-3}}{446 \times 10^{-9}}$$

$$T = 6497.085 \text{ K}$$

$$T = 6497.085 \text{ Kelvin}$$

$$\text{Intensity} = R = \sigma T^4$$

$$\text{where } \sigma = 5.67 \times 10^{-8}$$

$$R = (5.67 \times 10^{-8})(6497.085)^4$$

$$R = 101.0316 \times 10^6 \text{ W/m}^2$$

c)

$$T = 3300 \text{ K}$$

$$\lambda = \frac{b}{T} \rightarrow \text{from Wien's law}$$

$$\lambda = \frac{2.8977 \times 10^{-3}}{3300}$$

$$\lambda = 8.78091 \times 10^{-7} \text{ m}$$

$$\text{Intensity} = R = \sigma T^4, \sigma = 5.67 \times 10^{-8}$$

$$R = (5.67 \times 10^{-8}) (3300)^4$$

$$R = 6.72417207 \times 10^6 \text{ W/m}^2$$

$$\phi_2, D_1 = 80 \text{ nm}, D_2 = 110 \text{ nm}$$

$$K.E_1 = E_1 = 11.390 \text{ eV}, E_2 = 7.154 \text{ eV}$$

a) Find  $h = ?$

$$\frac{hc}{\lambda} = K.E + \phi$$

$$K.E = \frac{hc}{\lambda} - \phi$$

$$E_1 = 11.390 \times 1.6 \times 10^{-19} = \frac{hc}{80 \times 10^{-9}} - \phi \quad \textcircled{1}$$

$$E_2 = 7.154 \times 1.6 \times 10^{-19} = \frac{hc}{110 \times 10^{-9}} - \phi \quad \textcircled{2}$$

$\textcircled{1} - \textcircled{2} \rightarrow$

$$1.6 \times 10^{-19} [11.390 - 7.154] = hc \left[ \frac{10^9}{80} - \frac{10^9}{110} \right]$$

$$= hc \frac{(110 - 80)}{110 \times 80}$$

$$\Rightarrow h = 6.626986667 \times 10^{-34} \text{ J-s}$$

b)  $\phi$  - work function = ?

$v_0 = ?$ ,  $\lambda_0 = ?$

$$\frac{7.154 \times 1.6 \times 10^{-19} - hc}{110 \times 10^{-9}} = -\phi$$

$$-\phi = \frac{7.154 \times 1.6 \times 10^{-19} - 6.62648667 \times 10^{-34} \times 3 \times 10^8}{110 \times 10^{-9}}$$

$$\Rightarrow \phi = 6.627200091 \times 10^{-19} \text{ J}$$

$$\phi = 6.6272 \times 10^{-19} \text{ J}$$

$$h v_0 = \phi$$

$$v_0 = \frac{6.627200091 \times 10^{-19}}{6.62648667 \times 10^{-34}}$$

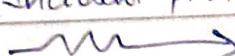
$$v_0 = 10^{15} \text{ Hz}$$

$$\lambda_0 = \frac{c}{v_0} = 3 \times 10^{-7} \text{ m}$$

Q3. a) ... Wavelength shift = ?

$$\lambda_f - \lambda_i = \Delta\lambda = \frac{h}{moc} (1 - w\delta\theta)$$

(Incident photon)



$\lambda_i$

$e^-$  (at rest)

$\phi$

$\theta$

Scattered photon  
 $\lambda_f$



$moc$

since the photons are back scattered,  $\theta = \pi$

$$\therefore \lambda_f - \lambda_i = \frac{h}{moc} (1 - (-1)) = \frac{2h}{moc}$$

$moc = \text{rest mass of } e^-$

$$\lambda_f - \lambda_i = \Delta\lambda = 2 \times \frac{6.626 \times 10^{-34}}{3 \times 10^8 \times 9.11 \times 10^{-31}}$$

$$\Delta\lambda = 4.8488401 \times 10^{-12} \text{ m}$$

$$\Delta\lambda = 4.849 \times 10^{-12} \text{ m}$$

b) Show That  $\rightarrow$  energy of scattered photons is half the rest mass energy of  $e^-$  regardless of energy of incident photons.

