Revised: 10/20/2017

AFM Imaging Request Form

Requestor:	James Rushing	_PI/Advisor:	Paul Simmonds
Project:	GaAs TSQD on InP(111)A		
-			

Overview

In order to obtain useful AFM images/data for you, we need to know a little bit about your sample(s) and what kind of data you're interested in. Additionally, you need to be aware of some of the important limitations and considerations associated with acquiring AFM images.

- AFM samples have to be scrupulously clean. This can't be emphasized enough if you can see even the smallest speck of dirt on your sample, there's even more the AFM will pick up that will almost certainly obscure what you actually want to observe.
- Maximum XY scan range depends on the particular AFM used (ranging from 90 µm for the lcon down to 15 µm for the MultiMode).
- XY resolution is limited by the larger of:
 - Probe radius (~2-25 nm depending upon probe type)
 - Pixel size (max of 5120 x 5120 pixels, regardless of scan size in μm)
- Maximum Z range depends on the particular AFM used (ranging from ~12 μm for the Icon down to ~3 μm for the FastScan and MultiMode).
- To ensure the safety/integrity of data, the SSL computers and associated data are isolated on an internal network and thus not externally accessible via the internet. Thus to obtain your data/images, you will need to come by the SSL with a virus free USB flash drive. It is mandatory that you run a virus scan on your flash drive in the presence of an SSL staff member before inserting it into any of the computers to protect both your data and that of everyone else who uses the SSL.
- Please plan ahead if you require rapid turnaround on imaging of your sample(s), you will
 need to work with SSL staff to make arrangements ahead of time so they can
 schedule/reserve the necessary equipment and create the time in their schedules to
 accommodate your request.

Sample Considerations

- 1. Are there any hazards associated with the materials/composition of your sample?
 - Will sublimate arsenic at extremely high temperatures
- 2. How was your sample prepared and cleaned?
 - Sample prepared by MBE.
- 3. Can we clean the sample further if necessary? If so, how (e.g., UHP N₂ gas, CO₂ "snow" cleaning, ultrapure water, soap and water, alcohols, ketones, ...)?
 - UHP N2 is fine but should not be necessary
- 4. Are there any special handling considerations for your sample?

InP(111) substrates are very fragile

Revised: 10/20/2017

5. When will you be coming to collect your sample (e.g., within 1 week, 2 weeks, 1 month)?

Imaging Needs

- Number of images with desired XY Scan size (in μm) and lateral image resolution (in nm).
 Note limitations above. 1x1,2x2,5x5. highest feasable resolution
- 2. Expected height/Z range (in nm or μm). Note limitations above.

5nm

3. Naming convention to be used for data. Include both an informative filename and a logical file directory structure for organization of data by project and sample type (*not* your name) for ease of future reference/long term data storage and management requirements.

M631-AFM

- 4. Type(s) of data needed (e.g., height/topography, surface roughness, conductivity, surface potential, etc.).

 Roughness
- 5. Do you have any literature references showing similar samples/images? If so, please attach them.

 attached
- 6. How soon do you need your sample(s) imaged (e.g., next few days, 1 week, 2 weeks, etc.)?

 ASAP
- 7. How do you want to receive the data (e.g., email, network drive, bring a flash drive, etc.)?

 email: jimmyrushing@boisestate.edu
- 8. If you want us to carry out image processing and data analysis, please indicate what you need (e.g., raw images, tabular numerical results, journal quality images, summary report, etc.).

 Journal quality, if possible, as one never knows.

Acknowledgment of the SSL

All written publications (e.g., posters, papers, conference proceedings) and oral presentations that utilize primary and/or secondary data obtained using the equipment housed in the SSL should acknowledge the SSL and *include the make(s) and model(s) of instrumentation employed*.¹ The SSL staff can assist you in crafting an appropriate experimental description. Depending upon the extent of involvement, we ask that you also consider including as co-authors SSL staff members if they contribute significantly to a project. By signing below, you acknowledge that you have read and agree to abide by the above as well as the SSL Use Request Form and Fee Schedule, which are incorporated herein by reference.

User Signature:

Date: 1-24-23

Date: 1/24/2023

¹ Examples of primary data include AFM data channel images or force curves. Secondary data include plots derived from the results of image or data analysis.

Semicond. Sci. Technol. 35 (2020) 105001 (9pp)

https://doi.org/10.1088/1361-6641/aba6e0

Impact of arsenic species on self-assembly of triangular and hexagonal tensile-strained GaAs(111)A quantum dots

Christopher F Schuck¹, Kevin D Vallejo¹, Trent Garrett², Qing Yuan³, Ying Wang³, Baolai Liang³ and Paul J Simmonds^{1,2,4}

- ¹ Micron School of Materials Science & Engineering, Boise State University, Boise, ID 83725, United States of America
- Department of Physics, Boise State University, Boise, ID 83725, United States of America
- ³ College of Physics Science & Technology, Hebei University, Baoding 071002, People's Republic of China

E-mail: p.j.simmonds.03@cantab.net

Received 5 June 2020, revised 14 July 2020 Accepted for publication 16 July 2020 Published 21 August 2020



Abstract

We use dimeric arsenic (As_2) or tetrameric arsenic (As_4) during molecular beam epitaxy to manipulate the structural and optical properties of GaAs(111)A tensile-strained quantum dots (TSQDs). Choice of arsenic species affects nucleation and growth behavior during TSQD self-assembly. Previously, epitaxial GaAs(111)A TSQDs have been grown with As₄, producing TSQDs with a triangular base, and 'A-step' edges perpendicular to the three $\overline{112}$ directions. We demonstrate that using As₂ at low substrate temperature also results in triangular GaAs(111)A TSQDs, but with 'B-step' edges perpendicular to the three $11\overline{2}$ directions. We can therefore invert the crystallographic orientation of these triangular nanostructures, simply by switching between As₄ and As₂. At higher substrate temperatures, GaAs(111)A TSQDs grown under As₂ develop with a hexagonal base. Compared with triangular dots, the higher symmetry of hexagonal TSQDs may reduce fine-structure splitting on this (111) surface, a requirement for robust photon entanglement. Regardless of shape, GaAs(111)A TSQDs grown under As₂ exhibit superior optical quality.

