Tanushree Mazumdor, Alysa Maduz, Noah Krooszewski, Khushi Patel, Rebecca Song (sparks)
5 A B (S)

Brisa Lopez Estrada / Muna Nnadonic
LTC ↳ (Beiberdorf)

Part I

① (A) $40\ g\ A \times \frac{1\ mol\ A}{60\ g\ A} \times \frac{5\ mol\ C}{3\ mol\ A} = 1.11\ mol\ C$

$35\ g\ B \times \frac{1\ mol\ B}{40\ g\ B} \times \frac{5\ mol\ C}{2\ mol\ B} = 2.1875\ mol\ C$

② $40\ g\ A \times \frac{1\ mol\ A}{60\ g\ A} \times \frac{1\ mol\ D}{3\ mol\ A} \times \frac{75\ g\ D}{1\ mol\ D} = 16.7\ g\ D$

③ $16.67\ g\ D \times \frac{1\ mol\ D}{75\ g\ D} \times \frac{2\ mol\ B}{1\ mol\ D} \times \frac{40\ g\ B}{1\ mol\ B} = 17.78\ g\ B$

$35 - 17.78 = 17.22\ g\ B$

Part II

① $Pb(NO_3)_2$ and KI

② Lead (II) nitrate and potassium nitrate

③ $Pb(NO_3)_{2(aq)} + 2KI_{(aq)} \rightarrow PbI_{2(s)} + 2KNO_{3(aq)}$

④ $9.22\ g\ PbI_2 \times \frac{1\ mol\ PbI_2}{461\ g} = 0.02\ mol\ PbI_2$

⑤

$0.22\ M\ Pb(NO_3)_2 = \frac{0.022\ mol\ Pb(NO_3)_2}{0.1\ L}$

$0.56\ M\ KI \times \frac{0.056\ mol}{0.1\ L} = 0.056\ mol\ KI$

$0.022\ mol\ Pb(NO_3)_2 \times \frac{1\ mol\ PbI_2}{1\ mol\ Pb(NO_3)_2} \times \frac{461\ g}{1\ mol\ PbI_2} = 10.142$

$0.056\ mol\ KI \times \frac{1\ mol\ PbI_2}{2\ mol\ KI} \times \frac{461\ g}{1\ mol} = 12.908$

$\frac{9.22}{10.142} \times 100 = 90.9\ \%$

Part III

① $735\ torr \times \frac{1\ atm}{760\ torr} \times \frac{1.0135\ bar}{1\ atm} = 0.98\ bar$

② Increase b/c the particles will now hit the 'walls' more

③ They have the same

④ The compressed sample