Let energy of scattered photons be  $E_2$

$$\text{then, we have } \rightarrow E_2 = \frac{hc}{\lambda_f}$$

$$\text{where } \lambda_f = \lambda_i + \frac{2h}{moc}$$

$$\lambda_f = 4.849 \times 10^{-12} + \lambda_i$$

$$\lambda_f = \lambda_i + \frac{2h}{moc}$$

$$E_2 = \frac{hc}{\lambda_i + \frac{2h}{moc}} \quad E_1 = \frac{hc}{\lambda_i}$$

$$E_2 = \frac{hc}{\frac{hc}{E_1} + \frac{2h}{moc}} = \frac{c}{\frac{c}{E_1} + \frac{2}{moc}}$$

$$E_2 = \frac{c}{c + \frac{2}{m_0 c}} = \frac{c [m_0 (E_1)]}{c^2 m_0 + 2 E_1}$$

$m_0$  = rest mass  
of  $e^-$

$$E_2 = \frac{m_0 c^2 E_1}{c^2 m_0 + 2 E_1}$$

$E_1$  = rest mass

given  $E \gg m_0 c^2$

energy of  $e^-$

$$\therefore E_2 = \frac{m_0 c^2}{2 + \frac{c^2 m_0 / E_1}{2}} = \frac{m_0 c^2}{2}$$

$$\frac{c^2 m_0}{E_1} \rightarrow 0$$

$$E_2 = \frac{m_0 c^2}{2}$$

We know the rest mass energy  
of  $e^-$  is  $E_1 = m_0 c^2$

$$\therefore E_2 = \frac{E_1}{2}$$

$\therefore$ , The energy of scattered photons is half  
the rest mass energy of the  $e^-$ .

$$c) E_2 = \frac{E_1}{2} = \frac{m_e c^2}{2} = \frac{(9.1 \times 10^{-31}) \times (3 \times 10^8)^2}{2}$$

$$E_2 = 256.21875 \times 10^3 \text{ eV}$$

$$E_2 = 0.256 \text{ MeV} = 0.256 \times 10^6 \text{ eV}$$

From

conservation of energy  $\rightarrow K_e = E - E_2$  where  $E = 150 \text{ MeV}$

$$E_2 = 0.256 \text{ MeV}$$

$$K_e = (150 - 0.256) \text{ MeV}$$

$$K_e = 149.744 \text{ MeV}$$

Q4.

The minimum energy  $E_{\min}$  if a photon required to produce an electron-positron pair must be equal to the rest mass energies of the electron & positron, this corresponds to the case where the kinetic energies of the electron & positron are zero.

The K.E of  $e^+$  &  $e^-$  will be zero for minimum case.

$$\begin{aligned} h\nu &= E_e^- + E_e^+ + E_N \\ &= (m_e c^2 + K_{e^-}) + (m_e c^2 + K_{e^+}) + K_N \end{aligned}$$

$$K_{e^-} = K_{e^+} = 0$$

$$\begin{aligned} \therefore h\nu_{\min} &= E_{\min} = 2 m_e c^2 \\ &= 2 \times 0.511 \text{ MeV} \\ &= 1.022 \text{ MeV} \end{aligned}$$

$$E_{\min} = 1.022 \text{ MeV}$$

$$\nu = \frac{E_{\text{min}} = h\nu}{6.626 \times 10^{-34}} = \cancel{1.022 \times 10^6 \times 1.6 \times 10^{-19}}$$

$$\nu = 2.4715 \times 10^{20} \text{ Hz}$$

$$\nu = 2.4715 \times 10^{20} \text{ Hz}$$

$$\lambda = \frac{c}{\nu} = \frac{3 \times 10^8}{2.4715 \times 10^{20}}$$

$$\lambda = 1.2138 \times 10^{-12} \text{ m}$$

Q5. e)  $K.E = 70 \text{ MeV} \rightarrow \text{proton}$

$$\lambda = ?$$

$$K.E = \frac{P^2}{2m_p} \quad \& \quad \lambda = \frac{h}{P}$$

$$\Rightarrow P = \frac{h}{\lambda}$$

$$K.E = \frac{h^2}{\lambda^2 2m_p}$$

$$70 \times 10^6 \times 1.6 \times 10^{-19} = \frac{h^2}{\lambda^2 2m_p}$$

$$\lambda^2 = \frac{h^2}{(K.E) \cancel{\times} (2m_p)}$$

$$\lambda = \frac{h}{\sqrt{2m_p(K.E)}} = \frac{6.626 \times 10^{-34}}{\sqrt{2m_p(K.E)}}$$

