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COMMON FIRST REVISION EX	AMINATION - JANUARY - 2020
Time: 3.00 Hours X STAN	
nstructions: (1) Check the question paper f	or fairness of printing. If there is any lack
or fairness, inform the Hall	Supervisor immediately write and pencil or draw diagrams.
Jote: (i) Anguer - II	Part - I
ode and the corresponding answer.	the four alternatives and write the option
1) Impulse is equals to	
a) rate of change of momentur	
c) change of momentum	d) rate of change of mass
2) The eye defect 'presbyopia' car	n be corrected by
a) Convex lens	b) Concave lens
c) Convex mirror	d) Bifocal lenses
3) The sound waves are reflected	
medium from which they were changes?	incident. Which of the following
a) Speed	b) Frequency
c) Wavelength	d) None of these
4) In the nuclear reaction $_{6}X^{12} \xrightarrow{decay}_{Z} Y$	A the value of A and Z.
a) 8, 6	b) 8, 4
c) 4, 8 d) cannot b	be determined with the given data
5) is an important metal	to form amalgam
a) Ag	b) Hg
c) Mg	d) Al
6) Deliquencence is due to	
a) strong affinity to water	h) loss offinity to wat
c) strong hatred to water	b) less affinity to water
Which of the following is not ap	d) Inertness to water
"element + element → compound"	
a) $C_{(s)} + O_{2(g)} \rightarrow CO_{2(g)}$	b) $2K_{(s)} + Br_{2(l)} \rightarrow 2KBr_{(s)}$
c) $2CO_{(g)} + O_{2(g)} \rightarrow 2CO_{2(g)}$	d) $4Fe_{(s)} + 3O_{2(g)} \rightarrow 2Fe_2O_{3(s)}$
	2 3(3)

- 19) Which hormone is known as 'time messenger'? Why?
- 20) Write the characteristic of insect pollinated flowers.
- 21) What is shale gas?
- 22) Three resistors of resistances 5 ohm, 3 ohm and 2 ohm are connected in series with 10V battery. Calculate their effective resis tance and the current flowing through the circuit.

## Part - III

## Note: Answer any seven Questions. Q.No. 32 is compulsory. 7×4=28

- 23) i) Define electric potential and potential difference.
  - ii) State ohm's law.
- 24) i) What do you understand by the term 'ultrasonic' vibration?
  - ii) State three uses of ultrasonic vibrations.
- 25) i) Give an example each i) gas in liquid
  - ii) solid in liquid
  - iii) Solid in solid
  - iv) gas in gas
  - ii) A hot saturated solution of copper sulphate forms crystals as it cools. Why?
- 26) i) What is the pH value of human saliva and milk of Magnesia?
  - ii) The hydroxide ion concentration of a solution is 1×10-11 M. What is the pH of the solution?
- 27) i) Explain the excretory system of leech.
  - ii) List out the parasitic adaptations in leech.
- 28) Illustrate the structure and functions of brain.
- 29) i) What do you understand by the term phenotype and genotype?
  - ii) What are allosomes?
- 30) i) How can you determine the age of the fossile?
  - ii) What are the types of variations?
- 31) Discuss the importance of biotecnology in the field of medicine.
- 32) The molecular formula of an alcohol is C4H10O. The locant number of its -OH group is 2.
  - i) Draw the structural formula

- ii) Give its IUPAC name.
- iii) Is it saturated or unsaturated?

## Part - IV

Note: Answer all the Questions

 $3 \times 7 = 21$ 

Draw diagram wherever necessary.

- 33) a) i) Give the applications of universal law of gravitation.
  - ii) A door is pushed, at a point whose distance from the hinger is 90 cm, with a force of 40N. Calculate the moment of the force about the hinges.

(or)

- b) i) Explain the process of controlled and uncontrolled chain reaction.
  - ii) Compare the properties of alpha, beta and gamma radiations (any 4 properties)
- 34) a) i) Explain froth floatation process.
  - ii) Metal A belongs to period 3 and group 13. A in red hot condition reacts with steam to form B. A with strong alkali forms C. Find A, B and C with reactions.