Keywords: quantum dots, (111), molecular beam epitaxy, arsenic species, III-V semiconductors, tensile strain

(Some figures may appear in colour only in the online journal)

1. Introduction and background

Self-assembled quantum dots (QDs) are routinely grown from compressively strained materials on (100)-oriented surfaces by molecular beam epitaxy (MBE). These QDs exhibit highly tunable structural and optoelectronic properties, high crystalline quality, formation via a single processing

step, and compatibility with semiconductor device architectures and processing lines [1–4]. Emerging research fronts and device applications have spurred efforts to extend the range of materials systems within which one can synthesize QDs. Example materials systems include (111)-oriented QDs, which have a low fine structure splitting (FSS) for quantum information applications [5–8], and tensile-strained QDs (TSQDs), which offer band structure engineering capabilities for long wavelength optoelectronics and a potential route towards direct band gap Ge [9–13].

⁴ Author to whom any correspondence should be addressed.

GaAs/InAlAs(111)A TSQDs were the first optically active nanomaterials to combine the benefits of epitaxial self-assembly, tensile strain, and (111)-orientation [8]. GaAs(111)A TSQDs are dislocation-free, structurally and optically tunable, emit light below the bulk GaAs band edge due to the residual tensile strain, and exhibit low FSS [14–17]. Recent studies describe how GaAs(111)A TSQD properties depend on various MBE growth parameters: deposition amount (in monolayers (MLs)), substrate temperature (T_{SUB}), growth rate, and group V/III flux [16, 17].

To date however, all GaAs(111)A TSQDs have been grown using tetrameric arsenic (As₄) [8, 12, 16–18]. Most growers today use As₂ rather than As₄ for MBE of III–V semiconductors. However, the original studies that identified the MBE conditions required to grow high quality material on (111) surfaces predated the widespread availability of valved arsenic crackers to generate dimeric As₂ [19–22]. As a result, those authors had no choice but to use As₄ [19–22]. The use of As₄ for existing work on (111)-oriented TSQDs followed directly from that historical precedent [8, 12, 16–18, 23].

Researchers have shown that choice of arsenic species can affect photoluminescence (PL) intensity [24, 25], PL linewidth, [26, 27] non-radiative interface recombination [28], the occurrence of deep electron traps [29], auto-compensation of amphoteric dopants like Si [29, 30], and surface morphology [31, 32]. For bulk GaAs(111)A, the As₂ incorporation coefficient (S_{As2}) is larger than that of As₄ (S_{As4}) [30]. PL emission from (100)-oriented QDs is also strongly influenced by choice of arsenic species [26, 33–36]. On (100), QDs grown with As₄ tend to be more elongated than QDs grown with As₂, with QD size and areal density also impacted [26, 33, 34, 36].

Clearly, the choice of arsenic species during MBE growth is an important growth variable. We therefore wanted to compare the impact of using As₄ or As₂ on the growth and properties of GaAs/InAlAs(111)A TSQDs, to ascertain whether there exists some benefit to using one molecular species over the other. In this paper we demonstrate that GaAs/InAlAs(111)A TSQD self-assembly still occurs when using As₂. We compare the resulting GaAs TSQDs to equivalent TSQDs grown with As₄. This simple change from As₄ to As₂ results in different nucleation and growth physics, and as a result, more control over TSQD structure, the option to grow either triangular or hexagonal TSQDs, and improved PL emission.

2. Methods

We use solid-source MBE to grow GaAs(111)A TSQDs within In_{0.52}Al_{0.48}As top and bottom barriers, lattice matched to semi-insulating Fe-doped InP(111)A substrates. First we grow 50 nm of lattice-matched In_{0.53}Ga_{0.47}As as a smoothing layer [23], followed by the 200 nm InAlAs bottom barrier. We have previously reported the optimal growth parameters for InAlAs and InGaAs on InP(111)A [16, 23]. GaAs(111)A TSQDs form spontaneously on InAlAs(111)A via a modified Stranski-Krastanov (SK) growth mode [17]. The critical thickness for the 2D-to-3D growth transition is 2.5 ML, but the wetting layer continues to grow even after the TSQDs have begun to self-assemble [17]. For TSQD growth, the GaAs deposition

rate is 0.075 ML s⁻¹, with an As:Ga beam equivalent pressure (BEP) ratio of 75, growth conditions favorable for producing high quality TSQDs [16]. We grow GaAs(111)A TSQD samples in the range 485 °C \leq T_{SUB} \leq 570 °C. For growth under As₄ and As₂ we used arsenic cracker temperatures of 600 °C and 900 °C, respectively. The GaAs TSQDs are capped with a 50 nm InAlAs top barrier, followed by another layer of GaAs TSQDs for structural analysis with tapping-mode atomic force microscopy (AFM). We use the buried layer of GaAs TSQDs for optical analysis by PL.

We use AFM to explore sample surface morphology, focusing in particular on the shape and size of the TSQDs. By analyzing multiple AFM images of each sample, we compile statistics of average TSQD height, diameter, and areal density as a function of the MBE growth conditions. Note that for these non-circular TSQDs, 'diameter' refers to twice the distance from dot center to vertex. We carry out PL measurements at 7 K, using a laser wavelength of 532 nm and an excitation density of 80 W cm⁻².

