

Homework I

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1.6-1. Find a the most probable distribution P_n such that,

$$\begin{aligned}\sum P_n &= N \\ \sum_n n P_n &= N n_a \\ \sum_n n^2 P_n &= N n_d^2\end{aligned}$$

Here the most probable distribution is defined as the distribution P_n of which the following quantity

$$\begin{aligned}\ln \Omega\{P_n\} &= \ln N_p! - \sum_n \ln P_n! \\ &= \ln N_p! - \sum_n P_n \ln P_n + P_n\end{aligned}$$

is maximized.

Solve:

Using the Langrari Multitlier

$$\begin{aligned}S(\alpha, \beta, \gamma, \{P_n\}) &= \ln \Omega\{P_n\} - \alpha \left(\sum_n P_n - N \right) - \beta \left(\sum_n n P_n - N n_a \right) - \gamma \left(\sum_n n^2 P_n - N n_d^2 \right) \\ \frac{\delta S}{\delta P_n} &= -\ln P_n - \alpha - n\beta - n^2\gamma = 0 \\ P_n &= e^{-\alpha - n\beta - n^2\gamma}\end{aligned}$$

Then apply P_n to the distribution property,

$$\begin{aligned}P_n &= \frac{1}{\sqrt{2\pi(n_d^2 - n_a^2)}} e^{-\frac{(n - n_a)^2}{2(n_d^2 - n_a^2)}} \\ P_n &= \frac{N}{\sqrt{2\pi(n_d^2 - n_a^2)}} e^{-\frac{(n - n_a)^2}{2(n_d^2 - n_a^2)}}\end{aligned}$$

1.6-2. For a ideal gas with distribution

$$P(v) = \prod_{i=1}^3 \sqrt{\frac{m}{2\pi\kappa_B T}} e^{-\frac{m(v_i - \bar{v}_i)^2}{2\kappa_B T}}$$

Please find out the state equation

Solve:

energy may not equated in each direction

We known

$$P = \frac{F}{A}$$

$$P = \frac{N1}{V3}m\bar{v}^2$$

Then, get the \bar{v}^2 from distribution.

$$\begin{aligned}\langle mv_i \rangle &= m\bar{v}_i \\ \langle mv_i^2 \rangle &= \kappa_B T + \langle mv_i \rangle \\ &= \kappa_B T + m\bar{v}_i \\ \langle mv^2 \rangle &= 3\kappa_B T + 3m\bar{v}\end{aligned}$$

Finally,

$$\begin{aligned}P &= \frac{N1}{V3}m\bar{v}^2 \\ &= \frac{N1}{V3}(3\kappa_B T + 3m\bar{v}) \\ &= \frac{N}{V}\kappa_B T + \frac{N}{V}m\bar{v} \\ \Rightarrow PV &= N(k_B T + m\bar{v})\end{aligned}$$

2.6-2. Find a the most probable distribution $P(x)$ such that,

$$\begin{aligned}\int P(x) &= 1 \\ \langle x \rangle &= \int P(x)x dx = \mu \\ \langle x^2 \rangle - \langle x \rangle^2 &= \int P(x)x^2 dx - \left(\int P(x)x dx\right)^2 = \sigma^2\end{aligned}$$

Here the most distribution is defined as the distribution $P(x)$ of which the entropy

$$S = - \int P(x) \ln P(x) dx$$

is maximized. Compared the result with the problem 1 of chapter 1. What do you find?

Solve:

Using the priciple of maximum entropy

$$\begin{aligned}\frac{\delta}{\delta P(x)} \left(S - \alpha \left(\int P(x) - 1 \right) - \beta \left(\int P(x)x dx - \mu \right) - \gamma \left(\int P(x)x^2 dx - \sigma^2 - \mu^2 \right) \right) &= 0 \\ - \int (\ln P(x) + 1) dx - \alpha \int 1 dx - \beta \int x dx - \gamma \int x^2 dx &= 0 \\ - \int 1 dx - \alpha \int 1 dx - \beta \int x dx - \gamma \int x^2 dx &= \int \ln P(x) dx \\ - \int (1 + \alpha + \beta x + \gamma x^2) dx &= \int \ln P(x) dx \\ -1 - \alpha - \beta x - \gamma x^2 &= \ln P(x) \\ e^{-1 - \alpha - \beta x - \gamma x^2} &= P(x)\end{aligned}$$

using the constrain to determine α, β, γ

$$P(x) = \frac{1}{\sqrt{2\pi\sigma^2}} e^{-\frac{(x-\mu)^2}{2\sigma^2}}$$

This problem get more a normalize factor than problem 1 of chapter 1.

2.6-3. For different $N = 1, 10, 100, 1000$, numerically calculate the following fuctions,

$$\sum_{n=1}^{100} \exp[N/n^2]$$

Compared it with the function $\exp[N]$ (then $n = 1$ term, the largest term, in the above function), what do you find?