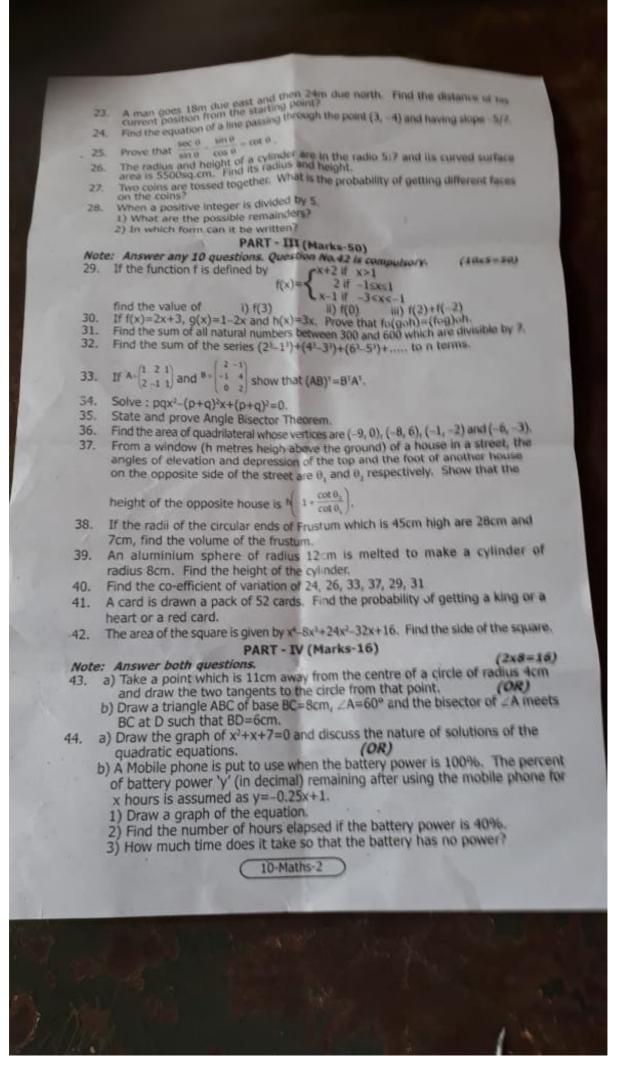
(or)

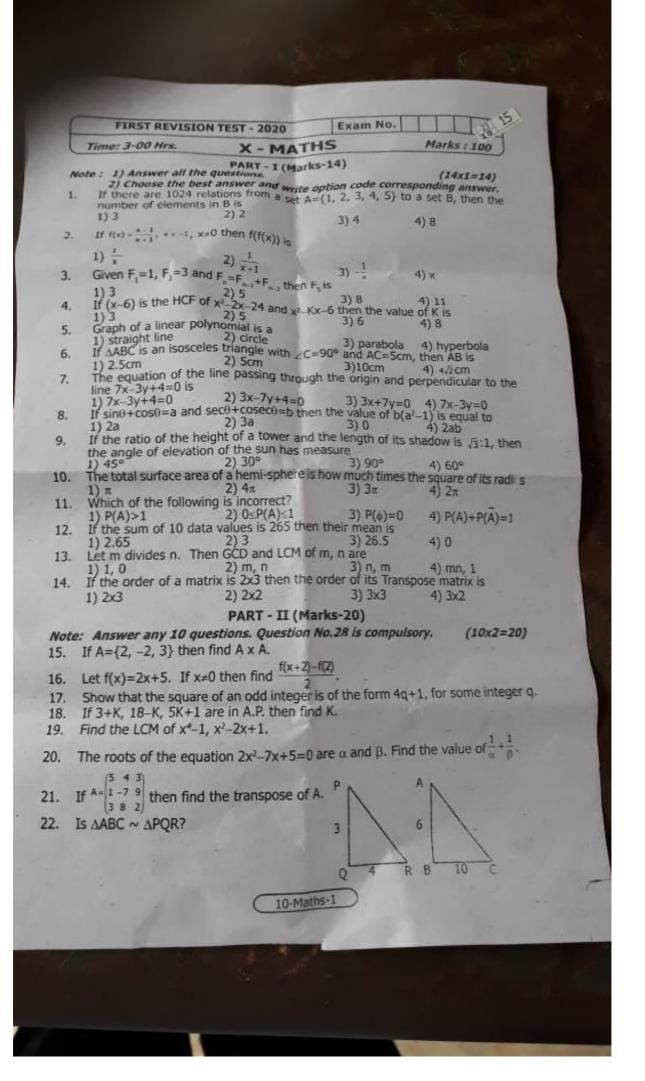
- b) i) Differentiate soaps and detergents.
  - ii) What is called homologous series? Give any three of its characteristics?
- 35) a) i) What is transpiration? Give the importance of transpiration.
  - ii) Differentiate Artery and Vein. (any 3)

(or)

- b) i) Suggest measures to overcome the problems of an alcoholic.
  - ii) Explain the types of cancers.

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Given series: 
$$(2^{3}-1)+(4^{3}-3^{3})+(6^{3}-5^{3})+\cdots$$
 $S_{N} = \sum [(2^{3}-1^{3})+(4^{3}-3^{3})+(6^{3}-5^{3})+\cdots]$ 
 $= \sum [(2^{3})^{3}-(2^{3}-1)^{3}]$ 
 $= \sum [(2^{3}-1)^{3}+(2^{3}-1)^{3}]$ 
 $= \sum [(2^{3}-1)^{3}+(2^{3}-1)^{3}+(2^{3}-1)^{3}]$ 
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