3. Results and discussion

We grew an initial pair of GaAs(111)A TSQD samples under identical MBE growth conditions ($T_{SUB} = 485$ °C, growth rate = 0.075 ML s⁻¹, As:Ga BEP ratio = 75), except that we used As₄ for one, and As₂ for the other. Both arsenic species give rise to GaAs TSQD self-assembly. When we compare the resulting GaAs TSQDs with conventional, compressively strained QDs grown on (100) surfaces, they have low average height (< 1.2 nm), large diameter (> 60 nm), and low areal density (< 5 × 10⁸ cm⁻²) (figures 1(a) and (c)). These size and density characteristics are consistent with previous research on GaAs(111)A TSQDs [16].

Figures 1(b) and (d) show the AFM amplitude (error) images corresponding to figures 1(a) and (c). AFM amplitude images, provide higher contrast for changes in surface height, for example at step edges and QD facets, and as a result allow us to better visualize the shapes of these self-assembled GaAs nanostructures [37]. Figures 1(a)–(b) show that the TSQDs grown under As₄ are always triangular, as seen in previous studies of GaAs(111)A TSQDs [8, 16, 17]. In contrast, figures 1(c)–(d) reveal that TSQDs grown under As₂ consist of a mixture of triangles and hexagons.

We were intrigued by the possibility that one could select whether to grow triangular or hexagonal TSQDs, simply by selecting a particular arsenic species (figure 1). Some examples of similar shape transitions exist in the literature for the homoepitaxial growth of metals [38–41], and unstrained GaAs(111)A QDs grown by droplet epitaxy [42]. However, there are no previous demonstrations of shape transitions like this in strain-driven self-assembled QD materials systems, and no existing reports of hexagonal TSQDs.

To investigate this effect further, we therefore grew a series of TSQD samples under As_4 and As_2 at 535 °C \leq $T_{SUB} \leq$ 570 °C, since TSQD crystal quality improves in this temperature range [16]. We found that TSQDs grown under As_4 at 535 °C are all triangular in shape, just as at 485 °C

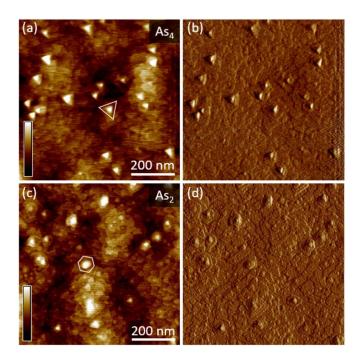


Figure 1. AFM images of GaAs (111)A TSQDs (1 μ m², 2 nm height bar). (a), (b) show 3 ML GaAs TSQDs deposited under As₄. (c), (d) show 4.5 ML GaAs TSQDs deposited under As₂. All other MBE growth conditions are the same. AFM height images (a), (c) provide information on QD size, shape, and spacing. AFM amplitude images (b), (d) enhance the appearance of QD edges. Using As₄ produces only triangular TSQDs (a), (b), while using As₂ produces a mixture of triangular and hexagonal TSQDs (c), (d).

Table 1. Summary of TSQD step-edge orientations as a function of arsenic species and substrate growth temperature. We did not categorize TSQDs that were not clearly triangular or hexagonal. Errors indicate the standard deviation in the percent of each type of TSQD observed in various AFM images for each sample.

T_{SUB}	As-species	A-step triangles (%)	Hexagons (%)	B-step triangles (%)
535 °C	As ₄ As ₂	100 0	0 35 ± 7	0 65 ± 9
570 °C	As ₄ As ₂	83 ± 5 10 ± 9	17 ± 5 90 ± 8	0

(figure 2(a)). At 570 °C, most TSQDs were still triangular (figure 2(b)), but we noted a small population that were hexagonal. For TSQDs grown under As₂ at 535 °C, most were triangular (figure 2(c)), but a significant minority were hexagonal in shape. By 570 °C, the vast majority of TSQDs were hexagonal, (figure 2(d)) with only occasional triangular TSQDs (table 1).

When we compared the crystallographic orientations of GaAs TSQDs grown at $T_{SUB} = 535$ °C, we found that the triangular dots formed under As_4 are inverted compared to those grown with As_2 . In figure 2(a), the triangular TSQDs grown under As_4 are bounded by 'A-steps,' perpendicular to the $[\bar{1}2\bar{1}]$, $[2\bar{1}\bar{1}]$, and $[\bar{1}\bar{1}2]$ directions. In figure 2(c), the triangular TSQDs grown under As_2 are bounded by 'B-steps,' perpendicular to $[11\bar{2}]$, $[1\bar{2}1]$, and $[\bar{2}11]$. As T_{SUB} is raised to

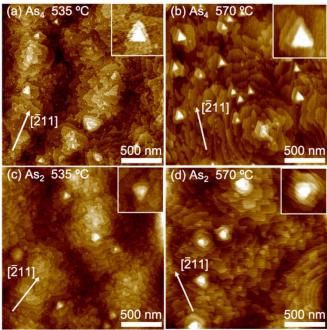


Figure 2. $2 \times 2 \mu m^2$ AFM images showing the appearance of 3.5 ML GaAs (111)A TSQDs as a function of T_{SUB} and arsenic species: (a) GaAs TSQDs grown at 535 °C under As₄; (b) GaAs TSQDs grown at 570 °C under As₄; (c) GaAs TSQDs grown at 535 °C under As₂; and (d) GaAs TSQDs grown at 570 °C under As₂. All other MBE growth conditions are the same. In (a)–(b), using As₄ produces only triangular TSQDs, while in (c)–(d), using As₂ produces a mixture of triangular and hexagonal TSQDs. Insets are $250 \times 250 \text{ nm}^2$ AFM images of individual QDs from each sample, oriented so that the [211] direction points vertically. All AFM images have a 2 nm height scale.

570 °C, the triangular TSQDs grown under As₄ maintain their A-step orientation (figure 2(c)). The hexagonal TSQDs that form at $T_{SUB} = 570$ °C under As₂ have both A- and B-step edges (figure 2(d)).