Solve:

N	1	10	100	1000
$\sum_{n=1}^{100} \exp[N/n^2]$	102.397	22141.87	2.688exp(43)	1.97exp(434)
$\exp[N]$	2.718	22026.46	2.688exp(43)	1.97exp(434)

The larger N , the closer $\exp[N]$ and $\sum_{n=1}^{100} \exp[N/n^2]$. It is a example of rule of large number

2.6-4. The probability for observing a closed thermally equilibrate system with given energy E is in forms of

$$P(E) = C \Omega(E) \exp(-\beta E)$$

where C is the normalization constant and $\Omega(E)$ is the number of microsates with enery E . Both $\ln \Omega(E)$ and E are of the order of the particle number N .

- 1). Using the steepest descent method to verify that $P(E)$ is a very narrow distribution centered on the most probable value of E (denoted as E_0).
- 2). Use the expansion to estimated the probability for observing a spontaneous fluctuation in E of the order of $10^{-6}E_0$ for 1 moles of gas. (Hint: Expanding $\ln P(E)$ in powers of $\delta E = E - E_0$. Formular $S(E) = \kappa_B \ln \Omega(E)$, $\partial E / \partial S = T$, $\partial E / \partial T = C_v$, $E_0 = 3N\kappa_B T / 2$ and $C_v = 3N\kappa_B / 2$ are needed in this calculation.

Solve:

- 1). Try to calculate the mean and variance to show $P(E)$ is narraow distribution.

$$\begin{aligned} \int E P(E) dE &= \int E C \Omega(E) \exp(-\beta E) dE \\ &= C \int \exp(\ln \Omega(E) + \ln E - \beta E) dE \\ &\cong \end{aligned}$$

$$\begin{aligned} \int E^2 P(E) dE &= \int E^2 C \Omega(E) \exp(-\beta E) dE \\ &= C \int \exp(\ln \Omega(E) + 2 \ln E - \beta E) dE \\ &= \end{aligned}$$

设 E_0 是极大值, 将 $\ln(P)$ 在 E_0 处展开,

$$\begin{aligned}
\ln P(E) &= \ln P(E_0) - (E - E_0) \frac{\partial \ln P(E)}{\partial E} \Big|_{E_0} + \frac{1}{2} (E - E_0)^2 \frac{\partial^2 \ln P(E)}{\partial E^2} \Big|_{E_0} \\
\ln P(E) &= \ln P(E_0) + \frac{1}{2} \delta E^2 \frac{\partial^2 \ln P(E)}{\partial E^2} \Big|_{E_0} \\
\frac{\partial \ln P(E)}{\partial E} &= \frac{\partial (C \Omega(E) \exp(-\beta E))}{\partial E} = C \frac{\partial (\exp(-\beta E + \ln \Omega(E)))}{\partial E} \\
&= C \left(-\beta + \frac{\partial \ln \Omega(E)}{\partial E} \right) \exp(-\beta E + \ln \Omega(E)) \\
\frac{\partial^2 \ln P(E)}{\partial E^2} &= C \left(\frac{\partial^2 \ln \Omega(E)}{\partial E^2} \right) \exp(-\beta E + \ln \Omega(E)) + C \left(-\beta + \frac{\partial \ln \Omega(E)}{\partial E} \right)^2 \exp(-\beta E + \ln \Omega(E)) \\
&= C \left(\frac{\partial^2 \ln \Omega(E)}{\partial E^2} + \left(-\beta + \frac{\partial \ln \Omega(E)}{\partial E} \right)^2 \right) \exp(-\beta E + \ln \Omega(E)) \\
&= ??? \\
&= \frac{1}{\kappa_B} \frac{\partial}{\partial E} \left(\frac{1}{T} \right) = \frac{-1}{\kappa_B T^2} \frac{\partial T}{\partial E} = \frac{-2}{\kappa_B T^2 3 N \kappa_B} = \frac{-2}{3 \kappa_B^2 T^2} \\
\Rightarrow \\
\ln P(E) &= \ln P(E_0) + \delta E^2 \frac{1}{2} \frac{-2}{3 \kappa_B^2 T^2} = \ln P(E_0) - \delta E^2 \frac{1}{3 \kappa_B^2 T^2} \\
P(E) &= P(E_0) e^{-\frac{\delta E^2}{3 \kappa_B^2 T^2}} = P(E_0) e^{-\frac{(E - E_0)^2}{3 \kappa_B^2 T^2}}
\end{aligned}$$

2).

$$\int_0^{10^{-6} E_0} P(E) dE =$$

$$\begin{aligned}
\delta E &= 10^{-6} E_0 \\
\frac{P(E)}{P(E_0)} &= e^{-\frac{10^{-6} E_0}{3 \kappa_B^2 T^2}} = ??? = \frac{1}{e^{10^9}}
\end{aligned}$$