$$m_p = 1.672 \times 10^{-27} \text{ kg}$$

$$K.E = 70 \times 10^6 \times 1.6 \times 10^{-19} \text{ J}$$

$$\lambda = \frac{6.626 \times 10^{-34}}{\sqrt{2 \times (70 \times 10^6 \times 1.6 \times 10^{-19}) \times (1.672 \times 10^{-27})}}$$

$$\lambda = 3.423807081 \times 10^{-15} \text{ m}$$

$$\lambda = 3.42381 \times 10^{-15} \text{ m}$$

b)  $\lambda = \frac{h}{P} = \frac{h}{mv} = \frac{6.626 \times 10^{-34}}{(100 \times 10^{-3})(900)}$

$$\lambda = 7.362 \times 10^{-36} \text{ m}$$

Q6.  $\Delta x = ?$

$$\Delta x \Delta p \geq \frac{h}{4\pi}$$

$$\Delta x \geq \frac{h}{m \Delta p}$$

$$\Delta x = \frac{6.626 \times 10^{-34}}{m \Delta p}$$

$$m_N = 1.65 \times 10^{-27} \text{ kg}$$

$$v = 5 \times 10^6 \text{ m/s}$$

$$\Delta x \approx \frac{6.626 \times 10^{-34}}{4\pi (1.65 \times 10^{-27} \times 5 \times 10^6)}$$