Figure 3(a) summarizes the crystallographic directions of the A- and B-steps for GaAs islands on the (111)A surface. Note that in the absence of any surface reconstruction, these GaAs(111)A islands are always bounded by As atoms. On this surface orientation, As atoms have three 'upward' pointing bonds with Ga atoms in the same layer, and one out-of-plane bond with a Ga atom in the layer below (figure 3(b)). In contrast, the Ga atoms form three bonds 'downwards' with As atoms in the same layer, and have one dangling bond pointing up out of the plane (not shown). A Ga atom at the periphery of the island would not form a bond with the surface below and so terminating the island edge with Ga atoms would be unstable. However, because MBE growth takes place under an excess flux of As, it is usually assumed that a Ga atom joining an island edge is stabilized immediately by the addition of an As atom that bonds with both that Ga atom and with a Ga atom on underlying surface. This As atom then marks the new edge of the island, explaining why these islands always have As atoms at the periphery.

Previous experimental and computational studies show that when growth is kinetically limited, which is usually the case

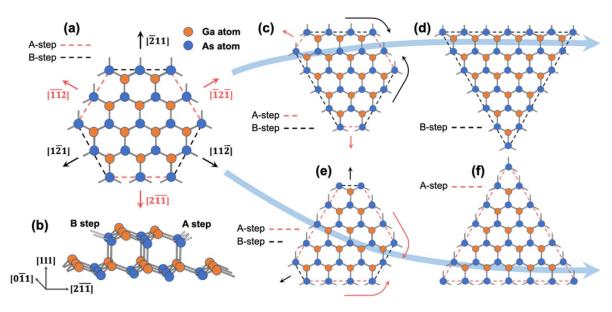


Figure 3. (a) Plan view of an unreconstructed GaAs(111)A island. The crystallographic orientation of the A-steps (red) and B-steps (black) are shown. (b) Cross-sectional view of an unreconstructed GaAs(111)A island in the $[0\bar{1}1]$ direction, showing the differences at the A- and B-steps in terms of dangling bonds (shown as dashed lines) and atomic arrangements. (c) If the net edge diffusion of adatoms is from B-steps to A-steps (black arrows), then the A-steps will grow faster (red arrows), getting shorter as they do so, while the B-steps get longer. (d) This process culminates in the formation of a triangular island terminated by B-steps. (e) In contrast, if net adatom diffusion is from A-steps to B-steps (red arrows), the result is the preferential growth at the B-steps (black arrows), and the lengthening of the A-steps. (f) Under this scenario, the outcome is the formation of a triangular island bounded by A-steps. Note the inverted orientation of the triangles in (d) and (f).

during MBE growth, islands bounded by B-steps often form preferentially [38–42]. Under these conditions, B-step islands are favored for three reasons [38–40, 42, 43]. First, each As atom at an A-step on the (111)A surface has two bonds to neighboring Ga atoms, one in the same atomic layer and one in the layer below (figure 3(b)). At a B-step however, each As atom has three bonds to neighboring Ga atoms, two in the same atomic layer and one in the layer below (figure 3(b)). Since the As atoms defining a B-step have more complete bonds than those at an A-step, the cohesive energy of the B-step (defined as the energy required to break the atoms of the solid into isolated atomic species [44]) is higher. B-steps are hence more stable, and so B-step islands are favored.

Second, B-steps have a higher edge diffusion coefficient, efficiently transporting adatoms to neighboring A-steps. As a Ga adatom diffuses along the edge of an island, it moves from one bonding site (local potential well) to another by crossing a potential barrier [45]. The lower this potential barrier, the higher the adatom diffusion coefficient. Each bonding site on a B-step corresponds to an As atom with a single dangling bond, while at an A-step the As atoms at the bonding sites have two dangling bonds (figure 3(b)). The result is a weaker interaction between Ga adatoms and the As atoms on B-steps, and so these bonding sites are associated with shallower potential wells than at A-steps. These shallower wells mean a lower potential barrier between bonding sites on a B-step, raising the edge diffusion coefficient compared with at an A-step. The outcome is that Ga adatoms diffuse rapidly along B-steps until they reach a neighboring A-step, where diffusion is slower.

Third, for diffusion to occur between A- and B-steps, adatoms must pass around a corner of the island. Several

studies have shown that this corner diffusion process is anisotropic: the probability of an adatom crossing from a B-step to an A-step is higher than from an A-step to a B-step (figure 3(c)) [39, 41, 42, 46]. Indeed, evidence exists that anisotropic corner diffusion may play the most important role of these three components in dictating final island shape [39, 41].

The net result of these three factors is that during kinetically limited growth, the A-steps grow faster than the B-steps. As a result, the A-steps grow out, get shorter (figure 3(c)), and eventually disappear. A triangular island is left behind, bounded by B-steps (figure 3(d)) [36, 38–40, 42, 43], just as we observe for GaAs(111)A TSQDs grown under As₂ at $T_{SUB} = 535$ °C (figure 2(c)).

There are however situations where triangular islands bounded by A-steps can form, just as we see for GaAs(111)A TSODs grown under As₄ (figures 2(a)–(b)) [38, 41]. Ovesson et al showed that simply reversing the corner diffusion anisotropy is sufficient to switch growth from B-step islands to Astep islands [41]. If the net edge diffusion of adatoms inverts from the A-steps to the B-steps, then the B-steps will grow preferentially, lengthening the A-steps (figure 3(e)). The eventual outcome is the formation of triangular islands bounded by A-steps (figure 3(f)). Michely *et al* show that one can use T_{SUB} to invert the A-step and B-step growth rate anisotropy [38]. At lower T_{SUB}, adatom diffusion conditions are such that Astep islands grow preferentially. However, we see the A-step islands in figures 2(a)-(b) forming at the same or higher substrate temperature to the B-step islands in figure 2(c). T_{SUB} cannot therefore be responsible for the observed change from A-step to B-step islands.

