Exp 6: Surface Analysis

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Introduction

Here, we will study the surface of Huaman hemoglobin, including hydrophobicity, conservation and electrostatic potential. We will use APBS to generate a electrostatic potential surface.

APBS, the Adaptive Poisson-Boltzmann Solver, is a freely available macromolecular electrostatics calculation program released under the GPL. It is a cost-effective but uncompromised alternative to GRASP, and it can be used within PyMOL. PyMOL can display the results of the calculations as an electrostatic potential molecular surface.

Methods

1. Load human hemoglobin structure (PDB id: 4HHB) in PyMOL

```
PyMOL> fetch 4HHB
```

2. Create new objecs containing beta-subunit.

```
PyMOL> select beta, chain B
```

- 3. Separate human hemoglobin and hemo group into two objects
- 4. Generate the surfaces to display the two different properties of the protein
 - Hydrophobicity

```
PyMOL> set surface_color, white, beta,
PyMOL> show surface, beta
PyMOL> select hydrophobicity, resn Ala+Val+IIe+Leu+Met+Phe+Tyr+Trp in beta
PyMOL> remove backbone in hydrophobicity
PyMOL> set surface_color, blue, hydrophobicity
PyMOL> show surface, hydrophobicity
PyMOL> set transparency, 0.2
```

Conservation

I aligned protein 4HHB-B, 4BJA-A,4MPM-A,4MPM-B using structural information by tcoffee. The identical residues were marked as red, and highly similar residues were marked as respherry, and similar residues were marked as warmpick.

5. Generate the electrostatic potential surfaces for the tetrameric hemoglobin

 $Action \longrightarrow generate \longrightarrow vacuum \ electrostatics \longrightarrow protein \ contact \ potential \ (local)$

- 6. Use APBS program to calculte the electrostatic potential surface
- 7. Study the muatation E6V.

 $Wizard \longrightarrow Mutagenesis \longrightarrow E6 \longrightarrow V6$

Results

1. Hydrophobicity

Save the result in Hydrophobicity.pse sence 003. The hole which heme insert is hydrophobicity, in oder to keep heme inside hemoglobin.(Figure 1) Many sunkens surface is hydrophobic(Figure 2), which suggestes show the folding occuar to decrease system energy. However, here are raised surface is hydrophobic.(Figure 3) The hydrophobic environment between subunit interacte edge may explain this exception.

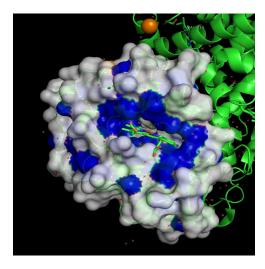
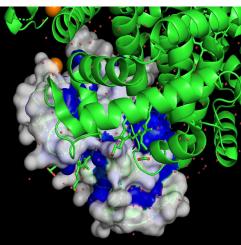


Figure 1. Heme insert to a hydrophobic hole



 ${\bf Figure~3.~} {\bf Subunit~interacte~surface~is~more~hydrophobic}$

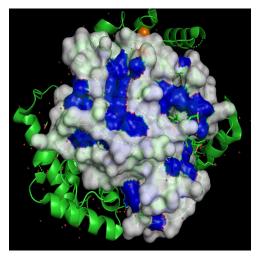


Figure 2. Much sunken surface is hydrophobic

2. Convsersion

I find following:

- i. The hemo binding sites are identical convservation.
- ii. Most interaction edge is similar residues.
- iii. At non subunit interaction surface, most loops and trun are conservation

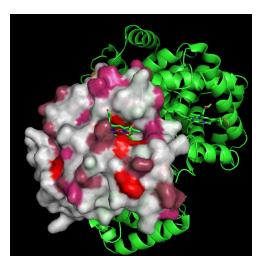


Figure 4. The Hemo binding site is

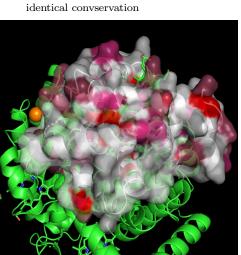


Figure 6. At no interaction surface, most loops and turn are conservation

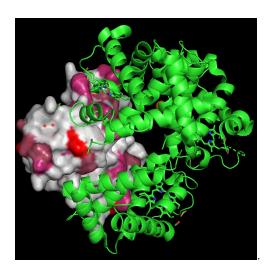


Figure 5. Most interaction edge is similar residues

3. Electrostatic in vacuum

Because 2,3-BPG is negative charge, I find a positive charged surface between two β sub-units.(Figure 4)(Reference 4)(electrostatic.pse sence 3). I find that four Heme interactioal surface on hemoglobin are positive charged (electrostatic.pse sence 4). However, Heme doesn't show much negative feactures.

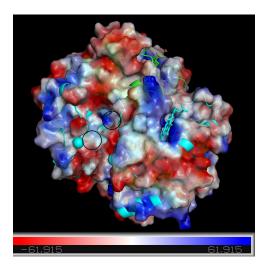


Figure 7. The positive charged pocket for 2,3-BPG binding. Butoom balck circle is a PO^{4-} , and top black circle is responding positive surface to hold PO^{4-} . (Two β subunits are colored by cyans.

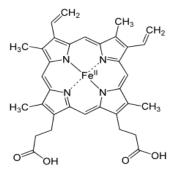


Figure 9. Heme b group (Ref 2)

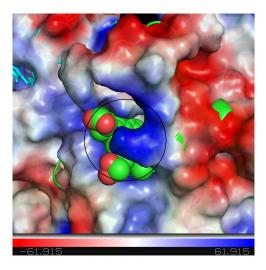


Figure 8. Here are positive charged surface on homoglobin around a Heme (Black circle).

4. Electrostatic potential surface using APBS program (APBS.pse)

The 2,3-BPG binding surface show more charged than surrounding, which show high charge constract and more reasonable biogical function than simple calculation. Also, the charge distruction is more smooth and continue than simple calculation.

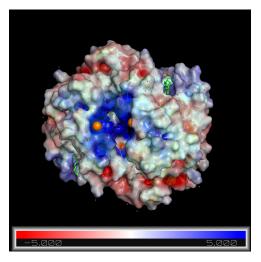


Figure 10. The 2,3-BPG site show smooth positive charged surface

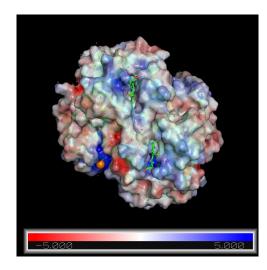


Figure 11. Surface arounds Heme

5.

Conclusions

References

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- 3. https://en.wikipedia.org/wiki/2,3-Bisphosphoglyceric acid
- $4.\ http://cbc.chem.arizona.edu/classes/bioc462/462a/NOTES/hemoglobin/hemoglobin_function.htm$
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