$$\Delta x \approx 6.39 \times 10^{-15} \text{ m}$$

$$b) \Delta n \geq \frac{h}{2\Delta p}$$

$$\Delta n \approx \frac{h}{2mv} = \frac{h}{4\pi m v}$$

$$\Delta n \approx \frac{6.626 \times 10^{-34}}{4\pi \times 50 \times 2}$$

$$\Delta n \approx 5.2728 \times 10^{-37} \text{ m}$$

Q7. a)  $E_n = ?$   $r_n = ?$

$$\text{Reduced mass } \mu = \frac{m_1 m_2}{m_1 + m_2}$$

$$\text{for } e^- + e^+, m_1 = m_2 = m_e$$

$$\mu = \frac{m_e^2}{2me} = \frac{me}{2}$$

$$\text{Bohr's radius } a_0 = \frac{4\pi \hbar^2 k^2}{me^2}$$

$$n^{\text{th}} \text{ orbit radius} = r_n = a_0 n^2$$

$$\text{for positronium} - a_0, P = \frac{4\pi \hbar^2 k^2}{me^2}$$

$$= 2 \left( \frac{4\pi \hbar^2 k^2}{me^2} \right)$$

$$= 2 \left( \frac{4\pi \hbar^2 k^2}{me^2} \right)$$

$$= 2a_0$$

$$r_n = 2a_0 n^2$$

$$r_n = \frac{8\pi^2 \hbar^2}{m_e e^2} n^2$$

$$r_n \approx 0.106 n^2 \text{ nm}$$

$$E_n = -\frac{2\pi^2 m_e e^4 z^2}{h^2 (4\pi\hbar)^2} \cdot \frac{1}{n^2}$$

$$= -\frac{13.6}{n^2}$$

For positronium  $\rightarrow$

$$E_n = -\frac{13.6}{2n^2}$$

$$E_n = \frac{\pi^2 m_e e^4 z^2}{h^2 (4\pi\hbar)^2} \cdot \frac{1}{n^2}$$

$$E_n = -\frac{6.8 \text{ eV}}{n^2}$$

$$r_p = 2r_e, E_p = E_e/2$$

b) Values of energy & radii of 3 lowest states

$$r_1 = 2r_e = 0.106 \text{ nm}$$

$$r_2 = 2r_e(2) = 8r_e = 0.424 \text{ nm}$$

$$r_3 = 2r_e(3) = 18r_e = 0.954 \text{ nm}$$

$$E_1 = -6.8 \text{ eV}$$

$$E_2 = -1.7 \text{ eV}$$

$$E_3 = -0.755 \text{ eV}$$

$$c) f = ?$$

$$\lambda = ?$$

energy of 1st excited state  
is  $E_2$

$$E_2 = -1.7 \text{ eV}$$

$$h\nu = E_\infty - E_2$$

$$= 0 - (-1.7 \text{ eV})$$

$$= 1.7 \times 1.6 \times 10^{-19}$$

$$h\nu = 1.7 \times 1.6 \times 10^{-19} \text{ J}$$

$$\nu = \frac{1.7 \times 1.6 \times 10^{-19}}{6.626 \times 10^{-34}}$$

$$\nu = 4.105 \times 10^4 \text{ Hz}$$

$$\lambda = \frac{c}{\nu} = \frac{3 \times 10^8}{4.105 \times 10^4}$$

$$\lambda = 7.308 \times 10^{-7} \text{ m}$$

$$\text{Q8. a) } P = 45 \text{ kW}$$

$$f = 4 \text{ MHz}$$

a) no. of photons emitted per second

$$n = \frac{E}{h\nu} = \frac{45 \times 10^3}{6.626 \times 10^{-34} \times 4 \times 10^6}$$

$$n = 1.797 \times 10^{31} \text{ photons per second.}$$

b) As the no. of photons emitted is very large, quantum effects can be ignored. We can ignore the quantum nature of the electromagnetic radiation emitted from the antenna.

$$\text{Q9. } m = 4 \text{ kg}, k = 196 \text{ N/m}$$

$$n = 25 \text{ cm} = 0.25 \text{ m}$$

$$\begin{aligned} \text{a) } v &= \frac{1}{2\pi} \sqrt{\frac{k}{m}} \\ &= \frac{1}{2\pi} \sqrt{\frac{196}{4}} = \frac{7}{2\pi} = 1.114 \text{ Hz} \end{aligned}$$

$$E = \frac{1}{2} k x_A^2 = \frac{1}{2} (196) (0.25)^2$$

$$E = 6.125 \text{ J}$$

$$\begin{aligned} \text{b) energy spacing } \Delta E &= h\nu \\ &= (6.626 \times 10^{-34}) (1.114) \end{aligned}$$

$$\Delta E = 7.386 \times 10^{-34} \text{ J}$$

$$n = \frac{E}{\Delta E} = \frac{6.125}{7.386 \times 10^{-34}}$$

$$n = 8.4619 \times 10^{33}$$

Since the energy of 1 quanta is very small compared to the total energy, the energy levels of the oscillators can be assumed as continuous. The quantum effects are not important.

Q10.

a)  $d = 0.27 \text{ nm} = \lambda$

$$P = \frac{2\pi h}{\lambda}$$

$$E = \frac{P^2}{2m_e} = \frac{4\pi^2 \hbar^2}{\lambda^2 (2m_e)}$$

$$E = 4\pi^2 \left( 1.055 \times 10^{-34} \right)^2 / (0.27 \times 10^{-9})^2 (2 \times 9.11 \times 10^{-31})$$

$$E = 3.308 \times 10^{-18} \text{ J}$$

$$E = 20.676 \text{ eV}$$

b)  $K.E = 2 \text{ eV}$  protons

5<sup>th</sup> max intensity observed at  $30^\circ$ , find crystal's plane separation.

Bragg's formula  $\rightarrow$

$$n\lambda = 2d \sin \theta$$

$$5\lambda = 2d \sin 30^\circ = d$$

$$E = \frac{P^2}{2m_p} = \frac{2\pi^2 \hbar^2}{m_p \lambda^2}$$

$$m_p = 1.672 \times 10^{-27} \text{ kg}$$

$$2 \times 1.672 \times 10^{-19} \times E = 2\pi^2 \left( 1.055 \times 10^{-34} \right)^2 \times 25$$

$$1.672 \times 10^{-27} \times d^2$$

$$\Rightarrow d = 1.013197732 \times 10^{-10} \text{ m}$$

Q11. a) Conservation of energy  $\rightarrow$

$$h\nu + m_e c^2 = h\nu' + (K_e + m_e c^2)$$

$\nu, \nu'$   $\rightarrow$  frequency of initial & scattered photons respectively.