The critical difference seems to be the arsenic species, and it is therefore critical to understand how the presence of As_4 or As_2 affects Ga adatom diffusion rates along A- and B-steps. While As_2 molecules dissociate directly on the epitaxial surface, incorporation of arsenic from As_4 is a second order process involving pairs of As_4 molecules at adjacent bonding sites [47, 48]. At low T_{SUB} on (111)A surfaces, the As_2 incorporation coefficient, $S_{As_2} = 1$, while for As_4 , $S_{As_4} = 0.5$. However, unlike the (100) surface, arsenic incorporation on (111) surfaces is temperature dependent. Both S_{As_2} and S_{As_4} fall as T_{SUB} is increased: at 560 °C, $S_{As_2} \sim 0.2$, and $S_{As_4} \sim 0.1$ (table 2) [30]. This decrease in S_{As_4} and S_{As_5} explains why at typical growth temperatures, a high S_{SS} and S_{SS} and S_{SS} for high epitaxial quality on (111) surfaces [23, 30].

In a study of InAsP growth, Zhang *et al* noted that when As incorporation is lower, for example when using As₄ instead of As₂, diffusion of the Group III adatoms is enhanced at A-steps. This leads to preferential bonding at B-steps and hence they grow faster than the A-steps [49]. We propose that this same mechanism is responsible for the appearance of A-step islands when we grow with As₄.

Hexagonal islands occur when T_{SUB} becomes high enough that the thermal energy eclipses the differences in cohesive energies and diffusion rates at the A- and B-steps. At such high temperatures, the atoms gain sufficient energy to detach from the A- and B-steps of existing islands, generating kinks. These kinks lower the chemical potential of the steps, and break down the competing edge diffusion processes [40]. The result is an island in equilibrium, bounded by both A- and B-steps, whose hexagonal shape reflects the underlying three-fold symmetry of the (111)A surface [38–40, 50]. We believe this is the mechanism responsible for higher percentage of hexagonal TSQDs that form under As_2 , as T_{SUB} is increased from 535 °C to 570 °C. This mechanism also explains the appearance of occasional hexagonal TSQDs at $T_{SUB} = 570$ °C for growth under As_4 (table 1).

Increasing T_{SUB} from 535 °C to 570 °C does not significantly change average TSQD height for samples grown with either As₄ or As₂ (figure 4(a)). For $T_{SUB} \le 560$ °C, the average diameter of TSQDs grown with As4, and As2 are essentially identical, with both increasing as we raise T_{SUB} in this range (figure 4(b)). At $T_{SUB} = 570$ °C however, we see a marked divergence in the response of TSQDs grown under As4 and As₂. The average diameter of TSQDs grown with As₂ continues to increase monotonically as we raise T_{SUB} (figure 4(b)). Raising T_{SUB} increases the mean adatom diffusion length, promoting adatom attachment to existing islands rather than the nucleation of new QDs [1, 14, 51, 52]. The result of raising T_{SUB} under As₂ is therefore to produce GaAs TSQDs with larger average size and lower areal density [53]. In contrast, TSQDs grown with As₄ at $T_{SUB} = 570$ °C undergo a marked reduction in their average diameter (figure 4(b)).

A possible explanation for this difference in TSQD selfassembly with As₄ and As₂ is again found in the disparity

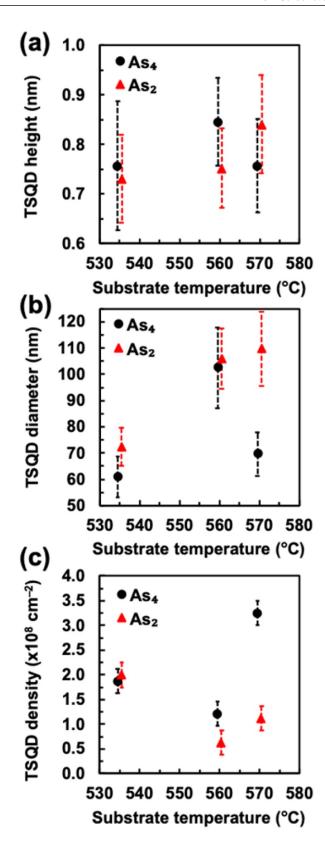


Figure 4. Average (a) height, (b) diameter and (c) areal density of GaAs/InAlAs(111)A TSQDs grown with As₄ (black) and As₂ (red), as a function of T_{SUB} (error bars represent one standard deviation). For clarity, data points at the same T_{SUB} are offset horizontally by 1 $^{\circ}$ C.

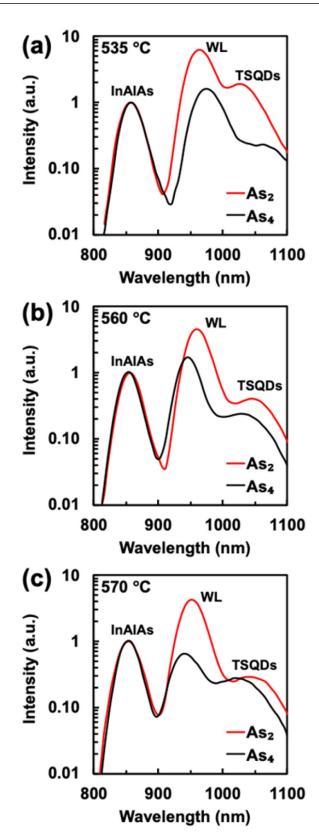


Figure 5. 7 K PL emission spectra from GaAs(111)A TSQD samples growth with As₂ (red) and As₄ (black) at $T_{SUB} = (a)$ 535 °C, (b) 560 °C, and (c) 570 °C. Spectral intensities are normalized to the bulk InAlAs peak.

between the values of S_{As4} and S_{As2} (table 2). $S_{As2}\approx 2S_{As4}$ at all values of T_{SUB} , meaning that for the same flux, the surface

Table 2. (111)A surface arsenic incorporation coefficients for As_4 and As_2 as a function of T_{SUB} , from Tok *et al* [30]. S_{As4} is roughly half that of S_{As2} , regardless of temperature.