$m_e c^2$   $\rightarrow$  rest mass energy of the initial electron

$K_e \rightarrow$  recoil kinetic energy.

$$K_e = h(\nu - \nu')$$

$$= h c \left( \frac{\lambda' - \lambda}{\lambda \cdot \lambda'} \right) = \frac{hc}{\lambda} \left( \frac{\lambda' - \lambda}{\lambda'} \right)$$

$$K_e = \frac{h\nu}{\lambda'} \frac{\Delta\lambda}{\lambda}$$

$$\text{wavelength shift} = \Delta\lambda = \lambda' - \lambda$$

$$= \frac{h(1-\cos\theta)}{m_e c}$$

$$= 2\lambda c \sin^2 \left( \frac{\theta}{2} \right)$$

$$\Delta\lambda = \frac{2\pi hc}{m_e c^2} (1-\cos\theta)$$

$$\Delta\lambda = 2\pi \times \frac{h}{2\pi} \cdot \frac{1}{m_e c} (1-\cos 60^\circ)$$

$$m_e = 9.11 \times 10^{-31} \text{ kg}$$

$$\Delta\lambda = 2\pi \times \frac{6.626 \times 10^{-34}}{2\pi} \cdot \frac{(1-\cos 60^\circ)}{9.11 \times 10^{-31} \times 3 \times 10^8}$$

$$\Delta\lambda = 0.0012 \text{ nm}$$

$$\lambda' = \Delta\lambda + \lambda = \Delta\lambda + \frac{2\pi hc}{h\nu}$$

$$\lambda' = 0.0012 + 0.414$$

$$\lambda' = 0.4152 \text{ nm}$$

$$k_e = (hv) \left( \frac{0.0012}{0.4152} \right)$$

$$k_e = (3 \times 10^3) \left( \frac{0.0012}{0.4152} \right)$$

$$k_e = 8.671 \text{ eV}$$

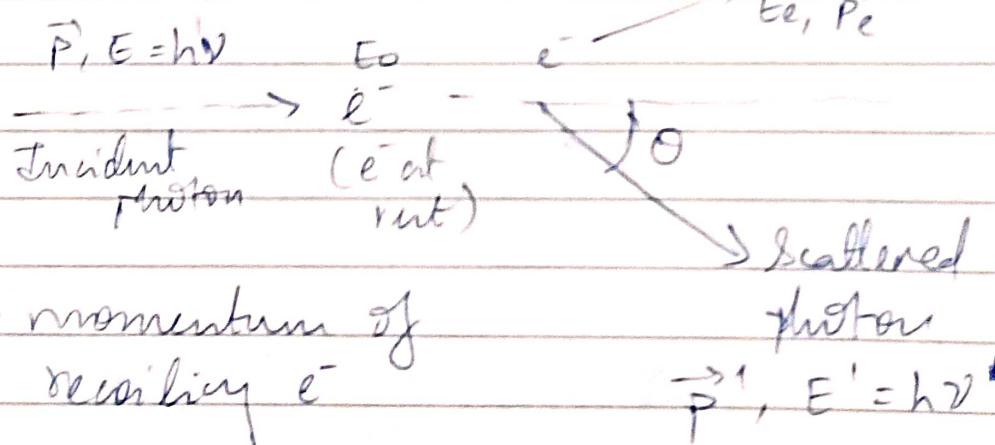
b) conservation of total momentum along  
n-t y-axis

$p, p'$  - initial & final momentum  
of photon respectively

$$p = p_e \cos\phi + p' \cos\theta \quad \text{--- (1)}$$

$$0 = p_e \sin\phi - p' \sin\theta \quad \text{--- (2)}$$

Recoil



$p_e \rightarrow$  momentum of  
describing  $e^-$

Scattered

photon

$$\vec{p}', E' = h\nu'$$

$$p_e \cos\phi = p - p' \cos\theta$$

$$p_e \sin\phi = p' \sin\theta$$

$$\tan\phi = \frac{\sin\theta}{\frac{p - p' \cos\theta}{p'}} = \frac{\sin\theta}{\frac{\lambda' - \cos\theta}{\lambda}}$$

$$p = h/\lambda, p' = h/\lambda'$$

$$\lambda = 0.414 \text{ nm}$$

$$\lambda' = 0.4152 \text{ nm}$$

$$\phi = \tan^{-1} \left( \frac{\sin \theta}{\frac{x'}{\lambda} - w \theta} \right)$$

$$= \tan^{-1} \left( \frac{\sin 60^\circ}{\frac{0.4152}{0.414} - 10860^\circ} \right)$$

$$\phi = 59.86^\circ$$

Q12.  $E = mc^2 = \underbrace{m_0 c^2}_{\sqrt{1 - v^2/c^2}}$

$$\phi = mv = \underbrace{m_0 v}_{\sqrt{1 - v^2/c^2}}$$

$$E = c \sqrt{p^2 + m_0^2 c^2}$$

$$c^2 (p^2 + m_0^2 c^2) = \frac{m_0^2 c^4}{1 - \frac{v^2}{c^2}}$$

We know  $\rightarrow$

$$\text{group velocity } v_g = \frac{dE(p)}{dp}$$

$$v_g = \frac{1}{2} \cdot \frac{1}{\sqrt{p^2 + m_0^2 c^2}} (c)(2p)$$

$$v_g = \frac{pc}{\sqrt{p^2 + m_0^2 c^2}} = v$$

$$\text{Phase velocity } v_p = \frac{E(p)}{p} = c \sqrt{1 + \frac{m_0^2 c^2}{p^2}} = \frac{c^2}{v}$$

Q3.

a) According to uncertainty principle  $\rightarrow$  $r \approx h$  where  $r \rightarrow$  radius of the  $e^-$  orbit $P \rightarrow e^-$  is momentum

$$P \approx \frac{h}{r}$$

To find ground state radius, minimize electron-proton energy  $\rightarrow$ 

$$E(r) = \frac{P^2}{2m_e} - \frac{e^2}{4\pi\epsilon_0 r}$$

$$= \frac{\frac{h^2}{r^2}}{2m_e} - \frac{e^2}{4\pi\epsilon_0 r}$$

$$\frac{dE(r)}{dr} = 0$$

$$-\frac{h^2}{m_e r_0^3} + \frac{e^2}{4\pi\epsilon_0 r_0^2} = 0$$

$$\Rightarrow r_0 = \frac{4\pi\epsilon_0 h^2}{m_e e^2} = 0.053 \text{ nm}$$

$$b) E(r_0) = \frac{h^2}{2m_e r_0^2} - \frac{e^2}{4\pi\epsilon_0 r_0}$$

$$= -\frac{m_e}{2h^2} \left( \frac{e^2}{4\pi\epsilon_0} \right)^2 = -13.6 \text{ eV}$$

Q14. Rest mass energy =  $m_N c^2 = 939.6 \text{ MeV}$

$$\Delta x_0 = 10^{-14} \text{ m}$$

$$\text{Time constant } T = \frac{2m_\mu^2 (\Delta x_0)^2}{\hbar c^2}$$

$$T = 2 \times (1.674 \times 10^{-27}) \times 10^{-28} \times 2 \times$$

$$6.626 \times 10^{-34}$$

$$T = 3.174 \times 10^{-21} \text{ s}$$

Time taken by a neutron packet to grow from an initial width  $\Delta x_0$  to a final size  $\Delta x(t)$  is -

$$t = T \sqrt{\left(\frac{\Delta x(t)}{\Delta x_0}\right)^2 - 1}$$

$$t = (3.174 \times 10^{-21}) \sqrt{\left(\frac{\Delta x(t)}{\Delta x_0}\right)^2 - 1}$$

a) 4 times its original size ( $\Delta x(t) = 4x_0$ )

$$t = 3.174 \times 10^{-21} \sqrt{4^2 - 1}$$

$$t = 3.174 \times 10^{-21} \sqrt{15}$$

$$t = 12.29 \times 10^{-21}$$

$$t = 1.229 \times 10^{-20} \text{ s}$$

b) size equal to earth's diameter  
(diameter =  $12.7 \times 10^6$ )

$$t = 3.174 \times 10^{-21} \sqrt{\left(\frac{12.7 \times 10^6}{10^{-14}}\right)^2 - 1}$$

$$t = 4.18$$

c) time equal to the distance between earth & moon ( $3.84 \times 10^8$ )

$$t = 3.174 \times 10^{-21} \sqrt{\left(\frac{3.84 \times 10^8}{10^{-14}}\right)^2 - 1}$$

$$t = 12.38$$