	280 °C	470 °C	535 °C	560 °C	570 °C
S_{As4}	0.5	0.24	0.11	0.09	0.08
$S_{As2} \\$	1.0	0.50	0.25	0.19	0.18

arsenic concentration will be higher when using As₂ than As₄. Growth under As₂ is therefore expected to result in (111)A surfaces with the arsenic-rich, As-trimer reconstruction [54, 55]. In contrast, at high T_{SUB} the lower incorporation of As₄ could push growth into a Ga-rich regime where the Gavacancy surface reconstruction is most stable [54]. Previous reports have shown bulk GaAs(111)A grown under As₄ transiting to a Ga-rich surface when heated above 550 °C [56]. Calculations suggest that the energy barrier to Ga adatom diffusion is four times larger on the Ga-vacancy reconstruction than the As-trimer reconstruction [45]. Lower adatom diffusion results in QDs with smaller diameter and higher areal density [1, 14, 16, 52]. A sudden transition at $T_{SUB} = 570$ °C from the As-trimer to Ga-vacancy reconstruction for growth under As₄ could therefore explain the abrupt reduction in TSQD diameter that we see in figure 4(b).

If this is the correct mechanism, the appearance of the Gavacancy reconstruction at $T_{SUB} = 570\,^{\circ}\text{C}$ should also increase TSQD areal density for samples grown under As_4 . This is indeed what we see (figure 4(c)). Regardless of As species, for $T_{SUB} \leq 560\,^{\circ}\text{C}$, we see a monotonic decrease in TSQD density as the temperature is raised, as expected for increased Ga adatom diffusion [14, 16]. However, at $T_{SUB} = 570\,^{\circ}\text{C}$, TSQDs grown under As_4 undergo a threefold increase in density, consistent with a sudden drop in adatom diffusion. For comparison, the density of TSQDs grown under As_2 does not change (to within error) between $T_{SUB} = 560\,^{\circ}\text{C}$ and $570\,^{\circ}\text{C}$ (figure 4(c)).

PL spectroscopy of the various samples consistently show enhanced emission intensity for TSQDs grown using As₂ compared with As₄ (figure 5). To aid comparison, in each case we have normalized the PL spectra to the intensity of the InAlAs barrier peak at 856 nm since this should be identical across samples. The peak at 941–975 nm corresponds to the tensile-strained GaAs wetting layer (WL), while the smaller peak at 1009–1040 nm comes from the GaAs TSQDs. Ref. [17] details how these peaks were assigned.

These spectra contrast with conventional InAs/GaAs(100) QDs whose areal densities are typically in the 10^{10} cm⁻² range [57]. Once the InAs dots begin to form, PL from the QDs dominates, while emission from the WL is difficult to resolve [57]. The opposite is true for the GaAs TSQD spectra in figure 5. For all samples, emission from the tensile-strained GaAs WL is more intense than from the GaAs TSQDs (figure 5). The reason is that the GaAs TSQD areal densities are in the 10^8 cm⁻² range (see figure 4(c)), i.e. 100×10^8 lower than for InAs/GaAs(100) QDs. PL emission from fewer TSQDs per unit area means that the TSQD peaks are weaker, and so the WL peaks dominate these spectra (figure 5).

However, our primary interest is to compare the effect of As_4 and As_2 on the optical properties of these samples. Plotting the intensity of the WL peaks in figure 5, shows that, regardless of T_{SUB} , the use of As_2 during growth results in brighter PL emission compared with the use of As_4 (figure 6(a)). We can perform a similar comparison for the PL intensity of the GaAs TSQD peaks, but we must first account for the variation in TSQD areal density from sample to sample. For a given laser spot size, the more TSQDs that are excited, the more photons will be emitted. This is particularly important for samples grown under As_4 , given the increase in TSQD density we see at $T_{SUB} = 570$ °C (figure 4(c)). We therefore use a metric we refer to as *emission efficiency*, which is given by the integrated area of the TSQD PL peak, divided by the areal density of TSQDs on that sample [16].

Plotting emission efficiency shows that regardless of T_{SUB} , PL emission from GaAs TSQDs grown with As_2 is more intense than from TSQDs grown with As_4 (figure 6(b)). Despite the higher areal density of the As_4 -grown TSQDs at $T_{SUB} = 570\,^{\circ}\text{C}$, the As_2 -grown TSQDs still have higher emission efficiency. TSQD emission efficiency falls off for both As_2 and As_4 samples as we raise T_{SUB} from 560 $^{\circ}\text{C}$ to 570 $^{\circ}\text{C}$. This behavior shows that material quality is compromised, most likely as a result of the very low As incorporation coefficients of both arsenic species at this high temperature (table 2) [30, 32].

We can therefore conclude that the use of As_2 during the growth of these TSQD samples significantly improves the optical quality of the GaAs in both the tensile-strained WL and in the TSQDs themselves. Specifically, the use of As_2 increases WL emission intensity by $3-7\times$ and TSQD emission efficiency by $3-4\times$, compared with the use of As_4 (figures 6(a)–(b)). In general, higher PL intensity is an indicator of lower non-radiative recombination and improved material quality. Researchers have long associated the pairwise incorporation mechanism for As_4 with increased levels of point defects [48], and shown that defect densities can be reduced by using As_2 instead [58].

The spectral linewidth of the WL PL peaks, defined as the full width at half maximum (FWHM), remains constant regardless of T_{SUB} and the arsenic species used (figure 6(c)). This tells us that the width uniformity of the WL quantum well is robust against these changes in the MBE growth parameters. For samples grown at $T_{SUB} \le 560$ °C, the FWHM of the TSQD PL peaks, is narrower for samples grown with As₂ than with As₄ (figure 6(c)). In general, for an array of QDs, a narrower FWHM is consistent with a more uniform dot size distribution, and is a positive attribute for many light emission applications. The overall trend for both As₄- and As₂-grown TSQDs is for larger FWHM as T_{SUB} is raised, consistent with an increase in Ostwald ripening as adatom diffusion is enhanced. However, at $T_{SUB} = 570$ °C the trend for TSQDs grown under As₄ is reversed, and a lower FWHM suggests that they are more uniform in size. The transition to the Ga-vacancy reconstruction and the accompanying reduction in Ga adatom diffusion

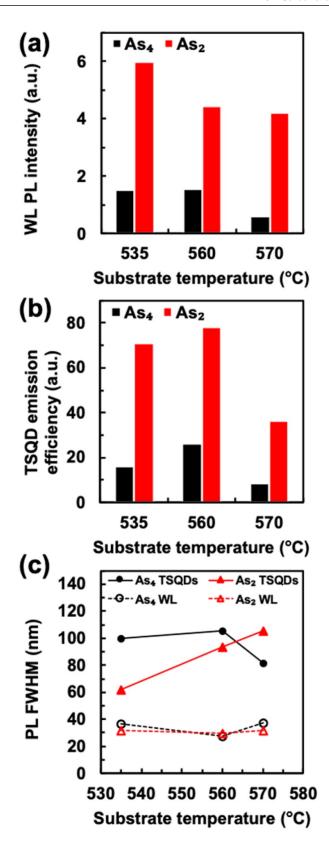


Figure 6. The impact of arsenic species on (a) tensile-strained GaAs WL PL intensity, and (b) GaAs(111)A TSQD PL emission efficiency as a function of T_{SUB} . (c) FWHM of the GaAs(111)A TSQD and WL PL peaks as a function of T_{SUB} and arsenic species.

hampers Ostwald ripening. At this high temperature, the result is a slight improvement in size uniformity for TSQDs grown under As_4 compared with As_2 .

4. Conclusions

We can manipulate the shape, size, areal density, and optical emission quality of GaAs/InAlAs(111)A TSQDs during MBE growth through our choice of arsenic species and T_{SUB}. GaAs(111)A TSQDs have three distinct morphologies: Aand B-step triangular TSQDs, and high symmetry hexagonal TSQDs. Which shape arises depends on the relative growth rates of the A- and B-steps. We demonstrate that we can influence these growth rates by growing under As4 or As2. At sufficiently high T_{SUB}, differences between the A- and Bstep growth rates become negligible, and hexagonal TSQDs appear. We know that higher structural symmetry helps to reduce FSS in semiconductor QDs, for example in triangular (111) TSQDs compared to elliptical (100) QDs [5, 6, 59, 60]. In future, we are therefore interested to see if the higher structural symmetry of the hexagonal GaAs(111)A TSQDs we can now grow using As₂, will offer even lower FSS than the triangular GaAs(111)A TSQDs we have investigated to date [8].

When using As_4 , we see evidence for a transition from the As-trimer to the Ga-vacancy surface reconstruction at $T_{SUB} = 570$ °C. This transition seems to be triggered by the particularly low arsenic incorporation coefficient for As_4 compared with As_2 at this high temperature. In all sample sets, growth under As_2 produces samples with better optical quality than As_4 . We therefore recommend that for optoelectronic applications, MBE growth of GaAs(111)A TSQDs should be carried out under As_2 instead of As_4 . The high symmetries available in these self-assembled nanostructures may prove to be beneficial for quantum optics applications.

Acknowledgments

This material is based upon work supported by the National Science Foundation under NSF CAREER Grant No. 1555270.

ORCID ID

Paul J Simmonds https://orcid.org/0000-0001-5524-0835

References

- [1] Joyce B A and Vvedensky D D 2004 *Mater. Sci. Eng. R* 46 127
- [2] Franchi S, Trevisi G, Seravalli L and Frigeri P 2003 Prog. Cryst. Growth Charact. Mater. 47 166
- [3] Ratto F and Rosei F 2010 Mater. Sci. Eng. R 70 243
- [4] Alivisatos A P 1996 Science **271** 933–7
- [5] Schliwa A, Winkelnkemper M, Lochmann A, Stock E and Bimberg D 2009 Phys. Rev. B 80 161307(R)
- [6] Schulz S, Caro M A, O'Reilly E P and Marquardt O 2011 Phys. Rev. B 84 125312
- [7] Juska G, Murray E, Dimastrodonato V, Chung T H, Moroni S T, Gocalinska A and Pelucchi E 2015 J. Appl. Phys. 117 134302

- [8] Yerino C D et al 2014 Appl. Phys. Lett. 105 251901
- [9] Signorello G, Karg S, Björk M T, Gotsmann B and Riel H 2013 Nano Lett. 13 917–24
- [10] El Kurdi M, Fishman G, Sauvage S and Boucaud P 2010 J. Appl. Phys. 107 013710
- [11] Li X, Maute K, Dunn M L and Yang R 2010 Phys. Rev. B 81 245318
- [12] Simmonds P J and Lee M L 2012 J. Appl. Phys. 112 054313
- [13] Sautter K E, Schuck C F, Garrett T A, Weltner A E, Vallejo K D, Ren D, Liang B, Grossklaus K A, Vandervelde T E and Simmonds P J 2020 J. Cryst. Growth 533 125468
- [14] Simmonds P J and Lee M L 2011 Appl. Phys. Lett. 99 123111
- [15] Simmonds P J, Laghumavarapu R B, Sun M, Lin A, Reyner C J, Liang B and Huffaker D L 2012 Appl. Phys. Lett. 100 243108
- [16] Schuck C F, McCown R A, Hush A, Mello A, Roy S, Spinuzzi J W, Liang B, Huffaker D L, Simmonds P J and Vac J 2018 J. Vac. Sci. Technol. B 36 031803
- [17] Schuck C F, Roy S K, Garrett T, Yuan Q, Wang Y, Cabrera C I, Grossklaus K A, Vandervelde T E, Liang B and Simmonds P J 2019 Sci. Rep. 9 18179
- [18] Simmonds P J, Simon J, Woodall J M, Lee M L and Vac J 2011 J. Vac. Sci. Technol. B 29 03C103
- [19] Okano Y, Shigeta M, Seto H, Katahama H, Nishine S and Fujimoto I 1990 Jpn. J. Appl. Phys. 29 L1357
- [20] Sato K, Fahy M R and Joyce B A 1994 Jpn. J. Appl. Phys. 33 L905
- [21] Yamamoto T, Inai M, Takebe T and Watanabe T 1993 *J. Vac. Sci. Technol.* A 11 631
- [22] Woolf D A, Westwood D I and Williams R H 1993 Semicond. Sci. Technol. 8 1075
- [23] Yerino C D, Liang B, Huffaker D L, Simmonds P J and Lee M L 2017 J. Vac. Sci. Technol. B 35 010801
- [24] Tsang W T, Ditzenberger J A and Olsson N A 1983 IEEE Electron Device Lett. 4 275–7
- [25] Hao Z-B, Ren Z-Y, Guo W-P and Luo Y 2001 *J. Cryst. Growth* **224** 224–9
- [26] Liang B et al 2018 Opt. Express 26 23107
- [27] Gilfert C, Pavelescu E-M and Reithmaier J P 2010 Appl. Phys. Lett. 96 191903
- [28] Duggan G, Dawson P, Foxon C T and 'T Hooft G W 1982 *J. Phys.* **43** 129
- [29] Jung H, Künzel H and Ploog K 1982 *J. Phys. Colloques* **43** C5-135-43
- [30] Tok E S, Jones T S, Neave J H, Zhang J and Joyce B A 1997 Appl. Phys. Lett. 71 3278
- [31] Hayakawa T, Nagai M, Morishima M, Horie H and Matsumoto K 1991 Appl. Phys. Lett. 59 2287
- [32] Holmes D M, Belk J G, Sudijono J L, Neave J H, Jones T S and Joyce B A 1995 Appl. Phys. Lett. 67 2848
- [33] Sugaya T, Amano T and Komori K 2006 J. Appl. Phys. 100 063107
- [34] Suraprapapich S, Panyakeow S and Tu C W 2007 Appl. Phys. Lett. 90 183112
- [35] Sugaya T, Furue S, Amano T and Komori K 2007 J. Cryst. Growth 301–2 801
- [36] Avery A R, Dobbs H T, Holmes D M, Joyce B A and Vvedensky D D 1997 Phys. Rev. Lett. 79 3938
- [37] Eaton P and West P 2010 *Atomic Force Microscopy* 1st edn (Oxford: Oxford University Press)
- [38] Michely T, Hohage M, Bott M and Comsa G 1993 *Phys. Rev.* Lett. 70 3943
- [39] Cox E, Li M, Chung P-W, Ghosh C, Rahman T S, Jenks C J, Evans J W and Thiel P A 2005 Phys. Rev. B 71 115414
- [40] Lü B, Almyras G A, Gervilla V, Greene J E and Sarakinos K 2018 Phys. Rev. Mater. 2 063401
- [41] Ovesson S, Bogicevic A and Lundqvist B I 1999 Phys. Rev. Lett. 83 2608

- [42] Jo M, Mano T, Abbarchi M, Kuroda T, Sakuma Y and Sakoda K 2012 *Cryst. Growth Des.* 12 1411
- [43] Xie M H, Seutter S M, Zhu W K, Zheng L X, Wu H and Tong S Y 1999 *Phys. Rev. Lett.* **82** 2749
- [44] Li X 2014 Nanotechnology 25 185702
- [45] Shapiro J N, Lin A, Huffaker D L and Ratsch C 2011 Phys. Rev. B 84 085322
- [46] Wang S C and Ehrlich G 1990 Surf. Sci. 239 301
- [47] Foxon C T and Joyce B A 1975 Surf. Sci. 50 434
- [48] Foxon C T and Vac J 1983 *Sci. Technol.* B 1 293
- [49] Zhang W, Yi C, Brown A and Vac J 2007 Sci. Technol. B 25 960
- [50] Schmid M, Lundgren E, Leonardelli G, Hammerschmid A, Stanka B and Varga P 2001 Appl. Phys. A 72 405

- [51] Shchukin V, Schöll E and Kratzer P 2008 *Semicond*.

 Nanostructures, ed D Bimberg (Berlin: Springer) pp 1–39
- [52] Meixner M, Kunert R and Schöll E 2003 Phys. Rev. B 67 195301
- [53] Lobo C and Leon R 1998 J. Appl. Phys. 83 4168
- [54] Kaxiras E, Bar-Yam Y, Joannopoulos J D and Pandey K C 1987 Phys. Rev. B 35 9625
- [55] Moll N, Kley A, Pehlke E and Scheffler M 1996 Phys. Rev. B 54 8844
- [56] Fahy M R, Sato K and Joyce B A 1994 Appl. Phys. Lett. 64 190
- [57] Leon R and Fafard S 1998 Phys. Rev. B 58 R1726
- [58] Künzel H and Ploog K 1980 Appl. Phys. Lett. 37 416
- [59] Seguin R, Schliwa A, Rodt S, Pötschke K, Pohl U W and Bimberg D 2005 Phys. Rev. Lett. 95 257402
- [60] Singh R and Bester G 2009 Phys. Rev. Lett. 103 063601